

Supplementary Material

Multi-gram synthesis of precursors of *bibrachial diaza-paracyclophanes*. Complexes with Zn²⁺, Cu²⁺ and Co²⁺ ions

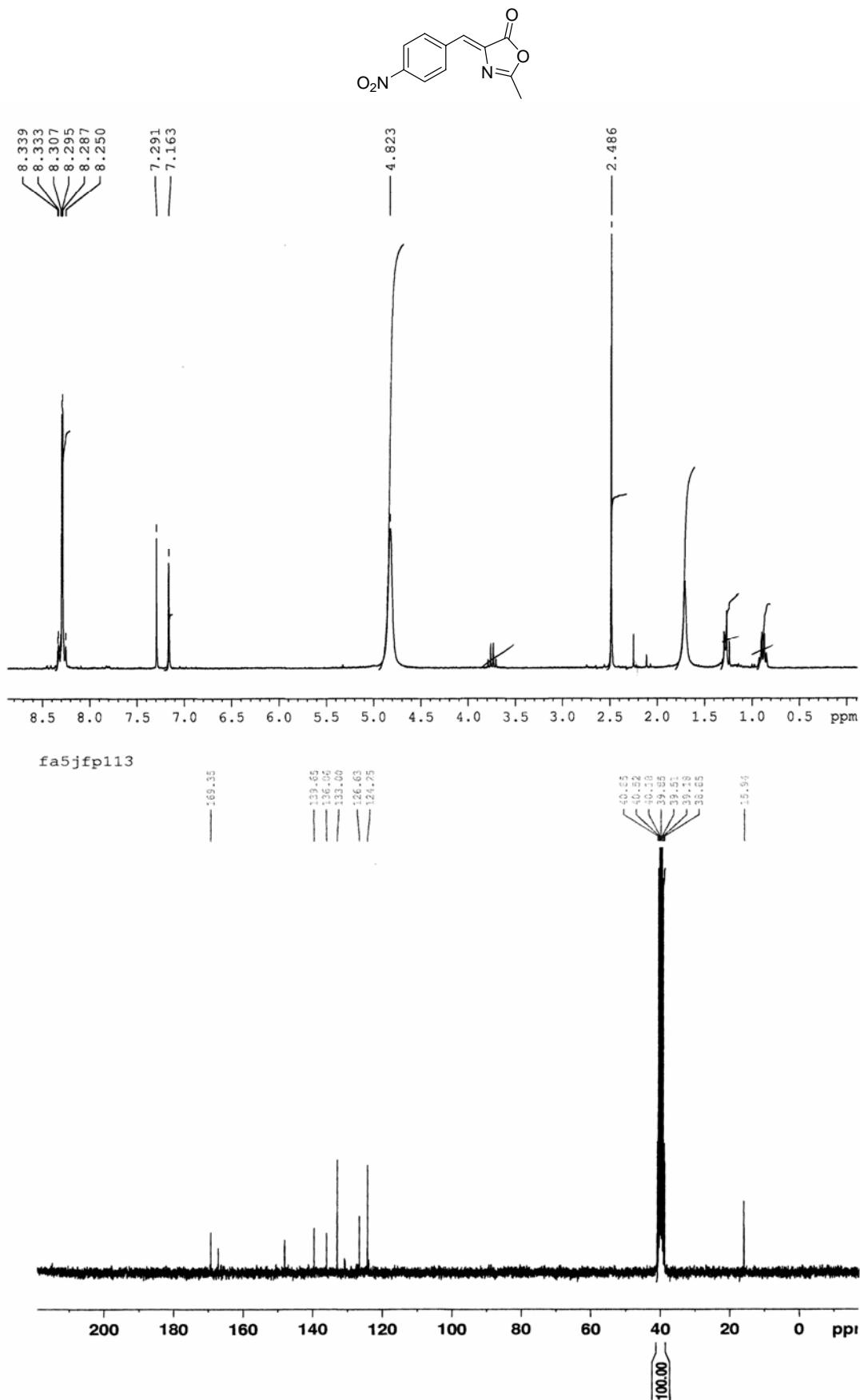
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J. Francisco González**

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Universidad Complutense, 28040 Madrid (Spain)*

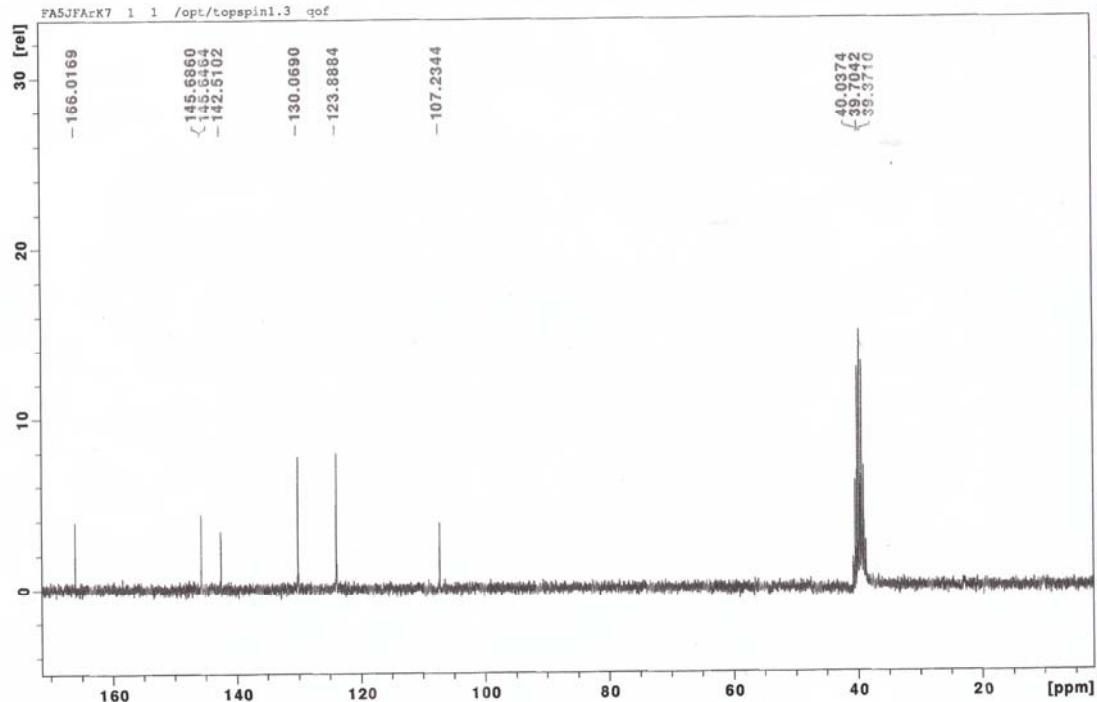
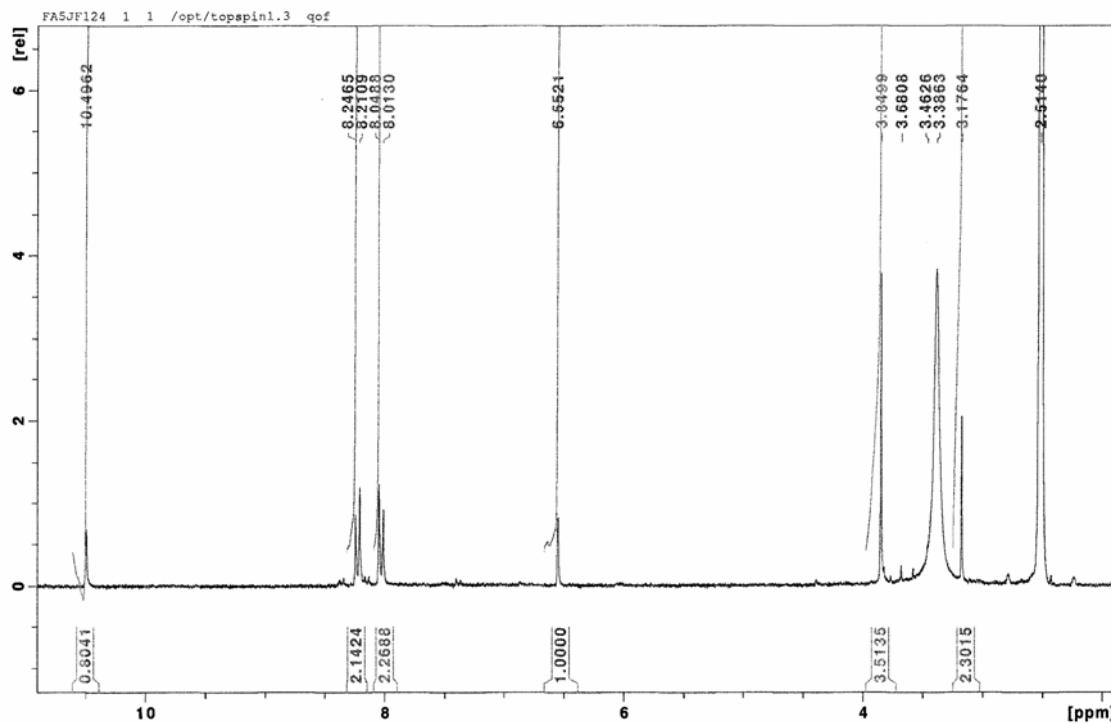
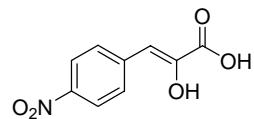
E-mail: avendano@farm.ucm.es

