Effective isomerization of 3',5'-O-(tetraisopropyldisiloxane-1,3diyl)nucleosides in the presence of trimethylsilyl trifluoromethanesulfonate

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Dedicated to Prof. Harri Lönnberg on his 60th birthday

Abstract

Trimethylsilyl trifluoromethanesulfonate catalyze effective isomerization of 3',5'-O-(tetraisopropyldisiloxane-1,3-diyl)nucleosides (1) in 1,2-dicloroethane at 0°C into 2',3'-O-(tetraisopropyldisiloxane-1,3-diyl)-derivatives (2), which can be obtained in 55-90% yields. On the other hand nucleosides 1 were found to be stable in the presence of tin tetrachloride. Only in the case of uridine derivative 1a a substantial amount of isomerization product 2a was formed.

Keywords: nucleoside protection, isomerization, 3',5'-O-(tetraisopropyldisiloxane-1,3-diyl)nucleosides, trimethylsilyl trifluoromethanesulfonate

Introduction

Tetraisopropyldisiloxane-1,3-diyl (TIPDS) group developed by Markiewicz^{1,2} is widely used for simultaneous protection of 3',5'-hydroxyl groups in ribonucleosides and subsequent manipulation with 2'-OH group of ribonucleosides: deoxygenation,^{3,4} oxidation,⁵ alkylation,⁶⁻¹⁰ glycosylation,¹¹⁻¹⁶ protection,¹⁷⁻²¹ preparation of 2'-amino-2'-deoxynuclesides,^{22,23} and so on. This group may be considered as one of the most popular and useful protecting group in the field of nucleoside chemistry. It is believed that the reaction of 1,3-dichloro-tetraisopropyldisiloxane with ribonucleoside starts with silylation of primary 5'-hydroxyl followed by the formation of an 8-membered ring. It was also shown that the reaction with 5'-O-protected ribonucleosides resulted in the formation of 2',3'-O-derivatives with 7 membered ring.^{1,24,25} It should be also mentioned that simultaneous protection of both 3'- and 5'-hydroxyls can be carried out with *N*-unprotected ribonucleosides and with base-protected N-acyl derivatives. TIPDS group is removed with fluoride ion, usually with Bu₄NF.3H₂O in tetrahydrofuran.²⁶

Results and Discussion

In the course of our studies²⁷⁻³⁰ on the synthesis of disaccharide nucleosides we have developed an efficient and simple synthesis of 2'-*O*- β -D-ribofuranosylnucleosides. The method consists of the condensation of a small excess of 1-*O*-acetyl-2,3,5-tri-*O*-benzoyl- β -D-ribofuranose with Nprotected 3',5'-*O*-TIPDS-nucleosides in 1,2-dichloroethane in the presence of tin tetrachloride. This reaction was carried out under mild conditions (0°C, 1,2-dichloroethane, 2 h for pyrimidine nucleosides, 7-16 h for purine derivatives) and the yields of the target compounds were 74 -82%. The presence of a participating 2-*O*-benzoyl group in sugar moiety leads exclusively to 1,2-trans-ribofuranosides. To study the broad applicability of the method, some other sugars such as fully acylated D-(and L)-arabinofuranose, D-ribopyranose and D-erythrofuranose were used in the *O*-glycosylation reaction.²⁷⁻³⁰

Ribosylation of pyrimidine 2',3'-di-O-acylnucleosides under the same conditions gave the desired disaccharide nucleosides in 74-78% yield.^{31,32} Analogous results were obtained, when the tin tetrachloride catalyst was substituted by an equal amount of trimethylsilyl trifluoromethanesulfonate (TMSOTf).^{31,32} Encouraged by these results we decided to examine the use of this catalyst in the preparation of $2'-O-\beta$ -D-pentafuranosylnucleosides. This catalyst is widely used in nucleoside chemistry and has some important advantages over tin tetrachloride.³³ Condensation of a small excess of 1-O-acetyl-2,3,5-tri-O-benzoyl-B-D-ribofuranose with 3',5'-O-TIPDS-uridine (1a) at 0°C in 1,2-dichloroethane in the presence of TMSOTf gave a mixture of two compounds. The main product was identical with the known blocked 2'-O-ribofuranosylnucleoside^{11,34} and the minor product was tentatively assigned as corresponding 5'-Oribofuranosyl isomer (3a). Analogous condensation of N^6 -benzoyl-3',5'-O-TIPDS-adenosine 1b with the same sugar gave 5'-O-ribofuranosyl derivative **3b** as a main product (up to 50% yield). The formation of 5'-O-ribofuranosylnucleosides may be explained by the isomerization of the TIPDS group from 3',5'- to 2',3'-positions in the presence of TMSOTf. It has been reported that condensation of 1-O-acetyl-2,3,5-tri-O-benzoyl-B-D-ribofuranose with 3',5'-O-TIPDS-uridine (1a) in 1,2-dichloroethane in the presence of tin tetrachloride at 20° for 0.5 hr resulted in a mixture of desired 2'-O-ribofuranosyluridine and 5'-O-ribofuranosyluridine derivatives (3a) in a ratio 4:1.¹² The structure of **3a** was proved by independent synthesis starting from 2',3'-O-TIPDS-uridine $(2a)^{12}$ which was prepared by reaction of 5'-O-monomethoxytrityluridine with 1,3-dichloro-tetraisopropyldisiloxane with the following removal of trityl group.¹



Scheme 1. TIPDS group migration in the presence of TMSOTf at 0°C in 1,2-dicloroethane under nitrogen for 1,5-4 hrs. **a** B=Ura, **b** B=Ade^{Bz}, **c** B=Cyt^{Bz}, **d** B=Gua^{iBu}, **e** B=Ade , **f** B=Cyt. (i) TMSOTf, 0°C, DCE, N₂; (ii) 1-*O*-acetyl-2,3,5-tri-*O*-benzoyl- β -D-ribofuranose, 20 hrs at 0°C; (iii) Bu₄NF/THF, 20 min at 20°C and 5 M NH₃/MeOH 3 days at 20°C.

