

Synthesis of novel dicyclohexeno-crown ethers with strategically incorporated double bonds for further functionalization

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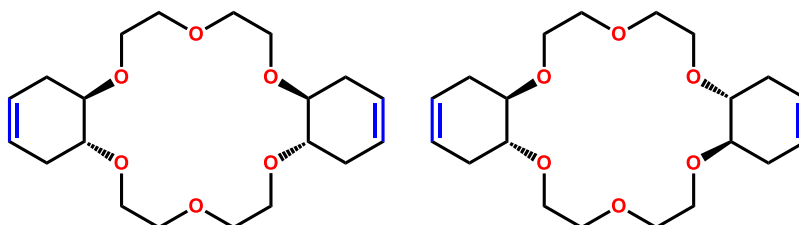
Received 02-13-2020

Accepted 03-06-2020

Published on line 03-29-2020

Abstract

The number of applications found for crown ether-based materials continues to expand. Many of these applications feature dicyclohexano-18-crown-6 (DC18C6). The number of its derivatives, however, is very limited. To provide a suitable starting structure for the preparation of diverse functionalized DC18C6, the *trans-syn-trans* and *trans-anti-trans* isomers of dicyclohexeno-18-crown-6 have been synthesized for the first time through a convenient three-step synthetic procedure, yielding both isomers simultaneously. An efficient chromatographic separation resulted in the isolation of the pure isomers, and single crystal X-ray crystallography determined the solid-state structure of these crown ethers.



Keywords: Crown ether, dicyclohexeno-18-crown-6, dicyclohexano-18-crown-6, crown ether functionalization

Introduction

Ever since their first discovery by Pedersen,¹ crown ethers have seen a tremendous increase in their stand-alone applications,^{2–5} as well as their utility as fundamental building blocks comprising higher order structures in the shape of cryptands and (pseudo)rotaxanes.^{6,7} Their unique suitability in host-guest chemistry has sparked the emergence of supramolecular chemistry and molecular machinery.^{2,4,5,8} Additionally, crown ether-based compounds are used for the selective binding of ions, in (phase-transfer) catalysis, membrane transport, and as antitumor agents, either in solution or attached to a solid support.^{3,9–11} Evidently, the range of purposes found for crown ether-based materials is both incredibly broad, and still expanding. Many applications feature dibenzo- (DB18C6) and dicyclohexano-18-crown-6 (DC18C6),^{1,3,9–11} and both their synthesis and physical properties are well described.^{1,12–16} Between these two similar structures, DB18C6 can be further functionalized through incorporation of additional substituents onto the catechol moiety. DC18C6, however, contains a relatively inert cyclohexane ring that severely limits any potential for subsequent functionalization. To this extent, we sought to develop a novel type of dicyclohexene-based crown ether that allows for additional modification of the 6-membered ring, opening up new possibilities for use in a wide range of applications. We envisioned the strategic incorporation of π -bonds that could further be modified in a variety of ways dependent on the final purpose of the target molecule. The well-defined conformational behavior of cyclohexane-based compounds makes these molecules ideal scaffolds for applications in supramolecular chemistry,^{4,5} where precise control over the shape of the structure is highly desirable. Our goal was to design a quick and convenient synthesis, without the use of any protecting-group strategies, to obtain two of the five possible isomers of dicyclohexeno-18-crown-6 (DCE18C6) (Figure 1) containing a single separation of isomers in the final step. To the best of our knowledge, the synthesis of this novel type of crown ether has not been reported before.

