The tandem radical route to indole alkaloids: an unusual rearrangement reaction

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Abstract
Cyclisation of the precursor 6 under standard radical conditions yields the tetracyclic structure 7 which represents the core of a number of indole alkaloids along with the novel tetracycle 13.

Keywords: Indoles, radical cyclisation, rearrangement, alkaloids

Introduction

The Strychnos and Aspidosperma alkaloids are structurally complex molecules that possess a range of biological activity and present a challenge to synthetic organic chemists.1 As part of our ongoing interest in these classes of alkaloids,2 we are particularly concerned with a synthesis of the core tetracyclic sub-structure 1, which forms the basis of these molecules as exemplified by strychnine 2 and aspidospermidine 3 (Figure 1).

Figure 1

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In our previous studies, we had shown that tandem radical cyclisation of the precursor 4 gave the tetracyclic structure 5 as a mixture of 4 diastereoisomers (Scheme 1).

Scheme 1

Whilst this route provided the core tetracyclic structure 5, it possesses limited functionality with which to undertake a synthesis of either the Strychnos or Aspidosperma alkaloids. We therefore embarked on a synthesis of a cyclisation precursor which, after cyclisation, would enable us to access suitable intermediates for the synthesis of these two classes of alkaloids. The proposed cyclisation precursor 6 is shown below. It possesses a terminal C/C triple bond, which on cyclisation under our standard conditions should provide an exocyclic double bond on the tetracyclic product 7 (Scheme 2). This would allow further elaboration of such molecules towards natural products such as strychnine and aspidospermidine.

Scheme 2

Results and Discussion

The required cyclisation precursor was synthesized following our previous approach from the sulfonamide 8 which utilized the work of Bowman and Fukuyama for the synthesis of the secondary amine. Reaction of 8 with 4-pentynol under Mitsunobu conditions gave the tertiary sulfonamide 9 in 96% yield (Scheme 3). Deprotection of the tertiary sulfonamide 9 using potassium carbonate and thiophenol gave the secondary amine 10 in 75% yield. Reaction of this with the acid chloride 11 following our previous work on oxindoles, gave the cyclisation precursor 6 in 86% yield as a 1:1:3 mixture of amide rotamers (Scheme 3).
Scheme 3. Reagents and Conditions: (i) PPh₃, DEAD, THF, 0 °C, 96%; (ii) K₂CO₃, MeCN, PhSH, 75%; (iii) iPr₂NEt, CH₂Cl₂, 0 °C, 86%.

Following the successful cyclisation of the N-pentenyl precursor 4 in our previous example,³ we decided to use the same conditions for cyclisation of the N-pentynyl precursor 6. Treatment of a solution of the tertiary amide 6 in t-butyl-m-xylene under reflux with syringe pump addition of a solution of tri-n-butyltin hydride and AIBN in t-butyl-m-xylene resulted in the formation of three isolable products. High-resolution mass spectrometry confirmed the same molecular formula for all three compounds, C₂₄H₂₃N₃O. The least polar compound was identified as the reduced product 12 owing to the presence of 9 aromatic protons, the terminal acetylene proton and an equal mixture of amide rotamers in the ¹H NMR spectrum. It was isolated in 45% yield. From the ¹³C NMR spectrum, the second product isolated in 16% yield clearly possessed an exocyclic double bond (resonance at δ 119.4) along with the distinctive spirocyclic center (δ 53.8), indicating that the precursor 6 had undergone the desired translocation-cyclisation-cyclisation sequence onto the exocyclic triple bond. This along with the other observable data such as two distinct AB quartets in the ¹H NMR clearly indicated that we had indeed synthesized the tetracyclic structure 7. The relative stereochemistry of this diastereoisomer cannot easily be determined owing to the three contiguous quaternary centers but it appears to be one single diastereoisomer.
The nature of the third isolable product presented a challenge, as it clearly possessed three AB quartets from the $^1$H NMR spectrum. The exocyclic double bond was also present indicating that the final cyclisation step had taken place. From the $^{13}$C NMR spectrum there were 5 CH$_2$ signals excluding the exocyclic methylene group. Eight quaternary carbons were also observed in the $^{13}$C NMR spectrum with no evidence for the expected N-methyl group in either the $^1$H or $^{13}$C NMR spectrum. On the basis of the evidence, we have assigned the tetracyclic structure 13 to this compound, which was isolated in 15% yield. The formation of this unexpected product is believed to have occurred by an additional 1,5-hydrogen atom abstraction by the vinylic radical 17 from the neighbouring N-methyl group of the indole (Scheme 5). The resultant primary radical 18 then undergoes a neophyl-type rearrangement by addition to the benzene ring to give cyclohexadienyl radical 19. Ring-opening of 19 to regenerate the aromatic ring gives an aminyl radical which is reduced by tri-$n$-butyltin hydride to give the final compound 13.