Previously it was shown that under anhydrous acidic conditions (mesitylenesulphonic acid, DMF, 20°C, 6-10 hours, 30-60% yield) 3',5'-O-TIPDS-nucleosides undergo the transformation from the thermodynamically less stable eight-membered ring to the more stable seven-membered ring 2',3'-O-TIPDS-nucleosides.³⁵ It was shown that isomerization may occur during refluxing in chloroform in the presence of p-toluenesulfonic acid. ³⁶ To study this reaction in detail we investigated the isomerization of 3',5'-O-TIPDS-nucleosides (**1a-f**) in the presence of TMSOTf. Nucleosides **1a-f** were treated with excess of TMSOTf in 1,2-dichloethane at 0°C under nitrogen atmosphere. Using 2-3 equivalents of TMSOTf the isomerization was rather fast and was completed in 1.5 to 4 hr. Corresponding 2',3'-O-TIPDS derivatives (**2 a-f**) were obtained in 55-90% yields (Table 1). It should be mentioned that in the case of purine derivatives the yields were much higher, up to 90%. Uridine derivative exhibited substantial decomposition.

В	equivalents	Yield of	Reaction	$R_{\rm f}$ of 1 in	$R_{\rm f}$ of 1 in	$R_{\rm f}$ of 2 in	$R_{\rm f}$ of 2 in
	of	2, %	time in hr	system A	system B	system A	system B
	TMSOTf						
a B=Ura	2	55	1.5	0.35		0.24	
b B=Ade ^{Bz}	3	90	3.0	0.36		0.41	
c B=Cyt ^{Bz}	2	60	4.0	0.34		0.43	
d B=Gua ^{iBu}	3	74	2.0	0.30		0.28	
e B=Ade	3	72	4.0		0.28		0.29
f B=Cyt	3	55	2.0		0.16		0.12

Table 1. Isomerization of 1 in the presence of TMSOTf in 1,2-dichloethane at 0° C, isolated yields of 2 and R_f values of 1 and 2

System A CH₂Cl₂ – EtOH 98:2; system B CH₂Cl₂ – EtOH 95:5.

The structures of **2a-f** were proved by ¹H and ¹³C NMR spectra and compared with NMR spectra of the starting 3',5'-O-TIPDS-nucleosides **1a-f**. The signal of primary 5'-hydroxyl groups

(see also supplementary data) in nucleosides **2a-f** in DMSO-*d*₆ appears as a doublet of doublets or a triplet at 5.15-5.33 ppm due to the coupling with both protons of the CH₂-group. The secondary 2'-hydroxyl groups in **1a-f** are located in lower field at 5.60-5.69 ppm with $J_{2',OH} = 3.9$ - 4.8 Hz. The coupling constants $J_{1',2'}$ in 3',5'-*O*-TIPDS-nucleosides **1a-f** are rather small,³⁷ on the contrary $J_{3',4'}$ is around 8 Hz. The sum of $J_{1',2'}$ and $J_{3',4'}$ for both nucleoside series **1a-f** and **2a-f** is nearly the same, 9.0-9.6 Hz.³⁸ TIPDS-group shifts neighboring protons to the lower field (0.3-0.6 ppm). The same tendency was observed in the ¹³C NMR spectra for C-2' (1.4-2.5 ppm) and C-5' (0.05-0.7 ppm). Moreover the signals of C-4' in compounds **1** are located in higher fields than the corresponding signals in nucleosides **2** and this difference is most pronounced in purine nucleosides (around 5.0 ppm). It should be mentioned that compounds **2a**¹, **2f**³⁶ and **2a,e,f**³⁵ were prepared earliar but were not completely characterized.

The influence of temperature, solvent, and catalyst on the isomerisation reactions was further investigated (Table 2). The isomerization of **1** in the presence of 1 equivalent of TMSOTf in 1,2-dichloroethane is much slower (entries 1-3) than in the presence of excess of catalyst. When the reaction is conducted using 3 equivalents of TMSOTf at 0 °C for 24 hrs a substantial amount of decomposition products (18%, entry 1) with lower $R_f = 0.25-0.15$ (system A) is formed which can be identified as 3'- and 2'-O-TIPDS-derivatives. It is known that treatment of 3',5'-O-TIPDS-nucleosides (**1**) with acids results in the formation of a mixture of 3'-O-TIPDS-derivative and its 2'-O-TIPDS-isomerization product.¹ The isomerization of **1** in acetonitrile is much slower than in 1,2-dichloroethane (entries 4 and 5). The conversion of $1 \rightarrow 2$ may be performed in the presence of trifluoromethanesulfonate (TfOH) (entries 6-8) or boron trifluoride etherate (entries 9 and 10) but these reactions are rather slower and accompanied by substantial decomposition.

We have also investigated the isomerization of **1** in the presence of tin tetrachloride, which is widely used in glycosylation reactions.²⁸⁻³⁰ From Table 2 it can be seen that only in the case of uridine derivative **1a** a substantial amount of **2a** and decomposition products were formed and the isomerization is faster at room temperature (entries 11-13). These results confirmed the above mentioned instability of uridine derivative **1a** in the presence of this catalyst.¹² With other tested nucleosides **1b-1f** only traces of products **2b-2f** may be detected using TLC (entries 14-19).

Entry	Substrate	Conditions	2, %	1, %	Decomp.%
1	1b	3 eq TMSOTf, DCE, 0°C, 24 hrs	80	2	18
2	1b	2 eq TMSOTf, DCE, 0°C, 24 hrs	85	2	13
3	1b	1 eq TMSOTf, DCE, 0°C, 24 hrs	50	45	5
4	1a	1 eq TMSOTf, MeCN, 20°C, 24 hrs	60	15	25
5	1b	1 eq TMSOTf, MeCN, 20°C, 24 hrs	50	45	5
6	1a	1 eq TfOH, DCE, 20°C, 3 hrs	30	55	15
7	1a	1 eq TfOH, DCE 20°C, 24 hrs	15	10	75
8	1c	1 eq TfOH, DCE, 20°C, 24 hrs	10	50	40
9	1a	1 eq BF ₃ ·EtOEt, DCE, 0°C, 24 hrs	7	75	18
10	1b	1 eq BF ₃ ·EtOEt, DCE, 0°C, 24 hrs	0	80	20
11	1a	1 eq SnCl ₄ , DCE, 0°C, 24 hrs	35	60	5
12	1a	1 eq SnCl ₄ , DCE, 20°C, 24 hrs	70	20	10
13	1a	1 eq SnCl ₄ , CH ₃ CN, 20°C, 24 hrs	65	20	15
14	1b	1 eq SnCl ₄ , DCE, 0°C, 20°C, 24 hrs	2	95	3
15	1b	1 eq SnCl ₄ , MeCN, 20°C, 24 hrs	5	90	5
16	1c	1 eq SnCl ₄ , DCE, 0°C, 20°C, 24 hrs	0	97	3
17	1d	1 eq SnCl ₄ , DCE, 0°C, 20°C, 24 hrs	2	95	3
18	1e	1 eq SnCl ₄ , DCE, 0°C, 20°C, 24 hrs	0	97	3
19	1f	1 eq SnCl ₄ , DCE, 0°C, 20°C, 24 hrs	0	97	3

Table 2. Isomerization of **1** in the presence of different catalysts

The ratio of starting 1, product 2 and decomposition products were determined using TLC and PMR spectroscopy.

