

Microwave-promoted Heck and Suzuki coupling reactions of new 3-(5bromobenzofuranyl)pyrazole in aqueous media

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B	D 111-1-1-1-1-00-04-204
Received 05-01-2018	Published on line U8-04-201

Abstract

Suzuki-Miyaura and Mizoroki-Heck cross-coupling reactions of 3-(5-bromobenzofuran-2-yl)-1*H*-pyrazole with various arylboronic acids and terminal olefins, respectively, were investigated using a benzothiazole-oxime palladium(II) complex, under both thermal and microwave-irradiation conditions, in an open vessel, using aqueous solvent. The benzothiazole-oxime-based Pd(II) complex was found to be an efficient, and highly active pre-catalyst for the cross-coupling reactions, and for the preparation of new C-C cross-coupled compounds of potential biological interest which are difficult to obtain using other synthetic routes.



Keywords: Benzofurans, pyrazoles, palladium catalysis, microwave, C-C cross couplings.

Introduction

The palladium-catalyzed Suzuki–Miyaura cross-coupling reaction represents one of the most widely used processes for arylation of aromatic and heteroaromatic compounds.^{1–6} In addition, palladium-catalyzed Heck reactions of aryl halides with alkenes are one of the most powerful tools in organic synthesis for the construction of carbon–carbon bonds.^{7–10} Interestingly, growing research work is focused on the use of microwave-irradiation methodology as a heating source because it markedly assists in achieving rapid incorporation of organic synthesis into diverse industrial applications.^{11–14} Furthermore, organic reactions that proceed well in aqueous media have several advantages over those that take place in organic solvents.^{15–17}

The benzofuran moiety constitutes the core of several pharmaceutically-active natural products, such as *Cicerfuran, Conocarpan, Ailanthoidol, Amiodarone* and *Bufuralol*.^{18–22} The synthetic benzofuran derivatives are reported to be effective drugs which have huge medicinal applications. ^{23–26} Pyrazolylbenzofuran derivatives, for example, have proven to have potent cytotoxic,²⁷ antimicrobial,^{28–30} antioxidant³¹ and insecticidal³² activities.

In continuation of our research work on the use of palladium(II) complexes under microwave irradiation for Suzuki–Miyaura, Heck–Mizoroki, and Sonogashira cross-coupling reactions,^{33–44} we report, herein, the use of the benzothiazole-oxime palladium(II) complex (**4**) as a pre-catalyst for the Suzuki–Miyaura and Heck–Mizoroki cross-coupling reactions of 3-(5-bromobenzofuran-2-yl)-1*H*-pyrazole (**3**) under thermal heating and microwave irradiation in aqueous media.

Results and Discussion

1-(5-Bromobenzofuran-2-yl)ethanone (1) was first prepared following a procedure reported in the literature.⁴⁵ Treatment of 1 with dimethylformamide-dimethylacetal (DMF-DMA) in dry xylene at reflux temperature afforded (*E*)-1-(5-bromobenzofuran-2-yl)-3-(dimethylamino)prop-2-en-1-one (2).⁴⁶ The enaminone derivative 2 was then reacted with hydrazine hydrate in refluxing ethanol to give 3-(5-bromobenzofuran-2-yl)-1*H*-pyrazole (3) as shown in Scheme 1.⁴⁶ The structures of the new 5-bromobenzofuran derivatives (2) and (3) were confirmed from their elemental analyses and spectroscopic data (IR, MS, ¹H and ¹³C NMR). The *E*-configuration of structure 2 was established based on the coupling constant *J* of 12.6 Hz.



Scheme 1. Synthesis of 3-(5-bromobenzofuran-2-yl)-1H-pyrazole (3).

Suzuki cross-coupling of 3-(5-bromobenzofuran-2-yl)-1*H*-pyrazole (3) with arylboronic acids (5a-f) under thermal heating and microwave irradiation

The catalytic activity of the Pd(II)-complex (**4**) (0.25 mol%) in the Suzuki-Miyaura cross-coupling reaction of a novel candidate, 3-(5-bromobenzofuran-2-yl)-1*H*-pyrazole (**3**), with arylboronic acids (**5a-f**) has been examined (Scheme 2).



Scheme 2. Suzuki cross-coupling of 3-(5-bromobenzofuran-2-yl)-1*H*-pyrazole (**3**) with arylboronic acids (**5a-f**) under thermal and μ w conditions.

