One-pot, three component reactions between isocyanides and dialkyl acetylenedicarboxylates in the presence of phenyl isocyanate: synthesis of dialkyl 2-(alkyl/arylimino)-2,5-dihydro-5-oxo-1-phenyl-1*H*-pyrrole-3,4-dicarboxylate

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Abstract

The reactive 1:1 adduct, generated from the reaction between alkyl/aryl iscyanides and dialkyl acetylenedicarboxylates, was trapped by phenyl isocyanate to yield dialkyl 2-(alkyl/arylimino)-2,5-dihydro-5-oxo-1-phenyl-1*H*-pyrrole-3,4-dicarboxylate in good yields.

Keywords: Phenyl isocyanate, 2,5-dihydro-5-oxo-1-phenyl-1*H*-pyrrole derivatives, three component reactions, isocyanides, acetylenic esters

Introduction

Pyrrole derivatives are an important class of heterocycles. Nitrogen heterocycles are of synthetic interest because they constitute an important class of natural and non-natural product, many of which exhibit useful biological activity.¹

Isocyanide-based multicomponent reactions (IMCR) now occupy a position of importance in synthetic organic chemistry, mainly due to the contributions of Ugi and co-workers.² Organic isocyanates are powerful tools in organic synthesis.³⁻⁵ Generally, isocyanates easily undergo polar cycloadditions with a large variety of unsaturated substrates.⁶⁻⁷

The reactivity of nucleophilic carbenes such as isocyanides towards dimethyl acetylenedicarboxylate (DMAD) is well documented.⁸⁻¹⁰ The reaction of isocyanides with carbon- carbon triple bonds occurs in a stepwise manner through a zwitterionic intermediate, the ultimate fate of which appears to be dictated by the nature of original triple-bonded substrate.¹¹⁻¹⁹ In continuation of our interest in the application of isocyanides in heterocyclic synthesis, ²⁰⁻²⁷ we

now report the reaction between alkyl/aryl isocyanides **1** and dialkyl acetylenedicarboxylate **2** in the presence of phenyl isocyanate. This one-pot, three component synthesis proceeded spontaneously at 38 °C in CH₂Cl₂ and leads to dialkyl 2-(alkyl/arylimino)-2,5-dihydro-5-oxo-1-phenyl-1*H*-pyrrole derivatives **3a-h** (Scheme 1).

Scheme 1. Synthesis of compounds 3a-h.

Result and Discussion

The structure of compounds **3a-h** was deduced from their IR, 1 H NMR, 13 C NMR, mass spectral data, HRMS data and elemental analysis. The mass spectra of these compounds **3a-h** displayed molecular ion peaks at appropriate m/e values. The 1 H NMR spectrum of compound **3a** exhibited a multiplet for the five CH₂ of cyclohexyl ring (δ 1.49-1.81 ppm), a multiplet for the N-CH cyclohexyl proton (δ 3.78 ppm), two singlet for the methoxy groups (δ 3.96 and 3.99 ppm). The aromatic hydrogens gave rise to characteristic multiplet signal in the aromatic region of the spectrum (δ 7.23-7.43 ppm).

The 13 C NMR spectrum of **3a** showed sixteen distinct resonances in agreement with proposed structure. The characteristic signal due to the amide carbonyl group was discernible at δ 148.0 ppm. Carbons of imino and two ester carbonyls resonated at δ 147.6, 160.3 and 160.6 ppm, respectively. Partial assignment of these resonances is given in the experimental data.

Dimethyl 2-(cyclohexylimino)-2,5-dihydro-5-oxo-1-phenyl-1*H*-pyrrole-3,4-dicarboxylate **3a** was obtained as a orange powder. Its molecular formula, $C_{20}H_{22}N_2O_5$, was determined on the basis of the positive HRESIMS at m/z 393.1435 [M+Na]⁺ (calcd 393.1426) and supported by the 1H , ^{13}C NMR and IR spectra.

The IR spectra of **3a** showed strong absorptions at 1718 and 1750 cm⁻¹ due to the ester carbonyls and also at 1656 cm⁻¹ because of the carbonyl of amide. The ¹H and ¹³C NMR spectra of **3b-h** are similar to **3a** and the results are described in experimental section.