The 2',3'-O-TIPDS-derivatives **2** may be used for the preparation 5'-substituted nucleosides. Thus **1b** was treated with 2 equivalents of TMSOTf in 1,2-dichloroethane at 0° C. After 12 hrs when isomerization according to TLC was complete 1-O-acetyl-2,3,5-tri-O-benzoyl- β -D-ribofuranose was added and the reaction mixture was kept at 0° C for 20 hr. Disaccharide nucleoside **3b** was obtained in 60 % yield, further deblocking gave the known 5'-O- β -D-ribofuranosylafenosine **4b**.³²

Conclusions

Trimethylsilyl trifluoromethanesulfonate catalyzes effective isomerization of 3',5'-O-TIPDS nucleosides (1) in 1,2-dicloroethane at 0°C into 2',3'-O-TIPDS-derivatives (2), which can be obtained in 55-90% yields. This reaction is in line with the proposed mechanism³⁶ of acid-catalysed cleavage of the silyl ether bond formed by primary 5'-hydroxyl group followed by

formation of 2',3'-O-TIPDS-nucleosides. As a result the thermodynamically less stable eightmembered ring transformed to the more stable seven-membered ring. On the other hand 3',5'-O-TIPDS-nucleosides (1) were found to be stable in the presence of tin tetrachloride. Only in the case of uridine derivative 1a substantial amount of isomerization product 2a was formed.

Experimental Section

General Procedures. Column chromatography was performed on silica gel Kieselgel 60 (0.063-0.200 mm, Merck). (0.040-0.063 mm), TLC was carried out on Alugram SIL G/UV₂₅₄ (Macherey-Nagel) with detection by UV and the following solvent systems (compositions expressed as v/v): methylene chloride – ethanol 98:2 (A); methylene chloride – ethanol 95:5 (B), detection by UV light. NMR Spectra: Bruker AMX 400 NMR spectrometer and Bruker Avance 300 NMR spectrometer; at 27°C. Chemical shifts δ in ppm were measured relative to the solvent signals (¹H and ¹³C). ¹³C NMR spectra were measured with proton decoupling and all carbon signals appeared as singlets. The coupling constants (J) are given in Hz. The signals were assigned using double resonance techniques. Mass spectrometry and exact mass measurements of the nucleoside intermediates were performed on a quadrupole/othogonal-acceleration time-of-flight tandem mass spectrometer (Q-Tof-2, Micromass, Manchester, UK) equipped with a standard electrospray ionization (ESI) interface.

3',5'-*O***-(1,1,3,3-Tetraisopropyldisiloxane-1,3-diyl)nucleosides (1)** were prepared according to literature.^{1,2}

3',5'-O-(1,1,3,3-Tetraisopropyldisiloxane-1,3-diyl)uridine (1a). R_f 0.35 (A). ¹H-NMR (300 MHz, DMSO-d₆): 11.36 brs (1H, NH), 7.68 d (1H, $J_{6,5} = 8.0$, H-6), 5.58 d (1H, $J_{2'-OH, 2'} = 4.4, 2'$ -OH), 5.54 d (1H, $J_{1',2'} = 0.6$, H-1'), 5.52 d (1H, $J_{5,6} = 8.0$, H-5), 4.16 dd (1H, $J_{3',2'} = 4.7, J_{3',4'} = 8.3$, H-3'), 4.13 m (2H, H-2', H-5'a), 3.98 ddd (1H, $J_{4',3'} = 8.3, J_{4',5'a} = 2.2, J_{4',5'b} = 2.5, H-4'$), 3.91 dd (1H, $J_{5'b,4'} = 2.5, J_{5'a,5'b} = -13.0, H-5'b$), 1.18-0.75 m (28H, iPr). ¹³C-NMR (75 MHz, DMSO-d₆): 163.20 (C-4); 150.14 (C-2); 139.80 (C-6); 100.95 (C-5); 90.52 (C-1'); 80.88 (C-4'); 73.52 (C-2'); 68.79 (C-3'); 60.22 (C-5'); 17.37, 17.24, 17.19, 17.12, 16.96, 16.87, 16.84, 16.78, 12.74, 12.37, 12.32, 11.95 (iPr).

*N*⁶-Benzoyl-3',5'-*O*-(1,1,3,3-Tetraisopropyldisiloxane-1,3-diyl)adenosine (1b). *R*_f 0.36 (A). ¹H-NMR (300 MHz, DMSO-d₆): 11.22 brs (1H, NH), 8.65 s (1H, H-8), 8.52 s (1H, H-2), 8.02 d (2H, *J* = 7.2, *o*-Bz), 7.63 t (1H, *J* = 7.2, *p*-Bz), 7.54 t (2H, *J* = 7.2, *m*-Bz), 5.99 d (1H, *J*_{1',2'} = 1.0, H-1'), 5.67 d (1H, *J*_{2'-OH, 2'} = 4.6, 2'-OH), 4.81 dd (1H, *J*_{3',2'} = 5.2, *J*_{3',4'} = 8.2, H-3'), 4.63 ddd (1H, *J*_{2',1'} = 1.0, *J*_{2',2'-OH} = 4.6, *J*_{2',3'} = 5.2, H-2'), 4.03 m (2H, H-4', H-5'a), 3.93 dd (1H, *J*_{5'b,4'} = 2.5, *J*_{5'a,5'b} = -12.5, H-5'b), 1.16-0.82 m (28H, iPr). ¹³C-NMR (75 MHz, DMSO-d₆): 165.67 (C=O), 151.43 (C-2), 151.43 (C-6), 150.53 (C-4), 143.08 (C-8), 133.38, 132.45, 128.50, 128.45 (Ph), 125.98 (C-5), 89.54 (C-1'), 80.97 (C-4'), 73.40 (C-2'), 69.88 (C-3'), 60.74 (C-5'), 17.35, 17.19, 17.16, 17.04, 16.92, 16.84, 12.76, 12.46, 12.27, 12.07 (iPr).

