

Synthesis of furano[2,3-*g*]indoles from activated indoles

Karin Pchalek, Naresh Kumar, and David StC. Black*

School of Chemistry, UNSW Sydney, Sydney, NSW 2052, Australia

Email: d.black@unsw.edu.au

Received mm-dd-yyyy

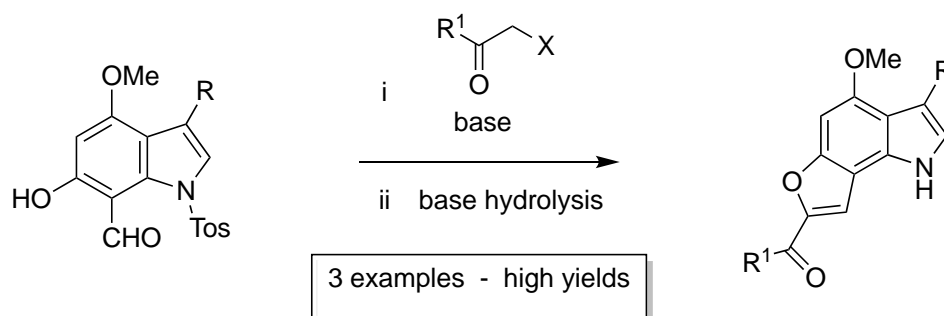
Accepted mm-dd-yyyy

Published on line mm-dd-yyyy

Dates to be inserted by editorial office

Abstract

3-(4-Chlorophenyl)-6-hydroxy-4-methoxy-1-(toluene-4-sulfonyl)indole-7-carbaldehyde undergoes alkylation at the phenolic OH group with α -halo ketones to give the corresponding ethers which can be cyclised with further base treatment to give *N*-tosylated furano[2,3-*g*]indoles. The indole NH can then be deprotected using potassium hydroxide in methanol to give furano[2,3-*g*]indoles.



Keywords: indoles, carbaldehydes, furan synthesis, phenols, alkylation, cyclisation

Introduction

Tricyclic structures such as 2*H*-furan[3,2-*f*]indoles **1** and 1*H*-furan[2,3-*g*]indoles **2** are not known, but are structurally closely related to the tetracyclic benzo[*b*]furanindoles **3** and **4**, as well as the indolobenzofurans **5** and furanocarbazole alkaloids of type **6** (Figure 1).

Compounds of type **3** are important antitumor agents against certain human breast cancer cells, epidermoid carcinoma cells and melanoma cells, and show low toxicity against normal cells.^{1,2} On the other hand, compounds of type **5** and **6** can be found in nature and are important alkaloids due to their pharmacological potential.³⁻¹²

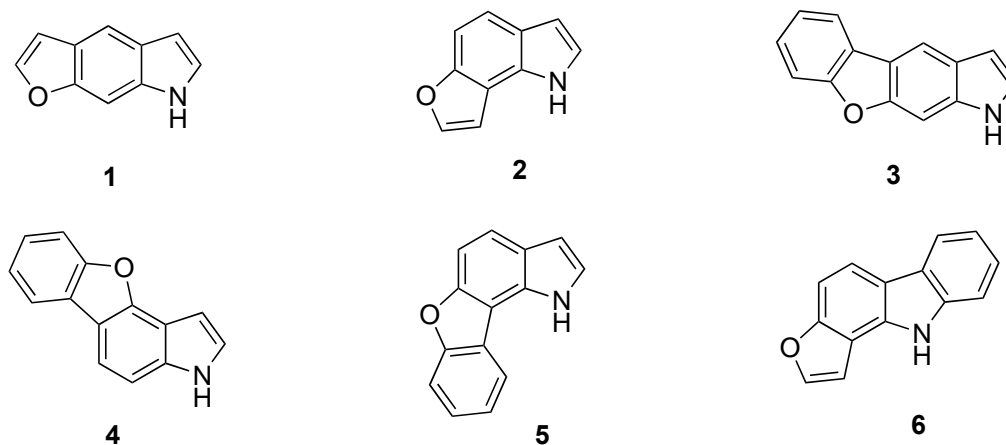
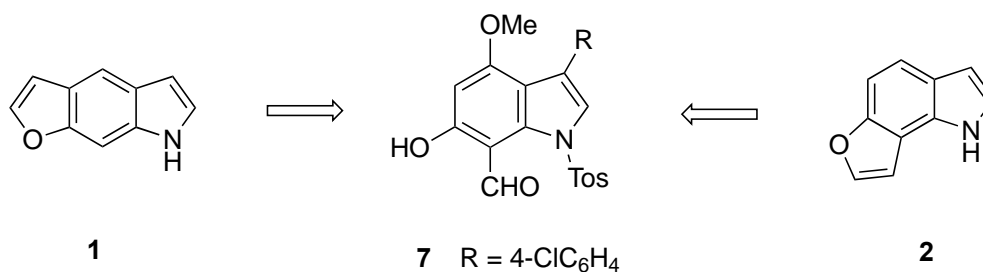


Figure 1. Some tri- and tetra-furanoindole structures **1-6**.

We have recently reported¹³ the preparation of the 6-hydroxy-4-methoxyindole-7-carbaldehyde **7**, and consequently the possible synthesis of examples of the novel tricyclic systems **1** and **2** was investigated (Scheme 1). While reaction at C5 is not usually observed in 4,6-dimethoxyindoles, it can occur in the less well protected 4-methoxy-6-hydroxyindoles. Several aspects of the following work have been mentioned in the report of a conference lecture.¹⁴



Scheme 1. Possible formation of furanoindoles from 6-hydroxyindole-7-carbaldehyde **7**.

