Stereoselective annelation of 3-substituted imidazo[4,5-b]pyridines with cyanoacetylenic alcohols and domino rearrangement of the adducts

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Abstract

Imidazo[4,5-*b*]pyridines are readily annelated (45-50 °C, 24-30 h, MeCN) with cyanoacetylenic tertiary α-alcohols to give stereoselectively functionalized 1,3-oxazolo[3,2-*a*]imidazo[4,5-*b*]pyridines in 50-88% yields. The adducts undergo a facile stereoselective hydrolytic domino rearrangement to functionalized derivatives of 2,3-diaminopyridines.

Keywords: Annelation, formamides, imidazopyridines, rearrangement, zwitterions

Introduction

Derivatives of imidazole and pyridine are of special interest in drug design. ^{1,2} The condensed heterocyclic systems consisting of imidazole and pyridine rings (imidazopyridines) often possess pharmaceutically valuable properties which are not typical for either of the separate moieties. For example, a popular soporific remedy, "Zolpidem" ("Ivadal"), is a functionalized imidazo[1,2-*a*]pyridine. ² Among other derivatives of imidazopyridines, are potent nitric-oxide synthase inhibitors of inflammatory diseases, ³ inhibitors of AKT kinase, ⁴ preventive and/or therapeutic agents for neutrophilic inflammation disease, ⁵ compounds with tuberculostatic ⁶ and hypoglycemic ⁷ activities, angiotensin II receptor antagonists, ⁸ and anticoccidial agents. ⁹ A series of imidazo[1,2-*a*]pyridines bearing sulfonylurea functions exhibit herbicidal activity. ¹⁰

Most reported syntheses of imidazopyridines are, as a rule, laborious and multi-step (up to 6 steps). Therefore, the search for novel straightforward approaches to the synthesis of functionalized imidazopyridines remains a challenge.

The objective of this present work was to develop a straightforward efficient approach to the synthesis of new condensed functionalized heterocyclic systems combining and imidazo[4,5-b]pyridine skeleton and a 1,3-oxazole moiety. For this, we have studied the reaction of substituted imidazo[4,5-b]pyridines with readily available cyanoacetylenic tertiary α -alcohols.¹¹

From out recent findings related to the annelation of the cyanoacetylenic alcohols with pyridines, ^{11b,12} quinoline and quinoxaline, ¹³ phenanthridines, ¹⁴ natural alkaloids ¹⁵ and substituted benzimidazoles ¹⁶ one might expect that either the pyridine or the imidazole (or both) components of an imidazo[4,5-*b*]pyridine would be involved in the reaction to furnish novel 1,3-oxazoloimidazopyridine tricyclic fused scaffolds. Therefore, a fundamental part of the objective is also the issue of competition between the imidazole and pyridine nitrogen atoms as nucleophiles towards the electron-deficient acetylenic bond.

Results and Discussion

The experimental results have shown that 3-substituted imidazo[4,5-b]pyridines **1a,b** are readily annelated with cyanoacetylenic tertiary α -alcohols **2a,b** (45-50 °C, 24-30 h, MeCN, **1:2** molar ratio equals 1:1) to give (Z)-3-cyanomethylene-1,3-oxazolo[3,2-a]imidazo[4,5-b]pyridines **3a-d** in 50-88% yields (Table 1). No *E*-isomers of the adducts **3a-d** have been discernible in the reaction mixture, i.e. the annelation is strictly stereoselective. Neither alternative adducts **4** nor possible diadducts with participation of both imidazole and pyridine counterparts have been detected among the reaction products thus indicating that imidazole nitrogen atom N(1) entirely wins the competition against the pyridine nitrogen N(4) in the nucelophilic addition to the triple bond. In other words, the annelation is also strongly regioselective. Notably, although under the same conditions the annelation of 1-benzylimidazo[4,5-b]pyridine **1c** with cyanoacetylenic alcohol **2a** proceeds reluctantly to afford the corresponding adduct 1,3-oxazoloimidazopyridine **3e** (also of Z-configuration only) just in 18% yield, no alternative adducts are formed (Table 1).

When carried out at room temperature (20-25 °C), completion of the annelation takes a much longer time (up to 147 h), though the yields of the adducts **3a-e** increase by approximately 10% (relative to the yields shown in Table 1), obviously due to the milder reaction conditions.

The pyridine nitrogen atom was not involved into the annelation even when adduct **3c** was allowed to react with cyanoacetylenic alcohol **2b** (20-25 °C, 147 h). Also, no bis-adduct was isolated even when the reactants **1a** and **2b** were taken in the 1:2 molar ratio.

The structure of adducts **3a-e** was established by NMR (¹H, ¹³C, 2D) and IR spectroscopy. Thus, in the ¹H NMR spectra of 1,3-oxazoloimidazopyridines **3a-e**, olefinic protons (H-10, the =CHCN fragment) appear as singlets in the region 4.43-4.55 ppm thus indicating to the

formation of only one isomer (NOESY). In the IR spectra, the absorption bands of the cyano groups on the double bond are observed at 2202-2213 cm⁻¹.

