Synthesis of 1-aroylmethylpyrroles as useful intermediates for further chemical transformation

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Abstract
The synthesis of 1-aroylmethylpyrroles from 2-bromo-1-(2-aminophenyl)ethan-1-one and mono-
tri-substituted 1H-pyrroles, has been investigated. The reactions take place at r.t. or 80 °C
in DMF containing potassium carbonate. Reduction of 1-[2-(2-aminophenyl)-2-oxoethyl]-4-
bromo-1H-pyrrole-2-carbaldehyde with sodium borohydride gave racemic 1-(2-aminophenyl)-2-
[4-bromo-2-(hydroxymethyl)-1H-pyrrol-1-yl]ethanol.

Keywords: Pyrroles, 1-aroylmethylpyrroles, alkylation, reduction

Introduction
The importance of 1-aroylmethylpyrroles stems from the fact that they constitute the central
structural component in marine natural products such as the Lamellarins1 and are precursors to
the closely related Lukianols.2 Moreover, 1-aroylmethylpyrroles are useful precursors in the
synthesis of pyrrolo[2,1-b][3]benzazepines1 and pyrrolo[2,1-b][1,3]benzothiazepines3,
compounds that possess interesting biological properties. Several syntheses of 1-
aroylmethylpyrroles have been reported. Belanger and co-workers3 synthesised methyl 1-(2-oxo-
2-arylethyl)-1H-pyrrole-2-carboxylates by reacting 2-bromo-1-arylethanones and methyl 1H-
pyrrole-2-carboxylates at ambient temperature in DMF containing potassium carbonate. Artico
and co-workers5 synthesised 1-aryl-2-(1H-pyrrol-1-yl)ethanones from 2-amino-1-arylethanone
hydrochlorides and 2,5-dimethoxytetrahydrofuran. These reactions were carried out by heating
briefly in DMF. En route to pyrrolobenzothiazepines, Campiani et al.4,6 synthesised several 1-
aryl-2-(1H-pyrrol-1-yl)ethanones by the same method, except that the reactions were carried out
in boiling acetic acid containing aqueous sodium acetate.
Results and Discussion

Herein we describe the preparation of 1-aroylmethylpyrroles \textbf{3a-h} by reacting 2-bromo-1-(2-aminophenyl)ethan-1-one \textbf{1} with the appropriately substituted pyrroles \textbf{2a-h} in DMF containing potassium carbonate (Scheme 1). The substrates used, reaction time, temperature and yield of products are given in Table 1.

![Scheme 1](image-url)

\textbf{Scheme 1.} Reagents and conditions: (i) K$_2$CO$_3$, DMF, r.t. or 80 °C.

\textbf{Table 1.} Conversion of 2-bromo-1-(2-aminophenyl)ethan-1-one \textbf{1} into 1-aroylmethylpyrroles \textbf{3a-h}

<table>
<thead>
<tr>
<th>Pyrrole</th>
<th>Product</th>
<th>(R^1)</th>
<th>(R^2)</th>
<th>(R^3)</th>
<th>(R^4)</th>
<th>Temp (°C)</th>
<th>Time (h)</th>
</tr>
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<tbody>
<tr>
<td>\textbf{2a}</td>
<td>\textbf{3a}</td>
<td>CHO</td>
<td>H</td>
<td>H</td>
<td>H</td>
<td>22</td>
<td>3</td>
</tr>
<tr>
<td>\textbf{2b}</td>
<td>\textbf{3b}</td>
<td>CHO</td>
<td>H</td>
<td>CHO</td>
<td>H</td>
<td>r.t.</td>
<td>3</td>
</tr>
<tr>
<td>\textbf{2c}</td>
<td>\textbf{3c}</td>
<td>CHO</td>
<td>H</td>
<td>Br</td>
<td>H</td>
<td>80</td>
<td>16</td>
</tr>
<tr>
<td>\textbf{2d}</td>
<td>\textbf{3d}</td>
<td>CHO</td>
<td>CH=CHCONMe$_2$</td>
<td>H</td>
<td>H</td>
<td>80</td>
<td>16</td>
</tr>
<tr>
<td>\textbf{2e}</td>
<td>\textbf{3e}</td>
<td>CO$_2$Me</td>
<td>H</td>
<td>H</td>
<td>H</td>
<td>r.t.</td>
<td>48</td>
</tr>
<tr>
<td>\textbf{2f}</td>
<td>\textbf{3f}</td>
<td>CO$_2$Me</td>
<td>H</td>
<td>Br</td>
<td>H</td>
<td>r.t.</td>
<td>48</td>
</tr>
<tr>
<td>\textbf{2g}</td>
<td>\textbf{3g}</td>
<td>CO$_2$Me</td>
<td>H</td>
<td>Br</td>
<td>Br</td>
<td>r.t.</td>
<td>48</td>
</tr>
<tr>
<td>\textbf{2h}</td>
<td>\textbf{3h}</td>
<td>CO$_2$Me</td>
<td>H</td>
<td>CH=CHCONMe$_2$</td>
<td>H</td>
<td>80</td>
<td>16</td>
</tr>
</tbody>
</table>