*N*⁴-Benzoyl-3',5'-*O*-(1,1,3,3-tetraisopropyldisiloxane-1,3-diyl)cytidine (1c). R_f 0.34 (A). ¹H-NMR (300 MHz, DMSO-d₆): 11.27 brs (1H, NH), 8.21 d (1H, $J_{6,5} = 7.5$, H-6), 8.00 d (2H, J = 7.2, *o*-Bz), 7.63 t (1H, J = 7.2, *p*-Bz), 7.52 t (2H, J = 7.2, *m*-Bz), 7.37 d (1H, $J_{5,6} = 7.5$, H-5), 5.80 d (1H, $J_{2'-OH, 2'} = 3.9$, 2'-OH), 5.64 s (1H, H-1'), 4.24 d (1H, $J_{5'a,5'b} = -13.2$, H-5'a), 4.11 m (3H, H-2', H-3', H-4'), 3.95 d (1H, $J_{5'b,5a'} = -13.2$, H-5'b), 1.15-0.88 m (28H, iPr). ¹³C-NMR(75 MHz, DMSO-d₆): 167.30 (C=O); 163.17 (C-4); 154.15 (C-2); 143.72 (C-6); 132.98, 132.75, 128.44, (Ph); 95.59 (C-5); 91.30 (C-1'); 80.91 (C-4'); 73.81 (C-2'); 68.05 (C-3'); 59.80 (C-5'); 17.39, 17.26, 17.20, 17.12, 16.91, 16.87, 16.78, 16.76, 12.71, 12.49, 12.35, 11.92 (iPr).

 N^2 -Isobutyryl-3',5'-*O*-(1,1,3,3-tetraisopropyldisiloxane-1,3-diyl)guanosine (1d). R_f 0.30 (A). ¹H-NMR (300 MHz, DMSO-d₆): 12.12 brs (1H, NH-1), 11.74 brs (1H, NH-2), 8.05 s (1H, H-8), 5.79 d (1H, $J_{1',2'}$ = 1.1, H-1'), 5.69 d (1H, $J_{2'-OH,2'}$ = 4.8, 2'-OH), 4.35 dd (1H, $J_{3',2'}$ = 4.9, $J_{3',4'}$ = 7.9, H-3'), 4.31 ddd (1H, $J_{2',1'}$ = 1.1, $J_{2',2'-OH}$ = 4.8, $J_{2',3'}$ = 4.9, H-2'), 4.12 dd (1H, $J_{5'a,4'}$ = 3.0, $J_{5'a,5'b}$ = -12.9, H-5'a), 4.04 ddd (1H, $J_{4',3'}$ = 7.9, $J_{4',5'a}$ = 3.0, $J_{4',5'b}$ = 2.6, H-4'), 3.95 dd (1H, $J_{5'b,4'}$ = 2.6, $J_{5'b,5'a}$ = -12.9, H-5'b), 2.78 sept (1H, J = 6.9, iBu), 1.12 d (6H, J = 6.9, iBu), 1.08-0.80 m (28H, iPr). ¹³C- NMR (75 MHz, DMSO-d₆): 180.21 (C=O), 154.83 (C-6), 148.28 (C-2), 148.01 (C-4), 136.22 (C-8), 120.39 (C-5), 88.17 (C-1'), 81.22 (C-4'), 74.01 (C-2'), 69.51 (C-3'), 60.61 (C-5'), 34.72 (iBu), 18.88, 18.83 (iBu) 17.36, 17.20, 17.15, 16.97, 16.90, 16.85, 16.80, 12.74, 12.44, 12.32, 12.02 (iPr).

3',5'-O-(1,1,3,3-Tetraisopropyldisiloxane-1,3-diyl)adenosine (1e). R_f 0.28 (B). ¹H-NMR (400 MHz, DMSO-d₆): 8.21 s (1H, H-8), 8.08 s (1H, H-2), 7.33 brs (2H, NH₂), 5.87 s (1H, H-1'), 5.62 d (1H, $J_{2'OH,2'}$ = 4.6, 2'-OH), 4.80 dd (1H, $J_{3',2'}$ = 5.1, $J_{3',4'}$ = 8.3, H-3'), 4.52 dd (1H, $J_{2',2'OH}$ = 4.6, $J_{2',3'}$ = 5.1, H-2'), 4.06 dd (1H, $J_{5'a,4'}$ = 3.1, $J_{5'a,5'b}$ = -12.3, H-5'a), 3.99 ddd (1H, $J_{4',3'}$ = 8.3, $J_{4',5'a}$ = 3.1, $J_{4',5'b}$ = 2.1, H-4'), 3.93 dd (1H, $J_{5'b,4'}$ = 2.1, $J_{5'b,5'a}$ = -12.3, H-5'b), 1.12-0.96 m (28H, iPr). ¹³C-NMR (100 MHz, DMSO-d₆): 156.09 (C-6), 152.48 (C-8), 148.62 (C-4), 139.20 (C-2), 119.25 (C-5), 89.33 (C-1'), 80.75 (C-4'), 73.64 (C-2'), 69.81 (C-3'), 60.80 (C-5'), 17.36, 17.23, 17.16, 17.00, 16.91, 16.82, 12.73, 12.44, 12.23, 12.06 (iPr).

3',5'-O-(1,1,3,3-Tetraisopropyldisiloxane-1,3-diyl)cytidine (1f). R_f 0.16 (B). ¹H-NMR (400 MHz, DMSO-d₆): 7.70 d (1H, $J_{6,5} = 7.5$, H-6), 7.16 brs (1H, NH₂), 7.07 brs (1H, NH₂), 5.68 d (1H, $J_{5,6} = 7.5$, H-5), 5.57 d (1H, $J_{2'-OH, 2'} = 4.4$, 2'-OH), 5.55 s (1H, H-1'), 4.15 dd (1H, $J_{5'a,4'} = 1.0$, $J_{5'a,5'b} = -12.8$, H-5'a), 4.08 dd (1H, $J_{3',2'} = 4.4$, $J_{3',4'} = 9.0$, H-3'), 4.00 ddd (1H, $J_{4',3'} = 9.0$, $J_{4',5'a} = 1.0$, $J_{4',5'b} = 2.5$, H-4'), 3.93 t (1H, $J_{2',2'-OH} = J_{2',3'} = 4.4$, H-2'), 3.91 dd (1H, $J_{5'b,4'} = 2.5$, $J_{5'b,5'a} = -12.8$, H-5'b), 1.08-0.92 m (28H, iPr). ¹³C- NMR (100 MHz, DMSO-d₆): 165.69 (C-4), 154.81 (C-2), 139.91 (C-6), 93.27 (C-5), 90.72 (C-1'), 81.51 (C-4'), 74.11 (C-2'), 68.52 (C-3'), 60.07 (C-5'), 17.37, 17.24, 17.18, 17.11, 16.84, 16.76, 12.77, 12.46, 12.37, 11.97 (iPr).