Table 1. Formation of 1,3-oxazoloimidazopyridines **3a-e** from substituted imidazo[4,5-*b*]pyridines **1a-c** and cyanoacetylenic alcohols **2a,b**

R ¹	\mathbb{R}^2	\mathbb{R}^3	Products 3		Yield (%)
Me	Me	Me	NC Me N Me N Me	3a	88
Bn	Me	Me	NC Me Me N Me Bn	3 b	70
Me	(CH ₂) ₅		NC N N N N N Me	3c	75

Table 1. Continued

R1	R2 R3	Products 3	Yield (%)
Bn	(CH ₂) ₅	NC N N N N Bn	50
Bn	Me Me	NC Me N Me Me 3e	18

According to the NOESY spectra, the 1,3-oxazoloimidazopyridines **3a-e** are Z-isomers: the cross-peaks between olefinic proton H-10 and the immediately adjacent protons of cyclohexyl (for **3c**) or Me group (for **3e**) are observed (Figure 1).

Figure 1. Cross-peaks in the 2D NOESY spectra of 1,3-oxazoloimidazopyridines 3c and 3e.

Apparently, the annelation is triggered by the addition of imidazopyridines $\mathbf{1a}$, \mathbf{b} via their imidazole nitrogen atom N(1) as neutral nucleophiles to cyanoacetylenic alcohols $\mathbf{2a}$, \mathbf{b} to generate zwitterions \mathbf{A} having a vinyl carbanion moiety which is then converted into the oxygencentered zwitterions \mathbf{B} the proton transfer (in either an intra- or intermolecular manner) from the hydroxyl group to the carbanionic center. The secondary zwitterions \mathbf{B} undergo the ring closure at the C(2) position of the imidazole ring to form the final annelated adducts $\mathbf{3a-d}$ (Scheme 1).

1a,b + 2a,b
$$\stackrel{NC}{\longrightarrow} \stackrel{H}{\stackrel{N}{\longrightarrow}} \stackrel{NC}{\stackrel{N}{\nearrow}} \stackrel{R^3}{\stackrel{R^2}{\longrightarrow}} \stackrel{NC}{\longrightarrow} \stackrel{NC}{\stackrel{NC}{\longrightarrow}} \stackrel{R^3}{\stackrel{NC}{\longrightarrow}} \stackrel{R^3}{\longrightarrow} \stackrel{NC}{\longrightarrow} \stackrel{N} \stackrel{NC}{\longrightarrow} \stackrel{NC}{\longrightarrow} \stackrel{NC}{\longrightarrow} \stackrel{NC}{\longrightarrow} \stackrel{NC}{\longrightarrow} \stackrel{NC}{\longrightarrow} \stackrel{NC}{\longrightarrow}$$

Scheme 1. Proposed mechanism of formation 1,3-oxazoloimidazopyridines **3a-d**.

The Z-configuration of the final adducts **3a-d** is the expected result of the nucleophilic addition to acetylenes, which is known to proceed in a concerted *trans*-fashion.¹⁷ Notably, the 1,3-oxazoloimidazopyridines **3a-d** retain a contribution of zwitterionic form **B** (an inner salt). A similar consideration is true also for the formation of adduct **3e**.

The regioselectivity of the annelation likely results from the steric constrains between the cyanomethylene group and the substituents R^1 at the imidazole N(3) atom in the possible alternative adducts **4** (cf. Table 1). Indeed, the nucleophilicities of N(1) of the imidazole ring and N(4) of the pyridine ring should not differ significantly since the basicities of pyridine and benzimidazole, an analog of the imidazopyridines **1a**,**b**, are close (p $K_a = 5.20$ and 5.50 for pyridine and benzimidazole, respectively¹). This conclusion is in agreement with the low yield (18%) of adduct **3e**, in which the cyanomethylene group faces the steric repulsion from the lone electron pair of the N(4) pyridine atom.

1,3-Oxazoloimidazopyridines **3a-d**, when passed through a column packed with neutral alumina, or in moist ethanol, undergo the hydrolytic domino rearrangement to functionalized derivatives of 2,3-diaminopyridines conjugated with a 2-aminodihydrofuran moiety -(E)-N-(3- $\{[5-amino-2,2-dialkyl-3(2H)-furanyliden]amino\}$ -2-pyridinyl)-N-alkylformamides **5a-d** in 83-98% yields (Table 2). This exceptionally facile multi-step rearrangement is rigorously stereoselective: the products **5a-d** are formed exclusively in the E-configuration (relative to the C=N bond). Likewise, 1,3-oxazoloimidazopyridine **3e** rearranges to a similar derivative of 2,3-diaminopyridine of E-configuration in 46% yield (Table 2).

Table 2. Rearrangement of 1,3-oxazoloimidazopyridines **3a-e** to pyridinyl-*N*-substituted formamides **5a-e**

Table 2. Continued

R1	R2	R3	Products 5		Yield (%)
Bn	(CH ₂) ₅		NH ₂ O N N O	5d	93
Bn	Me	Me	NH_2 NH_2 N	5e	46

(*E*)-Pyridinyl-*N*-substituted formamides **5a-e** were isolated as crystals. Their structures are based on analogy with that of formamide **5c**, which was established by X-ray diffraction analysis, and confirmed using ¹H and ¹³C NMR, UV and IR techniques.

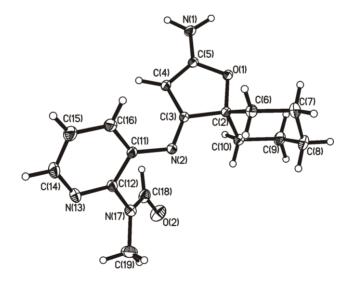


Figure 2. Molecular structure of (*E*)-pyridinyl-*N*-methylformamide **5c**.