The precursor to 2-bromo-1-(2-aminophenyl)ethan-1-one \textbf{1} is 2-bromo-1-(2-nitrophenyl)ethan-1-one. The latter was prepared by dropwise addition of bromine solution to 2-nitroacetophenone in chloroform following basically the procedure by Andreani et al.\textsuperscript{7} who did not isolate the material but reacted it straight with 2-aminothiazole. In our hands the compound was isolated in 83% yield. 2-Bromo-1-(2-nitrophenyl)ethan-1-one was reduced to compound \textbf{1} by copper powder in sulfuric acid according to the method of Grehm.\textsuperscript{8} Any attempt to reduce the compound by other means including Zn or Fe and acid, SnCl$_2$ or catalytic hydrogenation, failed. \(1H\)-Pyrrole-2-carbaldehyde \textbf{2a} was prepared from freshly distilled \(1H\)-pyrrole by Vilsmeier-Haack formylation.\textsuperscript{9} Freshly distilled \(1H\)-pyrrole was also used to prepare \(1H\)-pyrrole-2,4-dicarbaldehyde \textbf{2b}, by reacting with oxalyl chloride in DMF and then formylating the
intermediate \(N\)-[(1\(H\))-pyrrol-2-yl)methylene]-\(N\)-methylmethanaminium chloride by reacting with dichloro(methoxy)methane with aluminium trichloride as catalyst.\(^{10}\) 4-Bromo-2-formyl-1\(H\)-pyrrole \(2c\) was synthesised by the method of Sonnet\(^{11}\) that is, bromination of the ternary iminium salt, obtained from 1\(H\)-pyrrole-2-carbaldehyde \(2a\) and pyrrolidinium perchlorate in benzene, followed by mild hydrolysis. The synthesis of enamides \(2d\) and \(7\) was accomplished in two steps. In the first step Friedel-Crafts formylation of prop-2-enamide \(4\), prepared previously in our laboratory,\(^{12}\) with 1,1-dichloro-2-methoxyethane in the presence of aluminium trichloride gave a mixture that separated by column chromatography into (\(E\))-3-(2-formyl)-2-enamide \(5\) and (\(E\))-3-(5-formyl)-2-enamide \(6\) in 53 and 11% yield, respectively. A characteristic difference in the \(^1H\) NMR spectra of compounds \(5\) and \(6\) is the expected large coupling constant between H-4 and H-5 (\(J_{4,5} = 3.2\) Hz) of \(5\) and a much smaller coupling constant between H-2 and H-4 (not measured by the 400 MHz instrument) of \(6\). In the second step compounds \(5\) and \(6\) were detosylated by treatment with a methanolic solution of potassium carbonate, to yield corresponding enamides \(2d\) and \(7\) in 88 and 82% yield, respectively (Scheme 2).

Scheme 2. Reagents and conditions: (i) Cl\(_2\)CHOMe, ClCH\(_2\)CH\(_2\)Cl, AlCl\(_3\), 0\(^\circ\) C, 1 h., (ii) K\(_2\)CO\(_3\), MeOH, 1 h.

The synthesis of 1\(H\)-pyrrole-2-carboxylic acid ethyl ester \(2e\) took place according to Harbuck and Rapoport\(^{13a}\) by trichloroacetylating 1\(H\)-pyrrole with trichloacetyl chloride to 2,2,2-trichloro-1-(1\(H\)-pyrrol-2-yl)ethanone and then esterifying with a methanolic solution of sodium methoxide. Compound \(2e\) was brominated with one equivalent of bromine in DMF containing potassium carbonate to 4-bromo-1\(H\)-pyrrole-2-carboxylic acid ethyl ester \(2f\) in 85% yield.\(^{14}\) On the other hand, methyl 4,5-dibromo-1\(H\)-pyrrole-2-carboxylate \(2g\) was obtained by brominating 2,2,2-trichloro-1-(1\(H\)-pyrrol-2-yl)ethanone with two equivalents of bromine and then esterifying
2,2,2-trichloro-1-(4,5-dibromo-1H-pyrrol-2-yl)ethanone with methanolic sodium methoxide. Compound 2g was isolated in near quantitative yield from the last step. A literature report describes the preparation of 2g by the reaction of bromo ester 2f with NBS in DMF, albeit in only 29% yield. The Horner-Wadsworth-Emmons reaction was used to synthesise propenoate 2h by reacting phosphonate mono-carbanion, produced from dimethylcarbamoyl-methylphosphonic acid diethyl ester and slight excess of NaH in THF, with 4-formyl-1H-pyrrole-2-carboxylic acid methyl ester. The latter was prepared according to Rapoport and co-workers, in two steps, from 2,2,2-trichloro-1-(1H-pyrrol-2-yl)ethanone by Friedel-Crafts formylation with 1,1-dichloro-2-methoxyethane in the presence of aluminium trichloride and then esterification of 5-(2,2,2-trichloroacetyl)-1H-pyrrole-3-carbaldehyde with a methanolic solution of sodium methoxide.