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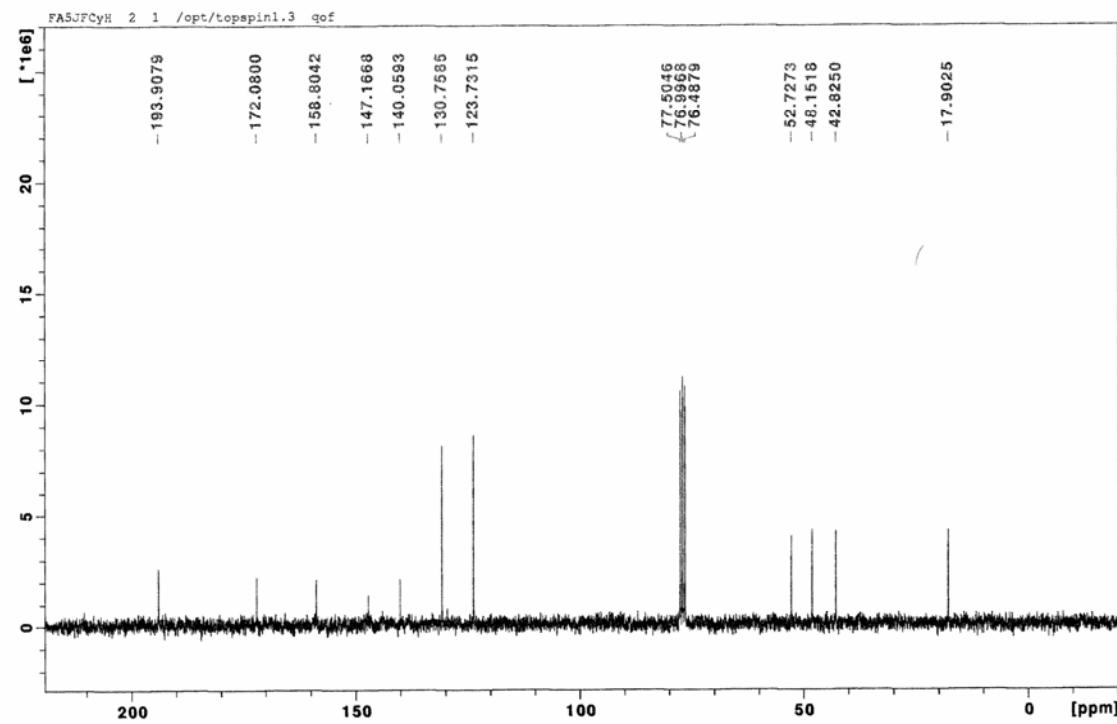
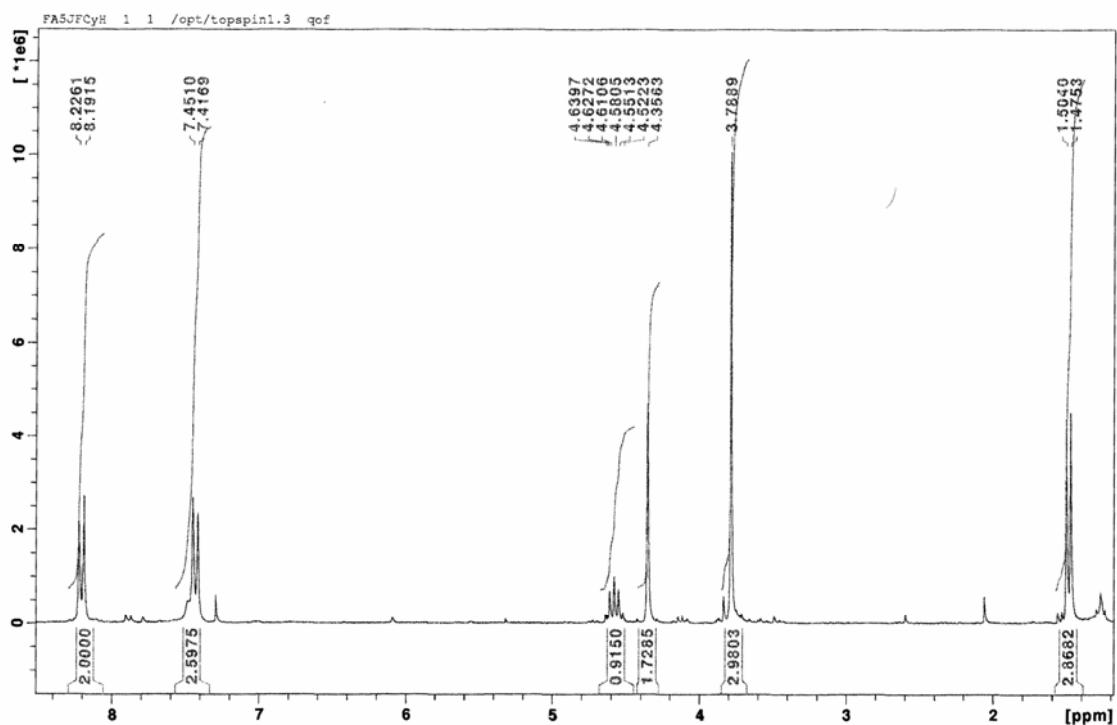
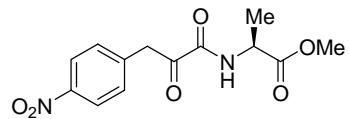
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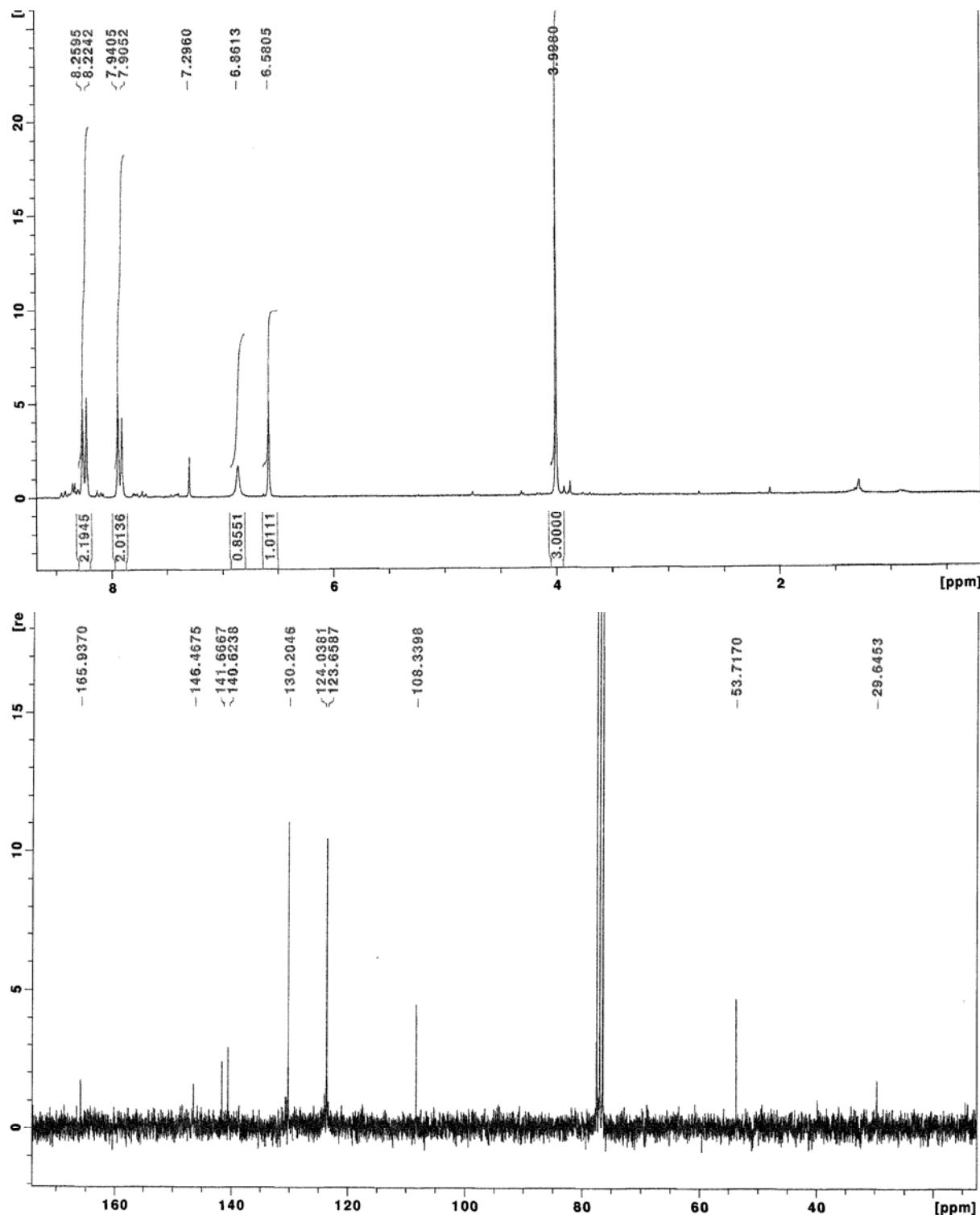
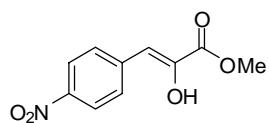
Compound 4



