Diacetylketene N,S-acetals in synthesis of new functionalized 2(1*H*)pyrimidinethiones

Vladimir Dorokhov,* Alexander Komkov, and Sergey Baranin

N.D.Zelinsky Institute of Organic Chemistry, Russian Academy of Sciences, 47 Leninsky prosp., 119991 Moscow, Russian Federation E-mail: <u>vador@ioc.ac.ru</u>

Dedicated to Professor Branko Stanovnik on the occasion of his 65th birthday (received 25 July 03; accepted 15 Dec 03; published on the web 29 Dec 03)

Abstract

New 5-acetyl-4-alkylthio-6-methyl-2(1H)-pyrimidinethiones were prepared from diacetylketene N,S-acetals and isothiocyanates. They were converted into 4-amino derivatives, which can be applied for the construction of functionalized pyrido[2,3-d]pyrimidines and pyrimido[4,5-d]pyrimidines.

Keywords: Diacetylketene N,S-acetals, isothiocyanates, heterocyclization, 2(1*H*)-pyrimidinethiones, pyrimido[4,5-*d*]pyrimidines, pyrido[2,3-*d*]pyrimidines

Introduction

Ketene N,S-acetals are known to be useful reagents in heterocyclic synthesis.^{1,2} Among these, particular attention has been given to oxoketene N,S-acetals as functionalized enaminones.^{3,4} Previously we reported a convenient procedure for the preparation of N-unsubstituted diacylketene N,S-acetals from β -diketones and alkyl thiocyanates in the presence of Ni(acac)₂.⁵ These compounds were shown to be suitable starting materials for synthesizing 4-acetyl-5-aminopyrazoles,⁶ pyrazolo[3,4-*d*]pyrimidines,⁶ functionalized 2(1*H*)pyrimidinones,⁷ and 3-cyano-4-pyridones.⁸ In continuation of our work on the synthetic utility of dioxoketene N,S-acetals and isothiocyanates. Although 2-pyrimidinethiones have been extensively investigated and different approaches to their preparation have been developed,⁹ new methods for the synthesis of 2-pyrimidinethiones carrying functional groups in the 5- and 6-positions are desirable, since compounds of this type may be used for constructing fused pyrimidines.

Monoaroylketene N,S-acetals are reported to react with benzoyl isothiocyanate as Cnucleophiles to give the corresponding adducts which undergo cyclization into 4-pyrimidinethiones.¹⁰ It is quite general that the C-C bond is formed by the attack of the enaminone nucleophilic C atom to the electrophilic C atom of isothiocyanate,¹¹ although enaminones as N-nucleophiles were found to react with phenylisothiocyanate in the presence of NaH affording 1-phenyl-4,6-disubstituted 2-pyrimidinethiones.¹²

Results and Discussion

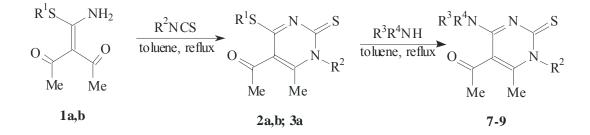
We have previously shown that the condensation of diacetylketene N,S-acetals with isocyanates occurs in the absence of basic catalysts and gives 4-alkylthiouracyl derivatives.⁷ It turned out that N,S-acetals **1a,b** react in similar manner with phenylisothiocyanate and allylisothiocyanate in boiling toluene providing the corresponding N-substituted 5-acetyl-4-alkylthio-6-methyl-2(1*H*)-pyrimidinethiones **2a,b** and **3a** (Scheme 1). The action of **1a** on the benzoylisothiocyanate in toluene at room temperature results in thiourea **4**, which is isolated as crude material. The structure of **4** is confirmed by ¹H NMR spectra (see the Experimental Section). When **4** is boiled with MeONa in MeOH, the closure of the pyrimidine ring is accompanied by debenzoylation, and the subsequent treatment with AcOH or MeI leads to pyrimidinethiones **5** or its S-methyl derivative **6**. Evidently, the formation of pyrimidinethiones **2**, **3** is also supposed to involve the attack by isothiocyanate at the N-nucleophilic center of acetals **1** but the intermediate thioureas formed easily undergo cyclization in the absence of MeONa.