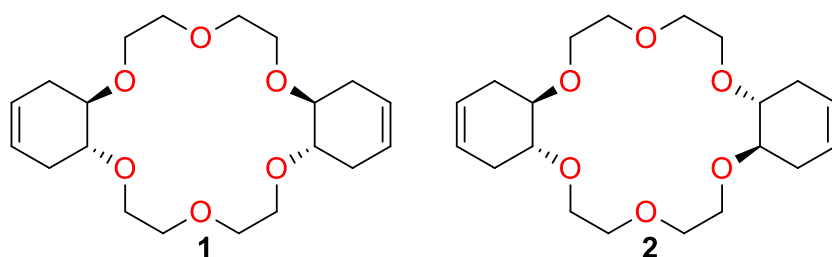
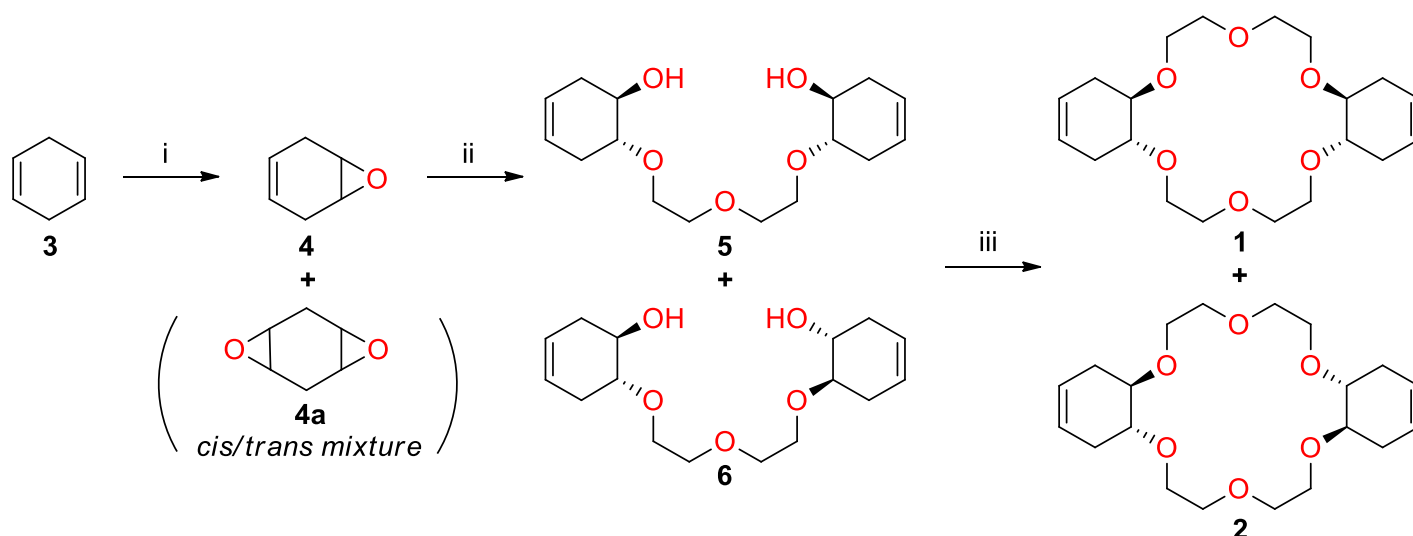


Figure 1. *Trans-syn-trans* (**1**) and *trans-anti-trans* (**2**) isomers of dicyclohexene-18-crown-6.

Results and Discussion

The synthesis of crown ethers **1** and **2** is based on a general strategy, as previously published for the synthesis of DC18C6,¹⁵ shown in Scheme 1. Starting from the commercially available 1,4-cyclohexadiene **3**, epoxide **4** was obtained through reaction with *meta*-chloroperoxybenzoic acid (*m*CPBA). In a first attempt, significant quantities of bis-epoxide **4a** (*cis/trans* mixture) were formed in addition to the desired product. The epoxidation of **4** to **4a** appears minimally hampered by increased steric hindrance in comparison to the diene **3**. To this extent, we conducted an optimization experiment by varying both temperature and the equivalents of *m*CPBA used. We monitored the ratios between the products formed using gas chromatography in an

attempt to establish the optimal conditions for this reaction. This effort led to an increase in yield from 50% to 70% of the pure epoxide **4** after vacuum distillation.



Scheme 1. Three-step simultaneous synthesis of *trans-syn-trans* (**1**) and *trans-anti-trans* (**2**) isomers of DCE18C6. (i) *m*CPBA, DCM, rt, 12 h, 70%. (ii) di(ethylene glycol), NaH, 60 °C, 42 h, 32%. (iii) di(ethylene glycol) ditosylate, 60% NaH, DMSO, rt, 18 h, 17% (**1**), 25% (**2**).