The 1-aroylmethylpyrroles 3a-h obtained as described above (Table 1) are useful precursors for further chemical transformation. One of the goals set forth at the beginning of this work was to study the intramolecular cyclisation of these compounds leading to pyrrolobenzodiazocines, analogues of the naturally occurring pyrrolo[2,1-c][1,4]benzodiazepine family of antibiotics. A preliminary study on the cyclodehydration of 3c to pyrrolobenzodiazocine 8 has been undertaken. Compound 3c was subjected to several literature methods for cyclising amino-aldehydes, including an intramolecular aza-Wittig reaction (dibromotriphenylphosphine and triethylamine), refluxing in ethanol or 2-methoxyethanol, in the presence of molecular sieves and Dean-Stark conditions (toluene and a catalytic amount of p-toluenesulfonic acid) all resulting in unreacted starting material (Scheme 3). The resistance of compound 3c to cyclodehydration may be due to its non-favourable conformation, possibly because of the keto group. One further attempt to cyclise compound 3c was by reduction with sodium borohydride. It was hoped that selective reduction of 3c would give the less conformationally constrained alcohol 10 where interaction between the amino and aldehyde groups of 10 would lead to pyrrolobenzodiazocine 11. Instead, non-selective reduction of 3c led to racemic dialcohol 9.
Scheme 3. Reagents and conditions: (i) (a) Ph₃PBr₂, Et₃N, CH₂Cl₂, (b) EtOH or MeO(CH₂)₂OH, 3Å molecular sieves, reflux or (c) toluene, TsOH, Dean-Stark, (ii) NaBH₄, Et₂O, 48 h.

At the present time we are investigating further the chemistry of the 1-aroylmethylpyrroles in order to synthesise a large variety of derivatives and prepare a profile of their biological activity. We are also examining the possibility of using the compounds as precursors to the pyrrolobenzodiazacine, 2,3-dihydro-1H-quinolin-4-one and 3,4-dihydropyrrolo[2,1-c][1,4]-oxazin-1-one ring systems.

Conclusions

We have described an efficient method of synthesising 1-aroylmethylpyrroles from 2-bromo-1-(2-aminophenyl)ethan-1-one and mono-, di- or tri-substituted 1H-pyrroles. These compounds are potentially useful as intermediates for further functional group transformation in heterocyclic synthesis and as scaffolds for the creation of libraries.

Experimental Section

General Procedures. Melting points were taken on a Büchi 510 apparatus and are uncorrected. Infrared spectra were recorded on a Perkin-Elmer 257 spectrometer, as Nujol mulls and liquids
between sodium chloride discs. Nuclear magnetic resonance spectra were measured at 300 MHz on Brüker AC 300 spectrometer, at 250 MHz on a Brüker AM 250 spectrometer or at 400 MHz on a Brüker AMX 400 spectrometer using tetramethylsilane as internal standard. Mass spectra were obtained using a JEOL JMS-AX 505W or a Bruker Apex III high resolution instruments. Analytical TLC was carried out on Fluka silica gel 60 F254. Preparative flash chromatography was carried out for all separations using Merck 9385 silica gel. Solvents and reagents were used as received from the manufacturers, except for dichloromethane, methanol, tetrahydrofuran, ethyl acetate and hexane that were purified and dried according to standard procedures.