2',3'-O-(1,1,3,3-Tetraisopropyldisiloxane-1,3-diyl)uridine (2a). To a cool solution (0°) of nucleoside **1a** (620 mg, 1.28 mmol) in 1,2-dichloroethane (20 ml) 2M solution of trimethylsilyl trifluoromethanesulfonate in 1,2-dichloroethane (1.28 ml, 2.56 mmol) was added under nitrogen atmosphere. The reaction mixture was stirred at 0°C for 1.5 hrs. Methylene chloride (50 ml) and 10% aqueous solution of sodium bicarbonate (20 ml) were added and the suspension was stirred at 0°C for 15 min. The organic layer was separated, washed with water (20 ml), dried over

anhydrous sodium sulfate, evaporated in vacuo and purified by column chromatography on silica gel (30 g). The column was washed with methylene chloride (300 ml), and then eluted with methylene chloride - ethanol 98:2. Fractions containing the product were evaporated and dried to give **2a** as a white amorphous powder. Yield 330 mg (55 %). R_f 0.24 (A). ¹H-NMR (300 MHz, DMSO-d_6): 11.33 s (1H, NH), 7.95 d (1H, $J_{6,5} = 8.1$, H-6), 5.79 d (1H, $J_{1',2'} = 4.1$, H-1'), 5.64 d (1H, $J_{5,6} = 8.1$, H-5), 5.25 t (1H, $J_{5'.OH, 5'a} = J_{5'.OH, 5'b} = 5.0$, 5'-OH), 4.52 dd (1H, $J_{2',1'} = 4.1$, $J_{2',3'} = 4.4$, H-2'), 4.37 dd (1H, $J_{3',2'} = 4.4$, $J_{3',4'} = 5.0$, H-3'), 3.94 ddd (1H, $J_{4',3'} = 5.0$, $J_{4',5'a} = 3.3$, $J_{4',5'b} = 3.0$, H-4'), 3.71 ddd (1H, $J_{5'a,4'} = 3.3$, $J_{5'a,5'-OH} = 5.0$, $J_{5'a,5'b} = -12.2$, H-5'a), 3.58 ddd (1H, $J_{5'b,4'} = 3.0$, $J_{5'b,5'-OH} = 5.0$, $J_{5'b,5'a} = -12.2$, H-5'b), 1.12-0.87 m (28H, iPr). ¹³C-NMR (75 MHz, DMSO-d_6): 163.62 (C-4); 150.95 (C-2); 140.57 (C-6); 102.10 (C-5); 88.94 (C-1'); 85.00 (C-4'); 76.18 (C-2'); 71.87 (C-3'); 60.22 (C-5'); 17.73, 17.58, 17.54, 17.45, 17.39, 17.28, 12.95, 12.69, 12.42 (iPr). LSI-MS: (C_{21}H_{38}N_2O_7Si_2 - H⁺) found 485.2140. Calc. 485.2139.

 N^{6} -Benzoyl-2',3'-O-(1,1,3,3-Tetraisopropyldisiloxane-1,3-diyl)adenosine (2b). Analogous conversion of 1b (620 mg, 1.01 mmol) was carried out in the presence of trimethylsilyl trifluoromethanesulfonate (2M solution in 1,2-dichloroethane, 1.51 ml, 3.03 mmol) in 1,2dichloroethane (20 ml) at 0° for 3 hrs. After column chromatography on silica gel (30 g) 2b was obtained as a white powder. The column was washed with methylene chloride (300 ml), and then eluted with methylene chloride - ethanol 99:1. Fractions containing the product were evaporated and dried to give **2b** as a white amorphous powder. Yield 558 mg (90 %). R_f 0.41 (A). 1H-NMR (300 MHz, DMSO-d₆): 11.23 s (1H, NH), 8.78 s (1H, H-8), 8.76 s (1H, H-2), 8.05 d (2H, J = 7.2, o-Bz), 7.66 t (1H, J = 7.2, p-Bz), 7.57 t (2H, J = 7.2, m-Bz), 6.13 d (1H, $J_{1',2'} = 5.7$, H-1'), 5.25 dd (1H, $J_{2',3'}$ = 4.9, H-2'), 5.15 t (1H, $J_{5'-OH, 5'a}$ = $J_{5'-OH, 5'b}$ = 5.2, 5'-OH), 4.68 dd (1H, $J_{3',2'}$ = 4.9, $J_{3',4'}$ = 4.0, H-3'), 4.09 ddd (1H, $J_{4',3'}$ = 4.0, $J_{4',5'a}$ = 4.8, $J_{4',5'b}$ = 4.0, H-4'), 3.73 ddd (1H, $J_{5'a,4'}$ =4.8, $J_{5'a,5'-OH} = 5.2$, $J_{5'a,5'b} = -12.0$, H-5'a), 3.62 ddd (1H, $J_{5'b,4'} = 4.0$, $J_{5'b,5'-OH} = 5.2$, $J_{5'b,5'a} = -12.0$, H-5'b), 1.17-0.95 m (28H, iPr). ¹³C- NMR (75 MHz, DMSO-d₆): 166.09 (C=O), 152.62 (C-2), 152.23 (C-6), 150.96 (C-4), 143.39 (C-8), 133.75, 132.96, 128.95, 128.92 (Ph), 126.29 (C-5), 88.04 (C-1'), 86.26 (C-4'), 76.07 (C-2'), 72.82 (C-3'), 61.16 (C-5'), 17.55, 17.49, 17.40, 17.24, 17.13, 13.11, 12.81, 12.71, 12.45 (iPr). LSI-MS: $(C_{29}H_{43}N_5O_6Si_2 + H^+)$ found 614.2828. Calc. 614.2830.