First, the catalytic condition [water/tetrabutylammonium bromide (TBAB)/KOH] was applied for the coupling of the 5-bromobenzofuran derivative (**3**) with phenylboronic acid (**5a**). No reaction was observed, however, after 15 min of microwave irradiation (μ w) or 6 h of thermal heating (entry 1, Table 1). Repeating the same reaction in DMF instead of water using Et₃N and TBAB under μ w irradiation for 20 min gave 3-(5-phenylbenzofuran-2-yl)-1*H*-pyrazole (**6**) in 70% isolated yield (entry 2, Table 1). Next, the solvent mixture of water-DMF (1:1, v/v) was applied using KOH and TBAB under thermal heating for 6 h as well as under μ w irradiation for 20 min. Under these conditions, full conversions were observed (based on TLC) with 90% and 88% isolated yields, respectively, to 3-(5-phenylbenzofuran-2-yl)-1*H*-pyrazole (**6**) (entry 3, Table 1).

The above reaction conditions were applied for further Suzuki-Miyaura cross-coupling reactions of the bromide (**3**) with aryl and heteroarylboronic acids (**5b-f**). The reaction-components molar ratios, in all cases, were: 1 mmol of bromide, 1.2 mmol of the arylboronic acids, 0.6 mmol TBAB, 2 mmoles of KOH and 0.25 mol% of the complex (**4**) in a water-DMF mixture (1:1, v/v) (3 mL). The arylboronic acids were coupled, accordingly, with the bromide (**3**) under thermal heating as well as microwave irradiation to give the corresponding arylated benzofuran products (**7-11**) in very good isolated yields (Table 1). The only exception was the coupling of bromide (**3**) with 2-thienylboronic acid (**5f**), in which the reaction did not proceed at all when water-DMF/TBAB/KOH was used (entry 8, Table 1). Therefore, KOH was replaced with cesium carbonate, a more reactive base. Using Cs₂CO₃ and 0.5 mol% of Pd-pre-catalyst (**4**), reasonable yields were obtained after 30 min of μw irradiation (53%), and after 14 h of thermal heating (42%), respectively (entry 9, Table 1).

Although, the microwave irradiation reaction times were generally ~ 20-25 min, the corresponding coupled products (6-10) were isolated mostly in high yields (~ 73-88%) as shown in Table 1. A rationale for the long reaction time required to accomplish C-C cross-coupling of bromide (3) might be due to the presence of two π -electron-excessive heterocycles (pyrazole and furan) in 3. Therefore, even though the 3-(5-bromobenzofuran-2-yl)-1*H*-pyrazole (3) is considered to be a deactivated bromide, the full conversion of 3 in water-DMF using 0.25 mol% of the pre-catalyst (4) reflects the high catalytic activity of this catalyst. The structures of the cross-coupled products were confirmed by spectral data (¹H and ¹³C NMR, MS and IR) and elemental analyses, as reported in the experimental section.

Entry	Droduct	Salvant	Thermal heating ^{a,b}		μw heating ^{a,b}	
Entry	Product	Solvent	Time (h)	Yield%	Time (min)	Yield%
1		H₂O/ KOH	6	0	15	0
2		DMF/ TEA	-	-	20	70
3	0 N-NH 6	H ₂ O-DMF/ KOH	6	90	20	88
4	Me V N-NH 7	H ₂ O-DMF/ KOH	8	80	25	84
5	MeO	H ₂ O-DMF/ KOH	8	72	25	80
6	CI C	H ₂ O-DMF/ KOH	6	95	20	87
7	о () () () () () () () () () ()	H₂O-DMF/ KOH	8	70 ^c	25	73 ^c
8 9	s 0 N-NH 11	H ₂ O-DMF/ KOH H ₂ O-DMF/Cs ₂ CO ₃	6 14	0 42 ^c	20 30	0 53 ^c

Table 1. Results of Suzuki cross-coupling of 3-(5-bromobenzofuran-2-yl)-1*H*-pyrazole (**3**) with arylboronic acids (**5a-f**) under thermal and μ w conditions

^aConditions. Bromide: 1 mmol; arylboronic acid: 1.2 mmol; TBAB: 0.6 mmol; base: 2 mmol; water or DMF: 3 mL, water-DMF: 3 mL (1:1, v/v), Pd-complex: 0.25 mol%, microwave heating (250 Watt) at 160 $^{\circ}$ C and at reflux for thermal heating. ^bThe values of yield% refer to the isolated yields. ^cThe starting substrate (**3**) was detected by TLC.

Heck cross-coupling of 3-(5-bromobenzofuran-2-yl)-1H-pyrazole (3) with olefins (12a-d)

The potential of the Pd-complex (4) towards Heck cross-coupling of the olefins (12a-d) with 3-(5-bromobenzofuran-2-yl)-1*H*-pyrazole (3) under both microwave and thermal heating was studied (Scheme 3).



Scheme 3. Heck cross-coupling reactions of 3-(5-bromobenzofuran-2-yl)-*1H*-pyrazole (**3**) with olefins **12a-d** under thermal and μ w conditions.