**General Procedure for the preparation of 1-arylmethylpyrroles (3a-h)**

To a stirred mixture of the appropriate pyrrole 2a-h (3 mmol) (Table 1) and potassium carbonate (0.83 g, 6 mmol) in dry DMF (5 mL) under an atmosphere of argon, was added 2-bromo-1-(2-aminophenyl)ethan-1-one 1 (0.64 g, 3 mmol). The reaction mixture was stirred at room temperature or at 80 °C for 3, 16 or 48 h, cooled, poured into ice-cold water (25 mL) and extracted with EtOAc (3 × 10 mL). The combined organic extracts were washed with brine (10 mL) and dried (Na2SO4). The solvent was evaporated under vacuum to give, either an oily residue which was triturated with diethyl ether to give a solid or a solid residue, which was then purified either by column chromatography followed by recrystallisation or by recrystallisation, to afford the product.

1-[2-(2-Aminophenyl)-2-oxoethyl]-1H-pyrrole-2-carbaldehyde (3a). Colourless microcrystals (recrystallised from toluene); (0.42 g, 62%); mp 159-160 °C. IR (Nujol, cm⁻¹): 3475, 3375, 1660, 1640; ¹H NMR (250 MHz, CDCl₃): δ = 5.81 (s, 2H, CH₂), 6.19 (br s, 2H, NH₂), 6.37 (dd, J = 4.0, 2.6 Hz, 1H, H-4), 6.65 − 6.73 (m, 2H, H-3’, H-5’), 6.93 (br s, 1H, H-5), 7.04 (dd, J = 4.0, 1.6 Hz, 1H, H-3), 7.31 (ddd, J = 8.3, 7.3, 1.2 Hz, 1H, H-4’), 7.75 (d, J = 8.3 Hz, 1H, H-6’), 9.54 (d, J = 0.9 Hz, 1H, CHO); ¹³C NMR (63 MHz, CDCl₃): δ = 54.74, 110.16, 115.64, 116.06, 117.48, 124.61, 129.83, 131.86, 132.65, 135.04, 150.69, 179.77, 194.03; MS (EI, 70 eV) m/z: 228 [M⁺, 86%], 133 (83), 120 (100), 92 (86), 65 (70%); HRMS −EI Calcd C₁₃H₁₂N₂O₂: 228.0899. Found: 228.0898.

1-[2-(2-Aminophenyl)-2-oxoethyl]-1H-pyrrole-2,4-dicarbaldehyde (3b). Colourless microcrystals (recrystallised from AcOEt −hexane); (0.41 g, 54%); mp 162-163 °C. IR (Nujol, cm⁻¹): 3475, 3375, 1720, 1700, 1680; ¹H NMR (400 MHz, CDCl₃): δ = 5.84 (s, 2H, CH₂), 6.19 (br s, 2H, NH₂), 6.64 − 6.70 (m, 2H, H-3’, H-5’), 7.34 (t, J = 7.4 Hz, 1H, H-4’), 7.39 (br s, 1H, H-3), 7.53 (br s, 1H, H-5), 7.71 (d, J = 8.2 Hz, 1H, H-6’), 9.63 (s, 1H, CHO−2), 9.88 (s, 1H, CHO−4); ¹³C NMR (100.6 MHz, CDCl₃): δ = 56.13, 109.32, 114.28, 118.39, 119.84, 126.45, 129.97, 133.06, 133.77, 135.28, 149.32, 173.94, 175.23, 191.56; MS (EI, 70 eV) m/z: 256 (12) [M⁺, 12%], 133 (83), 120 (100), 92 (86), 65 (70%); HRMS −EI Calcd C₁₄H₁₂N₂O₂: 228.0899. Found: 228.0898.

1-[2-(2-Aminophenyl)-2-oxoethyl]-4-bromo-1H-pyrrole-2-carbaldehyde (3c). Colourless solid after column chromatography [AcOEt−hexane, (1:4)] and recrystallisation (propan-1-ol); (0.50 g, 55%); mp 179-180 °C. IR (Nujol, cm⁻¹): 3480, 3380, 1720, 1695; ¹H NMR (400 MHz, CDCl₃): δ = 5.76 (s, 2H, CH₂), 6.19 (br s, 2H, NH₂), 6.68 (dd, J = 8.4, 1.1 Hz, 1H, H-3’), 6.70 (dd, J = 8.1, 7.1, 1.1 Hz, 1H, H-5’), 6.91 (dd, J = 1.8, 1.1 Hz, 1H, H-5), 7.01 (d, J = 1.8 Hz, 1H, H-3), 7.32
(dd, J = 8.4, 7.1, 1.2 Hz, 1H, H-4’), 7.69 (dd, J = 8.1, 1.2 Hz, 1H, H-6’), 9.47 (s, J = 1.1 Hz, 1H, CHO); $^{13}$C NMR (100.6 MHz, CDCl$_3$): δ = 54.82 (CH$_2$), 97.46 (C-4), 115.52 (C-1’), 116.20 (C-5’), 117.62 (C-3’), 125.22 (C-3), 129.76 (C-6’), 131.88 (C-5), 132.08 (C-2), 135.28 (C-4’), 150.82 (C-2’), 179.29 (CHO), 193.14 (CO); MS (EI, 70 eV) m/z: 306 [M$^+$, 11%], 133 (29), 120 (100), 92 (22), 65 (14%); HRMS–EI Calcd C$_{13}$H$_{13}$BrN$_2$O$_3$: 306.0004. Found: 306.0009.