Scheme 5
This unexpected product 13 possesses a novel ring structure which has arisen by a formal ring expansion of an indole to a tetrahydroisoquinoline. The formation of this product is surprising. 1,5-Hydrogen atom abstraction by a vinylic radical is well preceded as is the neophyl rearrangement involving an aromatic ring with concomitant ring expansion. However, the sequential occurrence of these two reactions as part of a tandem process is unprecedented.

In summary, cyclisation of the N-pentynyl precursor 6 results in the formation of the expected cyclised product 7, carrying suitable functionality for conversion into natural indole alkaloids. In addition, the novel tetracycle 13 has also been isolated. We suggest that this arises through a combination of hydrogen atom abstraction and radical rearrangement.

**Experimental Section**

**General Procedures.** $^1$H NMR and $^{13}$C NMR were recorded on a Bruker AM300 spectrometer operating at 300 MHz for proton and 75 MHz for carbon. Infrared spectra were recorded on a Perkin Elmer 1605 FT-IR spectrophotometer using NaCl plates. High resolution mass spectra were carried out at King’s College, London University or at Kingston University. Mass spectra carried out at King’s College using either a Kratos MS89MS with Kratos DS90 software or a Jeol AX505W with Jeol complement data system. Samples were ionised electronically (EI), with an accelerating voltage of $\approx$6kV or by low resolution fast atom bombardment (FAB) in a thioglycerol matrix. High resolution fast atom bombardment was carried out at the ULIRS mass spectrometry facility at the School of Pharmacy, University of London. Mass spectra carried out at Kingston University were recorded using a Micromass LCT time of flight mass spectrometer equipped with an ElectroSpray Ionisation (ESI) ion source.

**N-(2-Bromobenzyl)-2-nitrobenzenesulfonylamide (8).** A solution of 2-nitrobenzene sulfonyl chloride (25.00 g, 112.3 mmol) in dichloromethane (100 cm$^3$) was added dropwise to a stirred solution of 2-bromobenzylamine hydrochloride (25.00 g, 112.3 mmol) and triethylamine (50.1 cm$^3$, 36.40 g, 360.0 mmol) in dichloromethane (150 cm$^3$) at 0°C. The resulting solution was allowed to warm to room temperature and stirred for 16 hours, washed with water (3 x 200 cm$^3$), dried (MgSO$_4$), filtered and organic solvent removed under reduced pressure. The crude product was purified by recrystallisation (ethyl acetate/ hexane) to give N-(2-bromobenzyl)-2-nitrobenzenesulfonylamide (8) as a white crystalline solid (36.63 g, 89%); m.p. 101-103°C; $R_f$ (3:2, hexane: ethyl acetate) 0.57; (Found MH$^+$, 370.9694/ 372.9696, C$_{13}$H$_{12}$BrN$_2$O$_4$S requires 370.9701/ 372.9681); $\nu_{max}$/cm$^{-1}$ 3347.8 (NH), 1531.1 (Ar-NO$_2$), 1162.6 (NSO$_2$Ar); $\delta$H(300 MHz; CDCl$_3$) 4.34 (2H, d, $J$ 6.3, CH$_2$), 6.04 (1H, t, $J$ 6.3, NH), 7.00 (1H, td, $J$ 7.8 & 1.6, C-6H), 7.13 (1H, td, $J$ 7.8 & 1.6, C-5H), 7.29 (1H, dd, $J$ 7.8 & 1.6, C-6H), 7.31 (1H, dd, $J$ 7.8 & 1.6, C-3H), 7.51 (1H, td, $J$ 7.6 & 1.6, C-5'H), 7.58 (1H, td, $J$ 7.6 & 1.6, C-4'H), 7.34 (1H, dd, $J$ 7.6 & 1.6, C-6'H), 7.83 (1H, dd, $J$ 7.6 & 1.6, C-3'H); $\delta$C(75 MHz; CDCl$_3$) 48.20 (CH$_2$), 123.60 (C-2),
125.40 (Ar-C-H), 127.63 (Ar-C-H), 129.80 (Ar-C-H), 130.72 (Ar-C-H), 130.92 (Ar-C-H), 132.80 (Ar-C-H), 132.91 (Ar-C-H), 133.61 (Ar-C-H), 133.86 (quaternary C), 134.93 (quaternary C), 147.48 (C-2').

