# The homologation of carbonyl compounds by single carbon insertion reactions

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# Dedicated to Professor Alexander T. Balaban on the occasion of his 75<sup>th</sup> birthday

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### **Abstract**

This short review describes the methods of single carbon insertion reactions into carbonyl compounds.

**Keywords:** Insertion, rearrangement, carbonyl compounds, migration

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## 1. Introduction

Carbon chain extension or ring expansion of carbonyl compounds by one-carbon unit is a frequently encountered synthetic objective. Carbon insertion reactions are the most

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straightforward and most commonly used strategy for this purpose. Classical insertion reactions utilizing diazo compounds are known for diazoalkanes as well as functionalized diazo compounds such as ethyl diazoacetate etc. (Scheme 1(i)). However, the use of diazo compounds is not usually possible on the large scale and many attempts have been made to find other insertion methods avoiding diazo compounds.

 $\beta$ -Oxido carbenoid chemistry was initiated by Yamamoto using dihalomethyllithiums<sup>2a,b</sup> (Scheme 1(ii)) and then expanded significantly by Satoh and Yamakawa who utilized 1-chloroalkyl aryl sulfoxides as one carbon homologating agents.<sup>3a-c</sup> Trost introduced sulfone-assisted insertions: published examples have all involved the insertion of either CHSPh or CHOMe groups, utilizing  $\alpha$ -thio and  $\alpha$ -alkoxy sulfones<sup>4a</sup> (Scheme 1(iii)).

$$\begin{bmatrix} \text{LiO} & \text{CHBr}_2 \\ \text{R} & \text{R}^1 \end{bmatrix} \xrightarrow{\text{CH}_2 \text{Br}_2} \xrightarrow{\text{base}} (\text{ii}) \qquad \begin{bmatrix} \text{R}^1 & \text{O} \\ \text{R} & \text{R}^2 \end{bmatrix} \xrightarrow{\text{R}^2} \begin{bmatrix} \text{R}^1 & \text{O} \\ \text{R} & \text{R}^2 \end{bmatrix} \xrightarrow{\text{R}^2} \begin{bmatrix} \text{R}^1 & \text{O} \\ \text{R} & \text{R}^2 \end{bmatrix} \xrightarrow{\text{R}^2} \begin{bmatrix} \text{R}^1 & \text{O} \\ \text{R} & \text{R}^2 \end{bmatrix} \xrightarrow{\text{R}^2} \begin{bmatrix} \text{R}^1 & \text{O} \\ \text{R} & \text{R}^2 \end{bmatrix} \xrightarrow{\text{R}^2} \begin{bmatrix} \text{R}^2 & \text{CH}_2 \text{SO}_2 \text{Ph} \\ \text{Base} \end{bmatrix} \xrightarrow{\text{R}^2} \begin{bmatrix} \text{R}^1 & \text{O} \\ \text{R} & \text{R}^2 \end{bmatrix}} \xrightarrow{\text{R}^2} \begin{bmatrix} \text{R}^1 & \text{O} \\ \text{R} & \text{R}^2 \end{bmatrix}} \xrightarrow{\text{R}^2} \begin{bmatrix} \text{R}^1 & \text{O} \\ \text{R} & \text{R}^2 \end{bmatrix}} \xrightarrow{\text{R}^2} \begin{bmatrix} \text{R}^1 & \text{O} \\ \text{R} & \text{R}^2 \end{bmatrix}} \xrightarrow{\text{R}^2} \begin{bmatrix} \text{R}^1 & \text{O} \\ \text{R} & \text{R}^2 \end{bmatrix}} \xrightarrow{\text{R}^2} \begin{bmatrix} \text{R}^1 & \text{O} \\ \text{R} & \text{R}^2 \end{bmatrix}} \xrightarrow{\text{R}^2} \begin{bmatrix} \text{R}^1 & \text{O} \\ \text{R} & \text{R}^2 \end{bmatrix}} \xrightarrow{\text{R}^2} \begin{bmatrix} \text{R}^1 & \text{O} \\ \text{R} & \text{R}^2 \end{bmatrix}} \xrightarrow{\text{R}^2} \begin{bmatrix} \text{R}^1 & \text{O} \\ \text{R} & \text{R}^2 \end{bmatrix}} \xrightarrow{\text{R}^2} \xrightarrow{\text{R}^2} \begin{bmatrix} \text{R}^1 & \text{O} \\ \text{R} & \text{R}^2 \end{bmatrix}} \xrightarrow{\text{R}^2} \xrightarrow{\text{R}^2} \begin{bmatrix} \text{R}^1 & \text{O} \\ \text{R} & \text{R}^2 \end{bmatrix}} \xrightarrow{\text{R}^2} \xrightarrow{\text{R}^2} \begin{bmatrix} \text{R}^1 & \text{O} \\ \text{R} & \text{R}^2 \end{bmatrix}} \xrightarrow{\text{R}^2} \xrightarrow{\text{R}^2} \begin{bmatrix} \text{R}^1 & \text{O} \\ \text{R} & \text{R}^2 \end{bmatrix}} \xrightarrow{\text{R}^2} \xrightarrow{\text{R}^2} \xrightarrow{\text{R}^2} \begin{bmatrix} \text{R}^1 & \text{O} \\ \text{R} & \text{R}^2 \end{bmatrix}} \xrightarrow{\text{R}^2} \xrightarrow{\text{R}^2} \xrightarrow{\text{R}^2} \begin{bmatrix} \text{R}^1 & \text{O} \\ \text{R} & \text{R}^2 \end{bmatrix}} \xrightarrow{\text{R}^2} \xrightarrow{\text{R}^2} \xrightarrow{\text{R}^2}} \xrightarrow{\text{R}^2} \xrightarrow{\text{R}^2} \xrightarrow{\text{R}^2} \xrightarrow{\text{R}^2}} \xrightarrow{\text{R}^2} \xrightarrow{\text{R}^2} \xrightarrow{\text{R}^2} \xrightarrow{\text{R}^2}} \xrightarrow{\text{R}^2} \xrightarrow{\text{R}^2} \xrightarrow{\text{R}^2} \xrightarrow{\text{R}^2} \xrightarrow{\text{R}^2}} \xrightarrow{\text{R}^2} \xrightarrow{\text{R}^2} \xrightarrow{\text{R}^2} \xrightarrow{\text{R}^2}} \xrightarrow{\text{R}^2} \xrightarrow{\text{R}^2}$$