(2E)-3-[[1-[2-(2-aminophenyl)-2-oxoethyl]-2-formyl-1H-pyrrole-3-yl]-N,N-dimethylprop-2-enamide (3d). Brown-red microcrystals after column chromatography [EtOAc–hexane, (9:1)] and recrystallisation (AcOEt–hexane); (0.63 g, 65%); mp 148-150 °C; IR (Nujol, cm$^{-1}$): 3485, 3380, 1690, 1670, 1605; $^1$H NMR (300 MHz, CDCl$_3$): δ = 3.00 (s, 3H, CH$_3$), 3.10 (s, 3H, CH$_3$), 5.72 (s, 2H, CH$_2$), 6.12 (br s, 2H, NH$_2$), 6.49 (d, J = 2.7 Hz, 1H, H-4), 6.58-6.65 (m, 2H, H-3’, H-5’), 6.75 (d, J = 15.1 Hz, 1H, H-α), 6.80 (d, J = 2.7 Hz, 1H, H-5), 7.25 (dd, J = 8.2, 1.2 Hz, 1H, H-4’), 7.65 (dd, J = 8.1, 1.2 Hz, 1H, H-6’), 7.90 (d, J = 15.1 Hz, 1H, H-β), 9.87 (s, 1H, CHO); $^{13}$C NMR (75.5 MHz, CDCl$_3$): δ = 35.94, 37.40, 55.36, 107.28, 115.52, 115.32, 119.13, 128.85, 129.78, 131.76, 132.32, 132.93, 135.18, 150.73, 166.49, 178.71, 193.62; MS (EI, 70 eV) m/z: 325 [M$^+$, 67%], 167 (40), 154 (75), 149 (100), 136 (81), 113 (45), 91 (34), 73 (71), 57 (80%); HRMS–EI Calcd C$_{18}$H$_{16}$N$_3$O$_3$: 325.1426. Found: 325.1423.

1-[2-(2-Aminophenyl)-2-oxoethyl]-1H-pyrrole-2-carboxylic acid methyl ester (3e). Pale green needles after column chromatography [EtOAc–hexane, (1:1) and recrystallisation (propan-2-ol); (0.52 g, 67%); mp 157-158.5 °C; IR (Nujol, cm$^{-1}$): 3485, 3380, 1690, 1670; $^1$H NMR (400 MHz, CDCl$_3$): δ = 3.66 (s, 3H, CH$_3$), 5.69 (s, 2H, CH$_2$), 6.12 (br s, 2H, NH$_2$), 6.19 (dd, J = 4.0, 2.6 Hz, 1H, H-4), 6.58-6.64 (m, 2H, H-3’, H-5’), 6.75 (dd, J = 2.5, 1.9 Hz, 1H, H-5), 6.98 (dd, J = 4.0, 1.9 Hz, 1H, H-3), 7.22 (dd, J = 8.4, 7.1, 1.3 Hz, 1H, H-4’), 7.68 (dd, J = 8.1, 1.3 Hz, 1H, H-6’); $^{13}$C NMR (100.6 MHz, CDCl$_3$): δ = 51.52 (CH$_3$), 55.52 (CH$_2$), 109.06 (C-4) 116.06 (C-2), 114.46 (C-5’), 117.94 (C-3’), 118.56 (C-3), 130.17 (C-6’), 130.39 (C-5), 135.38 (C-4’), 151.09 (C-2’), 162.16 (CO ester), 194.96 (CO ketone); MS (EI, 70 eV) m/z: 258 [M$^+$, 25%], 133 (23), 120 (100), 92 (13%); HRMS–EI Calcd C$_{14}$H$_{14}$N$_2$O$_3$: 258.1004. Found: 258.1008.

