Synthesis and structure elucidation of novel pyrazolyl-2-pyrazolines obtained by the reaction of 3-(3-aryl-3-oxopropenyl)chromen-4-ones with phenylhydrazine

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Dedicated to Professor Furenc Fülöp on the occasion of his 60th birthday

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Abstract
Novel 3-aryl-5-[(4-[5-(2-hydroxyphenyl)-1-phenylpyrazolyl])]-2-pyrazolines 2a-g have been prepared by the treatment of 3-(3-aryl-3-oxopropenyl) chromen-4-ones 1a-g with phenylhydrazine in refluxing acetic acid. NMR studies on deuteriochloroform solutions of pyrazolyl-2-pyrazolines 2a-g at different temperatures showed that at room temperature a mixture of diastereomers are present. This diastereoselectivity arises from the combination of the pyrazoline C-4 stereocenter and two planar chiral subunits due to internal steric hindrance. The energy barriers of this steric hindrance were overcome in DMSO-d₆ solutions at 60°C. The acetylation of some pyrazolyl-2-pyrazoline derivatives 2a-c,e helped to confirm the presence of the referred mixture of diastereomers.

Keywords: Chromones, phenylhydrazine, bispyrazoles, 4-pyrazolyl-2-pyrazolines, diastereomers, planar chirality, NMR spectroscopy

Introduction

Pyrazoles are interesting five-membered nitrogen heterocyclic compounds due to their synthetic versatility and broad spectrum of biological properties. Numerous derivatives have been found...
to act as pharmacodynamic and chemotherapeutic agents and also playing an important role on the central nervous system. On the other hand, the syntheses and selective functionalization of pyrazoles have stimulated the research over the years. Classical methods for the synthesis of substituted pyrazoles involve approaches based either in condensations of hydrazines with 1,3-dicarboxyl compounds and in the intermolecular [3+2] cycloaddition reactions of 1,3-dipoles to alkynes. However, other methodologies have been developed for the preparation of novel substituted pyrazole-type compounds.

3-(3-Aryl-3-oxopropenyl)chromen-4-ones 1 are useful α,β-unsaturated ketones that can be used as starting materials in the synthesis of a wide variety of nitrogen-containing heterocyclic compounds. 4-Aryl-2-(3-chromonyl)-2,3-dihydro-1,5-benzothiazepines were synthesized by the reaction of 1 with 2-aminothiophenol, while the reaction with diazomethane afforded 3-aryloxypropenyl)-2-pyrazolines similarly to that observed with α,β-unsaturated ketones. Treating 3-(3-aryl-3-oxopropenyl)chromen-4-ones 1 with hydrazine hydrate provided several pyrazolyl-2-pyrazolines, an interesting new pyrazoline-type compounds that were oxidized to bispyrazoles.

These bisazole derivatives are important biologically active compounds, since several pharmacological activities, such as anti-allergic, antifungal, anti-inflammatory, antitumor as well as cytotoxic properties have already been described.

In the continuation of an on-going research program devoted to the synthesis and characterization of these bis-heterocyclic ring systems, it seemed to be a challenging task to study the reaction of 3-(3-aryl-3-oxopropenyl)chromen-4-ones 1 with substituted hydrazines. Our work in this chemical transformation and a detailed NMR analysis of the structural features of the novel pyrazolyl-2-pyrazolines will be presented and discussed.

**Results and Discussion**

*Synthesis*

In a previous work, we reported that pyrazolyl-2-pyrazoline derivatives were obtained as the major products in the reaction of 3-(3-aryl-3-oxopropenyl)chromen-4-ones 1 with hydrazine hydrate in refluxing acetic acid. However, 1-acetyl-3-aryl-5-(3-chromonyl)-2-pyrazolines were also isolated as minor products (2-6% yield). The isolation and characterization of these by-products led to the reaction mechanism rationalization. It was postulated that the α,β-unsaturated ketone moiety reacts first with one hydrazine molecule and then the chromone moiety reacts with a second molecule, leading to the formation of pyrazolyl-2-pyrazoline derivatives. Following this work and our interest on the synthesis and structural characterization of bis-1,2-azoles we treated 3-(3-aryl-3-oxopropenyl)chromen-4-ones 1a-g with an excess of phenylhydrazine in refluxing acetic acid and novel pyrazolyl-2-pyrazolines 2a-g were obtained in good yields (62-73%) (Scheme 1).
Here, the formation of pyrazolyl-2-pyrazolines 2a-g also implies that both chromone and α,β-unsaturated ketone moieties reacted with phenylhydrazine. Contrary to the referred previous work no by-products were isolated or detected even under careful chromatographic study. However, it can be postulated that (3-chromonyl)-2-pyrazoline-type compounds 4 are the primary reaction intermediates (Scheme 2). In a second reaction step these (3-chromonyl)-2-pyrazolines 4 react with another molecule of phenylhydrazine to provide the final products 2 (Scheme 2). In this transformation the reaction of the chromone moiety with phenylhydrazine, a non symmetric hydrazine derivative, in acidic medium can proceed in two different ways, due to the equilibrium between the non protonated and protonated form of the more nucleophilic phenylhydrazine amino group (Scheme 2). In the former case, the more nucleophilic amino group attacks the chromone C-2 carbon, in a conjugate-type addition, with consequent pyran ring opening leading to intermediates 5. Then, the intramolecular reaction between the other amino group (NHPh) and the carbonyl unit lead to pyrazolyl-2-pyrazoline derivatives 2. In the protonated hydrazine molecule NHPh becomes the nucleophile attacking the chromone C-2 carbon and after a pyran ring opening give rise to intermediates 6. The pyrazole ring may be formed by an intramolecular reaction between the other amino group of 7 and the carbonyl unit, resulting in the formation of compound 8. Our results indicate that the transformation proceeded by pathway a) of the reaction mechanism (Scheme 2). In fact, NHPh is a very week nucleophile implying that even in acidic medium, it is the more nucleophilic NH2 group that reacts with the chromone moiety.

Scheme 1
Scheme 2

To help the structural elucidation of the new synthesized 3-aryl-1-phenyl-5-{4-[5-(2-hydroxyphenyl)-1-phenylpyrazolyl]}-2-pyrazolines 2a-g, some of them 2a-c,e were acetylated with acetic anhydride in refluxing pyridine to afford acetylated derivatives 3a-c,e (Scheme 1).

Structure elucidation

The structures of all new compounds have been elucidated by using elemental analysis, mass spectrometry, infrared spectrophotometry and an exhaustive NMR study. Elemental analysis and mass spectrometric measurements unequivocally prove the elemental composition of compounds 2a-g and 3a-c,e.

The $^1$H NMR spectrum of pyrazolyl-2-pyrazoline 2c in CDCl$_3$ at room temperature (22ºC) showed extreme line broadening for almost all signals in the aliphatic and aromatic regions. This line broadening is especially evident in the proton resonances of the 5'-hydroxyphenyl group, H-3' and H-4, while H-5 appears as two broad signals centered at $\delta$ 4.99 and 5.10 ppm. Raising the temperature to 50ºC resulted in a clear sharpening of H-3' and both H-4 signals, while those of
5'-hydroxyphenyl group remained broad and there was the coalescence of both H-5 signals. Decreasing the temperature to 5°C, two sets of signals for the resonance of each proton were observed being easily identified in the aliphatic region (Figure 1). These data indicate that pyrazolyl-2-pyrazoline 2c exist as a mixture of diastereomers at room temperature in the chloroform solution. This diastereoselectivity arises from the combination of the pyrazoline C-4 stereocenter and two planar chiral subunits due to an internal possible hindered rotation involving the pyrazoline moiety and the 5'-hydroxyphenyl group. Under these conditions there is a slowly rotation between these moieties compared to the NMR time scale, reason why broad signals were observed in the 1H and 13C NMR spectra. At 5°C the two diastereomers were frozen and shape signals for the resonances of each one of the isomers were observed.