In all cases, full conversions into the corresponding disubstituted olefins (13-16) were confirmed by high isolated yields (Table 2). Styrene (12a) was easily coupled with the 5-bromobenzofuran (3) in the presence of

0.25 mol% of the Pd-complex (4) using DMF/Et₃N/TBAB as the catalytic system, to afford the 5styrylbenzofuran derivative (13) in excellent yield (entry 1, Table 2) after 8 hours of thermal heating or 20 min of microwave irradiation. In a similar trend, 3-(5-bromobenzofuran-2-yl)-1*H*-pyrazole (3) cross-coupled with acrylonitrile, ethyl acrylate, and *n*-butyl acrylate, respectively, under similar catalytic conditions, resulting in full conversions into the corresponding pyrazolyl benzofuranyl acrylates (14-16) under both thermal and microwave heating conditions.

Table 2.	Results of Heck cr	oss-coupling reaction	s of 3-(5-bromobe	enzofuran-2-yl)-1H-pyrazol	e (3) with	olefins
(12a-d) u	Inder thermal and I	μw conditions				

Entry	Ol No.	lefins R	Product	Thermal heating ^{a,b} Yield%	μw heating ^{a,b} Yield%
1	12a	Ph	13	84	87
2	12b	CN	14	79	82
3	12c	CO ₂ Et	15	94	90
4	12d	CO ₂ Bu ⁿ	16	91	85

^aConditions. Bromide: 1 mmol; olefin: 1.5 mmol; TBAB: 0.6 mmol; Et₃N: 3 mmol; DMF: 3 mL, Pd-complex: 0.25 mol%, microwave heating (250 Watt) at 160 °C for 20 min, and thermal heating at 130 °C for 8 h. ^bThe values of yield% refer to the isolated yields.

In all cases, the cross-coupling reaction was highly regio- and stereoselective, yielding only the *E*-isomer of the disubstituted olefins **13-16**, as substantiated by the GC, GC-MS and ¹H NMR-spectra of the crude reaction mixtures. The structures of the new products were confirmed from their elemental and spectral analyses. The *trans*-configuration of the products was estimated from spectral data, in which the butyl acrylate derivative **(16)** showed, for example, two characteristic doublets at δ 6.62 and 7.76 ppm with coupling constant *J* value 15.9 Hz, respectively, in its ¹H NMR spectrum.

Conclusions

In conclusion, the benzothiazole-oxime-based Palladium(II) complex (**4**) was found to be an efficient and highly active pre-catalyst for Suzuki-Miyaura and Heck-Mizoroki cross-coupling reactions of deactivated 3-(5-bromobenzofuran-2-yl)-1*H*-pyrazole (**3**) in aqueous medium, under thermal heating and microwave-irradiation conditions, for the preparation of new C-C cross-coupled products which are of potential biological and therapeutic interest and are difficult to obtain by other synthetic routes.

Experimental Section

General. Melting points were determined in open glass capillaries with a Gallenkamp apparatus. The infrared spectra were recorded in potassium bromide disks on a Pye Unicam SP 3-300 and Shimaduz FTIR 8101 PC

infrared spectrophotometer. NMR spectra were recorded using a Varian Mercury VXR-300 NMR spectrometer at 300 MHz (¹H NMR) and at 75 MHz (¹³C NMR) using DMSO- d_6 as solvent. Capillary GC analyses were performed with a Shimadzu GC-14A or GC-14B, a Shimadzu C-R6A Integrator, and a HP 5 column (25 m length, 0.25 mm i.d., 0.25 µm film) or recorded with an Agilent GC 6890N. Mass spectra (EI) were obtained at 70 eV using a type Shimadzu GCMQP 1000 EX Spectrometer. Analytical thin-layer chromatography (TLC) was performed using pre-coated silica gel 60778 plates (Fluka), and the spots were visualized with UV light at 254 nm. Fluka silica gel 60741 (70-230 mesh) was used for flash column chromatography. Microwave experiments were carried out using a CEM Discover LabmateTM Microwave (300 W with ChemDriverTM Software). The palladium complex (**4**) was prepared following our procedure previously reported in the literature.³⁸