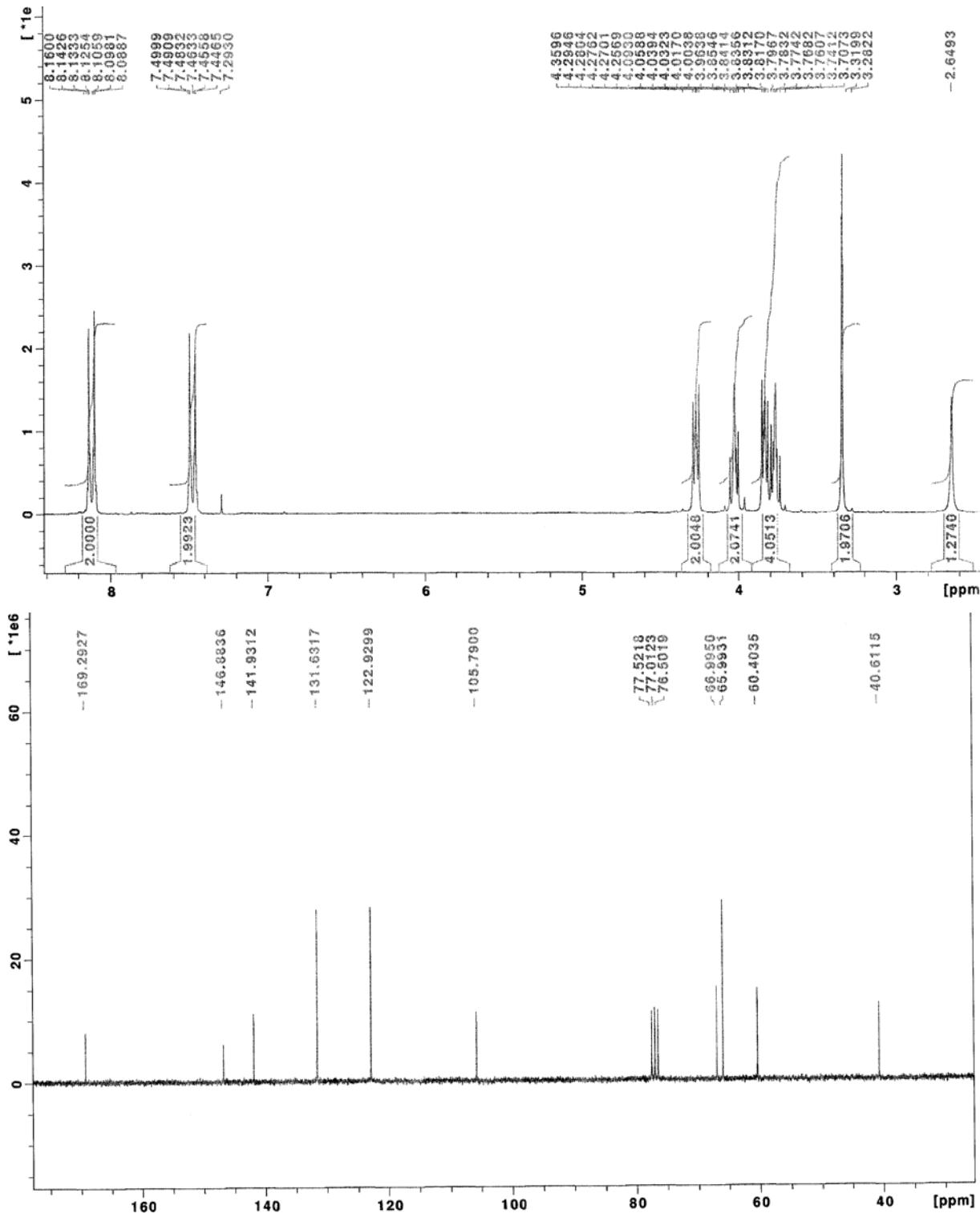
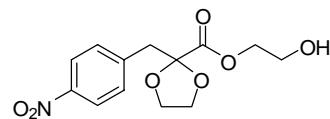
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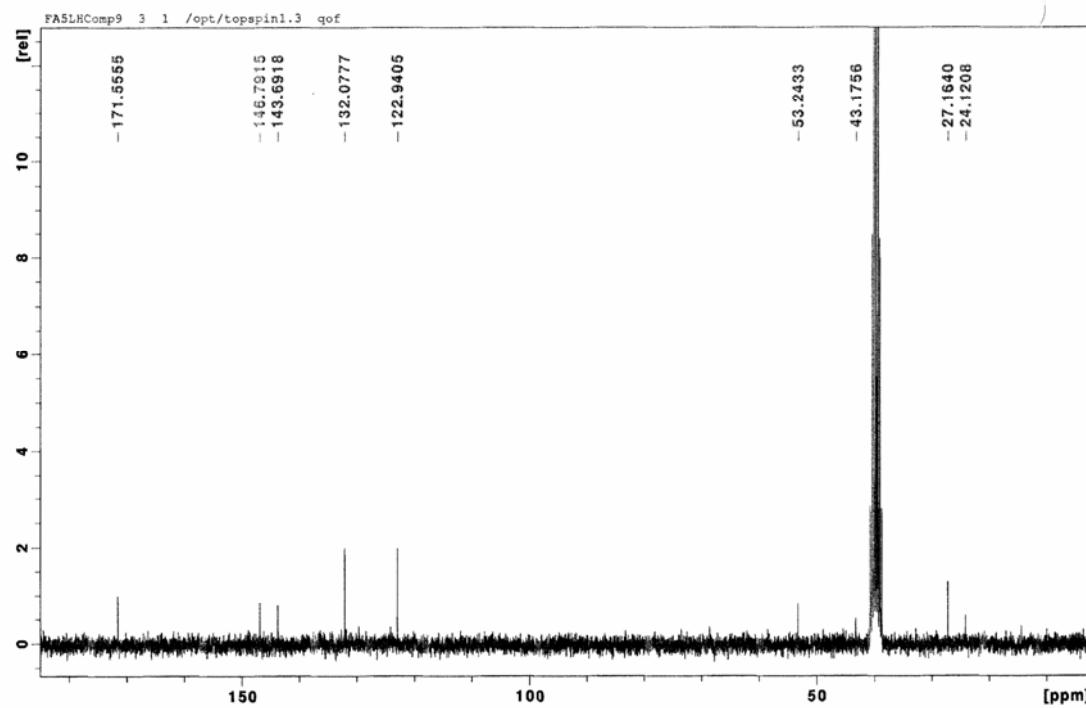
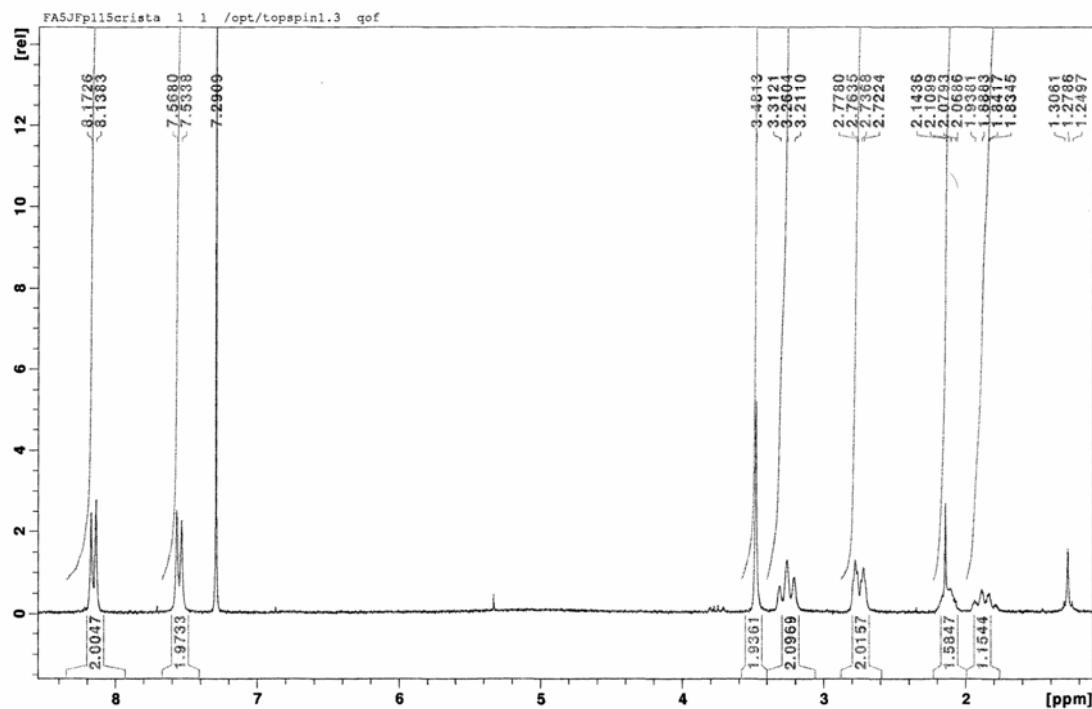
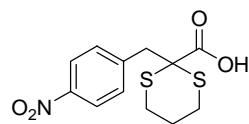
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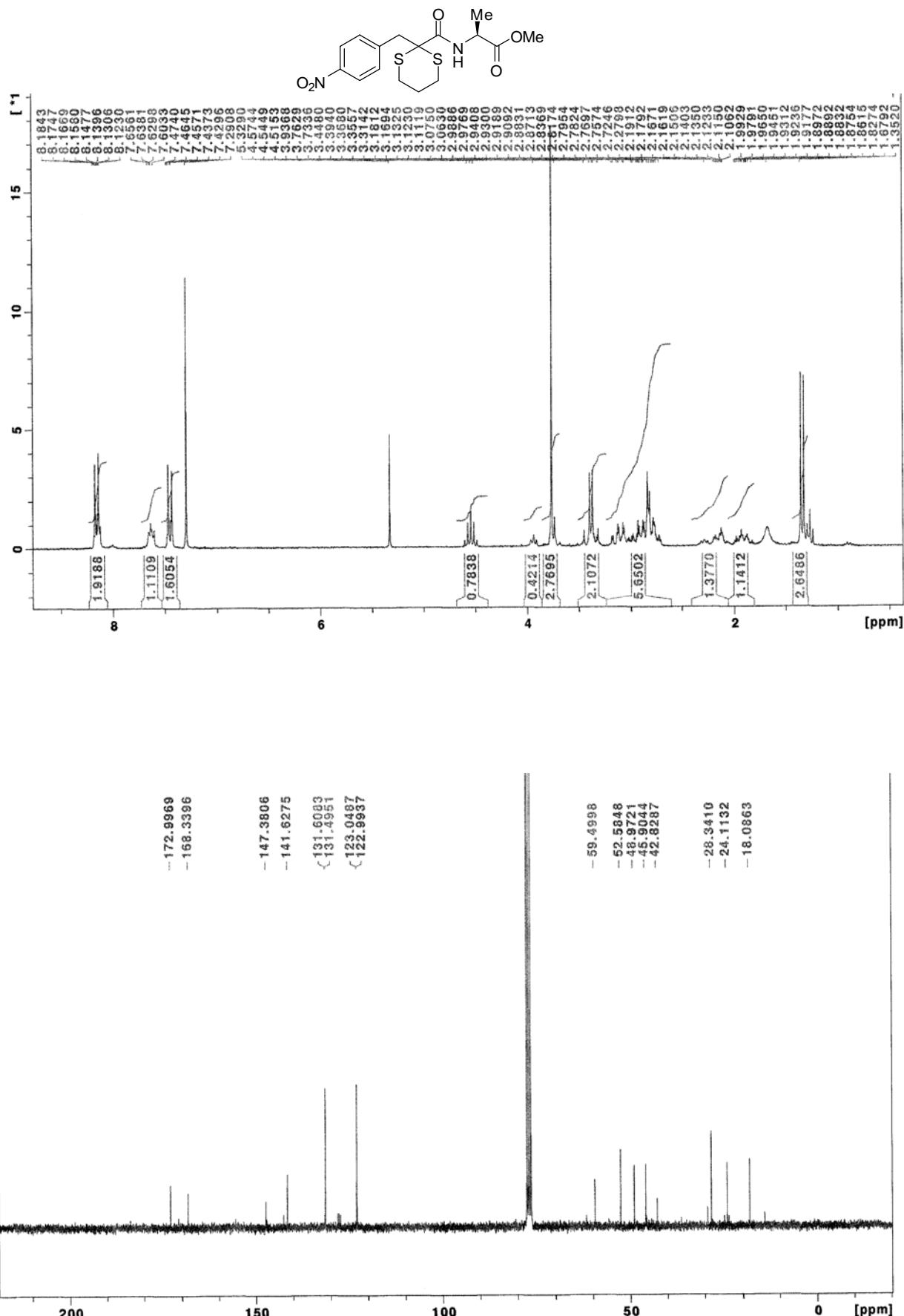