Since the temperature of 50°C was not enough to overcome the energy barrier of the referred hindered rotation of compound 2c and the temperature could not be further increased (b.p. CDCl3 = 61°C), DMSO-d6 was chosen as solvent to reach higher temperatures. In this case, once again, at temperatures between 22°C and 50°C some broad signals still appeared. Only at 60°C sharp resonances in the aliphatic and aromatic regions of both 1H and 13C NMR spectra were observed, indicating that the energy barrier was overcome and there was no more diastereomers, but a mixture of enantiomers. A maximum resolution spectrum of a DMSO-d6 solution of 2c was obtained at 80°C (Figure 2; vide experimental part).

From the described results it was decided to acquire the 1H and 13C NMR spectra of pyrazolyl-2-pyrazolines 2a-g at 5°C in CDCl3 allowing the structural characterization description of both diastereomers.

**Figure 1.** Aliphatic region of the 1H NMR spectrum of pyrazolyl-2-pyrazoline 2c in CDCl3: a) at 50°C; b) at room temperature (22°C); and c) at 5°C.
Figure 2. Aliphatic region of the $^1$H NMR spectrum of pyrazolyl-2-pyrazoline 2c in DMSO-d$_6$: a) at 80°C; b) at 60°C and c) at room temperature (22°C).

Although the aliphatic region of the $^1$H NMR spectra of pyrazolyl-2-pyrazolines 2a-e presents two groups of three signals, due to the H-4 and H-5 resonances of both diastereomers, it was not possible to unequivocally assign their structures and ratio, being thus designated as diastereomer A and diastereomer B. Based on the analysis of 2D COSY and NOESY experiments, the proton resonances of the 2-pyrazoline ring of both diastereomers have been assigned: diastereomer A - H-4$_{trans}^{15}$ (dd, $\delta = 3.07$-$3.14$ ppm, $J_{gem} = 16$-$17$ Hz, $J_{trans} \sim 7$ Hz), H-4$_{cis}$ (dd, $J_{gem} = 16$-$17$ Hz, $J_{cis} \sim 12$ Hz, $\delta = 3.47$-$3.55$ ppm) and H-5 (dd, $\delta = 5.10$-$5.17$ ppm); diastereomer B - H-4$_{trans}^{13}$ (dd, $\delta = 3.13$-$3.20$ ppm, $J_{gem} = 17$-$18$ Hz, $J_{trans} = 7$-$8$ Hz), H-4$_{cis}$ (dd, $J_{gem} = 17$-$18$ Hz, $J_{cis} = 12$-$13$ Hz, $\delta = 3.41$-$3.51$ ppm) and H-5 (dd, $\delta = 4.93$-$4.98$ ppm). In the case of compounds 2f,g, the presence of the naphthyl group shifted the H-4$_{cis}$ and H-4$_{trans}$ resonances for higher frequency values ($\Delta \delta \approx +0.2$ ppm) when compared with those of 2a-e. The proton resonance of pyrazole H-3’ of both isomers have been assigned to the two singlets at $\delta = 7.57$-$7.63$ ppm for compounds 2a-e and $\delta = 7.62$-$7.66$ ppm for compounds 2f,g (except for derivative 2c that appear as only one singlet at $\delta = 7.63$ ppm).

The complexity of the aromatic region in the $^1$H NMR spectra of pyrazolyl-2-pyrazolines 2a-g did not allow to unequivocally assign the proton resonances of each one of the diastereomers A and B. However, the 2D NMR experiments (COSY, NOESY, HSQC and HMBC spectra) permitted to assign each proton of both diastereomers. The most easily identified are the proton resonances to the 3-aryl group. In fact, in the case of compounds 2a-e H-2’’,6’’ appeared at high
values of frequency, $\delta = 7.55$-$7.66$ ppm, due to the deshielding anisotropic effect of the C=N pyrazoline bond. These assignments were also confirmed by the strong NOE cross peaks observed in the NOESY spectra between these protons and those of 4-CH$_2$. The remaining proton resonances of this 3-aryl ring were assigned based on the 2D COSY correlations. The proton assignments of the 1-phenyl group were based on the strong NOE cross peaks between the signal of H-5 and those of H-2”’,$6”’$, complemented by the correlations observed in the 2D COSY spectra.

It is also worth to mention the OH proton resonance appearing as a broad singlet ($\delta 5.08$-$5.20$ ppm) in the $^1$H NMR spectra of the para-substituted compounds 2b-e.

The HSQC spectra allowed the assignment of all protonated carbons of 2a-g, while the HMBC connectivities permitted to confirm some of them and to assign those of the quaternary carbons (Figure 3). In fact, the main HMBC connectivities of 2a-c were: 4-CH$_2$, H-5 and H-2”’,$6”’$ $\rightarrow$ C-3 ($\delta_C = 145.8$-$148.3$ ppm); H-5 $\rightarrow$ C-3’ ($\delta_C = 138.3$-$139.1$ ppm); 4-CH$_2$ and H-3’ $\rightarrow$ C-4’ ($\delta_C = 123.9$-$124.1$ ppm); H-5 and H-3’ $\rightarrow$ C-5’ ($\delta_C = 135.8$-$136.5$ ppm).

The OH acetylation of derivatives 2a-c,e promoted a better resolution of the $^1$H NMR spectra (mainly in the aromatic region) of compounds 3a-c,e and permitted to assign the individual protons of each one of the diastereomers. Thus, it was also possible to assign the ratio of both diastereomers, based in the area of the methyl signals of the acetyl group. The proportion of both isomers as $52.0$-$54.5\%$ and $45.5$-$48.0\%$ led us to designate them as major and minor diastereomers, respectively (vide experimental part).

The main features of the NMR data of these acetylated derivatives 3a-c,e were the resonances of the acetyl (CH$_3$ at $\delta_H = 2.04$-$2.13$ ppm and $\delta_C = 20.8$-$20.9$ ppm) and carbonyl at ($\delta_C = 168.3$-$168.6$ ppm) groups. Other important characteristics were the resonances of H-4$^{\text{trans}}$ ($\delta = 3.19$-$3.24$ ppm) and H-5 ($\delta = 5.15$-$5.23$ ppm) of the major diastereomers appearing at higher frequency than those of the minor ones ($\delta = 3.16$-$3.22$ and $5.07$-$5.15$ ppm, respectively). In the case of H-4$^{\text{cis}}$ the signal of the major diastereomer appeared at lower frequency ($\delta = 3.62$-$3.66$ ppm) than that of the minor one ($\delta = 3.68$-$3.74$ ppm). The NOE cross peaks between the 4-CH$_2$ signals and the doublets at $\delta = 7.60$-$7.75$ ppm and of the H-5 and the doublets at $\delta = 7.00$-$7.09$ ppm allowed to assign these chemical shifts to the resonances of H-2”’,$6”’$ and H-2”,$6”$, respectively. The COSY correlations permitted to assign the other protons of these two aromatic rings. The assignment of the remaining proton and carbon resonances in the NMR spectra of 3a-c,e unit was similar to that describe for compounds 2a-c,e and were based on the 2D NMR spectra (the main HMBC and NOESY correlations are presented in figure 3).
Figure 3. Main correlations observed in the HMBC and NOESY spectra of both diastereomers of pyrazolyl-2-pyrazolines 2a-g and 3a-c,e.