Mechanistically, it is conceivable that the reaction involves the initial formation of a 1:1 zwitterionic intermediate **4** in accord with reaction between alkyl/aryl isocyanide **1** and dialkyl acetylenedicarboxylate **2**. Nucleophilic addition of **4** to the carbonyl group of phenyl isocyanate formed a dipolar species **5**. Cyclization of the latter leads to the 2,5-dihydro-5-oxo-1-phenyl-1*H*-pyrrole derivatives **3** (Scheme 2).

In this work, we used two factors to generated pyrrole derivatives, first, temperature factor by refluxing in CH₂Cl₂ at 38 °C and the second, time factor between 24-48 hours. Therefore we have developed a one-pot three-component reaction between alkyl/aryl isocyanides 1 and dialkyl acetylenedicarboxylate 2 in the presence of phenyl isocyanate for the preparation of dialkyl 2-(alkyl/arylimino)-2,5-dihydro-5-oxo-1-phenyl-1*H*-pyrrole derivatives 3a-h.

$$R^{1} \longrightarrow \mathbb{Q}$$

$$1$$

$$CO_{2}R^{2}$$

$$R^{2}O_{2}C$$

$$R^{2}O_{2$$

Scheme 2. Proposed mechanism for the formation of compound **3**.

Conclusions

In summary, we have found that the reaction of alkyl/aryl isocyanides with dialkyl acetylenedicarboxylates in the presence of phenyl isocyanate leads to the one-pot and simple synthesis of highly functionalized 2,5-dihydro-5-oxo-1-phenyl-1*H*-pyrrole derivatives. The present method carries the advantages that, not only is the reaction performed under neutral conditions, but the substances can be mixed without any activation or modification.

Experimental Section

General. Cyclohexyl isocyanide, *t*-butyl isocyanide, 2,6-dimethylphenyl isocyanide, dialkyl acetylenedicaboxylates and phenyl isocyanate were purchased from Fluka and Aldrich and used without further purification. Melting points and IR spectra were measured on an Electrothermal 9100 apparatus and a JASCO FT-IR spectrometer, respectively. The ¹H and ¹³C NMR spectra were recorded on a BRUKER DRX–400 and 500 AVANCE instrument with CDCl₃ as solvent at 400.1 MHz (¹H NMR), 100.6 and 125.7 MHz (¹³C NMR). Mass spectra were recorded on a Shimadzu GC/MS QP 1100 EX mass spectrometer operating at an ionization potential of 70 eV. Elemental analyses for C, H and N using a Heraeus CHN-O-Rapid analyser were carried out at the Iranian Central Research of Petroleum Company. High resolution mass spectra (HRMS) were recorded on a Waters LCT Premier XW (ESI-TOF MS) spectrometer at Research School of Chemistry, Australian National University, Canberra ACT 0200, Australia.

General synthetic procedure (exemplified by 3a)

The solution of cyclohexyl isocyanide (0.13 g, 1.2 mmol) in 3 mL of CH_2Cl_2 solvent was slowly added dropwise to a mixture of phenyl isocyanate (0.12 g, 1mmol) and DMAD (0.17 g, 1.2 mmol) in 20 mL of CH_2Cl_2 solvent for 3 min at room temperature. After the addition, the solution was refluxed at 38 °C for 24 h. Then, the solvent was removed under reduced pressure, and the solid product washed with mixture of cold diethyl ether and n-hexane with 1: 3 ratio (2×3 mL). The liquid phase was filtered off and residual recrystallized in diethyl ether.