Synthesis of (E)-1-(5-bromobenzofuran-2-yl)-3-(dimethylamino)-prop-2-en-1-one (2). A mixture of 1-(5-bromobenzofuran-2-yl)ethanone **(1)** (4.78 g, 20 mmol) and dimethyl formamide-dimethylacetal (DMF-DMA) (2.66 ml, 20 mmol) was taken up in dried xylene (20 ml). The mixture was refluxed for 8 h, then left to cool to room temperature. The reddish-brown precipitated product was filtered off, washed with petroleum ether (60/80 °C), and dried. Recrystallization of the precipitate from benzene afforded 5.17 g of (*E*)-1-(5-bromobenzofuran-2-yl)-3-(dimethylamino)prop-2-en-1-one **(2)** (88% yield), mp 148-150 °C; IR (KBr) \cup 3434, 3100, 2915, 1638, 1566, 1539, 1353, 1109 cm⁻¹; ¹H NMR (300 MHz, DMSO-*d*₆) $\delta_{\rm H}$ 2.94 (s, 3H, CH₃), 3.18 (s, 3H, CH₃), 5.81 (d, 1H, *J* 12.6 Hz, CH olefinic), 7.48 (s, 1H, CH aromatic), 7.54 (d, 1H, *J* 8.7 Hz, CH aromatic), 7.64 (d, 1H, *J* 8.7 Hz, CH aromatic), 7.81 (d, 1H, *J* 12.6 Hz, CH olefinic), 7.93 (s, 1H, CH aromatic); ¹³C NMR (75 MHz, DMSO-*d*₆) δ 37.2, 44.6, 90.9, 107.8, 113.8, 115.5, 124.7, 128.9, 129.7, 153.0, 154.1, 156.5, 175.4; MS *m/z* (%) 294 (24.2, M⁺), 276 (18.7), 167 (25.3), 98 (58.2), 88 (40.7), 70 (57.1), 55 (100). Anal. Calcd for C₁₃H₁₂BrNO₂: C, 53.08; H, 4.11; N, 4.76. Found: C, 53.10; H, 4.07; N, 4.69%.

Synthesis of 3-(5-bromobenzofuran-2-yl)-1H-pyrazole (3). Hydrazine hydrate (2 ml) was added to a stirred solution of the 3-(dimethylamino)prop-2-en-1-one (2) (2.94, 10 mmol) in acetic acid (30 ml). Stirring was continued overnight at room temperature and the solid product obtained was filtered off, washed with ethanol, dried and recrystallized from DMF to afford 2.23 g (85% yield) of 3-(5-bromobenzofuran-2-yl)-1H-pyrazole (3) as yellow crystals; mp. 204-206 °C; IR (KBr) υ 3511, 3126, 2913, 1536, 1412, 1044 cm⁻¹; ¹H NMR (300 MHz, DMSO-*d*₆) δ 6.76 (d, 1H, *J* = 2.1 Hz, CH aromatic), 7.14 (s, 1H, CH aromatic), 7.41 (d, 1H, *J* 8.7 Hz, CH aromatic), 7.59 (d, 1H, *J* 8.7 Hz, CH aromatic), 7.84 (s, 1H, CH aromatic), 7.88 (s, 1H, CH aromatic), 13.25 (s, 1H, NH); ¹³C NMR (75 MHz, DMSO-*d*₆) δ 100.8, 103.2, 112.9, 115.3, 120.7, 123.3, 126.6, 130.0, 130.8, 131.8, 152.6; MS *m/z* (%) 263 (100, M⁺), 184 (24.7), 156 (31.2), 128 (37.7), 101 (58.4), 75 (76.6), 62 (87). Anal. Calcd for C₁₁H₇BrN₂O: C, 50.22; H, 2.68; N, 10.65. Found: C, 50.26; H, 2.61; N, 10.62%.

Suzuki cross-coupling reactions of 3-(5-bromobenzofuran-2-yl)-1*H*-pyrazole (3) with arylboronic acids in water-DMF under thermal heating. General procedure. A mixture of 3-(5-bromobenzofuran-2-yl)-1*H*-pyrazole (3) (263 mg, 1 mmol), and the appropriate arylboronic acids (5a-f) (1.2 mmol), tetrabutylammonium bromide (TBAB) (194 mg, 0.6 mmol), palladium complex (4) (1.14 mg, 0.25 mol %), KOH (112 mg, 2 mmol), and water-DMF (1:1) (3 mL) was stirred at 100 °C in an open reation vessel for ~ 4-10 hr (monitored by TLC) as listed in Table 1. The products were extracted with EtOAc (3 x 20 mL). The combined organic extracts were dried over anhydrous MgSO₄, filtered, and the solvent was evaporated under reduced pressure. The residue was then subjected to separation with flash column chromatography, using petroleum-distilled *n*-hexane/EtOAc (5:1) as an eluent, to give the corresponding pure products **6-11**.

Suzuki cross-coupling reactions of 3-(5-bromobenzofuran-2-yl)-1*H*-pyrazole (3) with arylboronic acids in water-DMF using microwave irradiation. General procedure. A mixture of the above components and solvent were mixed thoroughly in a process glass vial. The vial was capped properly, and the mixture was heated

under microwave-irradiating conditions at 160 °C and 250 Watts for the appropriate reaction times as shown in Table 1. After the reaction was almost complete (by TLC), the products were extracted with EtOAc (3 x 20 mL). The combined organic extracts were dried over anhydrous MgSO₄, filtered off, and the solvent was evaporated under reduced pressure. The residue was then subjected to separation with flash column chromatography, as described above, to give the corresponding pure cross-coupled products **6-11**.