Conclusion

Following our work on pyrazoles we synthesised a new series of 3-aryl-5-{4-[5-(2-hydroxyphenyl)-1-phenylpyrazolyl]}-2-pyrazolines 2a-g from the reaction of 3-(3-aryl-3-oxopropenyl)chromen-4-ones 1a-g with phenylhydrazine in refluxing acetic acid. A study on the structural characterisation of these compounds by NMR at different temperatures showed a mixture of diastereomers at room temperature. The intramolecular hindered rotation combined with the pyrazoline C-4 stereocenter induces this diastereoselectivity, situation that are to the best of our knowledge reported for the first time.

Experimental Section

General. Melting points were determined on a Kofler hot-stage apparatus and are uncorrected. NMR spectra were recorded on Bruker Avance 300 spectrometer (300.13 MHz for $^1$H and 75.47 MHz for $^{13}$C), at 5°C with CDCl$_3$ as solvent, if not stated otherwise. Chemical shifts (δ) are reported in ppm and coupling constants (J) in Hz; internal standard was TMS. $^1$H assignments were made by 2D $^1$H COSY and NOESY (mixing time 800 ms) experiments, while $^{13}$C assignments were made with the aid of 2D $^1$H HSQC and gHMBC (delays for one bond and long-range J C/H couplings were optimised for 145 and 7 Hz, respectively) experiments. The IR spectra were obtained with a Perkin-Elmer16 PC instrument. Mass spectra (CI) were recorded on a VG trio-2 apparatus. Elemental analyses (CHN) were measured in-house with a Carlo Erba 1106 instrument. TLC was performed on silica gel 60 F$_{254}$ (Merck) layer using toluene: ethyl acetate (4:1, v/v) as eluent. The starting materials 1a-g were synthesized according to known procedures.$^3$
General procedure for the synthesis of pyrazolyl-2-pyrazolines 2a-g

A mixture of the appropriate 3-(3-aryl-3-oxopropenyl)chromen-4-ones 1a-g (5.0 mmol), phenylhydrazine (4.9 mL, 50.0 mmol) and acetic acid (50 mL) was reflux for 3 hours and then poured into water (100 mL) and ice (50 g). The precipitate was filtered, washed with water and recrystallized from methanol to obtain crystalline 3-aryl-5-[4-[5-(2-hydroxy-phenyl)-1-phenylpyrazolyl]}-2-pyrazolines 2a-g (Scheme 1).

1,3-Diphenyl-5-[4-[5-(2-hydroxyphenyl)-1-phenylpyrazolyl]}-2-pyrazoline (2a). Obtained as white needles in 68% yield, m.p. 137-138°C.

Diastereomeric mixture – NMR at 5°C:

Diastereomer A: 1H NMR δ 3.14 (dd, 1 H, J = 6.7 and 16.0 Hz, H-4\text{cis}), 3.54 (dd, 1 H, J = 12.1 and 16.0 Hz, H-4\text{cis}), 5.16 (dd, 1 H, J = 6.7 and 12.1 Hz, H-5).

Diastereomer B: 1H NMR δ 3.20 (dd, 1 H, J = 7.1 and 17.5 Hz, H-4\text{trans}), 3.51 (dd, 1 H, J = 11.8 and 17.5 Hz, H-4\text{cis}), 4.98 (dd, 1 H, J = 7.1 and 11.8 Hz, H-5).

Diastereomers A+B: 1H NMR δ 6.66 (d, 1 H, J = 7.5 Hz, H-3'), 6.73-6.89 (m, 4H, 2xH-4'''), 6.91 (d, 1 H, J = 8.2 Hz, H-3'''), 6.92-7.00 (m, 2 H, 2xH-6''''), 7.01 (d, 2 H, J = 8.3 Hz, H-2'',6''), 7.09-7.39 (m, 25 H, 2xH-3',5''', 2xH-3'',4''',5''', 2xH-2'',3''',4''',5'',6'' and 2xH-4'''''), 7.59 and 7.62 (2s, 2 H, 2xH-3'), 7.660 and 7.664 (2d, 4 H, J = 7.8 and 8.1 Hz, 2xH-2'',6'''). 13C NMR δ 42.0 and 42.3 (C-4'), 55.6 and 56.3 (C-5), 113.6 and 113.8 (C-2'',6''), 116.2, 116.4 and 116.5 (C-1'''' and C-3'''''), 119.1 and 119.4 (C-4''), 120.4 and 120.7 (C-5''), 123.8 and 124.1 (C-2'',5'', 124.0 (C-4''), 125.6 and 125.7 (C-2'',6''), 127.3 (C-4''), 128.4, 128.6, 128.7 and 128.8 (C-3'',5''; C-3'',4'',5'' and C-3'',5'', 131.0 (C-4'''), 131.3 and 131.7 (C-6'''), 132.3 and 132.6 (C-1'''), 135.9 and 136.2 (C-5'), 138.51 and 138.52 (C-3'), 139.2 and 139.4 (C-1'''), 144.6 and 144.8 (C-1''), 147.1 and 148.0 (C-3), 153.9 and 154.5 (C-2'''''). IR (cm⁻¹): 1596, 1502, 1386, 1334, 1290, 1178, 1156, 1014, 758, 692; MS (EI 70 eV): m/z(%): 456 (M⁺, 78), 364 (38), 261 (47), 91 (100). Anal. Calcd. for C₃₀H₂₄N₄O: C, 78.93; H, 5.30; N, 12.27. Found: C, 78.84; H, 5.36; N, 12.35.

5-[4-[5-(2-Hydroxyphenyl)-1-phenylpyrazolyl]}-3-(4-methylphenyl)-1-phenyl-2-pyrazoline (2b). Prepared as white needles in 66% yield, m.p. 154-155°C.

Diastereomeric mixture – NMR at 5°C:

Diastereomer A: 1H NMR δ 2.37 (s, 3 H, CH₃), 3.13 (dd, 1 H, J = 6.6 and 16.4 Hz, H-4\text{trans}), 3.55 (dd, 1 H, J = 12.1 and 16.4 Hz, H-4\text{cis}), 5.13 (dd, 1 H, J = 6.6 and 12.1 Hz, H-5), 5.14 (br s, 1 H, OH). 

Diastereomer B: 1H NMR δ 2.37 (s, 3 H, CH₃), 3.19 (dd, 1 H, J = 7.2 and 16.6 Hz, H-4\text{trans}), 3.51 (dd, 1 H, J = 12.1 and 16.6 Hz, H-4\text{cis}), 4.97 (dd, 1 H, J = 7.2 and 12.1 Hz, H-5), 5.14 (br s, 1 H, OH).