**Scheme 1. (i)** insertion by diazo compounds. (ii) insertion via  $\beta$ -oxido carbenoid intermediates. (iii) insertion via  $\alpha$ -lithioalkyl sulfone intermediates.

# 2. Single carbon insertion via $\beta$ -oxido carbenoid intermediates

The Yamamoto method is presented in more detail in Scheme 2. This approach depends crucially on the generation in situ of dibromomethyllithium from methylene bromide and its addition to carbonyl compounds in the presence of lithium dialkylamide. Depending on the structure of the ketone the dibromomethyllithium carbonyl adduct may rapidly form the corresponding epoxide; to avoid this the reaction should be performed between -70 °C and -100°C. Although the dibromomethyllithium carbonyl adducts are extremely thermally unstable, several one-carbon homologated cyclic ketones were synthesized in high yields by this method (Scheme 2). <sup>2a,b</sup>

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O CH<sub>2</sub>Br<sub>2</sub> LiO CHBr<sub>2</sub> 
$$\xrightarrow{R^1}$$
 LiO CHBr<sub>2</sub>  $\xrightarrow{n\text{-BuLi}}$  LiO CHBrLi  $\xrightarrow{R^1}$  LiO CHBrLi  $\xrightarrow{R^1}$  Cyclic Ketones:

R=R<sup>1</sup>= -(CH<sub>2</sub>)<sub>4</sub>- (92%) R=R<sup>1</sup>= -(CH<sub>2</sub>)<sub>12</sub>- (96%)

R=R<sup>1</sup>= -(CH<sub>2</sub>)<sub>5</sub>- (70%) R=R<sup>1</sup>= -(CH<sub>2</sub>)<sub>13</sub>- (89%)

R=R<sup>1</sup>= -(CH<sub>2</sub>)<sub>6</sub>- (80%) R=R<sup>1</sup>= -(CH<sub>2</sub>)<sub>14</sub>- (79%)

R=R<sup>1</sup>= -(CH<sub>2</sub>)<sub>7</sub>- (87%)

### Scheme 2

The modification by Satoh and Yamakawa<sup>3a-c</sup> of this method for the one-carbon homologation of carbonyl compounds is based on the rearrangement of  $\beta$ -oxido carbenoids **4** generated via ligand exchange of the sulfinyl group of  $\alpha$ -chloro  $\beta$ -hydroxy sulfoxide **3** with t-BuLi to give homologated ketones **5** (Scheme 3): Table 1 shows examples where selective migration occurs to form a single regioisomer. Most insertions into cyclic ketones and aromatic or aliphatic aldehydes are regioselective with formation of single regioisomers; however, insertions into aryl-alkyl ketones lead to both aryl- and alkyl-migration.