Crystal structure of formamide 5c is formed by one crystallographically independent molecule (Figure 2) taking a general position. The cyclohexane ring has a chair conformation. The furan and pyridine heterocycles are almost planar, maximum deviation of atoms from their average planes do not exceed 0.02 Å (C(2) and C(12) atoms). The dihedral angle between these planes is 125.4°. Deviations of the N(1) and N(2) atoms from plane of the furan cycle are

insignificant (0.01 and 0.03 Å, respectively). The N(2) and N(17) atoms deviate from the plane of the pyridine ring by 0.23 and 0.10 Å. The fragment formed by the O(2)C(18)N(17)C(19) atoms is practically planar, maximum deviation of atoms from its average plane being 0.02 Å (C(18) atom). This plane forms with planes of the furan and pyridine moieties the dihedral angles of 117.4 and 130.5°, respectively.

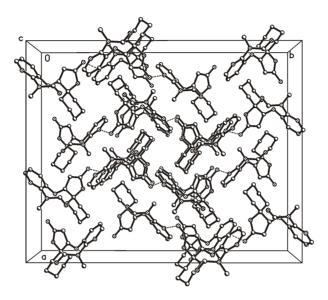


Figure 3. Crystal structure of formamide 5c.

Crystal structure of formamide **5c** is given in Figure 3. In this compound, the molecules are bonded in chains (shorten distances along the axis *b*) via hydrogen bonding *N*-H...*N*: N(1)...N(2) – 3.026(2) Å, N(1)-H(1A) – 0.93(2) Å, N(2)... H(1A) – 2.14(2) Å, N(1)-H(1A)...N(2) – 159(2)°, N(1')...N(13) – 2.999(2) Å, N(1')-H(1B') – 0.95(3) Å, N(13)...H(1B') – 2.09(3) Å, N(1')-H(1B')...N(13) – 160(2)° (sum of van-der-Waals radii N...H – 2.75 Å, N...N is 3.10 Å¹⁸).

Obviously, the rearrangement of 1,3-oxazoloimidazopyridines $\bf 3a-d$ starts with the cleavage of polar C(2)-O bond by water to deliver the intermediates $\bf C$ (Scheme 2) which further rearrange to the intermediates $\bf D$ having a formamide function, probably stabilized by intramolecular H-bonding between the NH moiety and the carbonyl oxygen. The latter, being an NH vinylamine, undergoes a 1,3-hydrogen shift to produce the imine intermediates $\bf E$, in which the free rotation of the CN function is allowed. By virtue of this, the ring closure by addition of hydroxyl function to the CN bond to furnish the iminodihydrofuran intermediates $\bf F$ becomes possible. The 1,3-hydrogen shift in the intermediates $\bf F$ to form iminodihydrofuranylidene moiety in final products $\bf 5a-d$ completes the domino sequence.

The rearrangement of **3e** to **5e** would pass through similar steps.

3a-d
$$\xrightarrow{Al_2O_3 (H_2O)_n}$$
 \xrightarrow{NC} \xrightarrow{H} \xrightarrow{NC} \xrightarrow{H} \xrightarrow{NC} $\xrightarrow{N$

Scheme 2. Probable steps of the rearrangement of (*Z*)-1,3-oxazoloimidazopyridines **3a-d** to (*E*)-pyridinyl-*N*-alkylformamides **5a-d**.

The driving force of this multi-step rearrangement is plausibly the formation of a long-range through-conjugated system of the push-pull type, accomplished in the final products **5a-d**. Despite the distorted coplanarity of the pyridine and furan rings in the crystal state (the corresponding dihedral angle is 125.4°, Figure 2), the conjugation can still be partly realized, particularly in solution, or in excited or transition states (during certain reactions).

Easrlier, a simlar deep rearrangement was observed for the adducts of benzimidazoles with cyanoacetylenic tertiary α -alcohols. ^{16a}

The rearrangement products **5a-e** can be prepared directly from reactants **1** and **2** avoiding the isolation of annelated species **3a-e** by passing the reaction mixture through a layer of alumina.

An attempt to obtain a monocrystal of 1,3-oxazoloimidazopyridine 3c [where $R^1 = Me$, $R^2-R^3 = (CH_2)_5$] from moist ethanol ends up with the formation of crystalline rearrangement product 5c.

In the ¹H NMR spectra of (*E*)-pyridinyl-*N*-substituted formamides **5a-e**, the proton of the formamide fragment (H-14) is observed in the region 8.23-8.78 ppm. In the ¹³C NMR spectra, carbonyl carbon resonates at 163.2-169.8 ppm. In the IR spectra, the absorption bands of the C=O group are present at 1659-1684 cm⁻¹.

The direct synthesis of the rearrangement products **5a-e** was monitored as an example of a three-component reaction, between imidazopyridine **1a**, cyanoacetylenic tertiary α -alcohol **2a** and water (1:1:1 molar ratio) in CD₃CN solution (room temperature, NMR tube). The reaction was followed by the methyl and olefin proton signals in the ¹H NMR spectra (Scheme 3).

Scheme 3. Monitoring of the three-component reaction of imidazopyridine **1a** with cyanoacetylenic alcohol **2a** and water.