Results and Discussion

Possible reaction paths

The benzofuran ring system, which is a substructure of the tricyclic structures **1** and **2**, can be synthesized from salicylaldehydes or activated phenols via α -aryloxyketones in two steps.^{15,16} The successful application of

those reactions to the 6-hydroxyindole **7** would possibly give derivatives of the heterocyclic scaffolds **1** and **2** (Scheme 1).

Base catalyzed alkylation and cyclisation

In order to avoid alkylation of the indole NH, the *N*-tosyl-protected 6-hydroxyindole **7** was chosen over a related unprotected indole. Alkylation of indole **7** with α -haloketones or esters would provide indolo-ethers, which could then potentially undergo intramolecular cyclisation to give derivatives of either the furano[2,3-*g*]indole **2** or the furano[3,2-*f*]indole **1**.

Preliminary experiments showed that the best results were obtained with potassium carbonate in boiling acetone. Other reaction conditions, such as sodium bicarbonate in acetone, potassium carbonate in ethanol, or sodium hydride in tetrahydrofuran, gave rather poor conversions or yielded product mixtures. Consequently, reaction of indole **7** with ethyl bromoacetate gave the indole ether **8** in 68% yield. A similar reaction with 2-bromopropiophenone gave the indole ether **9** in 62% yield. On the other hand, reaction of indole **7** with 2-bromoacetophenone and 2-bromo-4'-chloroacetophenone gave the furanoindoles **13** and **14** respectively, presumably via unstable indole ethers **10** and **11** respectively. The ether **10** could be isolated using a shorter reaction time, but was unstable and underwent rapid cyclisation to the furanoindole **13**: the ether **11** could not be isolated. Reaction of indole **7** with chloroacetone simply gave the furanoindole **15**, without any isolation or detection of the presumed intermediate **12** (Scheme 2).

Attempts to extend this cyclisation process to the indole ester **8** failed. The use of stronger bases, such as sodium ethoxide or sodium hydroxide, did not afford the cyclised product, but instead yielded the related free carboxylic acid **17** after work-up. Furthermore, base-catalysed cyclisation attempts with indole ether **9** resulted only in recovered starting material, since the intermediate cyclisation product is not able to aromatise to the furan ring.

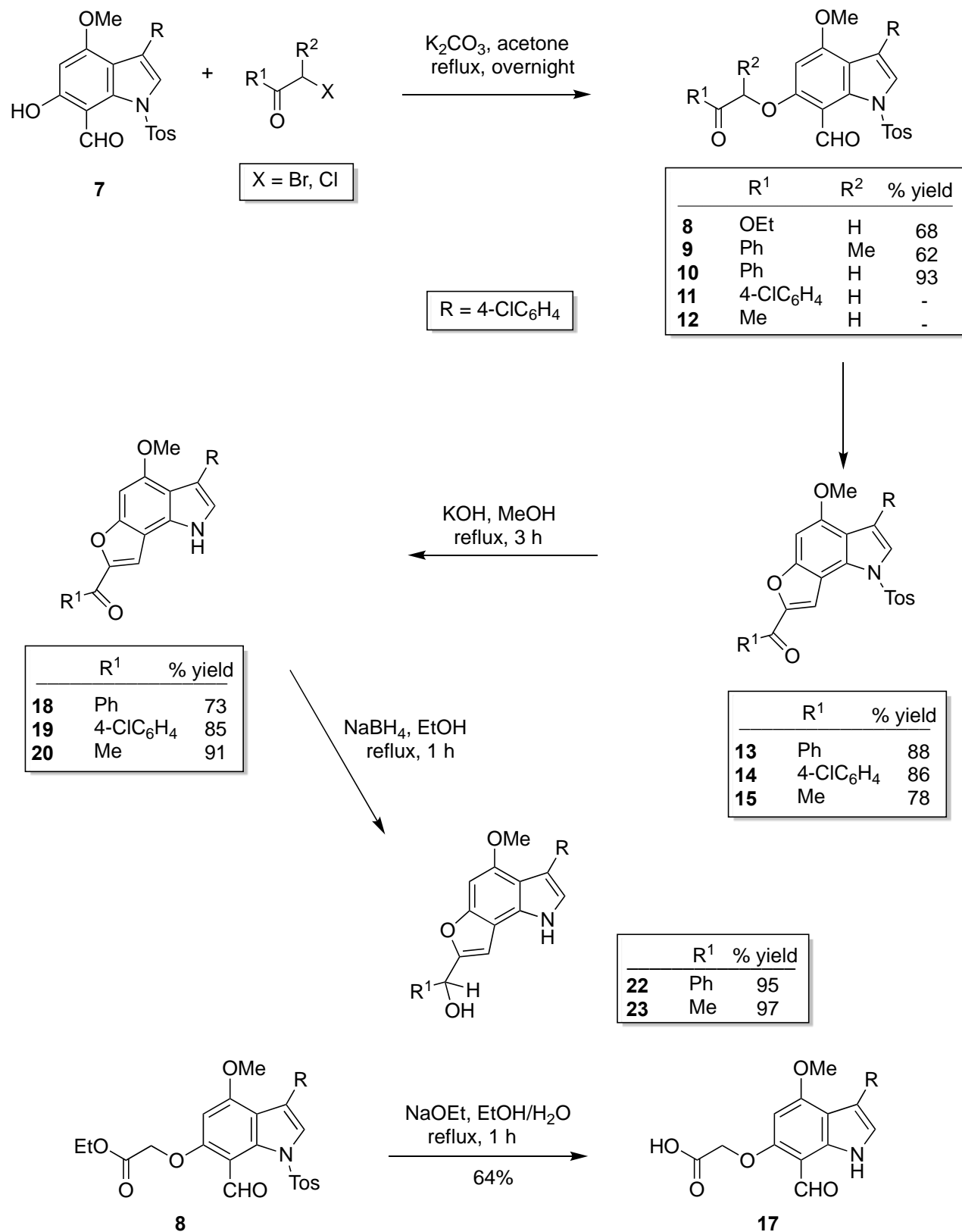
The tosyl group of the furanoindoles **13-15** could be removed easily with crushed potassium hydroxide in refluxing methanol, yielding the furanoindoles **18-20** respectively in 73-91% yield. The structures of these furanoindoles were confirmed by 1D and 2D NMR experiments.