3-(5-Phenylbenzofuran-2-yl)-1*H*-**pyrazole (6).** White crystals; mp 84-86 °C; R_f 0.47 (hexane:ethyl acetate, 5:1); IR (KBr) \cup 3375, 3139, 3036, 2922, 1600, 1442, 1256, 1072 cm⁻¹; ¹H NMR (300 MHz, DMSO- d_6) δ 6.76 (d, 1H, *J* 5.7 Hz, CH aromatic), 7.15 (s, 1H, CH aromatic), 7.26-7.47 (m, 3H, CH aromatic), 7.60 (d, 2H, *J* 7.8 Hz, CH aromatic), 7.69 (d, 1H, *J* 8.4 Hz, CH aromatic), 7.78 (d, 1H, *J* 8.1 Hz, CH aromatic), 7.88 (d, 1H, *J* 5.4 Hz, CH aromatic), 7.98 (s, 1H, CH aromatic), 13.81 (s, 1H, NH); ¹³C NMR (75 MHz, DMSO- d_6) δ 101.1, 102.9, 110.9, 111.2, 120.8, 123.0, 123.3, 124.0, 126.8, 127.2, 128.5, 128.8, 129.9, 133.9, 153.8; MS *m/z* (%) 260 (5.9, M⁺), 184 (100), 155 (33.1), 128 (40.7), 102 (55.9), 63 (74.6), 50 (83.1). Anal. Calcd for C₁₇H₁₂N₂O: C, 78.44; H, 4.65; N, 10.76. Found: C, 78.55; H, 4.58; N, 10.69%.

3-(5-(*p***-Tolyl)benzofuran-2-yl)-1***H***-pyrazole (7).** White crystals; mp 108-110 °C; *R_f* 0.44 (hexane:ethyl acetate, 5:1); IR (KBr) υ 3356, 3139, 2921, 2846, 1614, 1442, 1348 cm⁻¹; ¹H NMR (300 MHz, DMSO-*d*₆) δ 2.29 (s, 3H, CH₃), 6.75 (d, 1H, *J* 6.9 Hz, CH aromatic), 7.13 (d, 2H, *J* 7.5 Hz, CH aromatic), 7.26 (d, 2H, *J* 7.5 Hz, CH aromatic), 7.58 (d, 1H, *J* 8.1 Hz, CH aromatic), 7.61 (s, 1H, CH aromatic), 7.69 (d, 1H, *J* 7.8 Hz, CH aromatic), 7.87 (d, 1H, *J* 6.9 Hz, CH aromatic), 7.88 (s, 1H, CH aromatic), 13.19 (s, 1H, NH); ¹³C NMR (75 MHz, DMSO-d₆) δ 21.0, 101.1, 102.9, 110.9, 111.1, 112.9, 120.8, 123.0, 124.0, 126.6, 127.9, 129.4, 129.8, 134.1, 139.2, 153.8; MS *m/z* (%) 274 (100, M⁺), 245 (10.6), 184 (16.8), 155 (23.5), 88 (15.1), 63 (23.5), 51 (18.4). Anal. Calcd for C₁₈H₁₄N₂O: C, 78.81; H, 5.14; N, 10.21. Found: C, 78.72; H, 5.22; N, 10.17%.

3-(5-(4-Methoxyphenyl)benzofuran-2-yl)-1*H*-**pyrazole (8).** Yellow crystals; mp 104-106 °C; R_f 0.36 (hexane:ethyl acetate 5:1); IR (KBr) 2 3353, 3139, 2929, 2836, 1602, 1442, 1245, 1168 cm⁻¹; ¹H NMR (300 MHz, DMSO- d_6) δ 3.75 (s, 3H, OCH₃), 6.75 (d, 1H, *J* 6.3 Hz, CH aromatic), 6.88 (d, 2H, *J* 8.7 Hz, CH aromatic), 7.02 (d, 2H, *J* 9 Hz, CH aromatic), 7.59 (s, 1H, CH aromatic), 7.61 (d, 1H, *J* 8.4 Hz, CH aromatic), 7.74 (d, 1H, *J* 8.4 Hz, CH aromatic), 7.81 (s, 1H, CH aromatic), 7.86 (d, 1H, *J* 6.3 Hz, CH aromatic), 13.19 (s, 1H, NH); ¹³C NMR (75 MHz, DMSO- d_6) δ 55.1, 101.2, 102.9, 110.9, 111.1, 112.9, 114.2, 120.7, 120.9, 123.0, 124.1, 127.9, 128.5, 129.8, 135.7, 153.8; MS *m*/*z* (%) 290 (100, M⁺), 275 (41.5), 247 (16.5), 189 (16.0), 145 (22.9), 87 (15.4), 55 (18.6). Anal. Calcd for C₁₈H₁₄N₂O₂: C, 74.47; H, 4.86; N, 9.65. Found: C, 74.33; H, 4.81; N, 9.58%.