After 1 h, in the spectrum, along with the signals of starting materials [1.48 (Me in cyanoacetylenic α-alcohol 2a) and 3.83 (N-Me in imidazopyridine 1a) ppm], weak singlets at 2.96, 4.88 and 5.01 ppm assignable to N-Me, olefinic proton H-10 in cyanomethylene group (in annelated product 3a) and olefinic proton in the zwitterion B (Schemes 1, 3) respectively, appeared. During the next three days the intensities of these signals gradually increased and singlets at 1.29 and 1.38 ppm become observable. After 11 days, signals for the annelated adduct 3a were considerably augmented and weak signals at 3.16 and 4.38 ppm emerged attributable to N-Me and olefinic proton in the intermediate C (Schemes 2, 3). After 20 days, the intensities of these signals rose with a simultaneous decrease of the intensity of the olefinic proton signal in the zwitterion B (5.01 ppm). Finally, after 35 days, the singlet at 5.01 ppm disappeared and the signal intensities of annelated adduct 3a (1.29, 1.38, 2.96 and 4.88 ppm) fell sharply. In this spectrum, the signals of intermediates C (3.16 and 4.38 ppm) and D (4.35 ppm) dominated. When the reaction mixture was passed through the column packed with alumina (chloroform/benzene/ethanol, 20:4:1, mixture as eluent), the final rearrangement products 5a was isolated (Scheme 2).

Thus, the monitoring confirms Schemes 1 and 2, though under the conditions employed, the combined rearrangement annelation sequence proceeds only slowly and did not reach completion, so that even after 35 days the reaction mixture consisted of intermediates **C** and **D**.

Conclusions

In conclusion, a novel one-pot efficient methodology for the stereoselective synthesis of previously unknown imidazo [4,5-b] pyridine derived tricyclic fused functionalized systems – (Z)-

1,3-oxazolo[3,2-a]imidazo[4,5-b]pyridines, has been developed (yields up to 88%). The methodology represents an exceptionally facile (20-50 °C) metal-free stereoselective (relative to the cyanomethylene function) annelation of imidazo[4,5-b]pyridines with cyanoacetylenic tertiary α -alcohols, involving intermediate zwitterions. The 1,3-oxazolo[3,2-a]imidazo[4,5-b]pyridines, when passed through alumina, or in aqueous solvents at room temperature, undergo an easy hydrolytic multi-step domino rearrangement to functionalized derivatives of 2,3-diaminopyridines – (E)-pyridinyl-N-alkylformamides, in excellent to quantitative yields. The principal steps of the rearrangement were verified by NMR monitoring to show that initially the cleavage of the C(2)-O bond of the oxazole cycle by the molecule of water occurs. Then follows the rearrangement of the semi-aminal moiety forking the formamide function and a sequence of 1,3-prototropic shifts and the iminodihydrofuran ring closure. A straightforward three-component access to derivatives of 2,3-diaminopyridines conjugated with 2-aminodihydrofurans directly from imidazo[4,5-b]pyridine, cyanoacetylenic tertiary α -alcohols and water has been elaborated. The results contribute to the fundamental chemistry of imidazole, pyridine and acetylenes as well as to the development of novel imidazo[4,5-b]pyridines.

Experimental Section

General. NMR spectra were run on a Bruker DPX-400 spectrometer with HMDS as an internal standard. UV-VIS spectra were measured on a Perkin-Elmer Lambda 35 spectrometer at room temperature (EtOH, d = 0.1 cm). IR spectra were recorded on a IFS 25 instrument. Column and thin-layer chromatography was carried out on neutral Al₂O₃ with chloroform/benzene/ethanol (20:4:1) mixture as eluent (length of column = 35 cm, Ø = 0.8 cm). The reaction was monitored by the disappearance of absorption bands of initial acetylenes **2a,b** in the reaction mixture (IR spectroscopy). Alkylation of imidazo[4,5-*b*]pyridines **1a,b** was carried out according to the procedures: ¹⁹ mp **1a** 56 °C (hexane) [mp 76-78 °C²⁰]; mp **1b** 79-81 °C (hexane) [mp 83 °C²¹]; mp **1c** 134-135 °C (acetone) [mp 119 °C²¹]. Cyanoacetylenic tertiary α-alcohols **2a,b** were prepared by the method. ¹¹ Labeling of hydrogen and carbon atoms in compounds **3a-e** and **5a-e** are given in Figure 4.

Figure 4. Labeling of hydrogen and carbon atoms in compounds 3a-e and 5a-e.

X-ray diffraction. The X-ray diffraction study of compound 5c was carried out on an Enraf-Nonius CAD-4 diffractometer at room temperature ($\omega/2\theta$ -scans, Mo-K $_{\alpha}$ radiation, graphite monochromator). The crystal structure was solved by direct methods followed by Fourier-syntheses using SHELXS-97 programs. The structure refined by a full matrix least-squares anisotropic procedure using SHELXTL-97 programs. The parameters of hydrogen atoms were defined experimentally and refined anisotropically, except for parameters of hydrogen of the methyl group which were calculated geometrically. These data are available via www.ccdc.cam.uk/contsretrieving.html (or from CCDC, 12 Union Cambrige CB2 1EZ, UK, fax: +44 (0) 1223 336 033; or e-mail: deposit@ccdc.cam.ac.uk). Any request to the CCDC for data should quote the full literature citation and CCDC reference number CCDC 826322 (for 5c).