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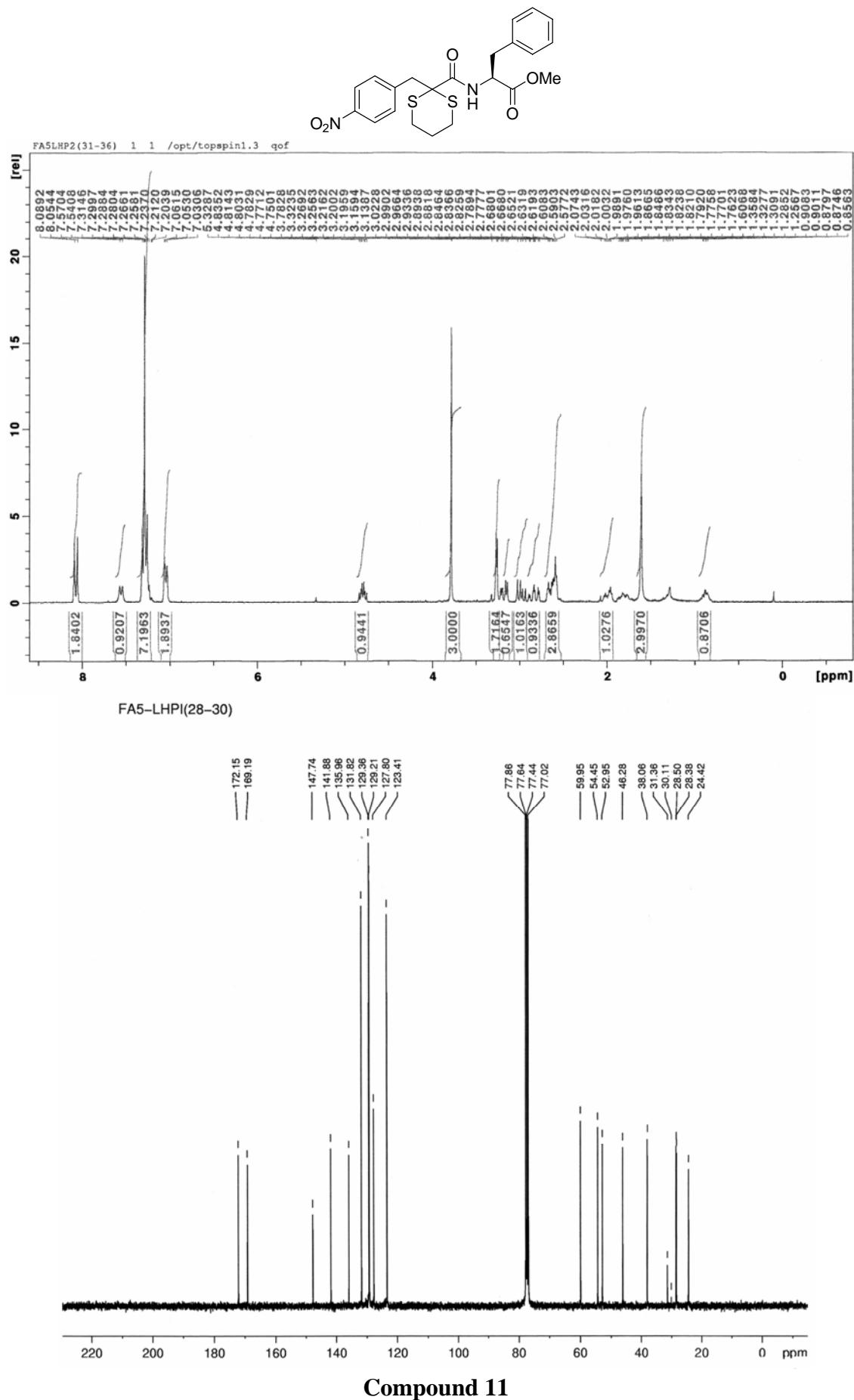