### Scheme 3

**Table 1.** Examples of carbon insertion via  $\beta$ -oxido carbenoid intermediates

Reagent	Carbonyl	Number of	Average	Examples of	Examples of	Ref.
Reagent	substrate	examples	yield, %	intermediates 2	products 5	ICI.
CIDCUSOAr	Cyclic ketones	9	68	Me(CH <sub>2</sub> ) <sub>9</sub> —SOAr	O (CH <sub>2</sub> ) <sub>9</sub> Me	3b,c
CIRCHSOAr R = Alk	Aliphatic and mixed ketones	2	58	CI Me OH SOAr	Me Me O	3b
ClCH <sub>2</sub> SOAr	Aromatic, aldehydes, cyclic and mixed ketones	7	52	HO SOAr CI MeO	COMe	3b

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Recently, Satoh and Miyashita reported one-carbon homologations of cyclic ketones by treatment of adducts **2** derived from the 1-chloroalkyl sulfoxides **1** with *t*-BuMgCl or LDA followed by reaction with *i*-PrMgCl. Then the enolate intermediates **7** generated in situ from the rearrangement of **6** were trapped with an electrophile to give the  $\alpha$ , $\alpha$ -disubstituted homologated ketones **8** (Scheme 4) (Table 2). <sup>3a</sup>

### Scheme 4

**Table 2.** Examples of carbon insertion via  $\beta$ -oxido carbenoid intermediates

Cyclic ketone	Electrophile	Number of examples	Average yield, %	Examples of intermediates 2	Examples of products 8	Ref
	EtCHO	1	38	HO SOTol Me Cl	O Me Et OH	3a
0	CICOOEt PhCHO PhCOCI	R = Me 3	58	HO SOTOI CI Me	O COPh Me	
	CD₃OD ClCOOEt EtCHO	$R = Et \text{ or}$ $(CH_2)_4 Ph$ 6	49	HO SOTol Cl Et	O D Et	3a
0	CICOOEt EtCHO PhCOCI	3	59	HO SOTol CI Me	COPh Me	3a

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# 3. Single carbon insertion via α-lithioalkyl sulfone intermediates

Trost introduced a useful insertion methodology by reaction of  $\alpha$ -lithioalkyl sulfones 10 with ketones 9 to form intermediates 11 followed by aluminium-based Lewis acid-induced ring expansion to  $\alpha$ -phenylthio and  $\alpha$ -methoxy ketones 12 as shown in Scheme 5 and Table 3 (entries 1 and 2). Published examples of this approach have all concerned cyclic or bicyclic ketones and  $\alpha$ -thio or  $\alpha$ -alkoxy sulfones.

#### Scheme 5

**Table 3.** Examples carbon insertion via  $\alpha$ -lithioalkyl sulfone intermediates

	Reagent 10	Ketone	Number of examples	Average yield, %	Examples of products 12	Ref
1	PhSCH <sub>2</sub> SO <sub>2</sub> Ph	Cyclic and bicyclic 4-, 5- and 6-ring	6	73	O SPh	4a
2	MeOCH <sub>2</sub> SO <sub>2</sub> Ph		5	68	OMe O	4a
3	MeOCH <sub>2</sub> SO <sub>2</sub> Ph	Aryl-alkyl ketones and dialkyl ketones	3	74	MeO Me	4b,c
4	MeOCH <sub>2</sub> SO <sub>2</sub> Ph	Cyclic	4	61	OMe	4b,c

Later, Trost's sulfone homologation procedure for the transformation of ketones 9 into their higher homologues 12 was extended by Taylor to aryl-alkyl ketones, dialkyl ketones and novel cycloalkanones utilizing ZrCl<sub>4</sub> promoted conditions in the rearrangement step (Scheme 5 and Table 3; entries 3 and 4).