Acid-catalysed cyclisation

On the basis that the synthesis of benzofurans can be achieved by intramolecular cyclisation from α -aryl(alkyl)oxyketones in the presence of trifluoroacetic acid,^{8,9} cyclisation of indole ethers **9** and **10** was attempted to give furano[3,2-*f*]indoles of type **1**. However, formation of the desired furano[3,2-*f*]indoles was not observed: indole ether **10** gave the furano[2,3-*g*]indole **13**, but indole ether **9** failed to react and only starting material was recovered.

Reduction of the carbonyl group

The carbonyl groups on the furanoindoles **18** and **20** were easily reduced using sodium borohydride in ethanol, giving excellent yields of the corresponding stable alcohols **22** and **23** (Scheme 2). These alcohols were of interest in that they incorporate a nucleophilic site at the indole C2 and an electrophilic site at the hydroxymethyl group of the furan ring. However, treatment of the alcohols **22** and **23** under a variety of acidic conditions led only to complex polymeric mixtures.



Scheme 2. Formation of furanoindoles **18-20**.

Conclusions

The activation of indoles leads to the synthesis of a 6-hydroxy-4-methoxyindole-7-aldehyde, which in turn can be converted by reaction with suitable α -haloketones to the previously unknown furano[2,3-*g*]indoles.

Experimental Section

General. Melting points were measured using a Mel-Temp melting point apparatus. Microanalyses were performed on Carlo Erba Elemental Analyser EA 1108 at the Campbell Microanalytical Laboratory, University of Otago, New Zealand. ^1H and ^{13}C NMR spectra were obtained on a Bruker DPX300 (300 MHz) spectrometer and internally referenced to solvent peaks. Mass spectra were recorded on either a Bruker Daltonics Bio Apex II FTICR MS (HRMS-ESI) at the School of Chemistry, University of New South Wales, or a Shimadzu LCMS QP 8000 (EI) at the University of Otago, New Zealand. Infrared spectra were recorded with a Thermo Nicolet 370 FTIR Spectrometer using KBr discs. Ultraviolet-visible spectra were recorded using a Varian Cary 100 Scan Spectrometer.

[3-(4-Chlorophenyl)-7-formyl-4-methoxy-1-(toluene-4-sulfonyl)indol-6-yloxy]acetic acid ethyl ester (8). A mixture of indole **7** (0.20 g, 0.44 mmol), ethyl bromoacetate (0.05 mL, 0.44 mmol) and potassium carbonate (0.12 g, 0.75 mmol) in acetone (20 mL) was heated under reflux overnight. The reaction mixture was diluted with dichloromethane (20 mL), washed with saturated aqueous ammonium chloride (2×10 mL), water (10 mL) and brine (10 mL), and dried over MgSO_4 . The solvent was evaporated and the product purified by column chromatography on silica gel ($\text{CH}_2\text{Cl}_2/\text{MeOH}$, 98:2) yielding indole **8** as an off-white solid (0.16 g, 0.30 mmol, 68%), mp (MeOH) 59–61 °C. IR (v max, cm^{-1}): 3419, 2980, 1755, 1693, 1590, 1369, 1199, 1171, 1089, 812, 666, 543. UV/Vis (λ max, nm, ϵ , $\text{cm}^{-1}\text{M}^{-1}$): 246 (37,100), 296 (12,900). ^1H NMR (300 MHz, CDCl_3): δ_{H} 1.29 (3H, t, J 7.1 Hz, CH_2Me), 2.36 (3H, s, Me), 3.74 (3H, s, OMe), 4.26 (2H, q, J 7.1 Hz, CH_2Me), 4.80 (2H, s, CH_2), 6.41 (1H, s, H5), 7.20 (2H, d, J 8.3 Hz, aryl), 7.32 (4H, br s, aryl), 7.40 (1H, s, H2), 7.57 (2H, d, J 8.7 Hz, aryl), 10.36 (1H, s, CHO). ^{13}C NMR (75 MHz, CDCl_3): δ_{C} 14.1 (CH_2Me), 21.5 (Me), 55.4 (OMe), 61.4 (CH_2), 69.1 (CH_2), 95.5 (C5), 111.4, 116.0 and 125.0 (aryl C), 126.6 (C2), 126.9, 128.7, 129.5 and 130.6 (aryl CH), 131.6, 133.4, 134.5, 137.2, 145.0, 157.6 and 158.4 (aryl C), 168.5 (CO), 187.0 (CHO). MS (+EI, m/z , %): 543 (42, MH^+), 388 (100). Anal. calcd for $\text{C}_{27}\text{H}_{24}\text{ClNO}_7\text{S}$: C, 59.83; H, 4.46; N, 2.58. Found: C, 59.67; H, 4.26; N, 2.52 %.