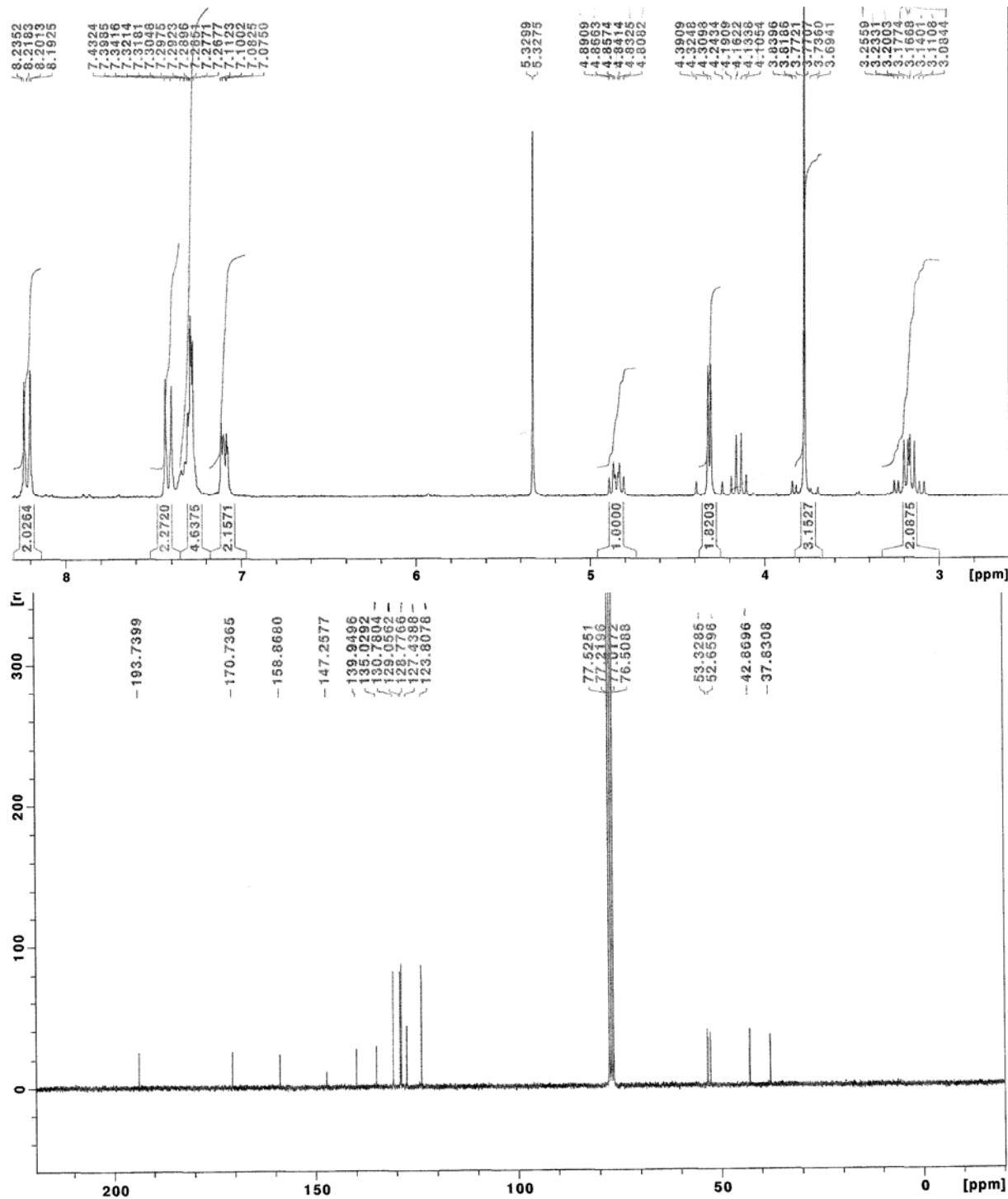
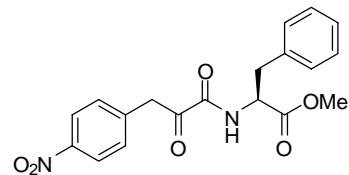
Compound 8