General synthetic procedure for the preparation of products 3, exemplified by (*Z*)-3-cyanomethylene-2,3,9,9a-tetrahydro-9-methyl-2,2-dimethyl-1,3-oxazolo[3,2-*a*]-imidazo[4,5-*b*]pyridine (3a)

A mixture of 3-methyl-3*H*-imidazo[4,5-*b*]pyridine (**1a**) (0.133 g, 1 mmol) and cyanoacetylenic alcohol **2a** (0.109 g, 1 mmol) in dry MeCN (0.2 mL) was stirred at 45-50 °C for 30 h. The solvent was removed and the residue was washed with anhydrous Et₂O (5 × 0.3 mL) to give 1,3-oxazoloimidazopyridine **3a**, yield 88%, 0.212 g, light-pink powder, mp 98-101 °C; IR (v_{max} , cm⁻¹): 1060, 1081, 1123 (C-O-C), 1652 (C=C), 2202 (=CH-*CN*), 3024 (=*CH*-CN). ¹H NMR (400.13 MHz, CDCl₃): δ_{H} 1.29, 1.38 (6H, 2 × s, 2 × CH₃), 2.96 (3H, s, NCH₃), 4.43 (1H, s, H-10), 6.28 (1H, s, H-9a), 6.48 (1H, dd, H-6), 7.63 (1H, d, ${}^{3}J_{H6,H7}$ = 5.2 Hz, H-7), 7.68 (1H, d, ${}^{3}J_{H5,H6}$ = 7.7 Hz, H-5). ¹³C NMR (100.62 MHz, CDCl₃): δ_{C} 26.5, 28.0 (2 × CH₃), 29.1 (NCH₃), 70.0 (C-10), 83.6 (C-2), 105.4 (C-9a), 113.8 (C-6), 118.6 (CN), 120.7 (C-5), 128.9 (C-4a), 142.3 (C-7), 153.8 (C-8a), 167.9 (C-3); Anal. Calcd for C₁₃H₁₄N₄O (242.20): C, 64.45; H, 5.82; N, 23.12%. Found: C, 64.60; H, 5.89; N, 23.10%.

(Z) - 3 - Cyanomethylene - 2, 3, 9, 9a - tetra hydro-9-benzyl - 2, 2 - dimethyl - 1, 3 - oxazolo [3, 2-a] - dimethylene - 2, 3, 9, 9a - tetra hydro-9-benzyl - 2, 2 - dimethyl - 1, 3 - oxazolo [3, 2-a] - dimethyl - 1, 3 - oxazolo [3

imidazo[4,5-*b*]pyridine (3b). Light-yellow powder, yield 70%, 0.256 g, mp 102-104 °C; IR (v_{max} , cm⁻¹): 1108, 1136, 1156 (C-O-C), 1650 (C=C), 2213 (=CH-CN), 3066 (=CH-CN). ¹H NMR (400.13 MHz, CDCl₃): $δ_H$ 1.39, 1.43 (6H, 2 × s, 2 × CH₃), 4.53, 4.97 (2H, 2 × d, $^2J_{H,H}$ = 15.6 Hz, CH₂ from N-benzyl), 4.54 (1H, s, H-10), 6.39 (1H, s, H-9a), 6.65 (1H, dd, H-6), 7.24-7.30 (5H, m, Ph from N-benzyl), 7.78 (1H, d, $^3J_{H6,H7}$ = 5.4 Hz, H-7), 7.84 (1H, d, $^3J_{H5,H6}$ = 7.7 Hz, H-5). ¹³C NMR (100.62 MHz, CDCl₃): $δ_C$ 26.5, 28.1 (2 × CH₃), 43.1 (CH₂ from N-benzyl), 70.7 (C-10), 83.3 (C-2), 103.4 (C-9a), 114.2 (C-6), 118.6 (CN), 120.8 (C-5), 127.7, 128.1, 128.7 (m,o,p-C, Ph from N-benzyl), 128.9 (C-4a), 136.4 (i-C, Ph from N-benzyl), 142.4 (C-7), 153.2 (C-8a), 168.2 (C-3); Anal. Calcd for C₁₉H₁₈N₄O (318.38): C, 71.68; H, 5.70; N, 17.60%. Found: C, 71.95; H, 5.42; N, 17.85%.

(Z)-Spiro-1,3-oxazolo[3,2-a]imidazo[4,5-b]pyridine (3c). Red powder, yield 75%, 0.212 g, mp 103-105 °C; IR (ν_{max} , cm⁻¹): 1052, 1074, 1125, 1148, 1167 (C-O-C), 1644 (C=C), 2206 (=CH-CN), 3077 (=CH-CN). ¹H NMR (400.13 MHz, CDCl₃): δ_{H} 1.10-1.90 (10H, m, cyclohexyl), 3.14 (3H, s, NCH₃), 4.55 (1H, s, H-10), 6.42 (1H, s, H-9a), 6.63 (1H, dd, H-6), 7.78 (1H, d, $^{3}J_{H6,H7}$ =

5.4 Hz, H-7), 7.83 (1H, d, ${}^3J_{H5,H6}$ = 7.8 Hz, H-5). 13 C NMR (100.62 MHz, CDCl₃): δ_C 21.3-36.2 (5 × CH₂, cyclohexyl), 28.6 (NCH₃), 70.3 (C-10), 84.9 (C-2), 105.2 (C-9a), 113.3 (C-6), 118.2 (CN), 120.3 (C-5), 128.7 (C-4a), 141.8 (C-7), 153.4 (C-8a), 167.6 (C-3); Anal. Calcd for $C_{16}H_{18}N_4O$ (282.34): C, 68.06; H, 6.43; N, 19.84%. Found: C, 68.33; H, 6.56; N, 19.55%.