3-(5-(4-Chlorophenyl)benzofuran-2-yl)-1*H*-**pyrazole (9)**. White powder; mp 78-80 °C; R_f 0.4 (hexane:ethyl acetate, 5:1); IR (KBr) \cup 3379, 3141, 2925, 2844, 1590, 1442, 1257 cm⁻¹; ¹H NMR (300 MHz, DMSO- d_6) δ 6.75 (d, 1H, *J* 8.7 Hz, CH aromatic), 7.39 (d, 2H, *J* 8.1 Hz, CH aromatic), 7.51 (d, 1H, *J* 8.4 Hz, CH aromatic), 7.58 (s, 1H, CH aromatic), 7.72 (d, 1H, *J* 8.7 Hz, CH aromatic), 7.79 (d, 2H, *J* 8.1 Hz, CH aromatic), 7.86 (d, 1H, *J* 8.7 Hz, CH aromatic), 8.13 (s, 1H, CH aromatic), 13.22 (s, 1H, NH); ¹³C NMR (75 MHz, DMSO- d_6) δ 102.9, 110.9, 111.3, 112.9, 120.7, 120.9, 123.0, 123.3, 127.3, 128.6, 128.7, 129.9, 135.8, 136.1, 153.8; MS *m/z* (%) 296 (33.3, M⁺+2), 295 (37.4, M⁺+1), 294 (100, M⁺), 202 (29.3), 150 (12.2), 133 (24.4), 56 (15.4). Anal. Calcd for C₁₇H₁₁ClN₂O: C, 69.28; H, 3.76; N, 9.50. Found: C, 69.31; H, 3.77; N, 9.45%.

3-(5-(3,4-Methylenedioxophenyl)benzofuran-2-yl)-1*H*-**pyrazole (10)**. Buff powder; mp 114-116 °C; R_f 0.39 (hexane:ethyl acetate, 5:1); IR (KBr) \cup 3427, 3139, 2921, 2845, 2361, 1437 cm⁻¹; ¹H NMR (300 MHz, DMSO- d_6) \cup 6.06 (s, 2H, OCH₂), 6.75 (d, 1H, *J* 6.6 Hz, CH aromatic), 6.88 (d, 1H, *J* 7.8 Hz, CH aromatic), 6.99 (d, 1H, *J* 7.8 Hz, CH aromatic), 7.16 (s, 1H, CH aromatic), 7.27 (s, 1H, CH aromatic), 7.60 (d, 1H, *J* 9 Hz, CH aromatic), 7.64 (d, 1H, *J* 8.7 Hz, CH aromatic), 7.84 (d, 1H, *J* 6.6 Hz, CH aromatic), 7.85 (s, 1H, CH aromatic), 13.29 (s, 1H, NH); ¹³C NMR (75 MHz, DMSO- d_6) δ 101.1, 101.4, 102.9, 103.2, 107.3, 108.5, 110.9, 112.9, 115.4, 120.9, 123.0, 123.3,

124.1, 126.6, 128.4, 130.8, 152.6, 153.8; MS *m*/*z* (%) 304 (100, M⁺), 264 (27.2), 184 (32.9), 128 (34.7), 92 (26.0), 51 (42.8). Anal. Calcd for C₁₈H₁₂N₂O₃: C, 71.05; H, 3.97; N, 9.21. Found: C, 70.92; H, 3.88; N, 9.15%.

3-(5-(Thien-3-yl)benzofuran-2-yl)-1*H***-pyrazole (11)**. Yellowish-white powder; mp 180-182 °C; R_f 0.41 (hexane:ethyl acetate, 5:1); IR (KBr) υ 3439, 3135, 2916, 2361, 1436, 1044 cm⁻¹; ¹H NMR (300 MHz, DMSO- d_6) δ 6.76 (d, 1H, *J* 6.3 Hz, CH aromatic), 7.15 (s, 1H, CH aromatic), 7.22-7.44 (m, 2H, CH aromatic), 7.54 (s, 1H, CH aromatic), 7.59 (d, 1H, *J* 8.7 Hz, CH aromatic), 7.62 (d, 1H, *J* 9 Hz, CH aromatic), 7.86 (d, 1H, *J* 6.3 Hz, CH aromatic), 7.88 (s, 1H, CH aromatic), 13.25 (s, 1H, NH); ¹³C NMR (75 MHz, DMSO- d_6) δ 100.7, 103.2, 110.9, 111.2, 112.9, 115.4, 120.2, 120.9, 123.3, 126.4, 126.6, 126.8, 129.9, 130.8, 152.6; MS *m/z* (%) 266 (16.2, M⁺), 184 (17.6), 155 (25.7), 127 (46.3), 98 (27.9), 74 (66.9), 63 (88.2), 50 (100). Anal. Calcd for C₁₅H₁₀N₂OS: C, 67.65; H, 3.78; N, 10.52; S, 12.04. Found: C, 67.59; H, 3.82; N, 10.46; S, 11.97%.