**N-(2-Bromobenzyl)-2-nitro-N-pent-4-ynylbenzenesulfonamide (9).** Diethyl azodicarboxylate (11.20 cm$^3$, 12.38 g, 71.1 mmol) was added dropwise to a solution of N-(2-bromobenzyl)-2-nitrobenzenesulfonamide (8) (20.30 g, 54.7 mmol), 4-pentyn-1-ol (6.1 cm$^3$, 5.52 g, 65.6 mmol) and triphenylphosphine (18.64 g, 71.1 mmol) in THF (150 cm$^3$) at 0 °C. The reaction was allowed to warm to room temperature and stirred overnight. Organic solvent was removed under reduced pressure and the residue taken up in diethyl ether (70 cm$^3$). The volume of the solution was reduced and the solution filtered to remove triphenylphosphine oxide. The filtrate was reduced in volume and the yellow oil purified by flash column chromatography (3: 2, hexane: ethyl acetate) to give the title compound (9) as a light yellow oil (22.96 g, 96%); R$_f$ (3: 2, hexane: ethyl acetate) 0.55; (Found MH$^+$, 437.0156/ 439.0169, C$_{18}$H$_{18}$BrN$_2$O$_4$S requires 437.0171/ 439.0150); $\nu_{\text{max}}$/cm$^{-1}$ 3300.4 (CCH), 3061.0, 2942.5 (Ar-H), 1544.7 (Ar-NO$_2$), 1371.4 (NSO$_2$Ar), 1163.8 (NSO$_2$Ar); $\delta$H(300 MHz; CDCl$_3$) 1.61 (2H, m, NCH$_2$C$_2$H$_2$), 1.87 (1H, t, J 2.6, CCH), 2.04 (2H, td, J 7.0 & 2.6, CH$_2$C$_2$H$_2$CCH), 3.39 (2H, t, J 7.6, NC$_2$H$_2$CH$_2$), 4.66 (2H, s, NC$_2$H$_2$Ar), 7.09 (1H, td, J 7.6 & 1.6, C-4H), 7.23 (1H, td, J 7.6 & 1.6, C-5H), 7.42 (1H, dd, J 7.6 & 1.6, C-6H), 7.47 (1H, dd, J 7.6 & 1.6, C-3H), 7.60-7.70 (3H, m, Ar-H); $\delta$C(75 MHz; CDCl$_3$) 15.73 (C$_2$H$_2$), 26.78 (C$_2$H$_2$), 47.28 (NCH$_2$CH$_2$), 51.48 (NCH$_2$Ar), 69.44 (CCH), 82.69 (CCH), 123.32 (C-2'), 124.33 (Ar-C-H), 127.91 (Ar-C-H), 129.47 (Ar-C-H), 129.80 (Ar-C-H), 130.86 (Ar-C-H), 132.04 (Ar-C-H), 132.96 (Ar-C-H), 133.09 (Ar-C-H), 135.02 (C-1'), 147.88 (C-2').