(Z)-Spiro-1,3-oxazolo[3,2-*a*]imidazo[4,5-*b*]pyridine (3d). Light-yellow powder, yield 50%, 0.197 g, mp 108-110 °C; IR (v_{max} , cm⁻¹): 1064, 1082, 1105, 1130, 1159, 1174 (C-O-C), 1647 (C=C), 2209 (=CH-*CN*), 3066 (=*CH*-CN). ¹H NMR (400.13 MHz, CDCl₃): δ_H 1.10-1.90 (10H, m, cyclohexyl), 4.57 (1H, s, H-10), 4.67, 4.92 (2H, 2 × d, $^2J_{H,H}$ = 15.5 Hz, CH₂ from *N*-benzyl), 6.41 (1H, s, H-9a), 6.65 (1H, dd, H-6), 7.25-7.40 (5H, m, Ph from *N*-benzyl), 7.80 (1H, d, $^3J_{H6,H7}$ = 5.5 Hz, H-7), 7.82 (1H, d, $^3J_{H5,H6}$ = 7.8 Hz, H-5). ¹³C NMR (100.62 MHz, CDCl₃): δ_C 21.6-24.7 (5 × CH₂, cyclohexyl), 46.2 (CH₂ from *N*-benzyl), 71.5 (C-10), 84.9 (C-2), 103.7 (C-9a), 114.0 (C-6), 118.5 (CN), 120.6 (C-5), 127.2, 128.2, 128.5 (*m*,*o*,*p*-C, Ph from *N*-benzyl), 129.2 (C-4a), 136.4 (*i*-C, Ph from *N*-benzyl), 142.2 (C-7), 153.1 (C-8a), 168.4 (C-3); Anal. Calcd for C₂₂H₂₂N₄O (358.44): C, 73.72; H, 6.19; N, 15.63%. Found: C, 73.45; H, 5.99; N, 15.37%.

(Z) - 2 - [9 - Benzyl - 2, 2 - dimethyl - 9, 9a - dihydro [1,3] oxazolo [2',3':2,3] imidazo [4,5-b] pyridin-dimethyl - 9, 9a - dihydro [1,3] oxazolo [2',3':2,3] imidazo [4,5-b] pyridin-dimethyl - 9, 9a - dihydro [1,3] oxazolo [2',3':2,3] imidazo [4,5-b] pyridin-dimethyl - 9, 9a - dihydro [1,3] oxazolo [2',3':2,3] imidazo [4,5-b] pyridin-dimethyl - 9, 9a - dihydro [1,3] oxazolo [2',3':2,3] imidazo [4,5-b] pyridin-dimethyl - 9, 9a - dihydro [1,3] oxazolo [2',3':2,3] imidazo [4,5-b] pyridin-dimethyl - 9, 9a - dihydro [1,3] oxazolo [2',3':2,3] imidazo [4,5-b] pyridin-dimethyl - 9, 9a - dihydro [1,3] oxazolo [2',3':2,3] imidazo [4,5-b] pyridin-dimethyl - 9, 9a - dihydro [1,3] oxazolo [2',3':2,3] imidazo [4,5-b] pyridin-dimethyl - 9, 9a - dihydro [1,3] oxazolo [2',3':2,3] imidazo [2,5-b] pyridin-dimethyl - 9, 9a - dihydro [1,3] oxazolo [2',3':2,3] imidazo [2,5-b] pyridin-dimethyl - 9, 9a - dihydro [1,3] oxazolo [2',3':2,3] imidazo [2,5-b] pyridin-dimethyl - 9, 9a - dihydro [1,3] oxazolo [2',3':2,3] imidazo [2,5-b] pyridin-dimethyl - 9, 9a - dihydro [1,3] oxazolo [2',3':2,3] imidazo [2,5-b] pyridin-dimethyl - 9, 9a - dihydro [1,3] oxazolo [2',3':2,3] imidazo [2,5-b] pyridin-dimethyl - 9, 9a - dihydro [1,3] oxazolo [2',3':2,3] imidazo [2,5-b] pyridin-dimethyl - 9, 9a - dihydro [1,3] oxazolo [2',3':2,3] imidazo [2,5-b] pyridin-dimethyl - 9, 9a - dihydro [1,3] oxazolo [2',3':2,3] imidazo [2,5-b] pyridin-dimethyl - 9, 9a - dihydro [2,5-b] pyridin-dimethyl - 9, 9a - dihydr