**Compound 9**

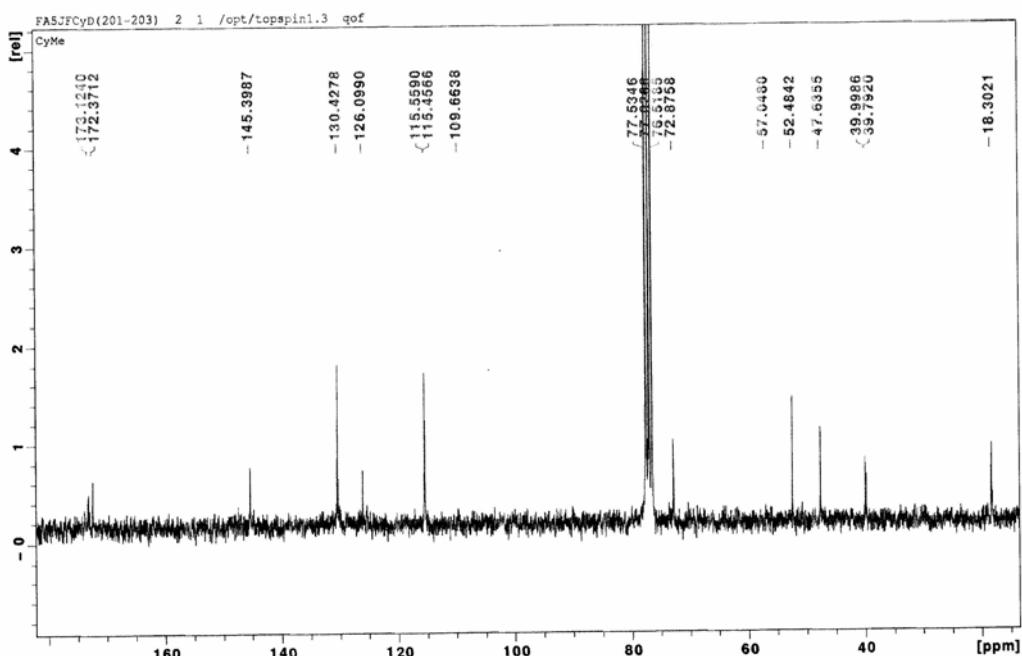
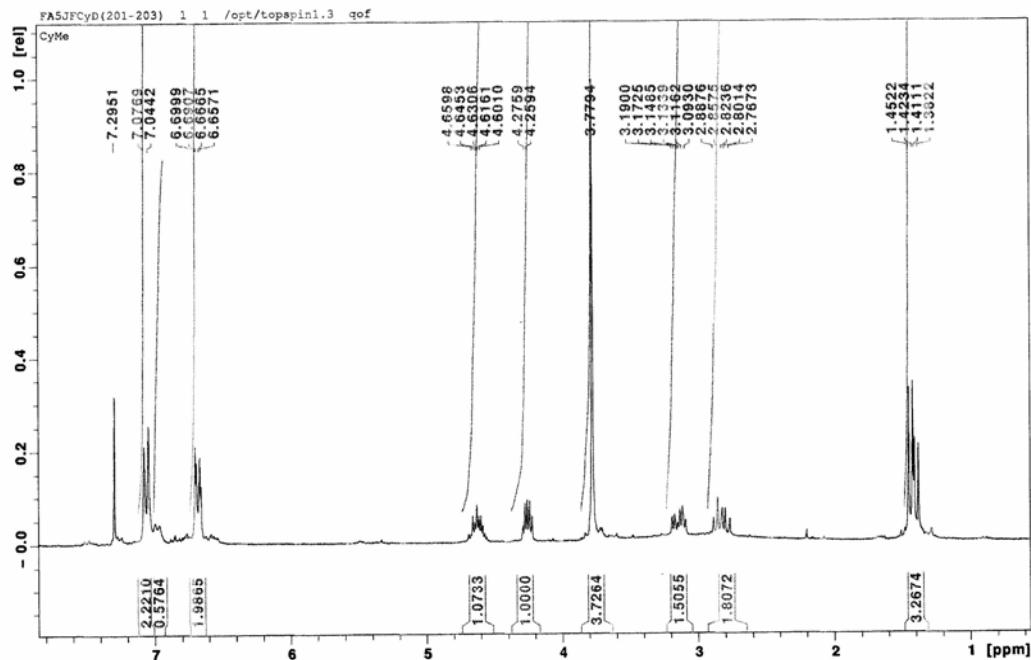
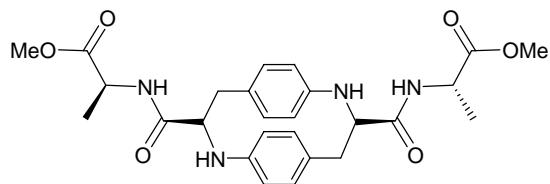