Dimethyl 2-(cyclohexylimino)-2,5-dihydro-5-oxo-1-phenyl-1*H*-**pyrrole-3,4-dicarboxylate** (**3a**). Orange powder, yield 93%, 0.34 g, mp 64-66 °C; IR (KBr) (v_{max} , cm⁻¹): 1750 and 1718 (2C=O of ester), 1656 (C=O). ¹H NMR (400.1 MHz, CDCl₃): δ_H 1.49-1.81 (10H, m, 5CH₂), 3.78 (1H, m, NCH), 3.96 and 3.99 (6H, 2s, 2OCH₃), 7.23-7.43 (5H, m, Ar-H). ¹³C NMR (125.7 MHz, CDCl₃): δ_C 24.4, 25.5 and 33.4 (5CH₂ of cyclohexyl), 53.1 and 53.2 (2OMe), 58.6 (N-CH), 124.8, 126.6, 128.7, 136.3, 137.9, 144.0 (C_{arom} and $C=C_{pyrrole\ ring}$), 147.6 (C=N), 148.0 (C=O), 160.3 and 160.6 (2C=O of ester). MS, m/e (%) = 371 (M⁺+1, 16), 370 (M⁺, 4), 338 (100), 308 (86), 252 (12), 197 (65), 119 (45), 83 (18), 77 (48), 55 (82); HRMS (ESI): $m/z = [M+Na]^+$ Calcd for $C_{20}H_{22}N_2O_5Na$, 393.1426; Found: 393.1435.

Diethyl 2-(cyclohexylimino)-2,5-dihydro-5-oxo-1-phenyl-1*H***-pyrrole-3,4-dicarboxylate (3b)**. Yellow powder, yield: 95%, 0.38 g, mp 63-65 °C; IR (KBr) (v_{max} , cm⁻¹): 1744 and 1703 (2C=O of ester), 1660 (C=O). ¹H NMR (400.1 MHz, CDCl₃): δ_H 1.38 and 1.41 (6H, 2t, ³J_{HH} = 7.1 Hz, 2CH₃), 1.11-1.93 (10H, m, 5CH₂), 3.77 (1H, m, NCH), 4.42 and 4.44 (2q, ³J_{HH} = 7.1 Hz, 2OCH₂), 7.23-7.54 (5H, m, Ar-H). ¹³C NMR (125.7 MHz, CDCl₃): δ_C 13.9 and 14.0 (2CH₃), 24.3, 25.6 and 33.4 (5CH₂ of cyclohexyl), 58.5 (N-CH), 62.4 and 62.4 (2OCH₂), 124.9, 126.6, 128.7, 135.9, 137.7, 144.1 (C_{arom} and C=C_{pyrrole ring}), 147.8 (C=N), 148.2 (C=O), 159.8 and 160.1 (2C=O of ester); MS, *m/e* (%) = 398 (M⁺, 2), 352 (100), 325 (19), 306 (93), 301 (6), 279 (31), 252 (19), 197 (59), 119 (58), 83 (19), 77 (62), 55 (75); HRMS (ESI): *m/z* = [M+Na]⁺ Calcd for C₂₂H₂₆N₂O₅Na, 421.1739; Found: 421.1740.

Di-*t*-butyl **2-**(cyclohexylimino)-2,5-dihydro-5-oxo-1-phenyl-1*H*-pyrrole-3,4-dicarboxylate (**3c**). Orange powder, yield 88%, 0.40 g, mp 66-68 °C; IR (KBr) (ν_{max}, cm⁻¹): 1755 and 1723 (2C=O of ester), 1660 (C=O). ¹H NMR (400.1 MHz, CDCl₃): δ_H 1.60 and 1.63 (18H, 2s, 2C(CH₃)₃), 1.34-1.78 (10H, m, 5CH₂), 3.77 (1H, m, N-CH), 7.11-7.53 (5H, m, Ar-H). ¹³C NMR (125.7 MHz, CDCl₃): δ_C 24.2, 25.6 and 33.3 (5CH₂ of cyclohexyl), 28.1 and 28.6 (2C(<u>C</u>H₃)₃), 58.5 (N-CH), 81.0 and 83.0 (2O<u>C</u>Me₃), 124.8, 126.3, 127.3, 128.7, 135.5 and 137.3 (C_{arom} and C=C_{pyrrole ring}), 148.1 (C=N), 148.8 (C=O), 159.1 and 159.4 (2C=O of ester). MS, *m/e* (%) = 454 (M⁺, 4), 397 (10), 324 (40), 280 (45), 251 (13), 198 (28), 119 (19), 83 (45), 77(19), 57 (100), 55 (80), 41 (48); Anal. Calcd for C₂₆H₃₄N₂O₅ (454.56): C, 68.70; H, 7.54; N, 6.16%. Found: C, 68.75; H, 7.60; N, 6.20%.