 $\alpha$ -Lithioalkyl aryl sulfoxides and selenoxides are effective reagents for the ring expansion of a variety of cyclobutanones to cyclopentanones. Intermediates 15 produced from 14 and cyclobutanones 13 undergo rapid ring expansion upon treatment with potassium hydride to give

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**16** in good yields (Scheme 6 and Table 4). Using **14** the carbon atom inserted into the cyclobutanone can be unsubstituted, monosubstituted, or disubstituted. <sup>4d</sup>

### Scheme 6

**Table 4.** Ring expansion of cyclobutanones to cyclopentanones via  $\alpha$ -lithioalkyl aryl sulfoxide and selenoxide intermediates

Cyclobutanone 13	Reagent 14	Number of examples <b>16</b>	Average yield, %	Ref
(i) and (ii)	2-ClPhS(O)R <sup>2</sup>	16	57	4d
(i) and (ii)	$ArS(O)R^2$ Ar = Ph, 2-ClPh	5	57	4d
(i), (iii) and (iv)	PhSe(O)R <sup>2</sup>	8	43	4d

# 4. General overview of benzotriazole-mediated single carbon insertion

Benzotriazole derivatives 17 may also be used in one-carbon homologation as shown in Scheme  $7.^{5a-d}$  Carbonyl compounds that can be utilized as starting materials include aliphatic, aromatic aldehydes and many types of ketones (Table 5). Monosubstituted benzotriazolylmethanes 17, which can be successfully inserted include 1-(arylmethyl)-, 1-(heteroarylmethyl)-, 1-(alkoxymethyl)-, and 1-[(phenylthio)methyl]-benzotriazoles, which allow the preparation of wide variety of  $\alpha$ -functionalized ketones. The generality of our methodology is also exemplified by successful insertions of disubstituted methylene groups into carbonyl compounds when disubstituted reagents of type Bt-CHXY are utilized.

Insertion into ketones can be performed by use of our procedure in a simple one-step operation. The lithium alcoholate **18** generated from **17** and a ketone undergoes a subsequent rearrangement catalyzed by zinc bromide to give the homologated ketones **19**. An alternative two-step procedure includes the formation of benzotriazolyl alcohols **20**, which undergo rearrangement *via* their lithium alcoholates **18** to give **19** (Scheme 7).

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## Scheme 7

**Table 5.** Examples of carbonyl compounds utilized in Bt-mediated insertion

Carbonyl substrate					
	K	Cetones		Aldehydes	Ref
Cyclic	Di-alkyl	Di-aryl	Alkyl-Aryl	Aluchyucs	
	<i>t</i> -BuCOMe	Z = 0, S	COMe Z Z = O, S	Ph(CH <sub>2</sub> ) <sub>2</sub> CHO	5a-d
0	Bn <sub>2</sub> CO	Z = O, S	Z COMe $Z = 0$ , S	t-BuCHO	5a-d
O Me	i-PrCOMe	S—COPh	S COMe	4-ClC <sub>6</sub> H <sub>4</sub> CHO	5a-d
0		4-PyCOPh	PhCOMe	<i>p</i> -TolCHO	5a-c
0			COMe	PhCHO	5b-d

# 5. Bt-mediated insertion with C-linked substituents on the carbon atom $\boldsymbol{\alpha}$ to Bt

A large variety of C-linked substituents can be carried by the carbon atom that is inserted next to carbonyl group. As shown in Scheme 7 and Table 6 these include vinyl groups, aryl groups and heteroaryl groups.

Insertion of aryl-linked<sup>6a,b</sup> and vinyl-linked<sup>6c</sup> carbons into carbonyl compounds has previously been achieved by direct insertion of the corresponding diazo compounds, but, this

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procedure is limited by low regioselectivity and handling difficulties. An advantage of the benzotriazole-mediated methodology is the possibility of introducing a heteroaryl group attached to the inserted carbon; the 1-methyl-1*H*-indol-3-yl and 5-methylthiophene-2-yl groups, for example, were introduced in this way. <sup>5a-c</sup>

17 X	C-linked group inserted	Carbonyl substrate	Number of examples	Average Yield, %	Examples of products 19	Re f.
Vinyl	CH-vinyl		1	60	Ph	5a
Ar	CH-Ar	Aromatic and aliphatic aldehydes; aliphatic, cyclic and mixed ketones	11	57	Me <sub>2</sub> N t-Bu	5a, c
Het	CH-Het	Aromatic and aliphatic aldehydes; aliphatic, cyclic and mixed ketones	10	76	O N Me	5a c

# 6. Bt-mediated insertion with Het-linked substituents on the carbon atom $\alpha$ to Bt

Similarly, a whole variety of heteroatom-linked substituents can be carried by the inserted carbon atom. As shown in Scheme 7 and Table 7 this can include various O-linked, S-linked and N-linked groups.