3(2*H***)-yliden]acetonitrile (3e).** Light-brown powder, yield 18%, 0.057 g, mp 121-124 °C; IR (v_{max} , cm⁻¹): 1079, 1155 (C-O-C), 1634 (C=C), 2204 (=CH-*CN*), 3063 (=*CH*-CN). ¹H NMR (400.13 MHz, CDCl₃): δ_{H} 1.39, 1.45 (6H, 2 × s, 2 × CH₃), 4.15, 4.55 (2H, 2 × d, ² $J_{H,H}$ = 15.5 Hz, CH₂ from *N*-benzyl), 4.51 (1H, s, H-10), 6.44 (1H, s, H-9a), 6.73 (1H, dd, H-7), 7.20-7.50 (5H, m, Ph from *N*-benzyl), 7.55 (1H, d, ³ $J_{H7,H8}$ = 7.7 Hz, H-8), 7.75 (1H, d, ³ $J_{H6,H7}$ = 5.5 Hz, H-6). ¹³C NMR (100.62 MHz, CDCl₃): δ_{C} 25.5, 27.5 (2 × CH₃), 44.1 (CH₂ from *N*-benzyl), 62.7 (C-10), 79.4 (C-2), 101.0 (C-9a), 114.0 (C-7), 118.5 (CN), 123.0 (C-8), 126.6, 127.9, 128.6 (*m*,*o*,*p*-C, Ph from *N*-benzyl), 130.2 (C-8a), 141.1 (*i*-C, Ph from *N*-benzyl), 142.1 (C-6), 145.1 (C-4a), 160.5 (C-3); Anal. Calcd for C₁₉H₁₈N₄O (318.38): C, 71.86; H, 5.70; N, 17.60%. Found: C, 71.89; H, 5.52; N, 17.68%.

General synthetic procedure for the preparation of products 5, exemplified by (E)-N-(3-[5-amino-2,2-dimethyl-3(2H)-furanyliden]amino}-2-pyridinyl)-N-methylformamide (5a)

1,3-Oxazoloimidazopyridine **3a** (0.145 g, 0.6 mmol) was dissolved in a mixture of CHCl₃-benzene-EtOH (20:4:1, 5 mL). Column chromatography on Al₂O₃ was employed to afford formamide **5a**, yield 83%, 0.130 g, light-yellow powder, mp 183-184 °C; IR (v_{max} , cm⁻¹): 1066, 1097, 1122, 1135 (C-O-C), 1676 (C=O), 3294, 3361 (NH₂). ¹H NMR (400.13 MHz, DMSO- d_6): δ_H 1.40 (6H, s, 2 × CH₃), 3.13 (3H, s, NCH₃), 4.31 (1H, s, H-3), 7.18 (1H, dd, H-9), 7.32 (1H, d, $^3J_{H8,H9}$ = 7.7 Hz, H-8), 7.50 (2H, s, NH₂), 8.01 (1H, d, $^3J_{H9,H10}$ = 4.6 Hz, H-10), 8.26 (1H, s, H-14). ¹³C NMR (100.62 MHz, DMSO- d_6): δ_C 25.3 (2 × CH₃), 30.2 (NCH₃), 69.6 (C-3), 89.0 (C-5), 122.1 (C-9), 130.7 (C-8), 141.0 (C-10), 142.8 (C-2), 147.1 (C-7), 163.2 (C-14), 173.9 (C-2), 178.1 (C-4). UV/Vis (EtOH): λ_{max} (log ε) = 298 (4.42) nm; Anal. Calcd for C₁₃H₁₆N₄O₂ (260.30): C, 59.99; H, 6.20; N, 21.52%. Found: C, 59.69; H, 6.47; N, 21.48%.

(E)-N-(3-{[5-Amino-2,2-dimethyl-3(2H)-furanyliden]amino}-2-pyridinyl)-N-benzyl-formamide (5b). White powder, yield 98%, 0.206 g, mp 160-161 °C; IR (v_{max}, cm⁻¹): 1076, 1106

(C-O-C), 1676 (C=O), 3331, 3356 (NH₂). ¹H NMR (400.13 MHz, DMSO- d_6): δ_H 1.45 (6H, s, 2 × CH₃), 4.41 (1H, s, H-3), 5.09 (2H, s, CH₂ from *N*-benzyl), 7.14 (1H, dd, H-9), 7.14 (1H, d, ${}^3J_{H8,H9} = 7.6$ Hz, H-8), 7.20-7.35 (5H, m, Ph from *N*-benzyl), 7.57 (2H, s, NH₂), 7.99 (1H, d, ${}^3J_{H9,H10} = 4.4$ Hz, H-10), 8.40 (1H, s, H-14). ¹³C NMR (100.62 MHz, DMSO- d_6): δ_C 25.4 (2 × CH₃), 45.5 (CH₂ from *N*-benzyl), 69.8 (C-3), 89.3 (C-5), 122.2 (C-9), 126.6, 126.9, 128.0 (m,o,p-C, Ph from *N*-benzyl), 130.3 (C-8), 137.9 (i-C, Ph from *N*-benzyl), 141.0 (C-10), 143.0 (C-12), 146.4 (C-7), 163.5 (C-14), 174.2 (C-2), 178.1 (C-4); Anal. Calcd for C₁₉H₂₀N₄O₂ (336.39): C, 67.84; H, 5.99; N, 16.66%. Found: C, 68.10; H, 5.87; N, 16.87%.