*N*⁴-Benzoyl-2',3'-*O*-(1,1,3,3-Tetraisopropyldisiloxane-1,3-diyl)cytidine (2c). Analogous conversion of 1c (564 mg, 0.96 mmol) was carried out in the presence of trimethylsilyl trifluoromethanesulfonate (2M solution in 1,2-dichloroethane, 0.96 ml, 1.92 mmol) in 1,2-dichloroethane (20 ml) at 0°C for 4 hrs. After column chromatography on silica gel (30 g) 2c was obtained as a white powder. The column was washed with methylene chloride (300 ml), and then eluted with methylene chloride - ethanol 99:1. Fractions containing the product were evaporated and dried to give 2c as a white amorphous powder. Yield 335 mg (60%) as a white powder. R_f 0.43 (A). ¹H-NMR (300 MHz, DMSO-d₆): 11.22 brs (1H, NH), 8.51 d (1H, $J_{6,5}$ = 7.4, H-6), 8.00 d (2H, J = 7.2, o-Bz), 7.63 t (1H, J = 7.2, p-Bz), 7.52 t (2H, J = 7.2, m-Bz), 7.34 d (1H, $J_{5,6}$ = 7.4, H-5), 5.86 d (1H, $J_{1',2'}$ = 2.6, H-1'), 5.32 dd (1H, $J_{5'-OH, 5'a}$ = 4.8, $J_{5'-OH, 5'b}$ = 5.0, 5'-OH), 4.56 dd (1H, $J_{2',1'}$ = 2.6, $J_{2',3'}$ = 4.3, H-2'), 4.38 dd (1H, $J_{3',2'}$ = 4.3, $J_{3',4'}$ = 7.1, H-3'), 4.01 ddd (1H, $J_{4',3}$ =

7.1, $J_{4',5'a} = 2.9$, $J_{4',5'b} = 3.0$, H-4'), 3.84 ddd (1H, $J_{5'a,4'} = 2.9$, $J_{5'a,5'OH} = 4.8$, $J_{5'a,5'b} = -12.4$, H-5'a), 3.64 ddd (1H, $J_{5'b,4'} = 3.0$, $J_{5'b,5'OH} = 5.0$, $J_{5'b,5'a} = -12.4$, H-5'b), 1.11-0.93 m (28H, iPr). ¹³C-NMR (75 MHz, DMSO-d_6): 167.35 (C=O); 163.10 (C-4); 154.38 (C-2); 144.87 (C-6); 133.22, 132.73, 128.45 (Ph); 96.09 (C-5); 90.46 (C-1'); 84.21 (C-4'); 76.30 (C-2'); 70.79 (C-3'); 59.13 (C-5'); 17.23, 17.17, 17.11, 17.07, 17.01, 16.95, 16.87, 12.74, 12.46, 12.31, 12.02 (iPr). LSI-MS: (C₂₈H₄₃N₃O₇Si₂ - H⁺) found 588.2564. Calc. 588.2561.

 N^2 -Isobutyryl-2',3'-O-(1,1,3,3-Tetraisopropyldisiloxane-1,3-diyl)guanosine (2d). Analogous conversion of 1d (603 mg, 1.01 mmol) was carried out in the presence of trimethylsilyl trifluoromethanesulfonate (2M solution in 1,2-dichloroethane, 1.51 ml, 3.03 mmol) in 1,2dichloroethane (20 ml) at 0°C for 2 hrs. After column chromatography on silica gel (30 g) 2d was obtained as a white powder. The column was washed with methylene chloride (300 ml), and then eluted with methylene chloride - ethanol 98:2. Fractions containing the product were evaporated and dried to give 2d as a white amorphous powder. Yield 446 mg (74 %). $R_f 0.28$ (A). ¹H-NMR (300 MHz, DMSO-d₆): 12.12 brs (1H, NH-1), 11.59 brs (1H, NH-2), 8.32 s (1H, H-8), 5.83 d (1H, $J_{1',2'}$ = 6.5, H-1'), 5.16 t (1H, $J_{5'-OH,5'a}$ = $J_{5'-OH,5'b}$ = 5.5, 5'-OH), 4.95 dd (1H, $J_{2',1'}$ = 6.5, $J_{2',3'} = 4.9$, H-2'), 4.58 dd (1H, $J_{3',2'} = 4.9$, $J_{3',4'} = 3.1$, H-3'), 4.01 ddd (1H, $J_{4',3'} = 3.1$, $J_{4',5'a} = 5.2$, $J_{4',5'b} = 4.6$, H-4'), 3.63 ddd (1H, $J_{5'a,4'} = 5.2$, $J_{5'a,5'-OH} = 5.5$, $J_{5'a,5'b} = -11.9$, H-5'a), 3.56 ddd (1H, $J_{5'b,4'} = 4.6, J_{5'b,5'-OH} = 5.5, J_{5'b,5'a} = -11.9, H-5'b), 2.78 \text{ sept (1H, } J = 6.9, iBu), 1.12 \text{ d (6H, } J =$ iBu), 1.08-0.98 m (28H, iPr). ¹³C-NMR (75 MHz, DMSO-d₆): 180.15 (C=O), 154.79 (C-6), 149.09 (C-2), 148.33 (C-4), 137.33 (C-8), 120.22 (C-5), 86.10 (C-1'), 85.87 (C-4'), 75.44 (C-2'), 72.22 (C-3'), 60.78 (C-5'), 34.70 (iBu), 18.87, 18.81 (iBu) 17.34, 17.17, 17.10, 17.07, 16.97, 16.81, 12.63, 12.37, 12.18, 12.13 (iPr). LSI-MS: (C₂₆H₄₅N₅O₇Si₂ - H⁺) found 594.2782. Calc. 594.2779.

2',3'-O-(1,1,3,3-Tetraisopropyldisiloxane-1,3-diyl)adenosine (2e). Analogous conversion of **1e** (550 mg, 1.08 mmol) was carried out in the presence of trimethylsilyl trifluoromethanesulfonate (2M solution in 1,2-dichloroethane, 1.51 ml, 3.03 mmol) in 1,2-dichloroethane (20 ml) at 0°C for 4 hrs. After column chromatography on silica gel (30 g) **2e** was obtained as a white powder. The column was washed with methylene chloride (300 ml), and then eluted with methylene chloride - ethanol 97:3. Fractions containing the product were evaporated and dried to give **2e** as a white amorphous powder. Yield 396 mg (72 %). R_f 0.29 (B). ¹H-NMR (400 MHz, DMSO-d_6): 8.38 s (1H, H-8), 8.13 s (1H, H-2), 7.23 brs (2H, NH₂), 5.96 d (1H, $J_{1',2'}$ = 5.9, H-1'), 5.33 dd (1H, $J_{5'}$ = 4.4, $J_{3',4'}$ = 4.4, H-3'), 4.06 ddd (1H, $J_{4',3'}$ = 4.4, $J_{4',5'a}$ = 4.0, $J_{4',5'b}$ = 4.4, H-2'), 4.63 dd (1H, $J_{5',4'}$ = 4.0, $J_{5'a,5'OH}$ = 5.3, $J_{5'a,5'DH}$ = 5.11.8, H-5'a), 3.58 ddd (1H, $J_{5'b}$ = 4.4, H-4'), 3.70 ddd (1H, $J_{5'a,4'}$ = 4.0, $J_{5'a,5'OH}$ = 5.3, $J_{5'a,5'D}$ = -11.8, H-5'a), 3.58 ddd (1H, $J_{5'b,4'}$ = 4.4, $J_{5'b,5'OH}$ = 5.9, $J_{5'b,5'a}$ = -11.8, H-5'b), 1.10-0.92 m (28H, iPr). ¹³C-NMR (100 MHz, DMSO-d_6): 156.14 (C-6), 152.56 (C-8), 149.16 (C-4), 139.71 (C-2), 119.25 (C-5), 87.79 (C-1'), 85.84 (C-4'), 75.45 (C-2'), 72.59 (C-3'), 61.08 (C-5'), 17.31, 17.16, 17.08, 16.93, 16.80, 12.67, 12.32, 12.29, 12.09 (iPr). LSI-MS: (C₂₂H₃₉N₅O₅Si₂ - H⁺) 508.2414. Calc. 508.2411.