1-[2-(2-Aminophenyl)-2-oxoethyl]-4-bromo-1H-pyrrole-2-carboxylic acid methyl ester (3f). Pale yellow microcrystals after recrystallisation (propan-2-ol); (0.70 g, 70%); mp 162-164 °C; IR (Nujol, cm$^{-1}$): 3465, 3352, 2924, 1705, 1650; $^1$H NMR (300 MHz, CDCl$_3$): δ = 3.74 (s, 3H, CH$_3$), 5.73 (s, 2H, CH$_2$), 6.20 (br s, 2H, NH$_2$), 6.66-6.72 (m, 2H, H-3’, H-5’), 6.85 (d, J = 1.8 Hz, 1H, H-3), 7.03 (d, J = 1.8 Hz, 1H, H-5), 7.30 (ddd, J = 8.5, 7.9, 1.5, Hz, 1H, H-4’), 7.72 (dd, J = 8.1, 1.5 Hz, 1H, H-6’); $^{13}$C NMR (75.5 MHz, CDCl$_3$): δ = 51.34, 55.05, 96.02, 115.39, 116.07, 117.54, 119.60, 122.99, 129.18, 129.60, 135.10, 150.70, 160.79, 193.60; MS (ESI) m/z: 337 [M + H$^+$]; HRMS–ESI: m/z [M + H]$^+$ Calcd C$_{14}$H$_{14}$BrN$_2$O$_3$: 337.0188. Found: 337.0192.

1-[2-(Aminophenyl)-2-oxoethyl]-4,5-dibromo-1H-pyrrole-2-carboxylic acid methyl ester (3g). Deep green microcrystals after recrystallisation (propan-2-ol); (0.80 g, 64%); mp 170-171 °C; IR (Nujol, cm$^{-1}$): 3485, 3380, 1690, 1670; $^1$H NMR (400 MHz, CDCl$_3$): δ = 3.66 (s, 3H, CH$_3$), 5.87 (s, 2H, CH$_2$), 6.12 (br s, 2H, NH$_2$), 6.58-6.69 (m, 2H, H-3’, H-5’), 7.04 (s, 1H, H-3), 7.24 (ddd, J = 8.4, 7.8, 1.3 Hz, 1H, H-4’), 7.68 (dd, J = 8.1, 1.3, Hz, 1H, H-6’); $^{13}$C NMR (100.6 MHz, CDCl$_3$): δ = = 51.01 (CH$_3$), 54.73 (CH$_2$), 99.94 (C-4) 114.70 (C-5), 115.77 (C-1’), 116.52 (C-3’),
117.99 (C-5′), 120.20 (C-3), 124.55 (C-2′), 135.62 (C-4′), 151.17 (C-2′′), 160.65 (CO ester), 193.15 (CO ketone); MS (EI, 70 eV) m/z: 414 [M+, 12%], 337 (24), 120 (100), 92 (15%); HRMS–EI Calcd for C_{14}H_{12}Br_{2}N_{2}O_{3}: 413.9215. Found: 413.9211.

1-[2-(2-Aminophenyl)-2-oxoethyl]-4-[(1E)-3-(dimethylamino)-3-oxoprop-1-enyl]-1H-pyrrole-2-carboxylic acid methyl ester (3h). Colourless microcrystals after column chromatography [EtOAc–hexane, (9:1)] and recrystallisation (AcOEt–hexane); (0.57 g, 54%); mp 207.5-209 °C. IR (Nujol, cm⁻¹): 3410, 3310, 1730, 1690, 1650; ¹H NMR (400 MHz, DMSO-d₆): δ = 2.96 (s, 3H, CH₃), 3.18 (s, 3H, CH₃), 3.72 (s, 3H, CH₃ ester), 5.86 (s, 2H, CH₂), 6.66 (ddd, J = 8.4, 8.2, 1.0 Hz, 1H, H-5′), 6.85 (dd, J = 8.4, 1.0 Hz, 1H, H-3′), 6.98 (d, J = 15.2 Hz, 1H, H-α), 7.20 (br s, 2H, NH₂), 7.36 (ddd, J = 8.4, 8.2, 1.3 Hz, 1H, H-4′), 7.40 (d, J = 15.2 Hz, 1H, H-β), 7.43 (d, J = 1.8 Hz, 1H, H-5), 7.52 (d, J = 1.8 Hz, 1H, H-3), 7.90 (dd, J = 8.2, 1.3 Hz, 1H, H-6′); ¹³C NMR (100.6 MHz, DMSO-d₆): δ = 35.61 (CH₃), 37.10 (CH₃), 51.43 (CH₂), 114.65 (C-4), 114.88 (C-5), 115.15 (C-α), 115.72 (C-5′), 117.42 (C-3′), 120.23 (C-1′), 123.35 (C-1′), 130.61 (C-6′), 132.68 (C-3), 134.56 (C-β), 135.03 (C-4′), 151.62 (C-2′′), 161.00 (CO, ester), 166.39 (CO, amide), 194.54 (CO); MS (EI, 70 eV) m/z: 355 [M+, 13%], 310 (29), 223 (32), 120 (100), 92 (17%); HRMS–EI Calcd C_{19}H_{21}N_{3}O_{4}: 355.1532. Found: 355.1536.