Compound 10

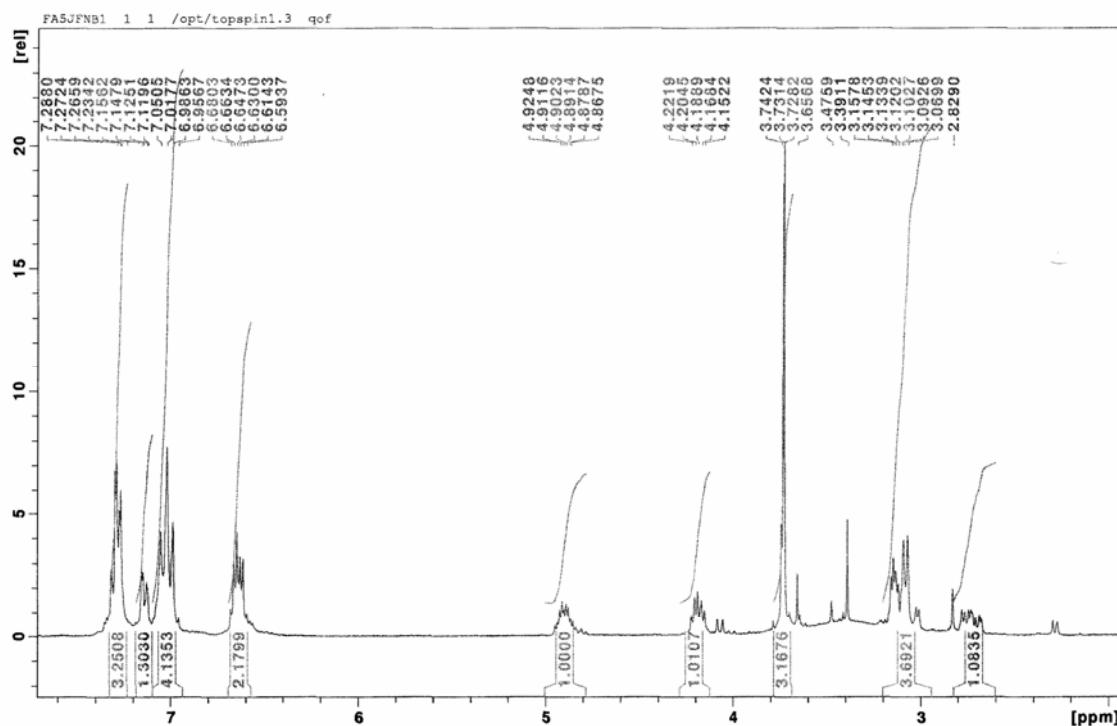
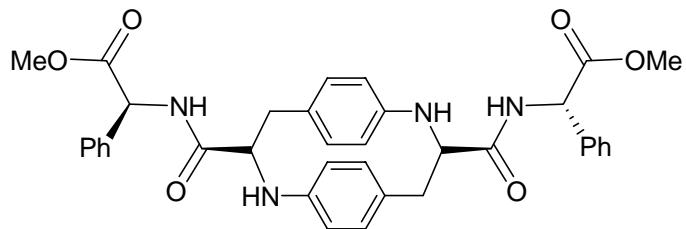




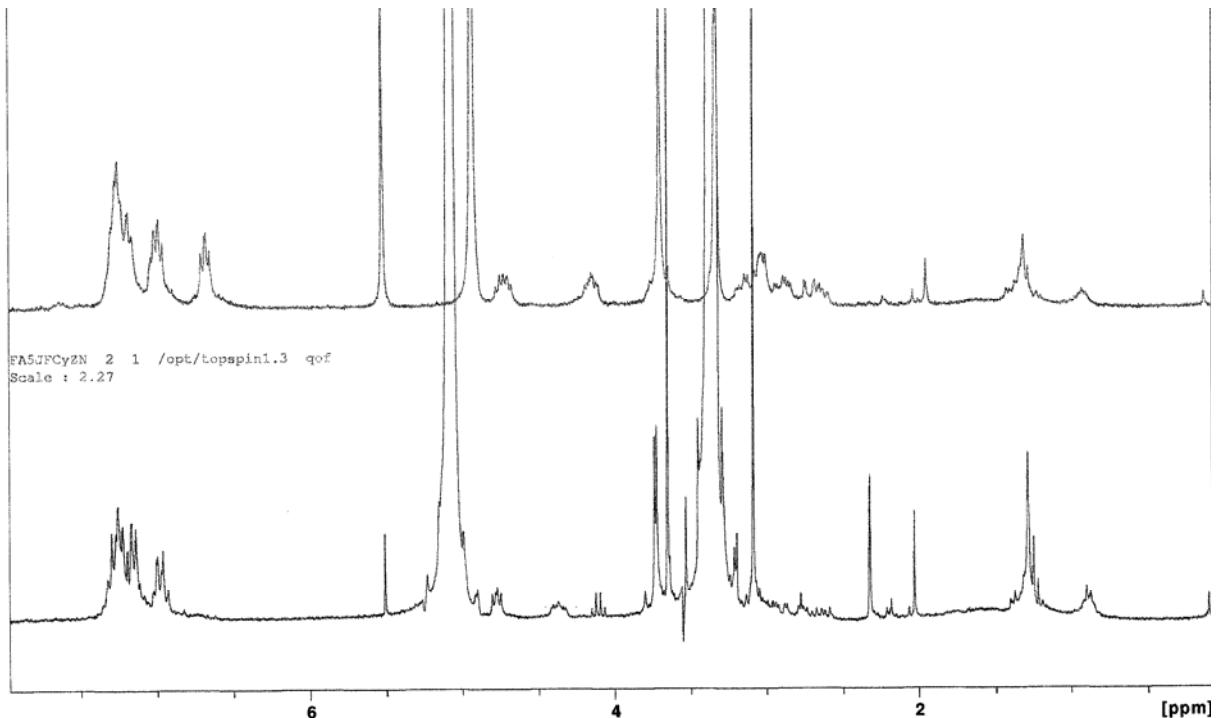
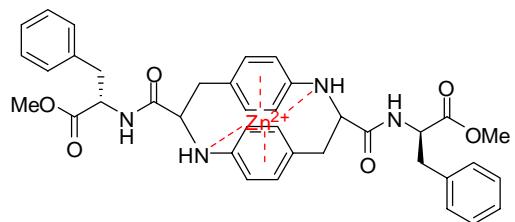
Compound 12