3-(4-Chlorophenyl)-4-methoxy-6-(1-methyl-2-oxo-2-phenylethoxy)-1-(toluene-4-sulfonyl)indole-7-carbaldehyde (9). A mixture of indole **7** (0.25 g, 0.55 mmol), 2-bromopropiophenone (0.1 mL, 0.55 mmol) and potassium carbonate (0.13 g, 0.93 mmol) in acetone (25 mL) was heated under reflux overnight. The mixture was diluted with dichloromethane (25 mL) and washed with both aqueous HCl (2M, 2×25 mL) and brine (2×25 mL) after which the organic layer was dried with Na_2SO_4 and the solvent removed under reduced pressure. The crude product was purified by column chromatography on silica gel ($\text{CH}_2\text{Cl}_2/\text{MeOH}$, 98:2) yielding indole **9** as an off-white solid (0.20 g, 0.34 mmol, 62%), mp 78–82 °C. IR (v max, cm^{-1}): 3010, 3065, 2937, 1694, 1590, 1369, 1171, 1089, 811, 703, 543. UV/Vis (λ max, nm, ϵ , $\text{cm}^{-1}\text{M}^{-1}$): 248 (44,700). ^1H NMR (300 MHz, CDCl_3): δ_{H} 1.83 (3H, br s, CHMe), 2.29 (3H, s, Me), 3.61 (3H, s, OMe), 5.68 (1H, br s, CHMe), 6.39 (1H, s, H5), 7.01–8.12 (14H, m, H2 and aryl), 10.34 (1H, s, CHO). ^{13}C NMR (75 MHz, $\text{DMSO}-d_6$): δ 19.0 (CHMe), 21.5 (Me), 55.3 (OMe), 81.5 (CHMe), 96.7 (C5), 111.2, 115.8 and 125.6 (aryl C), 126.5 (C2), 126.8, 127.7, 128.8, 128.9, 129.3 and 130.5 (aryl CH), 131.4 and 133.4 (aryl C), 133.8 (aryl CH), 133.9, 134.1, 138.0, 144.9, 157.7 and 157.9 (aryl C), 186.6 (CHO), 199.0 (CO). MS (+EI, m/z , %): 644 (19), 589 (32, MH^+), 434 (100), 312 (46), 301 (30). Anal. calcd for $\text{C}_{32}\text{H}_{26}\text{ClNO}_6\text{S}\cdot 0.5\text{CH}_2\text{Cl}_2$: C, 61.91; H, 4.32; N, 2.22. Found: C, 61.88; H, 4.25; N, 2.22 %.

3-(4-Chlorophenyl)-4-methoxy-6-(2-oxo-2-phenylethoxy)-1-(toluene-4-sulfonyl)indole-7-carbaldehyde (10). A mixture of indole **7** (0.35 g, 0.77 mmol), 2-bromoacetophenone (0.15 g, 0.77 mmol) and potassium

carbonate (0.18 g, 1.31 mmol) in acetone (35 mL) was heated under reflux overnight. The reaction mixture was diluted with aqueous HCl (2M, 40 mL), extracted with dichloromethane (2×35 mL), the combined organic layers washed with brine (25 mL) and dried over Na₂SO₄. The solvent was evaporated and the product purified by column chromatography on silica gel (CH₂Cl₂) yielding indole **10** as a yellow solid (0.41 g, 0.72 mmol, 93%), mp 206-208 °C. IR (ν max, cm⁻¹): 3429, 1699, 1599, 1494, 1373, 1175, 1134, 1090, 813, 679. UV/Vis (λ max, nm, ε, cm⁻¹M⁻¹): 237 (41,700), 369 (1,500). The compound was too unstable for ¹H NMR spectrum measurement, because of rapid cyclisation to compound **13**. MS (+EI, *m/z*, %): 573 (20, M⁺), 555 (59), 418 (100). Anal. calcd for C₃₁H₂₄ClNO₆S: C, 64.86; H, 4.21; N, 2.44. Found: C, 65.05; H, 4.16; N, 2.37 %.

[3-(4-Chlorophenyl)-4-methoxy-1-(toluene-4-sulfonyl)furano[2,3-g]indol-7-yl] phenyl ketone (13). Method A: A mixture of indole **7** (0.35 g, 0.77 mmol), 2-bromoacetophenone (0.15 g, 0.77 mmol) and potassium carbonate (0.18 g, 1.31 mmol) in acetone (35 mL) was heated under reflux for 24 h. The reaction mixture was diluted with aqueous HCl (2M, 40 mL), extracted with dichloromethane (2×35 mL), the combined organic layers washed with brine (25 mL) and dried over Na₂SO₄. The solvent was evaporated and the product purified by column chromatography on silica gel (CH₂Cl₂) yielding compound **13** as a bright yellow powder (377 mg, 0.678 mmol, 88%), mp 212-214 °C.

Method B: A solution of indole **10** (100 mg, 0.174 mmol) in trifluoroacetic acid (2.5 mL) was heated under reflux for 5 h and poured into ice-cold aqueous NaOH (0.2 M, 5 mL). Dichloromethane (5 mL) was added, the organic layer separated, dried with MgSO₄, evaporated *in vacuo* and the resulting green solid purified by flash chromatography on silica gel (CH₂Cl₂) yielding compound **13** as a bright yellow powder (79 mg, 0.143 mmol, 82%), mp 212-214 °C. IR (ν max, cm⁻¹): 3441, 2937, 1641, 1613, 1540, 1279, 1175, 1136, 1093, 677. UV/Vis (λ max, nm, ε, cm⁻¹M⁻¹): 230 (38,000), 369 (12,000). ¹H NMR (300 MHz, CDCl₃): δ_H 2.34 (3H, s, Me), 3.80 (3H, s, OMe), 6.98 (1H, s, H5), 7.22 (2H, d, *J* 7.9 Hz, tosyl), 7.36 (2H, d, *J* 6.4 Hz, aryl), 7.44 (2H, d, *J* 6.4 Hz, aryl), 7.54 (1H, s, H2), 7.55-7.70 (5H, m, aryl), 8.02 (2H, d, *J* 7.9 Hz, tosyl), 8.30 (1H, s, H8). ¹³C NMR (75 MHz, CDCl₃): δ_C 21.6 (Me), 55.5 (OMe), 90.2 (C5), 101.4 and 108.4 (aryl C), 117.1 (C8), 123.6 (C2), 123.9 (aryl C), 126.7, 127.7, 128.5, 129.3 and 130.1 (aryl CH), 130.9 (aryl C), 131.0 (aryl CH), 132.1 (aryl C), 132.7 (aryl CH), 133.2, 134.9, 137.5, 145.5, 150.7, 155.7, 156.9 (aryl C), 183.6 (CO). MS (+EI, *m/z*, %): 555 (26, M⁺), 400 (100), 350 (32). Anal. calcd for C₃₁H₂₂ClNO₅S.0.5CH₂Cl₂: C, 63.22; H, 3.87; N, 2.34. Found: C, 63.52; H, 3.96; N, 2.30 %.