Diastereomers A+B: 1H NMR δ 6.72-6.83 (m, 3 H, 2xH-4'' and H-3'''''), 6.83-6.93 (m, 3 H, H-3'''' and 2xH-5'''''), 6.94 (d, 1 H, J = 7.4 Hz, H-6'''''), 6.99-7.03 (m, 3 H, H-2'',6'' and H-6'''''), 7.08-7.25 (m, 22 H, H-2'',6''; 2xH-3'',5''; 2xH-3'',5''; 2xH-2'',3''',4'',5'',6'' and 2xH-4'').
7.57 (d, 4 H, J = 7.9 Hz, 2xH-2”",6””), 7.60 and 7.63 (2s, 2 H, 2xH-3’’). 13C NMR δ 21.4 (CH3), 42.0 and 42.4 (C-4), 55.6 and 56.3 (C-5), 113.5 and 113.9 (C-2”,6””), 116.2, 116.5 and 116.6 (C-1”’” and C-3”’”), 118.9 and 119.4 (C-4”), 120.5 and 120.7 (C-5”’”), 123.7 and 124.3 (C-2””,6””), 124.1 (C-4”), 125.6 and 125.7 (C-2””,6””), 127.2 (C-4”’”), 128.7, 128.8 and 129.2 (C-3”,5””; C-3”,5”” and C-3”’”,5”’”), 131.0 (C-4”’’”), 129.7 and 129.8 (C-1”’’”), 131.3 and 131.7 (C-6”’’’”), 135.8 and 136.1 (C-5”), 138.5 and 138.7 (C-4”’’”), 138.6 (C-3”), 139.2 and 139.4 (C-1”’’”), 144.7 and 145.0 (C-1”), 147.3 and 148.3 (C-3”), 153.9 and 154.4 (C-2”’’’”). IR (cm⁻¹): 1598, 1498, 1380, 1320, 1230, 1182, 1156, 1122, 1068, 1034, 854, 692, 662; MS (EI 70 eV): m/z(%) = 470 (M⁺, 81), 247 (19), 208 (19), 91 (100). Anal. Calcd. for C31H26N4O: C, 79.13; H, 5.57; N, 11.90. Found: C, 79.22; H, 5.63; N, 11.82.

5-{4-[5-(2-Hydroxyphenyl)-1-phenylpyrazolyl]-3-(4-methoxyphenyl)-1-phenyl-2-pyrazoline (2c). Isolated as white plates in 64% yield, m.p. 142-143°C.

1H NMR (DMSO-d₆, 80°C) δ 3.19 (dd, 1 H, J = 6.7 and 17.1 Hz, H-4trans), 3.63 (dd, 1 H, J = 11.6 and 17.1 Hz, H-4cis), 3.79 (s, 3 H, OCH3), 5.03 (dd, 1 H, J = 6.7 and 11.6 Hz, H-5), 6.70 (t, 1 H, J = 7.2 Hz, H-4”), 6.84 (dd, 1 H, J = 7.4 and 7.6 Hz, H-5”’’”), 6.92 (d, 1 H, J = 8.5 Hz, H-3”’’’”), 6.96 (d, 2 H, J = 8.9 Hz, H-3”’’’’), 7.02 (d, 2 H, J = 8.3 Hz, H-2”,6”), 7.05 (dd, 1 H, J = 1.8 and 7.6 Hz, H-6”’’’’), 7.12 (dd, 2 H, J = 7.2 and 8.3 Hz, H-3”’’’r,5”), 7.20-7.27 (m, 7 H, H-4”, H-2””,3”’”’”,4”’”,5”’”,6”’’’; H-4”’’’”), 7.45 (s, 1 H, H-3”), 7.61 (d, 2 H, J = 8.9 Hz, H-2””,6”), 9.55 (br s, 1 H, OH). 13C NMR (DMSO-d₆, 80°C) δ 41.8 (C-4”), 54.9 (OCH3), 55.5 (C-5”), 113.0 (C-2””,6”), 113.8 (C-3”’’’”,5”’’’”), 115.7 (C-1”’’’”), 116.3 (C-3”’’’”), 118.0 (C-4”), 118.8 (C-5”’’’”), 122.9 (C-4”’ and C-2”’’’’”,6”), 125.0 (C-1”), 126.2 (C-4”’””), 126.7 (C-2””,6”), 128.05 and 128.12 (C-3”,5””’” and C-3”’’’”,5”’’’”), 130.2 (C-4”’’’”), 131.2 (C-6”’’’”), 136.7 (C-5”), 137.3 (C-3”), 139.8 (C-1”’’’”), 144.7 (C-1”), 147.3 (C-3”), 155.2 (C-2”’’””), 159.6 (C-4”’””).

Diastereomeric mixture – NMR at 5°C:

Diastereomer A: 1H NMR δ 3.10 (dd, 1 H, J = 6.8 and 16.7 Hz, H-4trans), 3.51 (dd, 1 H, J = 12.4 and 16.7 Hz, H-4cis), 3.82 (s, 3 H, OCH3), 5.10 (dd, 1 H, J = 6.8 and 12.4 Hz, H-5), 5.20 (br s, 1 H, OH).

Diastereomer B: 1H NMR δ 3.15 (dd, 1 H, J = 7.5 and 16.9 Hz, H-4trans), 3.46 (dd, 1 H, J = 12.2 and 16.9 Hz, H-4cis), 3.82 (s, 3 H, OCH3), 4.93 (dd, 1 H, J = 7.5 and 12.2 Hz, H-5), 5.20 (br s, 1 H, OH).

Diastereomers A+B: 1H NMR δ 6.73-6.85 (m, 5 H, 2xH-4””, H-3”’’” and 2xH-5”’’’”), 6.85-6.90 (m, 4 H, 2xH-3”’’””,5”’’””), 6.90-6.93 (m, 2 H, H-3”’’” and H-6”’’’”), 6.99-7.01 (m, 3 H, H-2””,6”” and H-6”’’’”), 7.08-7.24 (m, 18 H, H-2”,6”; 2xH-3”””,5””; 2xH-2””,3”’”’”,4””’”,5””’”,6””’”; 2xH-4”’’’’”), 7.59 (d, 2 H, J = 8.6 Hz, H-2””,6”), 7.60 (s, 1 H, H-3”), 7.61 (d, 2 H, J = 8.7 Hz, H-2””,6”), 7.63 (s, 1 H, H-3”). 13C NMR δ 42.0 and 42.5 (C-4”), 55.3 (OCH3), 55.6 and 56.4 (C-5”), 113.4 and 113.9 (C-2””,6”), 113.8 (C-3”’’’”,5”’’’’”), 116.1, 116.4 and 116.6 (C-1”’’’” and C-3”’’’”), 118.8 and 119.3 (C-4”), 120.4 and 120.6 (C-5”’’’”), 123.7 and 124.1 (C-2””,6”), 124.0 and 124.3 (C-4”), 125.1 and 125.3 (C-1”’’”), 127.1 and 127.3 (C-2””,6”), 127.3 (C-4”’’”), 128.7 and 128.8 (C-3”,5”” and C-3”’’’”,5”’’’”),
131.0 (C-4’’’’), 131.3 and 131.6 (C-6’’’’), 135.8 and 136.1 (C-5’’), 138.4 and 139.3 (C-1’’’’), 138.6 and 139.1 (C-3’’), 144.9 and 145.1 (C-1’’), 147.2 and 148.3 (C-3), 153.9 and 154.4 (C-2’’’’), 159.8 and 159.9 (C-4’’’’). IR (cm⁻¹): 1596, 1498, 1386, 1252, 1110, 1068, 1032, 996, 692, 660; MS (EI 70 eV): m/z(%) = 486 (M⁺, 11), 394 (2), 247 (19), 91 (100). Anal. Calcd. for C₃₁H₂₆N₄O₂: C, 76.52; H, 5.39; N, 11.51. Found: C, 76.44; H, 5.43; N, 11.60.

3-(4-Fluorophenyl)-5-[4-[5-(2-hydroxyphenyl)-1-phenylpyrazolyl]-1-phenyl-2-pyrazoline (2d). Prepared as pale yellow needles in 71% yield, m.p. 136-137°C.

Diastereomeric mixture – NMR at 5°C:

**Diastereomer A**: ¹H NMR δ 3.08 (dd, 1 H, J = 6.7 and 17.0 Hz, H-4trans), 3.50 (dd, 1 H, J = 12.4 and 17.0 Hz, H-4cis), 5.08 (br s, 1 H, OH), 5.15 (dd, 1 H, J = 6.7 and 12.4 Hz, H-5).