2',3'-*O***-(1,1,3,3-Tetraisopropyldisiloxane-1,3-diyl)cytidine (2f).** Analogous conversion of 1f (500 mg, 1.03 mmol) was carried out in the presence of trimethylsilyl trifluoromethanesulfonate

(2M solution in 1,2-dichloroethane, 1.51 ml, 3.03 mmol) in 1,2-dichloroethane (20 ml) at 0°C for 2 hrs. After column chromatography on silica gel (30 g) **2f** was obtained as a white powder. The column was washed with methylene chloride (300 ml), and then eluted with methylene chloride - ethanol 95:5. Fractions containing the product were evaporated and dried to give **2f** as a white amorphous powder. Yield 275 mg (55 %). R_f 0.12 (B). ¹H-NMR (400 MHz, DMSO-d₆): 7.88 d (1H, $J_{6,5}$ = 7.5, H-6), 7.15 brs (1H, NH₂), 7.07 brs (1H, NH₂), 5.79 d (1H, $J_{1',2'}$ = 3.7, H-1'), 5.71 d (1H, $J_{5,6}$ = 7.5, H-5), 5.15 t (1H, $J_{5'-OH,5'a}$ = $J_{5'-OH,5'b}$ = 5.3, 5'-OH), 4.41 dd (1H, $J_{2',1'}$ = 3.7, $J_{2',3'}$ = 4.4, H-2'), 4.34 dd (1H, $J_{5'a,4'}$ = 3.1, $J_{5'a,5'-OH}$ = 5.3, $J_{5'a,5'b}$ = -12.1, H-5'a), 3.57 ddd (1H, $J_{5'b,4'}$ = 3.4, $J_{5'b,5'-OH}$ = 5.3, $J_{5'b,5'a}$ = -12.1, H-5'b), 1.05-0.92 m (28H, iPr). ¹³C-NMR (100 MHz, DMSO-d₆): 165.57 (C-4); 155.06 (C-2); 141.04 (C-6); 93.92 (C-5); 89.54 (C-1'); 83.96 (C-4'); 75.96 (C-2'); 71.45 (C-3'); 59.87 (C-5'); 17.23, 17.09, 17.04, 16.99, 16.93, 16.83, 12.63, 12.51, 12.30, 12.02 (iPr). LSI-MS: (C₂₁H₃₉N₃O₆Si₂ - H⁺) found 484.2302. Calc. 484.2299.

Investigation of isomerization of 3',5'-O-(tetraisopropyldisiloxane-1,3-diyl)nucleosides (1) in the presence of different catalysts. (see Table 2 for details). The catalyst was added to a solution of nucleoside 1 (0.5 mmol) in 1,2-dichloroethane or acetonitrile (10 ml) under nitrogen atmosphere. The reaction mixture was stirred at appropriate temperature for 24 hrs. Methylene chloride (20 ml) and 10% aqueous solution of sodium bicarbonate (10 ml) were added and the suspension was stirred for 15 min. The organic layer was separated, washed with water (20 ml), dried over anhydrous sodium sulfate and evaporated in vacuo to dryness. The ratio of 1:2 and the presence of decomposition products were determined with ¹H-NMR spectroscopy and TLC in systems A and B.

N⁶-Benzoyl-9-[2',3'-O-(1,1,3,3-tetraisopropyldisiloxane-1,3-diyl)-5'-O-(2,3,5-tri-O-benzoyl- β -D-ribofuranosyl)- β -D-ribofuranosyl]adenine (3b). To a cool solution (0° C) of nucleoside 1b (704 mg, 1.15 mmol) in 1,2-dichloroethane (25 ml) 2M solution of trimethylsilyl trifluoromethanesulfonate in 1,2-dichloroethane (1.15 ml, 2.29 mmol) was added under nitrogen atmosphere. The reaction mixture was stirred at 0°C for 12 hrs. After addition of 1-O-acetyl-2,3,5-tri-O-benzoyl-B-D-ribofuranose (752 mg, 1.49 mmol) the resulting solution was kept at 0°C for 20 hrs. Methylene chloride (50 ml) and 10% aqueous solution of sodium bicarbonate (25 ml) were added and the suspension was stirred at 0°C for 15 min. The organic laver was separated, washed with water (25 ml), dried over anhydrous sodium sulfate, evaporated in vacuo and purified by column chromatography on silica gel (40 g). The column was washed with methylene chloride (300 ml), and then eluted with methylene chloride - ethanol 99:1. Fractions containing the product were evaporated and dried to give **3b** as a foam. Yield 729 mg (60 %). R_f 0.66 (A). ¹H-NMR (400 MHz, CDCl₃): 8.80 s (1H, H-8), 8.36 s (1H, H-2), 8.05-7.85 m (8H, o-Bz), 7.64-7.23 m (12H, m,p-Bz), 6.09 d (1H, J_{1',2'} = 3.7, H-1', Ado), 5.77 dd (1H, J_{3',2'} = 4.7, H-3', Rib), 5.64 d (1H, $J_{2',3'}$ = 4.7, H-2', Rib), 5.35 s (1H, H-1', Rib), 5.14 dd (1H, $J_{2',1'}$ = 3.7, $J_{2',3'}$ = 4.4, H-2', Ado), 4.72 ddd (1H, $J_{4'3'} = 10.0$, $J_{4'5'a} = 4.4$, $J_{4'5'b} = 5.9$, H-4', Rib), 4.68 dd (1H, $J_{3'2'} = 4.4$, $J_{3',4'} = 5.6$, H-3', Ado), 4.60 dd (1H, $J_{5'a,4'} = 4.4$, $J_{5'a,5'b} = -11.8$, H-5'a, Rib), 4.56 dd (1H, $J_{5'b,4'} = -11.8$, H-5'a, Rib), 4.56 dd (1H, J_{5'b,4'} = -11.8, H-5'a, Rib), 4.56 dd (1H, Rib), 4.56 5.9, $J_{5'b.5'a}$ = -11.8, H-5'b, Rib), 4.35 ddd (1H, $J_{4',3'}$ = 5.6, $J_{4',5'a}$ = 2.8, $J_{4',5'b}$ = 5.0, H-4', Ado), 4.17