(4-Chlorophenyl) [3-(4-chlorophenyl)-4-methoxy-1-(toluene-4-sulfonyl)furano[2,3-g]indol-7-yl] ketone (14).

A mixture of indole **7** (0.25 g, 0.55 mmol), 2-bromo-4-chloroacetophenone (0.13 g, 0.55 mmol) and potassium carbonate (0.13 g, 0.93 mmol) in acetone (25 mL) was heated under reflux overnight. The mixture was diluted with dichloromethane (25 mL) and washed with both aqueous HCl (2M, 2×25 mL) and brine (2×25 mL), the organic layer dried with Na₂SO₄ and the solvent removed under reduced pressure. The crude product was purified by column chromatography on silica gel (CH₂Cl₂/light petroleum, 1:1) yielding compound **14** as a bright yellow powder (0.28 g, 0.47 mmol, 86%), mp 236-238 °C. IR (ν max, cm⁻¹): 3119, 2937, 2831, 1646, 1611, 1490, 1375, 1279, 1176, 1091, 1014, 816, 677, 544. UV/Vis (λ max, nm, ε, cm⁻¹M⁻¹): 232 (42,700), 267 (25,100), 374 (20,900). ¹H NMR (300 MHz, DMSO-*d*₆): δ_H 2.30 (3H, s, Me), 3.79 (3H, s, OMe), 7.34 (1H, s, H5), 7.41 (2H, d, *J* 8.3 Hz, aryl), 7.44 (2H, d, *J* 8.3 Hz, aryl), 7.55 (2H, d, *J* 8.3 Hz, aryl), 7.74 (2H, d, *J* 8.3 Hz, aryl), 7.79 (2H, d, *J* 8.7 Hz, aryl), 7.84 (1H, s, H2), 7.99 (2H, d, *J* 8.7 Hz, aryl), 8.15 (1H, s, H8). ¹³C NMR (75 MHz, DMSO-*d*₆): δ_C 21.4 (Me), 56.4 (OMe), 60.6 and 72.7 (aryl C), 91.6 (C5), 107.8 (aryl C), 116.5 (C8), 123.6 (aryl C), 124.8 (C2), 127.1 and 127.9 (aryl CH), 128.0 (aryl C), 129.3, 131.0, 131.2 and 132.0 (aryl CH), 132.4, 134.2, 136.1, 138.3, 146.5, 150.7, 156.1 and 156.9 (aryl C), 181.6 (CO). MS (+EI, *m/z*, %): 591 (95, MH⁺), 437 (100). Anal. calcd for C₃₁H₂₁Cl₂NO₅S: C, 63.06; H, 3.58; N, 2.37. Found: C, 62.81; H, 3.53; N, 2.38 %.

1-[3-(4-Chlorophenyl)-4-methoxy-1-(toluene-4-sulfonyl)furano[2,3-g]indol-7-yl]ethanone (15). A mixture of indole **7** (0.25 g, 0.55 mmol), chloroacetone (44 μL, 0.55 mmol) and potassium carbonate (0.13 g, 0.93 mmol)

in acetone (25 mL) was heated under reflux overnight. The mixture was diluted with dichloromethane (50 mL) and washed both with aqueous HCl (2M, 2×25 mL) and brine (2×25 mL), the organic layer dried with Na₂SO₄ and the solvent removed under reduced pressure. The crude product was purified by column chromatography on silica gel (CH₂Cl₂) yielding compound **15** as a white powder (0.21 mg, 0.43 mmol, 78%), mp 236-238 °C.

IR (ν max, cm⁻¹): 3437, 3137, 1667, 1611, 1493, 1372, 1276, 1179, 1142, 1093, 812, 680, 597. UV/Vis (λ max, nm, ε, cm⁻¹M⁻¹): 232 (44,700), 312 (20,000), 348 (20,100). ¹H NMR (300 MHz, CDCl₃): δ_H 2.34 (3H, s, Me), 2.62 (3H, s, Me), 3.77 (3H, s, OMe), 6.91 (1H, s, H5), 7.21 (2H, d, *J* 8.3 Hz, aryl), 7.34 (2H, d, *J* 8.7 Hz, aryl), 7.41 (2H, d, *J* 8.7 Hz, aryl), 7.51 (1H, s, H2), 7.68 (2H, d, *J* 8.3 Hz, aryl), 8.34 (1H, s, H8). ¹³C NMR (75 MHz, DMSO-*d*₆): δ_C 21.4 (Me), 26.7 (Me), 56.3 (OMe), 91.5 (C5), 107.8 (aryl C), 113.2 (C8), 116.3 and 123.6 (aryl C), 124.7 (C2), 127.2 and 127.8 (aryl CH), 128.1 (aryl C), 130.9 and 132.0 (aryl CH), 132.4, 132.4, 134.2, 146.4, 151.4, 155.6, 156.2 (aryl C), 187.2 (CO). MS (+EI, *m/z*, %): 495 (100, MH⁺), 340 (42). Anal. calcd for C₂₆H₂₀ClNO₅·0.25H₂O: C, 62.65; H, 4.15; N, 2.81. Found: C, 62.52; H, 4.02; N, 2.81 %.