Formylation of (E)-3-[1-(4-methylphenylsulfonyl)-1H-pyrrol-3-yl]-N,N-dimethylprop-2-enamide. A stirred mixture of prop-2-enamide 4¹² (0.50 g, 1.57 mmol) and AlCl₃ (0.49 g, 3.71 mmol) in dry 1,2-dichloroethane (10 mL) was cooled to –40°C under argon. Dichloromethylether (0.25 g, 2.20 mmol) was added dropwise and then the temperature was allowed to reach –10°C. Stirring was continued for 1 h and then the reaction mixture was added to ice-water, the organic layer separated and the aqueous layer extracted with CH₂Cl₂ (3 x 10 mL). The combined organic extracts were washed with a 10 % w/v aqueous solution of NaHCO₃ (3 x 10 mL), brine (20 mL) and dried (MgSO₄). The solvent was removed under reduced pressure and the residue was purified by column chromatography (CH₂Cl₂–acetone, 90:1 to 1:1) to give in the first fraction compound 6 (0.06 g, 11%) and in the second fraction compound 5 (0.29 g, 53%).

(E)-3-[2-formyl-[1-(4-methylphenylsulfonyl)-1H-pyrrol-3-yl]-N,N-dimethylprop-2-enamide (5). Pale yellow microcrystals after recrystallisation (AcOEt–hexane); (0.29 g, 53%); mp 153-154 °C (dec); IR (Nujol, cm⁻¹): 1680, 1660, 1330, 1120; ¹H NMR (300 MHz, CDCl₃): δ = 2.36 (s, 3H, CH₃), 3.07 (br s, 6H, 2 x CH₃), 6.58 (br s, 1H, H-4), 6.84 (d, J = 15.2 Hz, 1H, H-α), 7.31 (d, J = 8.1 Hz, 2H, H-3′, H-5′), 7.57 (br s, 1H, H-5), 7.76 (d, J = 8.1 Hz, H-2′, 2H, H-6′), 7.94 (d, J = 15.2 Hz, 1H, H-β), 10.20 (s, 1H, CHO); ¹³C NMR (75.5 MHz, CDCl₃): δ = 21.7, 35.8, 37.5, 110.2, 123.6, 127.5, 128.8, 129.2, 130.3, 131.8, 134.8, 135.0, 146.2, 166.1, 179.4; MS (EI, 70 eV) m/z: 346 [M⁺, 10%], 307 (25), 154 (100), 136 (89), 107 (54), 91 (38), 81 (36), 69 (54), 57 (73%); HRMS–EI Calcd for C_{17}H_{18}N_{2}O_{4}S: 346.0987. Found: 346.0985.

(E)-3-[5-formyl-[1-(4-methylphenylsulfonyl)-1H-pyrrol-3-yl]-N,N-dimethylprop-2-enamide (6). Pale yellow microcrystals after recrystallisation (AcOEt–hexane); (0.06 g, 11%); mp 153-154 °C (dec); IR (Nujol, cm⁻¹): 1690, 1670, 1340, 1150; ¹H NMR (300 MHz, CDCl₃): δ = 2.41 (s, 3H, CH₃), 3.05 (d, 6H, 2 x CH₃), 6.70 (d, J = 15.4 Hz, 1H, H-α), 7.29 (br s, 1H, H-4), 7.29-7.36 (m,
2H, H-3’, H-5’), 7.46 (d, J = 15.4 Hz, 1H, H-β), 7.69 (br s, 1H, H-2), 7.75-7.81 (m, 2H, H-2’, H-6’), 10.00 (s, 1H, CHO); 13C NMR (75.5 MHz, CDCl3): δ = 21.8, 36.1, 37.8, 118.8, 122.8, 127.5, 128.6, 130.2, 132.3, 134.5, 134.9, 146.3, 166.1, 179.2; MS (EI, 70 eV) m/z: 346 [M+, 52%], 307 (35), 289 (20), 154 (100), 136 (95), 120 (47), 97 (42), 84 (83), 77 (77), 63 (20%); HRMS–EI Calcd C17H18N2O4S: 346.0987. Found: 346.0983.