dd (1H, $J_{5'a,4'} = 2.8$, $J_{5'a,5'b} = -11.5$, H-5'a, Ado), 3.82 dd (1H, $J_{5'b,4'} = 5.0$, $J_{5'b,5'a} = -11.5$, H-5'b, Ado), 1.26-0.92 m (28H, iPr). ¹³C-NMR (400 MHz, CDCl₃): 166.23, 165.43, 165.28, 164.67 (C=O), 152.37 (C-2), 151.63 (C-6), 149.59 (C-4), 142.17 (C-8), 139.41, 133.79, 133.65, 133.53, 133.22, 132.91, 129.91, 129.87, 128.99, 128.60, 128.48, 128.15 (Ph), 114.20 (C-5), 106.23 (C-1', Rib), 90.15 (C-1', Ado), 83.66 (C-4', Ado), 79.51 (C-4', Rib), 75.99 (C-2', Rib), 75.48 (C-2', Ado), 72.71 (C-3', Rib), 72.59 (C-3', Ado), 67.41 (C-5', Rib), 65.24 (C-5', Ado), 17.57, 17.52, 17.35, 17.30, 17.26, 17.18, 17.11, 14.23, 13.41, 13.28, 13.04, 12.90 (iPr). LSI-MS: (C₅₅H₆₄N₅O₁₃Si₂ + H⁺) found 1058.4036. Calc.1058.4039.

9-(5'-O-β-D-Ribofuranosyl-β-D-ribofuranosyl)adenine (4b). Disaccharide nucleoside 3b (600 mg, 0.57 mmol) was dissolved in 0.5M solution of tetrabutylammonium fluoride trihydrate in tetrahydrofuran (3.4 ml, 1.70 mmol), kept at 20°C for 20 min and the solution was concentrated in vacuo and purified by column chromatography on silica gel (30 g). The column was washed with methylene chloride (300 ml), and then eluted with methylene chloride - ethanol 97.5:2.5. Fractions containing the product were evaporated and dried to give N⁶-benzoyl-9-[5'-O- $(2,3,5-\text{tri-O-benzoyl-}\beta-\text{D-ribofuranosyl})$ - β -D-ribofuranosyl]adenine as a foam. R_f 0.69 (B). ¹H-NMR (400 MHz, CDCl₃): 8.75 s (1H, H-8), 8.51 s (1H, H-2), 8.02-7.92 m (8H, o-Bz), 7.58-7.23 m (12H, m,p-Bz), 6.09 d (1H, $J_{1',2'}$ = 4.7, H-1', Ado), 5.62 dd (1H, $J_{3',2'}$ = 4.4, $J_{3',4'}$ = 5.9, H-3', Rib), 5.60 d (1H, $J_{2',3'}$ = 4.4, H-2', Rib), 5.29 s (1H, H-1', Rib), 4.80 dd (1H, $J_{2',1'}$ = 4.7, $J_{2',3'}$ = 5.3, H-2', Ado), 4.68 ddd (1H, $J_{4',3'}$ = 5.9, $J_{4',3'}$ = 10.0, $J_{4',5'a}$ = 4.4, $J_{4',5'b}$ = 5.6, H-4', Rib), 4.57 dd (1H, $J_{5'a,4'} = 4.4, J_{5'a,5'b} = -12.2, H-5'a, Rib), 4.53 dd (1H, J_{5'b,4'} = 5.6, J_{5'b,5'a} = -12.2, H-5'b, Rib), 4.49$ dd (1H, $J_{3',2'} = 5.3$, $J_{3',4'} = 4.7$, H-3', Ado), 4.37 ddd (1H, $J_{4',3'} = 4.7$, $J_{4',5'a} = 3.4$, $J_{4',5'b} = 4.1$, H-4', Ado), 4.11 dd (1H, $J_{5'a,4'} = 3.4$, $J_{5'a,5'b} = -11.2$, H-5'a, Ado), 3.76 dd (1H, $J_{5'b,4'} = 4.1$, $J_{5'b,5'a} = -11.2$, H-5'b, Ado). ¹³C- NMR (400 MHz, CDCl₃): 166.41, 165.60, 165.42, 164.65 (C=O), 152.24 (C-2), 151.23 (C-6), 149.19 (C-4), 142.30 (C-8), 129.94, 129.88, 129.82, 128.99, 128.64, 128.51, 128.24, 129.87, 128.99, 128.60, 128.48, 128.15 (Ph), 114.20 (C-5), 106.09 (C-1', Rib), 90.24 (C-1', Ado), 84.42 (C-4', Ado), 79.31 (C-4', Rib), 75.45 (C-2', Rib), 74.94 (C-2', Ado), 72.55 (C-3', Rib), 71.44 (C-3', Ado), 67.97 (C-5', Rib), 65.20 (C-5', Ado).

A solution of N⁶-benzoyl-9-[5'-O-(2,3,5-tri-O-benzoyl- β -D-ribofuranosyl)- β -D-ribofuranosyl]adenine (300 mg, 0.37 mmol) in 5M solution of ammonia in methanol (10 ml) was kept at 20°C for 3 days and then concentrated in vacuo. The residue was partitioned between methylene chloride (10 ml) and water (20 ml) and the water layer was washed with methylene chloride (3x10 ml). The water layer was concentrated in vacuo, the residue was evaporated with methanol to yield **4b** as a foam. Yield 159 mg (84 %). R_f 0.29 (methylene chloride –ethanol 6:4). The NMR, UV and mass spectra of **4b** were practically the same as those reported earlier.³²

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Supplementary Information Available

NMR spectra of compounds 1 and 2 are available as an attachment.

References

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