[3-(4-Chlorophenyl)-7-formyl-4-methoxyindol-6-yloxy]acetic acid (17). A solution of indole **8** (0.15 g, 0.28 mmol) in ethanolic sodium ethoxide (12.5%, 5 mL) was heated under reflux for 1 h, the suspension concentrated and the precipitate filtered off, washed with water until rinsings were neutral yielding indole **17** as a pale pink solid (0.64 g, 0.18 mmol, 64%), mp (CHCl₃) > 300 °C. IR (ν max, cm⁻¹): 3419, 2939, 1629, 1592, 1318, 1273, 1207, 1095, 935, 838, 791. UV/Vis (λ max, nm, ε, cm⁻¹M⁻¹): 252 (20,400), 328 (11,200). ¹H NMR (300 MHz, DMSO-*d*₆): δ_H 3.82 (3H, s, OMe), 4.33 (2H, s, CH₂), 6.30 (1H, s, H5), 7.13 (1H, d, *J* 1.9 Hz, H2), 7.35 (2H, d, *J* 8.6 Hz, aryl), 7.50 (2H, d, *J* 8.6 Hz, aryl), 10.36 (1H, s, CHO), 11.35 (1H, br s, NH). ¹³C NMR (75 MHz, DMSO-*d*₆): δ_C 55.7 (OMe), 69.9 (CH₂), 89.7 (C5), 104.8 and 116.3 (aryl C), 123.5 (C2), 127.6 (aryl C), 127.9 (aryl CH), 130.5 (aryl C), 131.0 (aryl CH), 134.9, 136.4, 160.7 and 163.7 (aryl C), 169.8 (CO₂H), 187.8 (CHO). MS (+EI, *m/z*, %): 361 (30, MH⁺), 315 (100), 299 (96), 287 (56). Anal. calcd for C₁₈H₁₄ClNO₅·0.75CHCl₃: C, 51.20; H, 3.64; N, 3.02. Found: C, 51.40; H, 3.32; N, 3.10 %.

[3-(4-Chlorophenyl)-4-methoxyfuro[2,3-*g*]indol-7-yl] phenyl ketone (18). A mixture of indole **13** (70 mg, 0.126 mmol) and crushed potassium hydroxide (140 mg, 2.5 mmol) in methanol (4 mL) was heated under reflux for 3 h, neutralised with acetic acid (2M) and the resulting precipitate filtered off, washed with water until rinsings were neutral and dried *in vacuo* yielding compound **18** as a bright yellow powder (37 mg, 0.092 mmol, 73%), mp (CHCl₃) 252-254 °C. IR (ν max, cm⁻¹): 3226, 2997, 1611, 1532, 1487, 1280, 1195, 1130, 485, 723. UV/Vis (λ max, nm, ε, cm⁻¹M⁻¹): 233 (30,200), 261 (13,500), 386 (18,200). ¹H NMR (300 MHz, CDCl₃): δ_H 3.89 (3H, s, OMe), 6.86 (1H, s, H5), 7.13 (1H, d, *J* 2.3 Hz, H2), 7.35 (2H, d, *J* 8.3 Hz, aryl), 7.50-7.65 (5H, m, aryl), 7.69 (1H, s, H8), 8.05 (2H, d, *J* 8.3 Hz, aryl), 8.70 (1H, br s, NH). ¹³C NMR (75 MHz, CDCl₃): δ_C 56.0 (OMe), 86.7 (C5), 107.8, 112.0, 116.4 and 117.9 (aryl C), 123.0 (C8), 127.8 (C2), 129.0 and 129.2 (aryl CH), 130.3 (aryl C), 130.6, 131.4 and 132.8 (aryl CH), 135.2, 138.1, 150.1, 155.9 and 157.2 (aryl C), 182.3 (CO). MS (+EI, *m/z*, %): 401 (90, M⁺), 351 (100). Anal. calcd for C₂₄H₁₆ClNO₃·0.25CHCl₃: C, 67.47; H, 3.79; N, 3.24. Found: C, 67.74; H, 3.82; N, 3.20 %.

(4-Chlorophenyl) [3-(4-chlorophenyl)-4-methoxyfurano[2,3-*g*]indol-7-yl] ketone (19). A mixture of indole **14** (100 mg, 0.169 mmol) and crushed potassium hydroxide (190 mg, 3.4 mmol) in methanol (5 mL) was heated under reflux for 3 h, neutralised with acetic acid (2M) and the resulting precipitate filtered off, washed with water until rinsings were neutral, and dried *in vacuo* yielding compound **19** as a bright yellow powder (63 mg, 0.144 mmol, 85%), mp 286-288 °C. IR (ν max, cm⁻¹): 3230, 1592, 1530, 1486, 1277, 1163, 1092, 1014, 887, 753. UV/Vis (λ max, nm, ε, cm⁻¹M⁻¹): 233 (36,300), 269 (21,900), 392 (36,300). ¹H NMR (300 MHz, DMSO-*d*₆): δ_H 3.86 (3H, s, OMe), 7.00 (1H, s, H5), 7.38 (2H, d, *J* 8.7 Hz, aryl), 7.41 (1H, d, *J* 2.6 Hz, H2), 7.53 (2H, d, *J* 8.7 Hz, aryl), 7.67 (2H, d, *J* 8.6 Hz, aryl), 7.87 (1H, d, *J* 0.8 Hz, H8), 8.00 (2H, d, *J* 8.6 Hz, aryl), 12.01 (1H, d, *J* 1.5 Hz, NH).