The residue consisted of crude bisepoxides **4a**, in a 57:43 ratio favoring the *cis* isomer, in accordance with the literature.^{17,18} The slight preference for the *cis*-isomer can possibly be explained by a hydrogen-bonding-effect between *m*CPBA and the epoxide oxygen on **4**. Next, we evaluated several reaction conditions to construct the half-crown intermediates **5** and **6**, and obtained our best results employing a modified solvent-free procedure based on the work of Yamato *et al.*¹⁵ Two equivalents of epoxide **4** were reacted with diethylene glycol in the presence of a catalytic amount of sodium hydride at 60°C. This resulted in a combined 32% yield for the *trans-syn-trans* (**5**) and *trans-anti-trans* (**6**) diols after chromatography. Cyclization to the desired dicyclohexeno-18-crown-6 was conducted following a procedure as described by Huber and Dietz.¹³ The diols were transformed into their corresponding alkoxides using stoichiometric quantities of sodium hydride, after which they were allowed to react with di(ethyleneglycol) ditosylate to give the *trans-syn-trans* (**1**) and *trans-anti-trans* (**2**) isomers of DCE18C6 (1:1 ratio by NMR). To our disappointment, attempts to separate the isomers through crystallization were unsatisfactory, in contrast to what had been reported for the analogous DC18C6 isomers.¹⁵ Luckily, the products could be separated by chromatography on silica and, in contrast to the literature,¹⁵ both compounds were obtained in pure form. The *meso* isomer **1** and racemic isomer **2** were isolated in 17% and 25% yields, respectively.

The relative stereochemistry of **1** and **2** were determined unequivocally by X-ray crystallography. Suitable crystals of both samples were obtained through slow evaporation of *n*-hexane. The *trans-syn-trans*- DCE18C6 (**1**) crystallizes in the triclinic space group P-1 with two half-molecule fragments in the asymmetric unit straddling an inversion center.¹⁹ The two fragments are nearly superimposable on each other with only slight deviations in the dihedral angles around the ethylene linkages. Both fragments contain an identical *trans-syn-trans* stereochemistry around the cyclohexene moieties. An ORTEP diagram for one of the molecules of *trans-syn-trans*-DCE18C6 (**1**) is shown on the left in Figure 2. The *trans-anti-trans* isomer, (**2**), crystallizes in the orthorhombic space group Pbca with one complete molecule in the asymmetric unit.¹⁹ The *trans-anti-trans* isomer's relative stereochemistry is clearly displayed in the ORTEP drawing shown on the right in Figure 2.

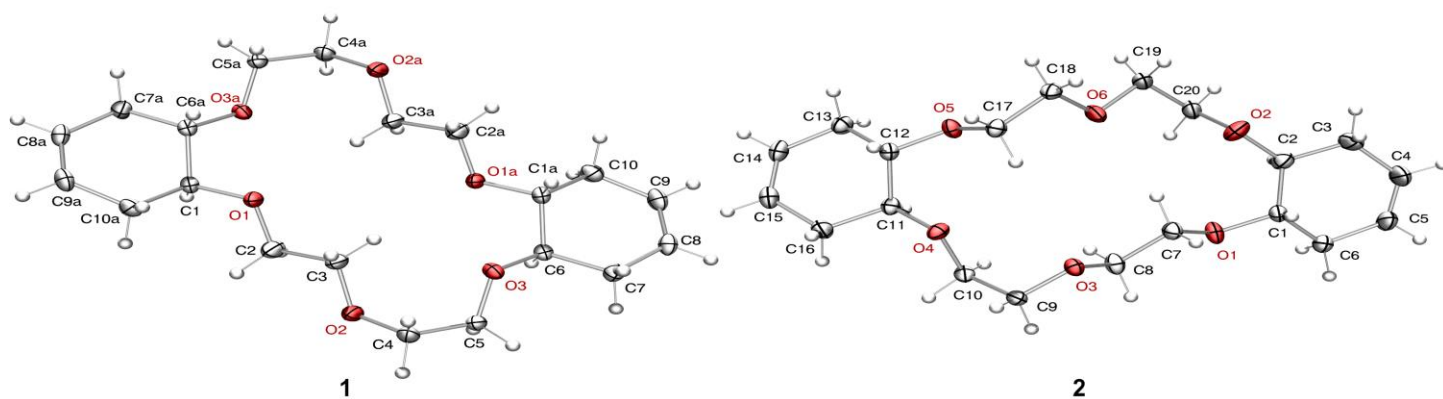


Figure 2. ORTEP diagram of *trans-syn-trans* DCE18C6 (**1**) and *trans-anti-trans* DCE18C6 (**2**).