1-Methoxymethyl-1*H*-benzotriazole, 1-phenoxymethyl-1*H*-benzotriazole, 1-methylsulfanylmethyl-1*H*-benzotriazole, 1-phenylsulfanylmethyl-1*H*-benzotriazole and 9-benzotriazol-1-ylmethyl-9*H*-carbazole (for 17 see Scheme 7 and Table 7) are readily available and versatile reagents. They enable the transformation of a wide range of aldehydes and ketones into corresponding functionalized one-carbon homologues 19.  $^{5a,b}$  Published examples of the insertions of  $\alpha$ -methoxymethylene and phenylthiomethylene groups to form  $\alpha$ -methoxy and  $\alpha$ -phenylthio alkyl ketones by the Trost  $^{4a}$  method all involve cyclic ketones.

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17 X	Het-linked group inserted	Carbonyl substrate	Number of examples	Average Yield, %	Examples of products <b>19</b>	Ref.
OMe	СНОМе	Aromatic and aliphatic aldehydes; aromatic, cyclic and mixed ketones	7	58	PhCH(OMe)CO Me	5a,b
OPh	CHOPh	Aromatic and aliphatic ketones	2	50	PhCOCH(OPh)Ph	5a,b
SMe	CHSMe	Aromatic aldehyde and mixed ketone	2	67	PhCOCH <sub>2</sub> SMe	5a,b
SPh	CHSPh	Aromatic and aliphatic aldehydes; aromatic, and mixed ketones	5	73	PhCH(SPh)COPh	5b
NRR'	CHNRR'	Aromatic aldehydes	3	76	PhCH(Cb)COMe	5b

**Table 7.** Het-linked substituents on the carbon atom  $\alpha$  to Bt

# 7. Insertion with C- and Het-linked substituents on the carbon α to Bt

and mixed ketone

It is possible to insert a carbon atom carrying two substituents with a reagent of type **21** (Scheme 8). In these examples one substituent is an alkyl or aryl group and the other is an *O*-linked, *S*-linked or *N*-linked group. Examples of such insertion reactions to aromatic aldehydes and cyclic ketones to give higher homologues **22** are shown in Scheme 8 and Table 8.

R<sup>2</sup>XCHBt i) n-BuLi 
$$R^1$$
  $R^2$   $R^3$   $R^4$   $R^2$   $R^4$   $R^2$   $R^4$   $R^2$   $R^4$   $R^2$   $R^4$   $R^2$   $R^4$   $R^2$ 

### Scheme 8

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21 X	Simultaneous insertion of two group	Carbonyl substrate	Number of examples	Average Yield, %	Examples of products 22	Ref
OPh	Alkyl-C-OPh	<i>p</i> -TolCHO	1	81	<i>p</i> -Tol OPh Me	5b
OEt	Ar-C-OEt	Aromatic aldehyde and cyclic ketone	2	71	CI—OCI	5a,b
SPh	$R^2$ -C-SPh $R^2$ = Alk, Ar	Aromatic aldehyde and cyclic ketone	2	70	PhCH(SPh)COPh	5b
NRR'	Alk-C-NRR'	PhCHO	1	56	PhCOCH(Cb)Me	5b

**Table 8.** C- and Het-linked substituents on the carbon atom  $\alpha$  to Bt

# 8. Examples of bis-Bt insertion

One example missing from the above is the insertion of carbon carrying a single alkyl group. This is not easy using an alkylbenzotriazole because the acidity of the  $\alpha$ -hydrogen is low and overcome yields poor. However, this limitation has been by (alkylidene)bisbenzotriazoles 23 (Scheme 9, Table 9).5d Insertion here gives the expected intermediate 24 containing two benzotriazole groups one of which is eliminated during the rearrangement to give ketone 25 and the other one can easily be eliminated by treatment with Zn metal to give functionalized ketone 26. <sup>7a</sup> Intermediates 25 with an α-benzotriazolyl group are of significant synthetic utility for transformations to diketones, 7b,c or olefins, 7d-g for directed regioselective α-alkylation. <sup>7h</sup> and for heterocyclic ring synthesis. <sup>7i</sup>