Mizoroki-Heck cross-coupling reactions of 3-(5-bromobenzofuran-2-yl)-1*H*-pyrazole (3) with olefins in DMF under thermal heating. A mixture of 3-(5-bromobenzofuran-2-yl)-1*H*-pyrazole (3) (263 mg, 1 mmol) and the appropriate olefin **12a-d** (1.5 mmol), TBAB (0.6 mmol), complex (4) (0.25 mol%), and TEA (303 mg, 3 mmol) in DMF (3 mL) was stirred at 130 °C in an open reaction vessel for 8h. After the reaction was almost complete (by TLC), the reaction mixture was left to cool to room temperature. The reaction mixture was then extracted three times with EtOAc (3 x 20 mL), and the organic fractions were combined, dried over MgSO₄, filtered, and the solvent was removed under vacuum. The residue was then subjected to purification via flash column chromatography with petroleum *n*-hexane/EtOAc (6:1) as eluent to give the corresponding pure products **13-16**.

Mizoroki-Heck cross-coupling reactions of 3-(5-bromobenzofuran-2-yl)-1*H*-pyrazole (3) with olefins in DMF under microwave heating. A mixture of 3-(5-bromobenzofuran-2-yl)-1*H*-pyrazole (3) (263 mg, 1 mmol) and the appropriate olefin **12a-d** (1.5 mmol), TBAB (0.6 mmol), complex (4) (0.25 mol%), and TEA (303 mg, 3 mmol) in DMF (3 mL) was mixed in a process glass vial. The vial was capped properly, and the mixture was heated under microwave-irradiating conditions at 160 °C and 250 Watt for 20 min. After the reaction was almost complete (by TLC), the reaction mixture was left to cool to room temperature. The reaction mixture was then extracted three times with EtOAc (3 x 20 mL), the organic fractions were combined, dried over MgSO₄, filtered and the solvent removed under vacuum. The residue was then subjected to purification via flash column chromatography with petroleum *n*-hexane/EtOAc (6:1) as eluent to give the corresponding pure products **13-16**.

(*E*)-3-(5-β-Styrylbenzofuran-2-yl)-1*H*-pyrazole (13). Pale yellow powder; mp 196-198 °C; R_f 0.20 (hexane:ethyl acetate, 6:1); IR (KBr) \cup 3376, 3140, 3021, 2917, 1448, 1254 cm⁻¹; ¹H NMR (300 MHz, DMSO- d_6) δ 6.76 (d, 1H, *J* 2.1 Hz, CH aromatic), 7.18 (s, 1H, CH aromatic), 7.24 (d, 1H, *J* 16.2 Hz, CH, olefinic), 7.28-7.38 (m, 3H, CH aromatic), 7.38 (d, 1H, *J* 15.9 Hz, CH, olefinic), 7.56-7.62 (m, 4H, CH aromatic), 7.86 (s, 2H, CH aromatic), 13.24 (s, 1H, NH); ¹³C NMR (75 MHz, DMSO- d_6) δ 101.4, 103.0, 111.1, 118.7, 120.8, 120.9, 123.1, 126.2, 127.3, 127.4, 127.8, 128.2, 128.6, 129.0, 132.5, 137.1, 153.6; MS *m/z* (%) 286 (100, M⁺), 257 (14.6), 227 (17.4), 189 (30.6), 163 (17.4), 129 (17.4), 95 (37.5), 63 (25.0). Anal. Calcd for C₁₉H₁₄N₂O: C, 79.70; H, 4.93; N, 9.78. Found: C, 79.76; H, 4.85; N, 9.68%.

(*E*)-3-(2-(1*H*-Pyrazol-3-yl)benzofuran-5-yl)acrylonitrile (14). Buff powder; mp 154-156 °C; R_f 0.24 (hexane:ethyl acetate, 6:1); IR (KBr) υ 3273, 3141, 3043, 2935, 2842, 2211, 1619, 1450, 1268 cm⁻¹; ¹H NMR (300 MHz, DMSO- d_6) δ 6.44 (d, 1H, *J* 16.5 Hz, CH, olefinic), 6.77 (s, 1H, CH aromatic), 7.20 (s, 1H, CH aromatic), 7.62 (d, 1H, *J* 9 Hz, CH aromatic), 7.68 (d, 1H, *J* 8.4 Hz, CH aromatic), 7.75 (d, 1H, *J* 16.8 Hz, CH olefinic), 7.89 (s, 1H, CH aromatic), 7.93 (s, 1H, CH aromatic), 13.24 (s, 1H, NH); ¹³C NMR (75 MHz, DMSO- d_6) δ 95.2, 101.2, 103.1, 111.6, 118.9, 120.7, 120.8, 120.9, 121.6, 123.8, 129.2, 129.9, 150.8, 155.2; MS *m/z* (%) 235 (100, M⁺),