**Diastereomer B**: ¹H NMR δ 3.14 (dd, 1 H, J = 7.1 and 17.6 Hz, H-4trans), 3.41 (dd, 1 H, J = 12.5 and 17.6 Hz, H-4cis), 4.97 (dd, 1 H, J = 7.1 and 12.5 Hz, H-5), 5.08 (br s, 1 H, OH).

**Diastereomers A+B**: ¹H NMR δ 6.71-6.85 (m, 5 H, 2xH-4”, H-3’’’’ and 2xH-5’’’’), 6.88-6.98 (m, 3 H, H-3’’’’ and 2xH-6’’’’), 6.98-7.01 (m, 2 H, H-2”’,6”), 7.04 (t, 4 H, J = 8.7 Hz, 2xH-3”’,5”’), 7.08-7.22 (m, 18 H, H-2”’,6”; 2xH-3”,5”; 2xH-2”’,3’’’’,4”’’,5”’’,6”’’; 2xH-4”’’’), 7.59 and 7.61 (2s, 2 H, 2xH-3”), 7.59-7.63 (m, 4 H, 2xH-2”’,6”’’). ¹³C NMR δ 42.0 and 42.3 (C-4), 55.8 and 56.3 (C-5), 113.5 and 113.7 (C-2”,6”), 115.5 (d, 2JC-F = 21.8 Hz, C-3”’,5”’), 116.1, 116.3 and 116.6 (C-1’’”’ and C-3’’’’), 119.1 and 119.4 (C-4”), 120.3 and 120.6 (C-5’’’’), 123.8 and 124.1 (C-2”’,6”’’), 124.0 and 124.1 (C-4”), 127.4 (d, 3JC-F = 4.0 Hz, C-2”’,6”’’), 127.4 and 127.5 (C-4”’”), 128.7 and 128.8 (C-3”’,5”’, C-1”’ and C-3’’’’,5”’’), 131.0 (C-4”’’’), 131.3 and 131.7 (C-6”’’’), 136.1 and 136.5 (C-5’), 138.4 (C-3”), 139.1 and 139.3 (C-1”’’), 144.6 and 144.7 (C-1”), 146.1 and 147.0 (C-3), 154.0 and 154.6 (C-2”’’’). 19F NMR δ 41.8 and 42.0 (C-4), 55.7 and 56.2 (C-5), 113.5 and 113.7 (C-2”’,6”’’), 116.0, 116.1, 116.2 and 116.4 (C-1”’’’’ and C-3’’’’), 119.2 and 119.5 (C-4”), 120.4 and 120.6 (C-5”’’’), 123.8

3-(4-Chlorophenyl)-5-[4-[5-(2-hydroxyphenyl)-1-phenylpyrazolyl]-1-phenyl-2-pyrazoline (2e). Obtained as yellow plates in 73% yield, m.p. 145-146°C.

Diastereomeric mixture – NMR at 5°C:

**Diastereomer A**: ¹H NMR δ 3.07 (dd, 1 H, J = 6.7 and 16.9 Hz, H-4trans), 3.47 (dd, 1 H, J = 12.2 and 16.9 Hz, H-4cis), 5.16 (br s, 1 H, OH), 5.17 (dd, 1 H, J = 6.7 and 12.2 Hz, H-5).

**Diastereomer B**: ¹H NMR δ 3.13 (dd, 1 H, J = 7.1 and 17.4 Hz, H-4trans), 3.42 (dd, 1 H, J = 12.1 and 17.4 Hz, H-4cis), 4.98 (dd, 1 H, J = 7.1 and 12.1 Hz, H-5), 5.16 (br s, 1 H, OH).

**Diastereomers A+B**: ¹H NMR δ 6.72 (d, 1 H, J = 8.3 Hz, H-3”’”), 6.73-6.87 (m, 4 H, 2xH-4”” and 2xH-5”’’’’’), 6.90 (d, 1 H, J = 8.3 Hz, H-3”’’’’’), 6.98 (d, 2 H, J = 8.1 Hz, H-2”’,6”), 7.10-7.25 (m, 20 H, H-2”,6”; 2xH-3”,5”; 2xH-2”’,3’’’’,4”’’,5”’’,6”’’; 2xH-4”’’’ and 2xH-6”’’’), 7.31 (d, 4 H, J = 8.6 Hz, 2xH-3”’,5”’), 7.55 (d, 4 H, J = 8.6 Hz, 2xH-2”’,6”’), 7.57 and 7.60 (2s, 2 H, 2xH-3”). ¹³C NMR δ 41.8 and 42.0 (C-4), 55.7 and 56.2 (C-5), 113.5 and 113.7 (C-2”’,6”’’), 116.0, 116.1, 116.2 and 116.4 (C-1”’’’’’ and C-3”’’’’’), 119.2 and 119.5 (C-4”), 120.4 and 120.6 (C-5”’’’’’), 123.8
and 124.1 (C-2””,6””), 123.9 and 124.0 (C-4”), 126.7 and 126.8 (C-2””,6””), 127.3 (C-4””), 128.6, 128.7 and 128.9 (C-3””,5””, C-3””,5””, and C-3””,5””), 131.0 (C-1””), 131.1 (C-4””), 131.3 and 131.7 (C-6””), 134.0 and 134.1 (C-4”), 136.1 and 136.4 (C-5”), 138.3 (C-3”), 139.0 and 139.3 (C-1””), 144.3 and 144.4 (C-1”), 145.8 and 146.6 (C-3), 153.9 and 154.5 (C-2””). IR (cm⁻¹): 1598, 1500, 1386, 1266, 1176, 1154, 1042, 1032, 938, 854, 708, 662; MS (EI 70 eV): m/z(%) = 490 (M⁺, 14), 247 (15), 91 (81), 77 (100). Anal. Calcd. for C₃₀H₂₃ClN₄O: C, 73.39; H, 4.72; N, 11.41. Found: C, 73.47; H, 4.67; N, 11.48.

5-[4-[5-(2-Hydroxyphenyl)-1-phenylpyrazolyl]-3-(1-naphthyl)-1-phenyl-2-pyrazoline (2f).

Prepared as white needles in 62% yield, m.p. 154-155°C.

Diastereomeric mixture – NMR at 5°C:

Diastereomer A: ¹H NMR δ 3.35 (dd, 1 H, J = 6.5 and 16.5 Hz, H-4trans), 3.76 (dd, 1 H, J = 12.0 and 16.5 Hz, H-4cis), 5.18 (dd, 1 H, J = 6.5 and 12.0 Hz, H-5).

Diastereomer B: ¹H NMR δ 3.41 (dd, 1 H, J = 7.2 and 17.0 Hz, H-4trans), 3.67 (dd, 1 H, J = 11.5 and 17.0 Hz, H-4cis), 4.99 (dd, 1 H, J = 7.2 and 11.5 Hz, H-5).