Dimethyl 2-(*t*-butylimino)-2,5-dihydro-5-oxo-1-phenyl-1*H*-pyrrole-3,4-dicarboxylate (3d). Yellow powder, yield 95%, 0.33 g, mp 59-61 °C; IR (KBr) (v_{max} , cm⁻¹): 1752 and 1729 (2C=O of ester), 1666 (C=O). ¹H NMR (400.1 MHz, CDCl₃): δ_{H} 1.32 (9H, s, CMe₃), 3.92 and 3.95 (6H, 2s, 2OCH₃), 7.28-7.36 (5H, m, Ar-H). ¹³C NMR (100.6 MHz, CDCl₃): δ_{C} 28.9 (CMe₃), 52.1 and 52.2 (2OMe), 55.8 (N-CMe₃), 123.1, 125.2, 127.6, 134.6, 137.3, 143.2 (C_{arom} and C=C_{pyrrole ring}), 147.6 (C=N), 148.0 (C=O), 159.4 and 159.6 (2C=O of ester). MS, m/e (%) = 345 (M⁺+1, 14), 344 (M⁺, 6), 329 (100), 313(75), 256 (58), 197 (91), 119 (29), 77 (17), 57 (43); HRMS (ESI): m/z = [M+Na]⁺ Calcd for C₁₈H₂₀N₂O₅Na, 367.1270; Found: 367.1267.

Diethyl 2-(*t*-butylimino)-2,5-dihydro-5-oxo-1-phenyl-1*H*-pyrrole-3,4-dicarboxylate (3e). Yellow powder, yield 82%, 0.31 g, mp 63-65 °C; IR (KBr) (v_{max} , cm⁻¹): 1744 and 1721 (2C=O of ester), 1668 (C=O). ¹H NMR (400.1 MHz, CDCl₃): δ_H 1.28 (9H, s, CMe₃), 1.38 and 1.40 (6H, 2t, ${}^3J_{HH}$ = 7.0 Hz, 2CH₃), 4.35 and 4.39 (4H, 2q, ${}^3J_{HH}$ = 7.0 Hz, 2OCH₂), 7.19-7.40 (5H, m, Ar-H). ¹³C NMR (100.6 MHz, CDCl₃): δ_C 13.9 and 14.0 (2CH₃), 30.04 (CMe₃), 56.8 (N-CMe₃), 62.3 and 62.4 (2OCH₂), 124.3, 126.2, 128.6, 135.6, 138.0, 144.5 (C_{arom} and C=C_{pyrrole ring}), 147.4 (C=N), 151.1 (C=O), 160.3 and 160.8 (2C=O of ester). MS, *m/e* (%) = 373 (M⁺+1, 14), 372 (M⁺, 5), 357 (80), 270 (62), 243 (30), 197 (75), 119 (28), 77 (24), 57 (40); Anal. Calcd for C₂₀H₂₄N₂O₅ (372.41): C, 64.50; H, 6.50; N, 7.52%. Found: C, 64.43; H, 6.58; N, 7.44%.

Di-*t*-butyl **2-**(*t*-butylimino)-**2,5-dihydro-5-oxo-1-phenyl-1***H*-pyrrole-**3,4-dicarboxylate** (**3f**). Pale yellow powder, yield 85%, 0.36 g, mp 57-59 °C; IR (KBr) (ν_{max}, cm⁻¹): 1744 and 1719 (2C=O of ester), 1662 (C=O). ¹H NMR (400.1 MHz, CDCl₃): δ_H 1.32 (9H, s, N-C(CH₃)₃), 1.61 and 1.68 (18H, 2s, 2OC(CH₃)₃), 7.03-7.39 (5H, m, Ar-H). ¹³C NMR (100.6 MHz, CDCl₃): δ_C 27.7 (N-C(CH₃)₃), 28.1 and 30.1 (2OC(CH₃)₃), 56.4 (N-CMe₃), 82.0 and 84.2 (2OCMe₃), 124.1, 125.8, 128.6, 129.1, 134.6, 138.4 (C_{arom} and C=C_{pyrrole ring}), 148.7 (C=N), 149.8 (C=O), 159.1 and 159.5 (2C=O of ester). MS, *m/e* (%) = 429 (M⁺+1, 27), 428 (M⁺, 5), 357 (7), 301 (77), 283 (5), 242 (79), 215 (13), 119 (9), 77 (10), 57 (100); Anal. Calcd for C₂₄H₃₂N₂O₅ (428.52): C, 67.27; H, 7.53; N, 6.54%. Found: C, 67.35; H, 7.61; N, 6.65%.