**General procedure for the detosylation of (E)-3-[2 or 5-formyl-[1-(4-methylphenylsulfonyl)-1H-pyrrol-4 or 3-yl]-N,N-dimethylprop-2-enamides (5,6)**

A mixture of prop-2-enamides 5 (0.22 g, 0.64 mmol) or 6 (0.06 g, 0.17 mmol) and K2CO3 (0.20 g, 1.45 mmol) in dry MeOH (15 mL) under argon was stirred for 1 h. The reaction mixture was poured into brine (30 mL), extracted with EtOAc (3 x 15 mL), and dried (Na2SO4). The solvent was removed under vacuo and the crude product 2d or 7 was purified by recrystallisation.

**(E)-3-[2-formyl-1H-pyrrol-3-yl]-N,N-dimethylprop-2-enamide (2d).** Brown microcrystals after recrystallisation (EtOAc–hexane); (0.11 g, 88%); mp 123-124 °C (dec); IR (Nujol, cm-1): 1690, 1670; 1H NMR (300 MHz, CDCl3): δ = 3.03 (s, 3H, CΗ3), 3.09 (s, 3H, CΗ3), 6.45-6.51 (m, 1H, H-4), 6.77 (d, J = 15.2 Hz, 1H, H-α), 7.03-7.12 (m, 1H, H-5), 7.90 (d, J = 15.2 Hz, 1H, H-β), 9.80 (s, 1H, CHO), 9.81 (br s, 1H, NH); 13C NMR (75.5 MHz, CDCl3): δ = 34.9, 36.7, 112.5, 124.7, 126.7, 129.2, 131.8, 135.0, 165.4, 181.6; MS (EI, 70 eV) m/z: 192 [M+, 25%], 148 (65), 120 (100), 92 (40), 65 (41%); HRMS–EI Calcd C10H12N2O2: 192.0899. Found: 192.0898.

**(E)-3-[5-formyl-1H-pyrrol-3-yl]-N,N-dimethylprop-2-enamide (7).** Brown microcrystals after recrystallisation (AcOEt–hexane); (0.027 g, 82%); mp 129-130 °C (dec); IR (Nujol, cm-1): 1680, 1670; 1H NMR (300 MHz, CDCl3): δ = 3.11 (s, 3H, CΗ3), 3.12 (s, 3H, CΗ3), 6.66 (d, J = 15.4 Hz, 1H, H-α), 7.14 (s, 1H, H-4), 7.31 (s, 1H, H-2), 7.61 (d, 1H, J = 15.4 Hz, H-β), 9.54 (s, 1H, CHO), 9.99 (br s, 1H, NH); 13C NMR (75.5 MHz, CDCl3): δ = 34.8, 36.7, 117.8, 122.6, 129.9, 133.3, 132.35, 136.3, 165.8, 181.8; MS (ESI) m/z: 193.1 [M + H]+; HRMS–ESI [M + H]+ Calcd C10H13N2O2: 193.0977. Found: 193.0981.

**(±)-1-(2-Aminophenyl)-2-[4-bromo-2-(hydroxymethyl)-1H-pyrrol-1-yl]ethanol (9).** To a stirred solution of 1-[2-(2-aminophenyl)-2-oxoethyl]-4-bromo-2-formyl-1H-pyrrole 3c (0.20 g, 0.66 mmol) in dry ether (10 mL), sodium borohydride (0.048 g, 1.27 mmol) was added and the reaction mixture stirred at room temperature for 48 h. The solvent was evaporated, water (15 mL) was added and acidified with glacial acetic acid to pH = 5. The mixture was extracted with dichloromethane (3 x 5 mL), the combined organic phases dried (Na2SO4) and evaporated to give an oil (0.14 g, 68%); IR (Nujol, cm-1): 3640, 3440, 3340; 1H NMR (250 MHz, DMSO-d6): δ = 3.99 (s, 1H, CH), 4.00 (s, 1H, CH), 4.37 (d, 2H, CH2), 5.11 (2H, CH2) 5.20 (1H, CH), 5.57 (br s, 1H, H-5), 5.99 (br s, 1H, H-3), 6.55 (t, 1H, H-5’), 6.65 (d, 1H, H-3’), 6.97 t, 1H, H-4’), 7.14 (d, J = 7.5 Hz, 1H, H-6’); 13C NMR (63 MHz, DMSO-d6): δ = 49.78, 52.88, 68.24, 90.70, 107.86, 113.63, 114.48, 120.60, 123.96, 125.08, 126.26, 132.35, 143.74; MS (EI, 70 eV) m/z: 310 [M+, 19%], 292 (18), 189 (32), 171 (46), 122 (100%); HRMS–EI Calcd C13H15BrN2O2: 310.0317. Found: 310.0312.
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References and Notes