Bt 1) *n*-BuLi HO 
$$R^{2}$$
  $ZnBr_{2}$  Bt  $R^{3}$   $ZnBr_{2}$  Bt  $R^{2}$   $R^{3}$   $R^{3}$   $R^{2}$   $R^{3}$   $R^{3}$   $R^{2}$   $R^{3}$   $R^{3}$   $R^{2}$   $R^{3}$   $R^{3}$   $R^{3}$   $R^{2}$   $R^{3}$   $R^{3}$ 

## Scheme 9

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**Table 9.** *Bis*-Bt examples

R <sup>3</sup>	Bt-CX is inserted	Carbonyl substrate	Number of examples	Average Yield, %	Examples of intermediate s	Examples of products 25	Ref
Н	Bt-CH	Aromatic, aliphatic, mixed and cyclic ketones	5	60	OH Bt	OBt	5d
Me	Bt-CMe	Aromatic, aliphatic, mixed and cyclic ketones	4	38	HO Me Ph Bt	Ph O Me Bt Me	5d

# 9. Intramolecular insertion

Intramolecular reaction of the carbonyl group of adducts **29** or **34** with a benzotriazole-activated nucleophilic α-carbon to give intermediates **30** or **35** is also possible (Schemes 10 and 11). Intermediates **30** and **35** can be isolated as corresponding alcohols or directly treated with Lewis acid to give inserted ketones **31** and **36** respectively. This method is valuable for the synthesis of 3-alkyl-3-aryl-2,3-dihydrobenzofuran-2-ones, which are important intermediates for the synthesis of the anti-cancer compound, diazonamide A, <sup>8b,c</sup> analgesics and antidepressants. <sup>8d</sup>

### Scheme 10

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### Scheme 11

# 10. Selectivity in the rearrangement steps

If an unsymmetrical ketone is used as the starting material, then obviously two products **37** and **38** could occur from the rearrangement (Scheme 12). The mechanism of rearrangement was discussed in previous publications, <sup>5a-c</sup> and involves zinc bromide-promoted oxirane ring-closure-ring-opening followed by migration of the group that can best stabilize an electron deficiency (Scheme 12).

H>Ar>Alk; sec-Alk>n-Alk

## Scheme 12

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The selectivity of benzotriazole-mediated insertions is notable. <sup>5a-c</sup> In most cases, single regioisomers **37** were produced by migration of the R<sup>1</sup> shown in Scheme 12 and Table 10. Similar migration aptitudes were found in other pinacol-type rearrangements H>Ar>Alk; *sec*-Alk>*n*-Alk. <sup>9a,b</sup>

Table 10.	Selectivity	of mig	ration
-----------	-------------	--------	--------

$R^1$	$R^2$	R	X
Migrating group	Non- migrating group	K	Λ
Н	$Ph(CH_2)_2$ -	Н	H, OMe, SPh, Ar
<i>i</i> -Pr	Me	Н	OPh, Ar
Н	$4-C1C_6H_4$	Н	OEt, SPh, Ar
Ph	Me	Н	SPh, OMe, Ar
Ph	Me	Ph, Me	SPh
Н	Ph	Н	SPh, SMe, OMe, Ar
Ph	4-pyridyl	Н	OMe
Н	$4-MeC_6H_4$	Н	$C_5H_{11}$

To extend the synthetic utility of benzotriazolyl-mediated one carbon insertion, the migratory aptitude of  $\pi$ -electron-rich heterocycles of 2-benzotriazolyl alcohols **40** in the presence of alkyl and aryl groups has been investigated recently. <sup>5e</sup> It was demonstrated that electron rich heteroaryl groups migrate more easily than the methyl group but less easily than a phenyl group (Scheme 13 and Table 11).

### Scheme 13

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Migrating	Non-migrating	Carbonyl	Number of	Average	Examples of
group (R)	group (R <sup>1</sup> )	substrate	examples	Yield, %	products 41
2-thienyl					^
3-thienyl	Me	O Het Me	6	53	
2-furyl					O SMe
2-benzofuryl					
2-benzothienyl					Me
3-benzothienyl					O
Ph	2-thienyl	O Het Ph	5	41	
	3-thienyl				//\\ _O
	2-furyl				S
	2-benzofuryl				Ph SMe
	2-benzothienyl				

Table 11. Examples of migration

# 11. Conclusions

Results compiled and discussed in this short review demonstrate the value of single carbon-insertion methods for the homologation of carbonyl compounds in organic synthesis. The benzotriazole-mediated carbon-insertion method described in this review seems to be general, highly regioselective and applicable to most aldehydes and ketones allowing the introduction of a variety of substituents attached to the inserted carbon atom.

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