**N-(2-Bromobenzyl)pent-4-ynylamine (10).** Thiophenol (5.9 cm$^3$, 6.36 g, 57.8 mmol) was added to a stirred solution of sulfonamide (9) (22.95 g, 52.5 mmol) and potassium carbonate (23.58 g, 170.6 mmol) in acetonitrile (150 cm$^3$) at room temperature. The resulting solution was left to stir overnight at room temperature. Organic solvent was removed under reduced pressure and the residue was taken up in diethyl ether (200 cm$^3$). Hydrochloric acid (2.0 M solution) was added until the solution remained acidic and the resulting solution stirred for ten minutes. The organic layer was separated and washed with dilute hydrochloric acid (100 cm$^3$ of a 1.0 M solution). The combined aqueous phases were washed with diethyl ether (200 cm$^3$) and the resulting solution stirred for ten minutes. The organic layer was separated and washed with dilute hydrochloric acid (100 cm$^3$ of a 1.0 M solution). The combined aqueous phases were washed with diethyl ether (200 cm$^3$) and made basic by the addition of solid potassium carbonate. The aqueous solution was extracted with diethyl ether (4 x 150 cm$^3$) and the combined extracts washed with brine (200 cm$^3$), dried (MgSO$_4$), filtered and evaporated to dryness under reduced pressure to give the title compound (10) as a pale yellow oil (9.94 g, 75%); R$_f$ (3: 2, hexane: ethyl acetate) 0.21; (Found MH$^+$, 252.0377/ 254.0381; C$_{12}$H$_{15}$BrN requires 252.0388/ 254.0367); $\nu_{\text{max}}$/cm$^{-1}$ 3422.5 (NH), 3301.6 (CCH); $\delta$H(300 MHz; CDCl$_3$) 1.39 (1H, br s, NH), 1.64 (2H, quin, J 7.0, NHCH$_2$CH$_2$), 1.86 (1H, t, J 2.6, CCH), 2.19 (2H, td, J 7.0 & 2.6, CH$_2$CCH), 2.64 (2H, t, J 7.0, NCH$_2$CH$_2$), 3.76 (2H, s, NCH$_2$Ar), 7.01 (1H, td, J 7.5 & 1.0, C-4H), 7.17 (1H, td, J 7.5 & 1.0, C-5H), 7.29 (1H, dd, J 7.5 & 1.0, C-6H), 7.43 (1H, dd, J 7.5 & 1.0, C-3H); $\delta$C(75 MHz; CDCl$_3$) 16.61 (CH$_2$), 28.75 (CH$_2$), 47.28 (NCH$_2$CH$_2$), 51.48 (NCH$_2$Ar), 69.44 (CCH), 82.69 (CCH), 123.32 (C-2'), 124.33 (Ar-C-H), 127.91 (Ar-C-H), 129.47 (Ar-C-H), 129.80 (Ar-C-H), 130.86 (Ar-C-H), 132.04 (Ar-C-H), 132.96 (Ar-C-H), 133.09 (Ar-C-H), 135.02 (C-1'), 147.88 (C-2').
47.92 (NCH₂CH₂), 53.65 (NCH₂Ar), 68.61 (CCH), 84.08 (CCH), 123.94 (C-2), 127.39 (C-5), 128.52 (C-4), 130.19 (C-6), 132.75 (C-3), 139.34 (C-1).