^{13}C NMR (75 MHz, DMSO- d_6): δ_{C} 56.0 (OMe), 86.7 (C5), 107.8 and 112.0 (aryl C), 116.3 (C8), 117.9 (aryl C), 122.9 (C2), 127.8 and 129.1 (aryl CH), 130.1 and 130.6 (aryl C), 131.1 and 131.4 (aryl CH), 135.1, 136.7, 137.7, 150.0, 156.0 and 157.3 (aryl C), 181.0 (CO). MS (+EI, m/z , %): 437 (100, MH^+), 325 (18). Anal. calcd for $\text{C}_{24}\text{H}_{15}\text{Cl}_2\text{NO}_3 \cdot 0.5\text{H}_2\text{O}$: C, 64.73; H, 3.62; N, 3.15. Found: C, 64.91; H, 3.40; N, 3.10 %.

1-[3-(4-Chlorophenyl)-4-methoxyfurano[2,3-*g*]indol-7-yl]ethanone (20). A mixture of indole **15** (60 mg, 0.117 mmol) and crushed potassium hydroxide (136 mg, 2.34 mmol) in methanol (3 mL) was heated under reflux for 3 h, neutralised with acetic acid (2M), and the resulting precipitate filtered off, washed with water until rinsings were neutral, and dried *in vacuo* yielding compound **20** as a pale brown powder (36 mg, 0.106 mmol, 91%), mp (CH_2Cl_2) 280-284 °C. IR (v max, cm^{-1}): 3276, 1634, 1544, 1486, 1355, 1279, 1160, 1127, 938, 838. UV/Vis (λ max, nm, ϵ , $\text{cm}^{-1}\text{M}^{-1}$): 233 (30,200), 358 (20,400). ^1H NMR (300 MHz, DMSO- d_6): δ_{H} 2.51 (3H, s, Me), 3.83 (3H, s, OMe), 6.94 (1H, s, H5), 7.37 (2H, d, J 8.3 Hz, aryl), 7.41 (1H, s, H2), 7.53 (2H, d, J 8.3 Hz, aryl), 7.85 (1H, s, H8), 12.03 (1H, br s, NH). ^{13}C NMR (75 MHz, DMSO- d_6): δ_{C} 26.4 (Me), 55.9 (OMe), 86.7 (C5), 107.5 and 112.0 (aryl C), 112.4 (C8), 117.8 (aryl C), 122.8 (C2), 127.8 (aryl CH), 130.2 and 130.6 (aryl C), 131.4 (aryl CH), 135.2, 150.8, 155.1 and 156.7 (aryl C), 186.4 (CO). MS (+EI, m/z , %): 341 (100, MH^+). Anal. calcd for $\text{C}_{19}\text{H}_{14}\text{ClNO}_3 \cdot 0.25\text{CH}_2\text{Cl}_2$: C, 64.05; H, 4.05; N, 3.88. Found: C, 63.92; H, 4.04; N, 3.88 %.

[3-(4-Chlorophenyl)-4-methoxyfurano[2,3-*g*]indol-7-yl](phenyl)methanol (22). A mixture of indole **18** (60 mg, 0.149 mmol) and sodium borohydride (56.4 mg, 1.49 mmol) was heated under reflux in ethanol (3 mL) for 1 h and solvent removed *in vacuo*. The residue was suspended in water (10 mL), filtered off, washed with water until rinsings were neutral yielding compound **22** as a yellow powder (0.57 mg, 0.141 mmol, 95%), mp 82-84 °C. IR (v max, cm^{-1}): 3425, 2942, 1642, 1601, 1489, 1399, 1194, 1110, 1015, 787, 696. UV/Vis (λ max, nm, ϵ , $\text{cm}^{-1}\text{M}^{-1}$): 236 (37,200), 298 (23,400). ^1H NMR (300 MHz, DMSO- d_6): δ_{H} 3.73 (3H, s, OMe), 5.82 (1H, d, J 4.5 Hz, CHOH), 6.11 (1H, s, J 4.5 Hz, OH), 6.69 (1H, s, H8), 6.79 (1H, s, H5), 7.27 (1H, s, H2), 7.33-7.52 (9H, m, aryl), 11.67 (1H, br s, NH). ^{13}C NMR (75 MHz, DMSO- d_6): δ_{C} 55.6 (OMe), 69.2 (CHOH), 87.0 (C5), 101.2 (C8), 107.0, 111.3 and 117.1 (aryl C), 122.1 (C2), 127.0, 127.6, 127.7 and 128.5 (aryl CH), 130.0 and 130.2 (aryl C), 131.3 (aryl CH), 137.7, 142.9, 152.4, 152.8 and 157.6 (aryl C). MS (+EI, m/z , %): 402 (23, $[\text{M}-\text{H}]^+$), 386 (100, $[\text{M}-\text{OH}]^+$). Anal. calcd for $\text{C}_{24}\text{H}_{18}\text{ClNO}_3 \cdot 0.5\text{H}_2\text{O}$: C, 69.82; H, 4.64; N, 3.39. Found: C, 69.73; H, 4.69; N, 3.20 %.