$(E)-N-\{3-[(2-Amino-1-oxaspiro[4,5]dec-2-en-4-yliden)amino]-2-pyridinyl\}-N-methyl-$

formamide (**5c**). Red powder, yield 91%, 0.272 g, mp 219-221 °C; IR (v_{max} , cm⁻¹): 1072, 1107 (C-O-C), 1684 (C=O), 3216, 3352 (NH₂). ¹H NMR (400.13 MHz, DMSO- d_6): δ_H 1.50-1.70 (10H, m, cyclohexyl), 3.11 (3H, s, NCH₃), 4.29 (1H, s, H-3), 7.16 (1H, dd, H-9), 7.28 (1H, d, ${}^3J_{H8,H9} = 7.8$ Hz, H-8), 7.48 (2H, s, NH₂), 8.01 (1H, d, ${}^3J_{H9,H10} = 4.5$ Hz, H-10), 8.23 (1H, s, H-14). ¹³C NMR (100.62 MHz, DMSO- d_6): δ_C 21.8-33.8 (5 × CH₂, cyclohexyl), 30.4 (NCH₃), 70.3 (C-3), 90.5 (C-5), 122.3 (C-9), 131.0 (C-8), 141.1 (C-10), 142.9 (C-12), 147.2 (C-7), 163.3 (C-14), 174.3 (C-2), 178.2 (C-4); Anal. Calcd for C₁₆H₂₀N₄O₂ (300.36): C, 63.98; H, 6.43; N, 18.65%. Found: C, 63.68; H, 6.74; N, 18.36%.

$(E)-N-\{3-[(2-Amino-1-oxaspiro[4,5]dec-2-en-4-yliden)amino]-2-pyridinyl\}-N-benzyl-new amino ami$

formamide (**5d**). Similar to the synthesis of formamide **5b**, from 1,3-oxazolidino-imidazopyridine **3d** (0.093 g, 0.3 mmol) was prepared formamide **5d**, yield 93%, 0.094 g, light-brown powder, mp 158-160 °C; IR (v_{max} , cm⁻¹): 1059, 1112 (C-O-C), 1659 (C=O), 3284, 3335 (NH₂). ¹H NMR (400.13 MHz, DMSO- d_6): δ_H 1.55-1.80 (10H, m, cyclohexyl), 4.39 (1H, s, H-3), 5.06 (2H, s, CH₂ from *N*-benzyl), 7.13 (1H, d, H-8), 7.17 (1H, dd, ${}^3J_{H8,H9} = 7.6$ Hz, H-9), 7.30-7.40 (5H, m, Ph from *N*-benzyl), 7.55 (2H, s, NH₂), 7.98 (1H, d, ${}^3J_{H9,H10} = 4.4$ Hz, H-10), 8.37 (1H, s, H-14). ¹³C NMR (100.62 MHz, DMSO- d_6): δ_C 21.7-33.8 (5 × CH₂, cyclohexyl), 45.6 (CH₂ from *N*-benzyl), 70.3 (C-3), 90.6 (C-5), 122.2 (C-9), 126.6, 126.9, 128.0 (m,o,p-C, Ph from *N*-benzyl), 130.4 (C-8), 137.8 (i-C, Ph from *N*- benzyl), 140.9 (C-10, C-12), 146.3 (C-7), 163.4 (C-14), 174.4 (C-2), 178.1 (C-4); Anal. Calcd for C₂₂H₂₄N₄O₂ (376.46): C, 70.19; H, 6.43; N, 14.88%. Found: C, 69.99; H, 6.62; N, 14.56%.

$(E)-N-(2-\{[5-Amino-2,2-dimethyl-3(2H)-furanyliden]amino\}-3-pyridinyl)-N-benzyl-$

formamide (**5e**). Analogously, from 1-benzyl-1*H*-imidazo[4,5-*b*]pyridine **1c** (0.080 g, 0.25 mmol) was prepared formamide **5e**, yield 46%, 0.039 g, light-brown powder, mp 107-110 °C; IR (ν_{max}, cm⁻¹): 1076, 1181 (C-O-C), 1664 (C=O), 3283, 3338 (NH₂). ¹H NMR (400.13 MHz, DMSO- d_6): δ_H 1.37 (6H, s, 2 × CH₃), 4.59 (1H, s, H-3), 5.39 (2H, s, CH₂ from *N*-benzyl), 6.54 (1H, dd, H-10), 7.03 (1H, d, ${}^3J_{\text{H10,H11}} = 7.2$ Hz, H-11), 7.30-7.40 (5H, m, Ph from *N*-benzyl), 7.50 (2H, s, NH₂), 7.68 (1H, d, ${}^3J_{\text{H9,H10}} = 5.9$ Hz, H-9), 8.78 (1H, s, H-14). ¹³C NMR (100.62 MHz, DMSO- d_6): δ_C 25.0 (2 × CH₃), 54.3 (CH₂ from *N*-benzyl), 70.4 (C-3), 89.4 (C-5), 110.4 (C-10), 126.2 (C-11), 127.5, 127.6, 128.5 (*m*,*o*,*p*-C, Ph from *N*-benzyl), 132.1 (C-9), 136.8 (*i*-C, Ph from *N*-benzyl), 143.8 (C-7), 154.5 (C-12), 169.8 (C-14), 174.0 (C-2, C-4); Anal. Calcd for C₁₉H₂₀N₄O₂ (336.39): C, 67.84; H, 5.99; N, 16.66%. Found: C, 68.08; H, 5.67; N, 16.56%.

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