Dimethyl 2-(2,6-dimethylphenylimino)-2,5-dihydro-5-oxo-1-phenyl-1*H***-pyrrole-3,4-dicarboxylate (3g)**. Yellow powder, yield 84%, 0.33 g, mp 107-109 °C; IR (KBr) (v_{max} , cm⁻¹): 1745 and 1688 (2C=O of ester), 1594 (C=O). ¹H NMR (400.1 MHz, CDCl₃): δ_H 2.13 (6H, s, 2CH₃), 3.99 and 4.02 (6H, 2s, 2OCH₃), 6.95-7.36 (8H, m, 2Ar-H). ¹³C NMR (100.7 MHz,

CDCl₃): δ_C 17.8 (2CH₃), 53.3 and 53.4 (2OCH₃), 118.1, 121.7, 122.2, 123.8, 127.6, 128.8, 128.9, 130.0 (C_{arom}), 136.9 and 139.8 (C=C_{pyrrole ring}), 148.8 (C=N), 152.6 (C=O), 158.8 and 159.9 (2C=O of ester). MS, m/e (%) = 393 (M⁺+1, 16), 392 (M⁺, 62), 377 (5), 361 (8), 301 (51), 274 (6), 212 (64), 167 (6), 119 (17), 93 (100), 91 (14), 77 (61), 59 (8); HRMS (ESI): $m/z = [M+Na]^+$ Calcd for $C_{22}H_{20}N_2O_5Na$, 415.1270; Found: 415.1271.

Di-*t*-butyl **2-**(2,6-dimethylphenylimino)-2,5-dihydro-5-oxo-1-phenyl-1*H*-pyrrole-3,4-dicarboxylate (3h). Red powder, yield 81%, 0.39 g, mp 92-94 °C; IR (KBr) (v_{max} , cm⁻¹): 1735 and 1696 (2C=O of ester), 1594 (C=O). ¹H NMR (400.1 MHz, CDCl₃): δ_H 1.47 and 1.51 (18H, 2s, 2C(CH₃)₃), 2.25 (6H, s, 2CH₃), 7.01-7.61 (8H, m, 2Ar-H). ¹³C NMR (100.6 MHz, CDCl₃): δ_C 18.3 (2CH₃), 27.3 and 27.5 (2C(<u>C</u>H₃)₃), 82.5 and 82.7 (2O<u>C</u>Me₃), 118.1, 119.4, 121.7, 123.6, 126.8, 127.7, 128.8, 130.8 (C_{arom}), 135.5 and 139.8 (C=C_{pyrrole ring}), 150.0 (C=N), 152.6 (C=O), 160.2 and 161.3 (2C=O of ester). MS, m/e (%) = 477 (M⁺+1, 7), 476 (M⁺, 20), 420 (6), 375 (6), 347 (13), 275 (14), 245 (13), 212 (37), 119 (12), 105 (10), 93 (100), 91 (12), 77 (33), 57 (49); HRMS (ESI): $m/z = [M+Na]^+$ Calcd for C₂₈H₃₂N₂O₅Na, 499.2209; Found: 499.2208.