Diastereomers A+B: ¹H NMR δ 6.73-6.76 (m 1 H, H-3””), 6.80-6.85 (m, 2 H, 2xH-5”’”), 6.92-6.99 (m, 3H, 2xH-4”” and H-3””), 7.08 (d, 2 H, J = 9.1 Hz, H-2””,6””), 7.12-7.25 (m, 20 H, , H-2””,6””, 2xH-3””,5””; 2xH-2””,3””,4””,5””,6””; 2xH-4”” and 2xH-6””), 7.35-7.46 (m, 2 H, H-6”” and 2xH-7””), 7.46-7.53 (m, 1 H, H-6””), 7.56 (d, 2 H, J = 7.1 Hz, 2xH-4””), 7.64-7.69 (m, 2 H, 2xH-3””), 7.64 and 7.66 (2s, 2 H, H-3”), 7.79 (br d, 2 H, J = 6.8 Hz, 2xH-8””), 7.87 (d, 2 H, J = 8.5 Hz, 2xH-5””), 9.43 and 9.45 (2d, 2 H, J = 8.2 Hz, 2xH-2””). ¹³C NMR δ 44.4 and 44.7 (C-4”), 54.4 and 55.0 (C-5), 113.6 and 113.8 (C-2””,6””), 116.1, 116.4, 116.5 and 116.6 (C-1”””, and C-3”””), 119.1 and 119.5 (C-4”), 120.5 and 120.7 (C-5”””), 123.8 and 124.0 (C-4’), 123.9 and 124.1 (C-2””,6””), 124.8 (C-4”” and C-6””), 126.1 (C-7””), 127.2 (C-2””, C-3”” and C-4””), 128.5, 128.7, 128.8 and 128.9 (C-5”””, C-3””,5””, and C-3””,5””), 129.4 and 129.5 (C-8””), 130.2 and 130.3 (C-1””), 131.0 (C-4”””), 131.4 and 131.5 (C-6”””), 133.6 and 134.0 (C-4a”” and C-8a””), 136.0 and 136.3 (C-5”), 138.5 (C-3”), 139.2 and 139.3 (C-1””), 144.4 and 144.7 (C-1””), 147.6 and 148.5 (C-3), 153.9 and 154.5 (C-2”””). IR (cm⁻¹): 1596, 1498, 1380, 1292, 1176, 1142, 1032, 1020, 996, 910, 856, 718, 656; MS (EI 70 eV): m/z(%) = 506 (M⁺, 2), 247 (44), 127 (74), 77 (100). Anal. Calcd. for C₃₄H₂₆N₄O: C, 80.61; H, 5.17; N, 11.05. Found: C, 80.53; H, 5.11; N, 11.10.

5-[4-[5-(2-Hydroxyphenyl)-1-phenylpyrazolyl]-3-(2-naphthyl)-1-phenyl-2-pyrazoline (2g).

Isolated as white needles in 69% yield, m.p. 160-161°C.

Diastereomeric mixture – NMR at 5°C:

Diastereomer A: ¹H NMR δ 3.27 (dd, 1 H, J = 6.6 and 16.6 Hz, H-4trans), 3.67 (dd, 1 H, J = 11.9 and 16.9 Hz, H-4cis), 5.21 (dd, 1 H, J = 6.6 and 11.9 Hz, H-5).

Diastereomer B: ¹H NMR δ 3.32 (dd, 1 H, J = 7.1 and 16.9 Hz, H-4trans), 3.61 (dd, 1 H, J = 11.5 and 16.6 Hz, H-4cis), 5.05 (dd, 1 H, J = 7.1 and 11.5 Hz, H-5).
Diastereomers A+B: 1H NMR δ 6.74 (d, 1 H, J = 7.8 Hz, H-3''), 6.76-6.91 (m, 4 H, 2xH-4'' and 2xH-5''), 6.93 (d, 1 H, J = 7.9 Hz, H-3''), 7.05 (d, 4 H, J = 8.1 Hz, H-2'',6''), 7.13-7.24 (m, 18 H, 2xH-3'',5''; 2xH-2'',3'',4'',5'',6'''; 2xH-4''' and 2xH-6'''''); 7.45-7.50 (m, 4 H, 2xH-6'''' and 2xH-7'''), 7.62 and 7.65 (2s, 2 H, H-3'), 7.72 (br s, 2 H, 2xH-1'''), 7.70-7.83 (m, 4 H, 2xH-5'' and 2xH-8''), 7.82 (d, 2 H, J = 8.7 Hz, 2xH-4'''), 8.10 (dd, 2 H, J = 8.7 and 1.5 Hz, 2xH-3'''). 13C NMR δ 41.9 and 42.1 (C-4), 55.7 and 56.2 (C-5), 113.6 and 113.8 (C-2'',6''), 116.0, 116.1, 116.4 and 116.5 (C-1'''' and C-3'''''), 119.1 and 119.5 (C-4''), 120.5 and 120.7 (C-5'''''), 123.4 and 123.7 (C-3''), 123.95 and 124.09 (C-4''), 123.98 and 124.08 (C-2'',6'''), 125.1 (C-1''), 126.3 and 126.4 (C-6'''' and C-7'''''), 127.3 (C-4''''), 127.8 and 128.0 (C-4'', C-5'''' and C-8'''''), 128.7, 128.9 and 129.0 (C-3'',5'' and C-3''''',5'''''), 131.0 (C-4'''''), 131.3 and 131.7 (C-6'''''), 133.0 and 133.2 (C-2'', C-4a'' and C-8a'''), 135.8 and 136.2 (C-5''), 138.5 and 138.6 (C-3''), 139.1 and 139.4 (C-1'''''), 144.4 and 144.6 (C-1''), 147.1 and 147.9 (C-3), 153.9 and 154.4 (C-2''''). IR (cm⁻¹): 1596, 1498, 1382, 1274, 1178, 1032, 1020, 996, 856, 718, 662; MS (EI 70 eV): m/z(%) = 506 (M⁺, 18), 247 (18), 91 (87), 77 (100). Anal. Calcd. for C₃₄H₂₆N₄O: C, 80.61; H, 5.17; N, 11.05. Found. C, 80.67; H, 5.22; N, 11.12.

Acetylation of 3-aryl-5-[4-5-(2-hydroxyphenyl)-1-phenylpyrazolyl]-1-phenyl-2-pyrazolines 2a-c,e
A mixture of 3-aryl-5-[4-5-(2-hydroxyphenyl)-1-phenylpyrazolyl]-1-phenyl-2-pyrazolines 2a-c,e (1.0 mmol), acetic anhydride (10.0 mL) and anhydrous pyridine (5.0 mL) was refluxed for 6 hours, then poured onto crushed ice (20 g). The precipitate was filtered, washed with water (3 x 20 mL) and recrystallized from methanol to afford 5-[4-5-(2-acetoxyphenyl)-1-phenylpyrazolyl]-3-aryl-1-phenyl-2-pyrazolines 3a-c,e (Scheme 1).

5-[4-5-(2-Acetoxyphenyl)-1-phenylpyrazolyl]-1,3-diphenyl-2-pyrazoline (3a). Isolated as white needles in 82% yield, m.p. 157-158°C.