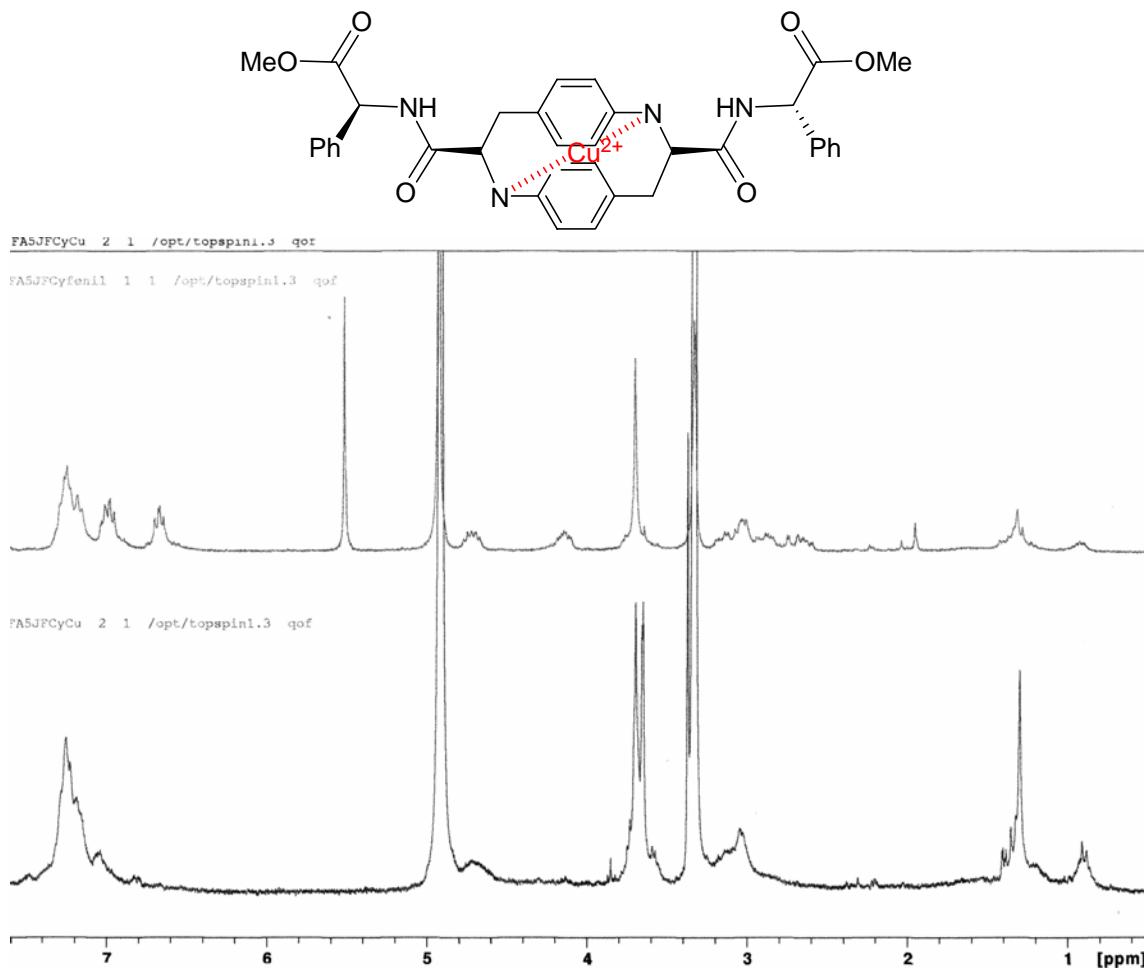
Compound 13



Compound 14



Compound [Zn²⁺ 14]
¹H NMR spectra of 14 in CD₃OD in the absence and presence of Zn²⁺



Compound $[\text{Cu}^{2+} \mathbf{14}]$
 ^1H NMR spectra of $\mathbf{14}$ in CD_3OD in the absence and presence of Cu^{2+}

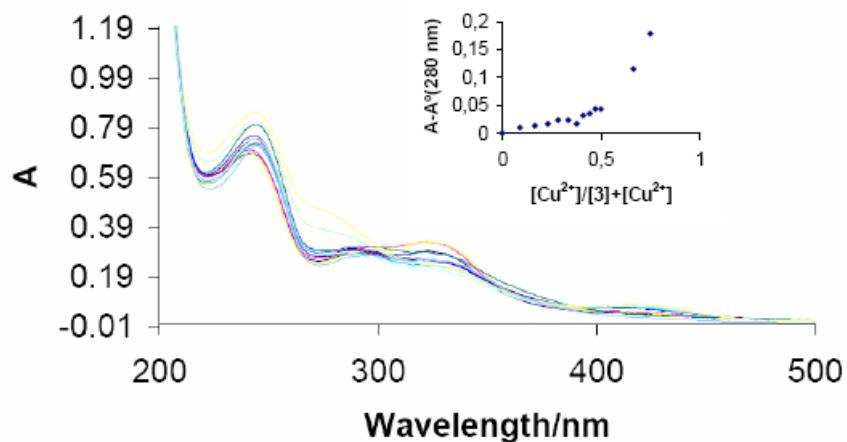
UV/Vis spectrum titration with Cu²⁺ and Co²⁺

Figure S. UV-vis spectral changes of ligand **3** (3.33×10^{-5} M in methanol) in response to addition of Cu²⁺ ion at 298 K. The concentrations of Cu²⁺ for curves were: 0, 0.33, 0.66, 1.00, 1.33, 1.66, 2.00, 2.33, 2.66, 3.00, 3.33, 6.66, and 9.99×10^{-5} M in methanol

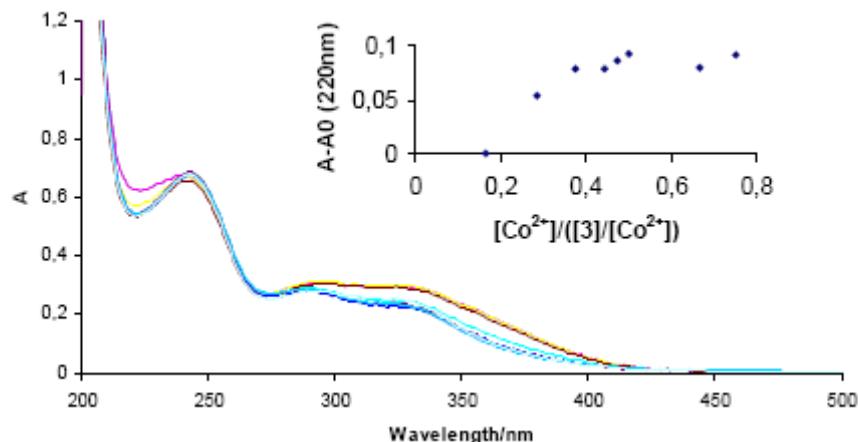


Figure S. UV-vis spectral changes of ligand **3** (3.33×10^{-5} M in methanol) in response to addition of Co²⁺ ion at 298 K. The concentrations of Co²⁺ for curves were: 0.0, 0.66, 1.33, 2.00, 2.66, 3.00, 3.33, 6.66, and 9.99×10^{-5} M in methanol