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References

- 1. Swinbourne, J. F.; Hunt, H. J.; Klinkert, G. Adv. Heterocycl. Chem. 1987, 23, 103.
- 2. (a) Domling, A.; Ugi, I. *Angew. Chem., Int. Ed.* **2000**, *39*, 3168. (b) Ugi, I.; Werner, B.; Domling, A. *Molecules* **2003**, *8*, 53.
- 3. Ullmanns Encycl. Tech. Chem. 1957, 9, 1.
- 4. Ozaki, S. Chem. Rev. 1972, 72, 457.
- 5. L'abbé, G. Synthesis 1987, 525.
- 6. Ulrich, H. Cycloaddition Reactions of Heterocumulenes; Academic: New York: London, 1967; pp 122.
- 7. Ulrich, H. Acc. Chem. Res. 1969, 2, 186.
- 8. Winterfeldt, E.; Schumann, D.; Dillinger, H. J. Chem. Ber. 1969, 102, 1656.
- 9. Oakes, T. R.; David, H. G.; Nagel, F. J. Am. Chem. Soc. 1969, 91, 4761.
- 10. Takisawa, T.; Obata, N.; Suzuki, Y.; Yanagida, T. Tetrahedron Lett. 1969, 10, 3407.
- 11. Ugi, I. Isonitrile Chemistry; Academic: London, 1971.
- 12. Ugi, I. Angew. Chem., Int. Ed. Engl. 1982, 21, 810.
- 13. Yavari, I.; Maghsoodlou, M. T. J. Chem. Res. 1998, 386.

- 14. Nair, V.; Rajesh, C.; Vinod, A. U.; Bindu, S.; Sreekanth, A. R.; Mathen J. S.; Balagopal, L. *Acc. Chem. Res.* **2003**, *36*, 899.
- 15. Shaabani, A.; Hossein-Rezayan, A.; Ghasemi, S.; Sarvary, A. Tetrahedron Lett. 2009, 50, 1456.
- 16. Teimouri, M. B.; Mansouri, F.; Bazhrang, R. Tetrahedron 2010, 66, 259.
- 17. Alizadeh, A.; Hosseinpour, R. Synthesis 2009, 16, 2733.
- 18. Shaabani, A.; Sarvary, A.; Hossein-Rezayan, A.; Keshipour, S. Tetrahedron 2009, 65, 1.
- 19. Alizadeh, A.; Rostamnia, S.; Zohreh, N.; Bijanzadeh, H. Monatsh Chem. 2008, 139, 49.
- 20. Maghsoodlou, M. T.; Hazeri, N.; Habibi-Khorassani, S. M.; Marandi, G.; Nassiri, M. *Synth. Commun.* **2005**, *35*, 2771.
- 21. Maghsoodlou, M. T.; Hazeri, N.; Habibi-Khorassani, S. M.; Marandi, G.; Nassiri, M. *J. Heterocyclic Chem.* **2006**, *43*, 481.
- 22. Maghsoodlou, M. T.; Habibi-Khorasani, S. M.; Hazeri, N.; Heydari, R.; Marandi, G.; Nassiri, M. J. Chem. Res. 2006, 220.
- 23. Maghsoodlou, M. T.; Hazeri, N.; Habibi-Khorassani, S. M.; Solimani, V.; Marandi, G.; Razmjoo, Z. *J. Chem. Res.* **2008**, 198.
- 24. Hazeri, N.; Maghsoodlou, M. T.; Habibi-Korassani, S. M.; Marandi, G.; Khandan-Barani, K.; Ziyadini, M.; Aminkhani, A. *Arkivoc* **2007**, (*i*), 173.
- 25. Hazeri, N.; Maghsoodlou, M. T.; Habibi-Korassani, S. M.; Ziyadini, M.; Marandi, G.; Khandan-Barani, K.; Bijanzadeh, H. R. *Arkivoc* **2007**, (*xiii*), 34.
- 26. Maghsoodlou, M. T.; Marandi, G.; Hazeri, N.; Habibi-Korassani, S. M.; Mirzaei, A. A. *Mol. Divers* **2010**, *14*, 1.
- 27. Marandi, G.; Maghsoodlou, M. T.; Hazeri, N.; Heydari. R.; Habibi-Korassani, S. M.; Ebrahimi, A.; Poor, S. M.; Mahdiabad, H. H.; Nassiri, M.; Kabiri, R. *Heteroatom. Chem.* **2010**, *21*, 228.