Conclusions

We have developed the first synthesis of two stereoisomers of dicyclohexeno-18-crown-6 in order to provide suitable starting structures for the future preparation of diverse functionalized DC18C6. A convenient 3-step route combined with a chromatographic separation of the final products resulted in the isolation of the pure crown ethers. Future work includes further functionalization to more advanced materials, and optimization of the synthetic protocol. The possible changes of metal complexation by dicyclohexeno-18-crown-6 compared to the saturated analogues is also of substantial interest.

Experimental Section

General. Starting reagents were purchased from commercial sources (Sigma Aldrich, TCI, Alfa Aesar) and used without additional purification. All solvents were dried/purified by conventional techniques. The reactions were monitored by TLC on silica gel (8 x 2 cm) plates from Analtech Inc. (visualization by UV light and/or staining with I_2). Column chromatography was performed on silica gel (40-75 μ m, Sorbent Technologies). 1H NMR and ^{13}C NMR spectra were acquired on JEOL ECA-600 NMR-spectrometer (600 MHz for 1H and 151 MHz for ^{13}C). High resolution mass spectra (HRMS) were obtained on a JEOL AccuTOF time-of-flight mass spectrometer (Peabody, MA) coupled with an Ionsense DART open-air ionization source (Saugus, MA). Crystallographic data were collected at 150K using a Bruker Smart Apex II single crystal diffractometer with MoK_{α} ($\lambda = 0.71073$ Å) using omega and psi scans. Frames were integrated with the Bruker SAINT software package using a narrow-frame algorithm. Data were corrected for absorption effects using the Multi-Scan method (SADABS). The structures were solved and refined using the Bruker SHELXTL Software Package.

Synthesis of 7-oxa-bicyclo[4.1.0]hept-3-ene (4**).** To a stirred solution of 1,4-cyclohexadiene (20.0 mL, 205 mmol) in dichloromethane (1 L) at 0°C was added *m*CPBA (55.4 g, 238 mmol) in a portion-wise fashion. The mixture was subsequently stirred at 0°C for 1h, followed by stirring at RT for 16h. Next, 5% w/w Na_2SO_3 (aq.) (400 mL) was added, followed by vigorous stirring for 1h at RT. The water layer was extracted with DCM (2x200 mL). The combined organic extracts were washed with saturated $NaHCO_3$ (aq.) (2x250 mL) and brine

(1x100 mL), dried over anhydrous Na_2SO_4 , and concentrated *in vacuo*. The residual yellow oil was subjected to vacuum distillation (20 mbar, 47–49°C), yielding the pure epoxide **4** as a colorless oil (13.73 g, 70%). $^1\text{H-NMR}$ (600 MHz, CDCl_3) δ 5.44 (dd, 2H), 3.24–3.25 (m, 2H), 2.58–2.60 (m, 1H), 2.54–2.56 (m, 1H), 2.46–2.47 (m, 1H), 2.43 (m, 1H). $^{13}\text{C-NMR}$ (151 MHz, CDCl_3) δ 121.67, 51.15, 25.11. HRMS: calculated for $\text{C}_6\text{H}_8\text{O}$ $[\text{M}+\text{H}]^+$ m/z 97.0653; observed m/z 97.0646.

Synthesis of (1S,6S)-6-(2-(2-(((1R,6R)-6-hydroxycyclohex-3-en-1-yl)oxy)ethoxy)ethoxy) cyclohex-3-enol (5) and (1S,6S)-6-(2-(2-(((1R,6R)-6-hydroxycyclohex-3-en-1-yl)oxy)ethoxy) ethoxy)cyclohex-3-enol (6).