Diastereomeric mixture 53.5:46.5:
Major diastereomer: 1H NMR δ 2.05 (s, 3 H, CH₃), 3.24 (dd, 1 H, J = 6.8 and 16.7 Hz, H-4trans), 3.66 (dd, 1 H, J = 12.16 and 16.7 Hz, H-4cis), 5.21 (dd, 1 H, J = 6.8 and 12.1 Hz, H-5), 6.78-6.84 (m, 1 H, H-4''), 7.09 (d, 2 H, J = 7.7 Hz, H-2'',6''), 7.20 (t, 2 H, J = 7.7 Hz, H-3'',5''), 7.20-7.42 (m, 11 H, Ar), 7.43-7.52 (m, 1 H, H-4'''''), 7.63 (s, 1 H, H-3'), 7.69-7.72 (m, 2 H, H-2'',6'''); 13C NMR δ 20.8 (CH₃), 42.6 (C-4), 55.9 (C-5), 113.6 (C-2'',6''), 119.2 (C-4''), 123.37 (C-2'',6'''), 123.42 (C-1'''''), 123.8 (C-4'''''), 124.3 (C-4''), 125.7 (C-2'',6'''), 126.2 (C-3'''''), 127.17 (C-4'''''), 128.4 (C-3'',5''), 128.52 (C-5'''''), 128.77 (C-3'',5'''), 128.79 (C-3'',5'''), 130.7 (C-4'''''), 131.8 (C-6'''''), 132.69 (C-1'''''), 134.7 (C-5''), 139.0 (C-3'), 139.9 (C-1'''''), 144.7 (C-1''), 147.0 (C-3), 148.3 (C-2'''''), 168.3 (C=O).
Minor diastereomer: 1H NMR δ 2.13 (s, 3 H, CH₃), 3.22 (dd, 1 H, J = 6.9 and 16.9 Hz, H-4trans), 3.74 (dd, 1 H, J = 11.9 and 16.9 Hz, H-4cis), 5.13 (dd, 1 H, J = 6.9 and 11.9 Hz, H-5), 6.78-6.84 (m, 1 H, H-4''), 7.02 (d, 2 H, J = 7.7 Hz, H-2'',6''), 7.18 (t, 2 H, J = 7.7 Hz, H-3'',5''),
7.20-7.42 (m, 11 H, Ar), 7.43-7.52 (m, 1 H, H-4”), 7.67 (s, 1 H, H-3’), 7.71-7.75 (m, 2 H, H-2”,6”). 13C NMR δ 20.9 (CH3), 42.5 (C-4), 56.2 (C-5), 113.9 (C-2”,6”), 119.4 (C-4”), 122.9 (C-1”), 123.5 (C-2”,6”), 123.8 (C-4”), 124.2 (C-4”), 125.7 (C-2”,6”), 126.6 (C-3””), 127.15 (C-4”), 128.53 (C-3”,5”), 128.6 (C-5””), 128.77 (C-3””,5””), 128.81 (C-3”,5””), 130.8 (C-4”), 131.6 (6”), 132.73 (C-1”), 134.5 (C-5”), 138.6 (C-3’), 139.6 (C-1”), 144.9 (C-1”), 147.3 (C-3), 149.0 (C-2”), 168.6 (C=O).


5-[4-[5-(2-Acetoxyphenyl)-1-phenylpyrazolyl]-3-(4-methoxyphenyl)-1-phenyl-2-pyrazoline (3b). Prepared as white needles in 83% yield, m.p. 197-198°C.

Diastereomeric mixture 54:45:6:

Major diastereomer: 1H NMR δ 2.05 (s, 3 H, CH3), 2.38 (s, 3 H, 4’”-CH3), 3.22 (dd, 1 H, J = 6.9 and 16.7 Hz, H-4trans), 3.64 (dd, 1 H, J = 12.1 and 16.7 Hz, H-4cis), 5.17 (dd, 1 H, J = 6.9 and 12.1 Hz, H-5), 6.76-6.83 (m, 1 H, H-4”), 7.07 (dd, 2 H, J = 1.1 and 8.9 Hz, H-2”,6”), 7.14-7.34 (m, 12 H, Ar), 7.44-7.49 (m, 1 H, H-4’”), 7.66 (s, 1 H, H-3’), 7.60 (d, 2 H, J = 8.3 Hz, H-2”,6”). 13C NMR δ 20.8 (CH3), 21.4 (4’”-CH3), 42.8 (C-4), 55.8 (C-5), 113.6 (C-2”,6”), 119.1 (C-4”), 123.44 (C-1’”), 123.45 (C-2””,6””), 123.8 (C-4’”), 124.4 (C-4’”), 125.7 (C-2””,6””), 126.2 (C-3””), 127.1 (C-5””), 128.75, 128.77 and 128.81 (C-3””,5”” and C-3””,5””), 129.17 (C-3’”,5’”), 129.90 (C-1’”), 130.6 (C-4’”), 131.8 (C-6””), 134.7 (C-5’), 138.6 (C-4’”), 139.0 (C-3’), 139.9 (C-1’”), 144.9 (C-1’), 147.3 (C-3), 148.3 (C-2””), 168.3 (C=O).

Minor diastereomer: 1H NMR δ 2.13 (s, 3 H, CH3), 2.40 (s, 3 H, 4’”-CH3), 3.20 (dd, 1 H, J = 6.9 and 16.9 Hz, H-4trans), 3.72 (dd, 1 H, J = 11.9 and 16.9 Hz, H-4cis), 5.10 (dd, 1 H, J = 6.9 and 11.9 Hz, H-5), 6.76-6.83 (m, 1 H, H-4”), 7.01 (d, 2 H, J = 1.2 and 8.8 Hz, H-2”,6”), 7.14-7.34 (m, 12 H, Ar), 7.44-7.49 (m, 1 H, H-4’”), 7.627 (d, 2 H, J = 8.3 Hz, H-2”,6”), 7.628 (s, 1 H, H-3’). 13C NMR δ 20.9 (CH3), 21.4 (4’”-CH3), 42.6 (C-4), 56.1 (C-5), 113.9 (C-2””,6””), 119.2 (C-4”), 122.9 (C-1’”), 123.5 (C-2””,6””), 123.8 (C-4’”), 124.2 (C-4’”), 125.7 (C-2””,6””), 126.6 (C-3””), 127.2 (C-5””), 128.75, 128.77 and 128.81 (C-3””,5”” and C-3””,5””), 129.23 (C-3’”,5’”), 129.94 (C-1’”), 130.7 (C-4’”), 131.6 (C-6””), 134.5 (C-5’), 138.6 (C-4’”), 138.7 (C-3’), 139.6 (C-1’”), 145.0 (C-1’”), 147.5 (C-3), 149.0 (C-2””), 168.6 (C=O).

IR (cm⁻¹): 1763, 1597, 1499, 1382, 1325, 1188, 1131, 1069, 910, 816, 764, 693; MS (El 70 eV): m/z(%) = 512 (M⁺, 24), 469 (4), 247 (15), 91 (100). Anal. Calcd. for C33H28N4O2: C, 77.32; H, 5.51; N, 10.92. Found: C, 77.23; H, 5.46; N, 10.84.

5-[4-[5-(2-Acetoxyphenyl)-1-phenylpyrazolyl]-3-(4-methoxyphenyl)-1-phenyl-2-pyrazoline (3c). Isolated as white needles in 85% yield, m.p. 112-113°C.

Diastereomeric mixture 52:04:48:0: Major diastereomer: 1H NMR δ 2.05 (s, 3 H, CH3), 3.20 (dd, 1 H, J = 6.9 and 16.7 Hz, H-4trans), 3.62 (dd, 1 H, J = 12.0 and 16.7 Hz, H-4cis), 3.85 (s, 3 H, OCH3), 5.15 (dd, 1 H, J = 6.9
and 12.0 Hz, H-5), 6.79 (t, 1 H, J = 7.1 Hz, H-4”), 6.91 (d, 2 H, J = 8.7 Hz, H-3‴,5‴), 7.06 (d, 2 H, J = 8.2 Hz, H-2”,”6”), 7.14-7.34 (m, 10 H, Ar), 7.43-7.51 (m, 1 H, H-4‴), 7.63 (s, 1 H, H-3’), 7.64 (d, 2 H, J = 8.7 Hz, H-2”,”6”). 13C NMR δ 20.8 (CH3), 42.8 (C-4), 55.3 (OCH3), 55.9 (C-5), 113.5 (C-2’,”6”), 113.8 (C-3″,5″), 118.9 (C-4″), 122.9 (C-1‴), 123.36 (C-2‴,”6”), 123.8 (C-4‴), 124.5 (C-4”), 125.0 (C-1”‴), 126.2 (C-3‴), 127.2 (C-2″,”6”), 128.76, 128.81 and 128.82 (C-3″,”5”,” and C-5‴), 130.6 (C-4‴), 131.8 (C-6‴), 134.7 (C-5’), 139.0 (C-3’), 139.9 (C-1”″), 145.1 (C-1”‴), 147.1 (C-3), 148.3 (C-2‴), 160.1 (C-4‴), 168.3 (C=O).