1-[3-(4-Chlorophenyl)-4-methoxyfurano[2,3-*g*]indol-7-yl]ethanol (23). A mixture of indole **20** (80 mg, 0.235 mmol) and sodium borohydride (89 mg, 2.35 mmol) was heated under reflux in ethanol (4 mL) for 1 h and the solvent removed *in vacuo*. The residue was suspended in water (15 mL), filtered off, washed with water until rinsings were neutral yielding compound **23** as a white powder (78 mg, 0.227 mmol, 97%), mp 223-225 °C. IR (v max, cm^{-1}): 3485, 3240, 2980, 2821, 1643, 1488, 1403, 1338, 1198, 1130, 1091, 1067, 1014, 793. UV/Vis (λ max, nm, ϵ , $\text{cm}^{-1}\text{M}^{-1}$): 236 (37,200), 296 (20,000). ^1H NMR (300 MHz, DMSO- d_6): δ_{H} 1.46 (3H, d, J 6.4 Hz, Me), 3.76 (3H, s, OMe), 4.79-4.87 (1H, m, CHOH), 5.37 (1H, d, J 5.3 Hz, CHOH), 6.79 (1H, s, H5), 6.82 (1H, s, H2), 7.30 (1H, d, J 2.6 Hz, H8), 7.35 (2H, d, J 8.7 Hz, aryl), 7.53 (2H, d, J 8.7 Hz, aryl), 11.70 (1H, br s, NH). ^{13}C NMR (75 MHz, DMSO- d_6): δ_{C} 22.5 (Me), 55.6 (OMe), 62.7 (CHOH), 87.0 (C5), 99.2 (C8), 107.1, 111.2 and 117.1 (aryl C), 122.1 (C2), 127.7 (aryl CH), 130.1 and 130.2 (aryl C), 131.3 (aryl CH), 135.8, 152.3, 152.4 and 159.3 (aryl C). MS (+EI, m/z , %): 340 (8, $[\text{M}-\text{H}]^+$), 324 (100, $[\text{M}-\text{OH}]^+$). Anal. calcd for $\text{C}_{19}\text{H}_{16}\text{ClNO}_3$: C, 66.76; H, 4.72; N, 4.10. Found: C, 66.37; H, 5.01; N, 3.86 %.

Acknowledgements

Financial support from the Australian Research Council is gratefully acknowledged. K.P. also acknowledges receipt of an International Postgraduate Research Scholarship from the Australian Government and UNSW.

Supplementary Material

References

1. Chunchatprasert, L.; Rao, K. R. N.; Shannon, P. V. R. *J. Chem. Soc., Perkin Trans. 1* **1992**, 1779-83.
<https://doi.org/10.1039/P19920001779>
2. Rao, N. K. R.; Stables, J. N.; Shannon, P. V. R.; Rao, N. K. R.; Chunchatprasert, L. WO. Patent 9402483, 1994.
3. Knölker, H-J.; Fröhner, W. *Tetrahedron Lett.* **1996**, 37, 9183-9186.
[https://doi.org/10.1016/S0040-4039\(96\)02181-8](https://doi.org/10.1016/S0040-4039(96)02181-8)
4. Hagiwara, H.; Choshi, T.; Fujimoto, H.; Sugino, E.; Hibino, S. *Chem. Pharm. Bull.* **1998**, 46, 1948-1949.
<https://doi.org/10.1248/cpb.46.1948>
5. Soos, T.; Timari, G.; Hajos, G. *Tetrahedron Lett.* **1999**, 40, 8607-8609.
[https://doi.org/10.1016/S0040-4039\(99\)01803-1](https://doi.org/10.1016/S0040-4039(99)01803-1)
6. Knölker, H-J.; Fröhner, W. *Synthesis* **2000**, 2131-2136.
<https://doi.org/10.1055/s-2000-8714>
7. Hagiwara, H.; Choshi, T.; Nobuhiro, J.; Fujimoto, H.; Hibino, S. *Chem. Pharm. Bull.* **2001**, 49, 881-886.
<https://doi.org/10.1248/cpb.49.881>
8. Yasuhara, A.; Suzuki, N.; Sakamoto, T. *Chem. Pharm. Bull.* **2002**, 50, 143-145.
<https://doi.org/10.1248/cpb.50.143>
9. Knölker, H-J.; Krahl, M. P. *Synlett* **2004**, 528-530.
<https://doi.org/10.1055/s-2004-815417>
10. Fröhner, W.; Krahl, M. P.; Reddy, K. R.; Knölker, H-J. *Heterocycles* **2004**, 63, 2393-2407.
<https://doi.org/10.3987/REV-04-585>
11. Krahl, M. P.; Schmidt, A. W.; Knölker, H-J. *Heterocycles* **2012**, 86, 357-370.
[https://doi.org/10.3987/COM-12-S\(N\)24](https://doi.org/10.3987/COM-12-S(N)24)
12. Spindler, B.; Kataeva, O.; Knölker, H-J. *J. Org. Chem.* **2018**, 83, 15136-15143.
<https://doi.org/10.1021/acs.joc.8b02426>
13. Pchalek, K.; Condie, G. C.; Kumar, N.; Black, D. StC. *Arkivoc* **2020**,(viii), 1.
<https://doi.org/10.24820/ark.5550190.p011.405>
14. Black, D. StC.; Channon, M. F.; Clayton, K. A.; Condie, G. C.; Harper, J. B.; Kumar, N.; Pchalek, K.; Wahyuningsih, T. D. *Arkivoc* **2006**, 67-75.
<https://doi.org/10.3998/ark.5550190.0007.707>
15. Burgstahler, A. W.; Worden, L. R. *Org. Synth.* **1966**, 46, 28-31.
<https://doi.org/10.15227/orgsyn.046.0028>
16. Black, D. StC.; Rezaie, R. *Tetrahedron Lett.* **1999**, 40, 4251-4254.
[https://doi.org/10.1016/S0040-4039\(99\)00699-1](https://doi.org/10.1016/S0040-4039(99)00699-1)

This paper is an open access article distributed under the terms of the Creative Commons Attribution (CC BY) license (<http://creativecommons.org/licenses/by/4.0/>)