Diethylene glycol (5.58 g, 52.0 mmol) and 95% w/w sodium hydride (263 mg, 10.4 mmol) were stirred in an oven-dried vessel under argon atmosphere for 0.5h at 60°C. Next, 7-oxa-bicyclo[4.1.0]hept-3-ene **4** (10.0 g, 104 mmol) was added to the mixture followed by stirring at 60°C for 42h. After cooling to RT, the mixture was diluted with water (10 mL) and the pH was adjusted to 7–8 using 1.1M HCl. The aqueous mixture was extracted with DCM (1x50 mL, 2x30 mL). The combined organic extracts were washed with brine (1x20 mL), dried over anhydrous Na_2SO_4 , and concentrated *in vacuo* to yield a dark red oil. The crude product was purified by column chromatography (SiO_2 , DCM/MeOH 39:1) yielding half-crowns **5** and **6** as a yellow oil (4.97 g, 32%). $^1\text{H-NMR}$ (600 MHz, CDCl_3) δ 5.47–5.54 (m, 4H), 5.09 (s, 1H, OH), 5.04 (s, 1H, OH), 3.56–3.90 (m, 10H), 3.36–3.43 (m, 2H), 2.47–2.52 (m, 4H), 1.97–2.11 (m, 4H). $^{13}\text{C-NMR}$ (151 MHz, CDCl_3) δ 125.15, 125.12, 124.04, 124.01, 81.56, 81.14, 70.92, 70.85, 70.61, 70.42, 68.99, 68.49, 33.28, 31.22, 31.06. HRMS: calculated for $\text{C}_{16}\text{H}_{26}\text{O}_5$ $[\text{M}+\text{H}]^+$ m/z 299.1859; observed m/z 299.1820.

Synthesis of (4aR,11aS,15aS,22aR)-1,4,4a,6,7,9,10,11a,12,15,15a,17,18,20,21,22a-hexadecahydrodibenzo[b,k][1,4,7,10,13,16]hexaoxacyclooctadecine (1) and (4aR,11aR,15aR,22aR)-1,4,4a,6,7,9,10,11a,12,15,15a,17,18,20,21,22a-hexadecahydrodibenzo[b,k][1,4,7,10,13,16]

hexaoxacyclooctadecine (2). A mixture of half-crown **5** and **6** (3.50 g, 11.7 mmol) in DMSO (14 mL) was added to a stirred suspension of 60% w/w NaH in mineral oil (1.41 g, 35.2 mmol) in DMSO (30 mL) under argon atmosphere. After stirring for 0.5h, di(ethylene glycol) ditosylate (5.12 g, 11.7 mmol) was added to the mixture followed by stirring for 18h at RT, after which full consumption of the starting material was observed on TLC. Brine (36 mL) was subsequently added to break down excess NaH, followed by stirring for 5 minutes. Formed solids were filtered off and the solution was extracted with diethyl ether (5x70 mL). The combined organic extracts were washed with brine (1x35 mL), dried over anhydrous Na_2SO_4 , and concentrated *in vacuo* yielding an orange oil containing isomers **1** and **2** (1:1 ratio by NMR). Column chromatography with gradient elution (SiO_2 , DCM/MeOH/ conc. NH_3 (aq.) 15:1:1 \rightarrow 10:1:1) was utilized to yield the pure *trans-syn-trans* crown ether **1** (white solid, 743 mg, 17%) and pure *trans-anti-trans* crown ether **2** (white solid, 1.09 g, 25%). **1:** $^1\text{H-NMR}$ (600 MHz, CDCl_3) δ 5.52 (t, J 1.2 Hz, 4H), 3.83–3.87 (m, 4H), 3.64–3.76 (m, 12H), 3.57–3.59 (m, 4H), 2.48 (dd, J 19.5, 3.7 Hz, 4H), 2.06 (ddd, J 15.5, 5.3, 3.2 Hz, 4H). $^{13}\text{C-NMR}$ (151 MHz, CDCl_3) δ 124.43, 78.59, 71.07, 69.48, 31.34. HRMS: calculated for $\text{C}_{20}\text{H}_{32}\text{O}_6$ $[\text{M}+\text{H}]^+$ m/z 369.2277; observed m/z 369.2239. **2:** $^1\text{H-NMR}$ (600 MHz, CDCl_3) δ 5.52 (t, J 1.5 Hz, 4H), 3.95–3.99 (m, 4H), 3.64–3.73 (m, 12H), 3.52–3.55 (m, 4H), 2.47 (dd, J 19.5, 3.7 Hz, 4H), 2.07 (ddd, J 15.6, 5.5, 3.2 Hz, 4H). $^{13}\text{C-NMR}$ (151 MHz, CDCl_3) δ 124.50, 79.26, 71.26, 70.35, 31.66. HRMS: calculated for $\text{C}_{20}\text{H}_{32}\text{O}_6$ $[\text{M}+\text{H}]^+$ m/z 369.2277; observed m/z 369.2249.