N-(2-Bromobenzyl)-2-(2-cyano-1-methyl-1H-indol-3-yl)-N-pent-4-ynylacetamide (6). A solution of acid chloride (11) (3.25 g, 14.0 mmol) in dichloromethane (50 cm³) was added dropwise via cannula to a stirred solution of (2-bromobenzyl)-pent-4-ynylamine (10) (4.58 g, 18.15 mmol) and N, N-diisopropylethylamine (8.5 cm³, 6.31 g, 48.9 mmol) in dichloromethane (100 cm³) at 0 °C under nitrogen. The resulting mixture was left to stir at 0 °C for one hour, allowed to warm to room temperature and then left to stir overnight. Organic solvent was removed under reduced pressure and the residue taken up in dichloromethane (200 cm³), washed with HCl (200 cm³ of a 0.5 M solution), saturated sodium hydrogen carbonate (200 cm³) and brine (200 cm³). The organic phase was dried (MgSO₄), filtered and solvent removed under reduced pressure to give a light green solid which was purified by flash column chromatography (2: 1, hexane: ethyl acetate) to give the tertiary amide (6) as a light yellow solid and as a 1: 1.3 ratio of amide rotamers (5.40 g, 86%); m.p. 119-120 °C; Rf (2: 1, hexane: ethyl acetate) 0.83; (Found M⁺, 447.0956/ 449.0916, C₂₄H₂₂BrN₃O requires 447.0947/ 449.0927); ν max /cm⁻¹ 3303.1 (CCH), 3012.8 (Ar-H), 2217.6 (CN), 1653.9 (CONR₂); δ H(300 MHz; CDCl₃) 1.96 (2H, quin, J 6.8, C-2''H₂ (major + minor)), 2.08 (0.4H, t, J 2.4, C-5''H (minor)), 2.17 (0.6H, t, J 2.4, C-5''H (major)), 2.34 (1.2H, td, J 6.8 & 2.4, C-3''H₂ (major)), 2.40 (0.8H, td, J 6.8 & 2.4, C-3''H₂ (minor)), 3.64 (2H, m, C-1''H₂ (major + minor)), 3.83 (1.8H, s, NCH₃ (major + minor)), 4.02 (1.2H, s, C₂H₂C (major + minor)), 4.25 (0.8H, s, CH₂CO (minor)), 4.82 (1.2H, s, NCH₂Ar (major)), 4.88 (0.8H, s, NCH₂Ar (minor)), 6.92 (0.4H, m, Ar-H (minor)), 7.04-7.42 (5.6H, m, Ar-H (major + minor)), 7.49 (0.4H, d, J 8.0, Ar-H (minor)), 7.70 (0.6H, m, Ar-H (major)), 7.76 (0.6H, d, J 8.0, Ar-H (major)), 7.89 (0.4H, d, J 8.0, Ar-H (minor)); δ C(75 MHz; CDCl₃) 15.86 (CH₂), 16.17 (CH₂), 26.33 (CH₂), 27.36 (CH₂), 30.76 (CH₂), 31.09 (CH₂), 31.41 (NCH₃), 31.62 (NCH₃), 46.50 (CH₂), 46.78 (CH₂), 48.65 (CH₂), 52.57 (CH₂), 69.12 (C-4''), 70.00 (C-4''), 82.72 (C-5''), 83.40 (C-5''), 109.82 (C-3), 110.06 (C-7), 110.14 (C-7), 112.93 (CN), 113.32 (CN), 120.09 (quaternary C), 120.69 (quaternary C), 121.08 (Ar-C-H), 121.18 (Ar-C-H), 121.43 (Ar-C-H), 122.60 (quaternary C), 123.50 (quaternary C), 126.07 + 126.22 + 126.77 + 127.66 + 127.33 + 128.85 + 128.99 + 129.21 + 132.75 + 133.08 (Ar-C-H), 135.33 (quaternary C), 136.25 (quaternary C), 137.96 (quaternary C), 138.14 (quaternary C), 169.43 (CO), 169.84 (CO); m/z 449.09 (18.8%, M⁺ 81Br), 447.10 (20.6, M⁺ 79Br), 368.17 (62.9), 183.09 (59.2), 170.99 (91.8), 170.10 (66.6), 169.09 (100.0), 168.08 (43.3), 91.06 (49.9), 90.05 (56.0).

Radical cyclisation of N-(2-bromobenzyl)-2-(2-cyano-1-methyl-1H-indol-3-yl)-N-pent-4-ynylacetamide (6). A solution of tri-n-butyltin hydride (0.75 cm³, 0.81 g, 2.78 mmol) and azobis-isobutyrionitrile (AIBN) (0.048 g, 2.78 mmol) in t-butyl-m-xylene (30 cm³) were added dropwise via syringe pump (6.37 cm³/hour) to a solution of the amide (6) (1.039 g, 2.32 mmol) in t-butyl-m-xylene (90 cm³) at reflux. Once addition was complete, heating was continued for 2 hours. The reaction was allowed to cool and organic solvent removed under reduced pressure and the residue taken up in acetonitrile (50 cm³) and extracted with hexane (4 x 50 cm³). The
acetonitrile was removed under reduced pressure and the residue purified by multiple flash column chromatography (3: 1, hexane: ethyl acetate) to give a mixture of three products:

**N-benzyl-2-(2-cyano-1-methyl-1H-indol-3-yl)-N-(pent-4-ynyl)acetamide (12).** White solid as a 1: 1 mix of amide rotamers (0.38 g, 45%); m.p. 82-84 °C; (Found M^+, 369.18510, C_{24}H_{23}N_{3}O requires 369.18411); ν_max /cm^{-1} 3296.7 (C=O), 3059.2, 2938.7 (Ar-H), 2217.5 (CN), 1648.6 (CONR_2); δ_H(300 MHz; CDCl_3) 1.81 (2H, quin, J 7.0, C-2''H_2), 2.05 (0.5H, t, J 2.5, CCH), 2.19 (1H, td, J 7.0 & 2.5, C-3''H_2), 2.26 (1H, td, J 7.0 & 2.5, C-3''H_2), 3.51 (2H, m, C-1''H_2), 3.72 (1.5H, s, NCH_3), 3.85 (1.5H, s, NCH_3), 3.96 (1H, s, CH_2CO), 4.09 (1H, s, CH_2CO), 4.65 (1H, s, CONCH_2Ar), 4.72 (1H, s, CONCH_2Ar), 7.09-7.44 (8H, m, Ar-H), 7.69 (0.5H, d, J 8.1, Ar-H), 7.77 (0.5H, d, J 8.1, Ar-H); δ_C(75 MHz; CDCl_3) 15.80 (CH_3), 16.14 (CH_3), 26.28 (CH_2), 27.15 (CH_2), 30.77 (CH_2), 31.27 (CH_2), 31.44 (NCH_3), 31.62 (NCH_3), 46.15 (CH_2), 46.37 (CH_2), 48.72 (CH_2), 51.91 (CH_2), 69.12 (CCH), 69.94 (CCH), 82.84 (CCH), 83.54 (CCH), 109.76 (C-3), 110.05 (C-7), 110.15 (C-7), 111.23 (CN), 113.38 (CN), 120.48 (quaternary C), 120.91 (quaternary C), 121.20 (Ar-C-H), 121.29 (Ar-C-H), 121.43 (Ar-C-H), 125.93 (Ar-C-H), 126.05 (Ar-C-H), 126.14 (Ar-C-H), 126.20 (Ar-C-H), 127.43 (Ar-C-H), 127.48 (Ar-C-H), 128.18 (Ar-C-H), 128.35 (quaternary C), 128.60 (Ar-C-H), 128.82 (Ar-C-H), 136.53 (quaternary C), 137.39 (quaternary C), 138.02 (quaternary C), 138.15 (quaternary C), 169.27 (CO), 169.71 (CO); m/z 369.19 (43.6%, M^+), 321.23 (11.4), 273.13 (77.2), 169.08 (90.9), 168.08 (56.6), 121.06 (79.9), 91.04 (81.4), 77.04 (37.6).

**1-Benzyl-8-methyl-9-methylene-2-oxo-2,3,8,8a,9,10,11,11a-octahydro-1H-pyrrolo[2,3-d]carbazole-8a-carbonitrile (7).** Colourless oil (0.14 g, 16%); (Found M^+, 369.18666, C_{24}H_{23}N_{3}O requires 369.18411); ν_max /cm^{-1} 2924.4, 2854.8 (Ar-H), 2218.2 (CN), 1694.1 (CONR_2); δ_C(75 MHz; CDCl_3) 21.94 (0.5H, t, J 5.3, NCH_3), 2.35 (2H, m, NCH_2CO), 2.84 & 2.98 (2H, ABq, J 17.5, CH_2CO), 3.37 (1H, dd, J 12.6 & 3.3, NCH_2), 3.52 (1H, d, J 12.6, NCH_2), 3.54 (1H, t, J 5.3, NCH), 4.04 (1H, d, J 14.9, CONCH_2Ar), 4.19 (1H, br s, NH), 4.96 (1H, d, J 14.9, CONCH_2Ar), 5.31 (1H, s, CCH_2), 5.62 (1H, s, CCH_2), 6.54 (1H, d, J 7.5, C-7H), 6.69 (1H, td, J 7.5 & 1.0, C-
5H), 7.01-7.07 (2H, m, Ar-H), 7.13-7.24 (5H, m, Ar-H); δC(75 MHz; CDCl3) 24.57 (CH2), 25.38 (CH2), 42.91 (CH2), 43.01 (quaternary C), 44.70 (CH2), 45.65 (quaternary C), 48.91 (CH2), 62.13 (NCH), 115.54 (C-7), 116.99 (CCH2), 119.20 (Ar-C-H), 120.12 (quaternary C), 125.53 (quaternary C), 126.58 (Ar-C-H), 127.75 (Ar-C-H), 128.35 (2 x Ar-C-H), 128.55 (Ar-C-H), 128.62 (2 x Ar-C-H), 135.67 (quaternary C), 140.27 (quaternary C), 140.87 (quaternary C), 172.60 (CO); m/z 369.18 (48.0%, M⁺), 303.15 (21.7), 266.99 (27.8), 245.11 (29.2), 169.08 (53.2), 91.06 (28.8).

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References and Footnotes

11. Intramolecular addition of an alkyl radical to an aromatic ring has been shown to occur, see