179 (19.1), 151 (19.1), 128 (31.9), 76 (57.4), 62 (44.7), 50 (29.8). Anal. Calcd for C₁₄H₉N₃O: C, 71.48; H, 3.86; N, 17.86. Found: C, 71.41; H, 3.80; N, 17.83%.

(*E*)-Ethyl 3-(2-(1*H*-pyrazol-3-yl)benzofuran-5-yl)acrylate (15). White crystals; mp 180-182 °C; R_f 0.14 (hexane:ethyl acetate 7:1); IR (KBr) \cup 3238, 3115, 3062, 2986, 1682, 1367, 1294, 1237, 1038 cm⁻¹; ¹H NMR (300 MHz, DMSO- d_6) δ 1.27 (t, 3H, *CH*3CH2-, *J* 7.2 Hz), 4.20 (q, 2H, CH3*CH2*-, *J* 7.2 Hz), 6.62 (d, 1H, *J* 16.2 Hz, CH olefinic), 6.77 (s, 1H, CH aromatic), 7.17 (s, 1H, CH aromatic), 7.64 (d, 1H, *J* 8.4 Hz, CH aromatic), 7.69 (d, 1H, *J* 9 Hz, CH aromatic), 7.76 (d, 1H, *J* 15.9 Hz, CH, olefinic), 7.88 (s, 1H, CH aromatic), 8.00 (s, 1H, CH aromatic), 13.22 (s, 1H, NH); ¹³C NMR (75 MHz, DMSO- d_6) δ 14.1, 59.8, 101.2, 103.1, 111.5, 116.8, 121.4, 124.5, 129.1, 129.4, 129.9, 141.9, 144.7, 152.5, 154.9, 166.2; MS *m/z* (%) 282 (100, M⁺), 237 (61.2), 210 (26.0), 152 (15.3), 77 (18.0), 63 (27.3). Anal. Calcd for C₁₆H₁₄N₂O₃: C, 68.07; H, 5.00; N, 9.92. Found: C, 68.11; H, 4.95; N, 9.89%.

(*E*)-Butyl 3-(2-(1*H*-pyrazol-3-yl)benzofuran-5-yl)acrylate (16). White powder; mp 146-148 °C; R_f 0.26 (hexane:ethyl acetate 6:1); IR (KBr) \cup 3423, 3142, 3043, 2954, 2864, 1709, 1631, 1447, 1260, 1170 cm⁻¹; ¹H NMR (300 MHz, DMSO- d_6) δ 0.92 (t, 3H, *J* 7.5 Hz, *CH*₃CH₂CH₂CH₂), 1.35-1.43 (m, 2H, CH₃*CH*₂CH₂CH₂), 1.59-1.66 (m, 2H, CH₃CH₂*CH*₂), 4.16 (t, 2H, *J* 6.6 Hz, CH₃CH₂CH₂*CH*₂), 6.62 (d, 1H, *J* 15.9 Hz, CH, olefinic), 6.77 (d, 1H, *J* = 2.1 Hz, CH aromatic), 7.18 (s, 1H, CH aromatic), 7.64 (d, 1H, *J* 8.7 Hz, CH aromatic), 7.70 (d, 1H, *J* 9 Hz, CH aromatic), 7.76 (d, 1H, *J* 15.9 Hz, CH olefinic), 5.88 (s, 1H, CH aromatic), 8.00 (s, 1H, CH aromatic), 13.24 (s, 1H, NH); ¹³C NMR (75 MHz, DMSO- d_6) δ 13.5, 18.6, 30.2, 63.5, 101.2, 103.1, 111.4, 116.8, 121.5, 124.5, 129.2, 129.4, 130.0, 141.9, 144.7, 152.5, 154.9, 166.3; MS *m/z* (%) 310 (89.5, M⁺), 254 (100), 237 (75.3), 210 (26.0), 154 (14.2), 105 (25.3), 63 (31.8). Anal. Calcd for C₁₈H₁₈N₂O₃: C, 69.66; H, 5.85; N, 9.03. Found: C, 69.54; H, 5.77; N, 8.93%.

Supplementary Materials

Copies of ¹H and ¹³C spectra for compounds **2**, **3**, **6-11**, and **13-16** associated with this article can be found in the online version of the text.

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