Acknowledgements

This work was supported by the University of the Pacific's Office of Graduate Studies and College Research Fund. We would also like to acknowledge the NSF Major Research Instrumentation Grant NSF-MRI-0722654.

References

1. Pedersen, C. J. *J. Am. Chem. Soc.* **1967**, *89*, 7017–7036.
<https://doi.org/10.1021/ja01002a035>
2. Di Stefano, S.; Biagini, C. *Angew. Chemie Int. Ed.* **2020**.
<https://doi.org/10.1002/anie.201912659>
3. Gokel, G. W.; Leevy, W. M.; Weber, M. E. *Chem. Rev.* **2004**, *104*, 2723–2750.
<https://doi.org/10.1021/cr020080k>
4. Samoshin, V. V. *Rev. J. Chem.* **2011**, *1*, 250–274.
<https://doi.org/10.1134/S2079978011020026>
5. Samoshin, V. *Mini. Rev. Org. Chem.* **2005**, *2*, 225–235.
<https://doi.org/10.2174/1570193054368882>
6. Han, Y.; Jiang, Y.; Chen, C.-F. *Tetrahedron* **2015**, *71*, 503–522.
<https://doi.org/10.1016/J.TET.2014.11.006>
7. Xue, M.; Yang, Y.; Chi, X.; Yan, X.; Huang, F. *Chem. Rev.* **2015**, *115*, 7398–7501.
<https://doi.org/10.1021/cr5005869>
8. Zhang, M.; Yan, X.; Huang, F.; Niu, Z.; Gibson, H. W. *Acc. Chem. Res.* **2014**, *47*, 1995–2005.
<https://doi.org/10.1021/ar500046r>
9. Kralj, M.; Tušek-Božić, L.; Frkanec, L. *ChemMedChem* **2008**, *3*, 1478–1492.
<https://doi.org/10.1002/cmdc.200800118>
10. Stillwell, W.; Stillwell, W. E. *An Introduction to Biological Membranes* Elsevier, 2016.
11. Albanese, D. *Catal. Rev.* **2003**, *45*, 369–395.
<https://doi.org/10.1081/CR-120025538>
12. Izatt, R. M.; Haymore, B. L.; Bradshaw, J. S.; Christensen, J. J. *Inorg. Chem.* **1975**, *14*, 3132–3133.
<https://doi.org/10.1021/ic50154a059>
13. Huber, V. J.; Dietz, M. L. *Tetrahedron Lett.* **2001**, *42*, 2945–2948.
14. Yamato, K.; Fernandez, F. A.; Vogel, H. F.; Bartsch, R. A.; Dietz, M. L. *Tetrahedron Lett.* **2002**, *43*, 5229–5232.
[https://doi.org/10.1016/S0040-4039\(02\)01040-7](https://doi.org/10.1016/S0040-4039(02)01040-7)
15. Yamato, K.; Bartsch, R. A.; Dietz, M. L.; Rogers, R. D. *Tetrahedron Lett.* **2002**, *43*, 2153–2156.
[https://doi.org/10.1016/S0040-4039\(02\)00221-6](https://doi.org/10.1016/S0040-4039(02)00221-6)
16. Yamato, K.; Bartsch, R. A.; Broker, G. A.; Rogers, R. D.; Dietz, M. L. *Tetrahedron Lett.* **2002**, 5805–5808.
[https://doi.org/10.1016/S0040-4039\(02\)01187-5](https://doi.org/10.1016/S0040-4039(02)01187-5)
17. Clique, B.; Ironmonger, A.; Whittaker, B.; Colley, J.; Titchmarsh, J.; Stockley, P.; Nelson, A. *Org. Biomol. Chem.* **2005**, *3*, 2776–2785.
<https://doi.org/10.1039/B505865A>
18. Kavadias G, Belleau, B. US Patent 4003922; (1977)
19. CCDC 1963312 (1) and CCDC 1963313 (2) contain the supplementary crystallographic data for this paper. These data can be obtained free of charge at www.ccdc.cam.ac.uk/conts/retrieving.html [or from the Cambridge Crystallographic Data Center (CCDC), 12 Union Road, Cambridge CB2 1EZ, UK fax: +44(0) 1223 336 033; e-mail: deposit@ccdc.cam.ac.uk].