**Minor diastereomer:** 1H NMR δ 2.12 (s, 3 H, CH3), 3.17 (dd, 1 H, J = 7.1 and 16.8 Hz, H-4trans), 3.70 (dd, 1 H, J = 11.8 and 16.8 Hz, H-4cis), 3.86 (s, 3 H, OCH3), 5.07 (dd, 1 H, J = 7.1 and 11.8 Hz, H-5), 6.79 (t, 1 H, J = 7.1 Hz, H-4”), 6.94 (d, 2 H, J = 8.7 Hz, H-3‴,5‴), 7.00 (d, 2 H, J = 8.2 Hz, H-2”,”6”), 7.14-7.34 (m, 10 H, Ar), 7.43-7.51 (m, 1 H, H-4‴), 7.68 (s, 1 H, H-3’), 7.65 (d, 2 H, J = 8.7 Hz, H-2”,”6”). 13C NMR δ 20.9 (CH3), 42.7 (C-4), 55.3 (OCH3), 56.2 (C-5), 113.9 (C-2’,”6”), 114.0 (C-3″,5″), 119.1 (C-4″), 122.8 (C-1”‴), 123.44 (C-2‴,”6”), 124.0 (C-4‴), 124.3 (C-4”), 126.0 (C-1”‴), 126.6 (C-3‴), 127.2 (C-2″,”6”), 128.76, 128.81 and 128.82 (C-3″,”5”,” and C-5‴), 130.7 (C-4‴), 131.6 (C-3″), 134.5 (C-5’), 138.6 (C-3’), 139.6 (C-1”″), 145.2 (C-1”‴), 147.4 (C-3), 148.6 (C-2‴), 160.0 (C-4‴), 168.6 (C=O).

IR (cm⁻¹): 1766, 1597, 1499, 1384, 1251, 1190, 1111, 1034, 910, 832, 762, 693; MS (EI 70 eV): m/z(%) = 528 (M+), 247 (89), 191 (100). Anal. Calcd. for C33H28N4O3: C, 74.98; H, 5.34; N, 10.59. Found: C, 74.89; H, 5.38; N, 10.50.

5-{4-[5-(2-Acetoxyphenyl)-1-phenylpyrazolyl]-3-(4-chlorophenyl)-1-phenyl-2-pyrazoline (3e). Obtained as pale yellow plates in 76% yield, m.p. 123-124°C.

**Diastereomeric mixture 54.5:45.5:**

**Major diastereomer:** 1H NMR δ 2.04 (s, 3 H, CH3), 3.19 (dd, 1 H, J = 6.9 and 16.8 Hz, H-4trans), 3.62 (dd, 1 H, J = 12.3 and 16.8 Hz, H-4cis), 5.23 (dd, 1 H, J = 6.9 and 12.3 Hz, H-5), 6.79-6.85 (m, 1 H, H-4’), 7.08 (d, 2 H, J = 8.2 Hz, H-2”,”6”), 7.15-7.33 (m, 10 H, Ar), 7.34 (d, 2 H, J = 8.5 Hz, H-3″,”5”), 7.42-7.51 (m, 1 H, H-4‴), 7.61 (d, 2 H, J = 8.5 Hz, H-2”,”6”), 7.62 (s, 1 H, H-3’). 13C NMR δ 20.76 (CH3), 42.4 (C-4), 56.0 (C-5), 113.6 (C-2″,”6”), 119.4 (C-4″), 123.35 (C-1”‴), 123.37 (C-2″,”6”), 123.8 (C-4‴), 124.1 (C-4”), 126.2 (C-3″), 126.8 (C-2”,”6”), 127.24 (C-5‴), 128.6 (C-3″,”5”), 128.80 and 128.82 (C-3″,”5” and C-3″,”5″), 130.7 (C-4″), 131.22 (C-1”″), 131.8 (C-6″), 134.16 (C-4‴), 134.8 (C-5’), 139.0 (C-3’), 139.8 (C-1”″), 144.4 (C-1”‴), 145.9 (C-3), 148.3 (C-2″), 168.3 (C=O).

**Minor diastereomer:** 1H NMR δ 2.10 (s, 3 H, CH3), 3.16 (dd, 1 H, J = 7.1 and 16.9 Hz, H-4trans), 3.68 (dd, 1 H, J = 12.0 and 16.9 Hz, H-4cis), 5.15 (dd, 1 H, J = 7.1 and 12.0 Hz, H-5), 6.79-6.85 (m, 1 H, H-4’), 7.02 (d, 2 H, J = 8.2 Hz, H-2”,”6”), 7.15-7.33 (m, 10 H, Ar), 7.36 (d, 2 H, J = 8.4 Hz, H-3″,”5”), 7.42-7.51 (m, 1 H, H-4‴), 7.64 (d, 2 H, J = 8.4 Hz, H-2”,”6”), 7.65 (s, 1 H, H-3’). 13C NMR δ 20.82 (CH3), 42.3 (C-4), 56.4 (C-5), 114.0 (C-2″,”6”), 119.6 (C-4″), 122.8 (C-1”‴), 123.5 (C-2″,”6”), 123.8 (C-4‴), 124.9 (C-4”), 126.6 (C-3″), 126.8 (C-2”,”6”), 127.22 (C-5‴), 128.7 (C-3″,”5”), 128.80 and 128.82 (C-3″,”5” and C-3″,”5″), 130.8
(C-4‴), 131.24 (C-1‴), 131.5 (6‴), 134.24 (C-4‴), 134.6 (C-5‴), 138.5 (C-3‴), 139.6 (C-1‴), 144.6 (C-1″), 146.1 (C-3), 148.9 (C-2‴), 168.6 (C=O).

IR (cm⁻¹): 1766, 1598, 1501, 1453, 1386, 1322, 1189, 1090, 1010, 910, 762, 692; MS (EI 70 eV): m/z (%): 532 (M⁺, 16), 489 (6), 247 (16), 91 (100); Anal. Calcd. for C₃₂H₂₅ClN₄O₂: C, 72.11; H, 4.73; N, 10.51. Found: C, 72.21; H, 4.78; N, 10.43.

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References


13. The TLC control of the reaction mixture showed only the presence of the obtained 3-aryl-5-{4-[5-(2-hydroxyphenyl)-1-phenylpyrazolyl]}-2-pyrazolines 2a-g and some degradation products.

14. The broadening of $^1$H and $^{13}$C NMR signals due to intramolecular hindered rotation is a well-known situation (e.g. Pihlaja, K.; Kleinpeter, E. *Carbon-13 NMR chemical shifts in structural and stereochemical analysis*, VCH, New York, **1994**, Chapter 6, p. 295-322), that led to atropisomers. In the present case, this phenomenon combined with the C-4 stereocenter led to the formation of two diastereomers, situation that are to the best of our knowledge reported for the first time.

15. The H-4$_{cis}$ and H-4$_{trans}$ configurations are referred to their relative position to H-5.