Crystalline pyrimidinethiones **2a,b** and **3a** are easily soluble in DMF, CHCl₃, EtOH, and acetone, moderately soluble in benzene and toluene, and insoluble in petroleum ether and water. Compound **5** is soluble only in DMF and DMSO. The structures of **2a,b**, **3a**, and **5** were confirmed by spectral data (mass spectrometry, ¹H and ¹³C NMR, IR spectroscopy).

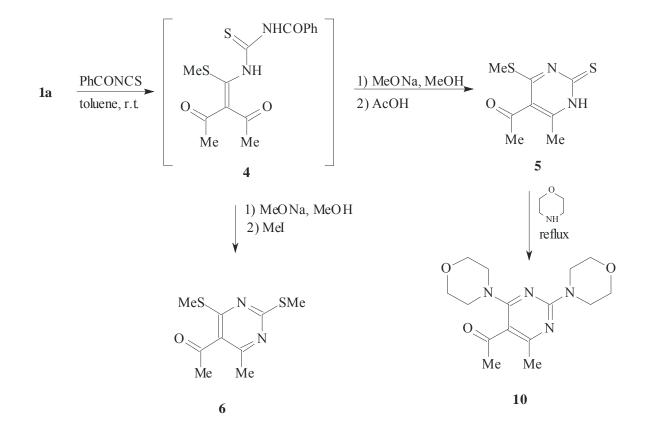
The MeS group in pyrimidinethiones 2, 3 can be substituted by primary and secondary amines, and compound 2a was thus converted into the corresponding 4-amino-2(1*H*)-pyrimidinethione derivatives 7-9. It should be noted that the yields of 7, 8 (42-43%) appear to be lower than the yield of 9 (69%), because the reaction of 2a with primary amines is accompanied by the partial cleavage of the pyrimidine ring. Indeed, the N-benzyl-N`-phenylthiourea was isolated as by-product when 2a reacted with benzylamine.

In the case of **5**, the double substitution by morpholine can be achieved, and dimorpholinopyrimidine **10** was obtained in 79% yield.

The presence of vicinal MeCO and NH groups in the molecules of compounds **7**, **8** is favorable for the annelation of the second nitrogen-containing ring to the pyrimidine cycle.

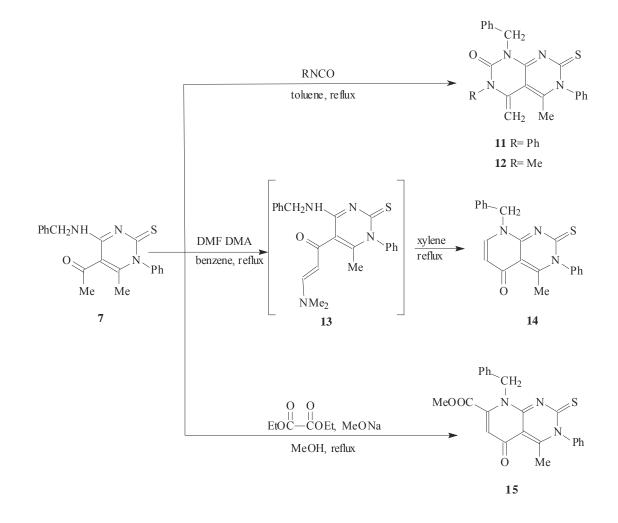


 $\begin{array}{ll} R^{1} = Me(\mathbf{a}); \ Et(\mathbf{b}); & R^{2} = Ph(\mathbf{2,7.9}); & All(\mathbf{3}); & R^{3} = PhCH_{2}, & R^{4} = H(\mathbf{7}); & R^{3} = Bu, & R^{4} = H(\mathbf{8}); \\ R^{3}, R^{4} = -(CH_{2})_{2}O(CH_{2})_{2}-(\mathbf{9}) \end{array}$



Scheme 1

We have chosen pyrimidinethione **7** to demonstrate the selected examples for fused pyrimidines construction. Earlier the series of 4-methylene-3,4-dihydro-2(1H),7(6*H*)-pyrimido[4,5-*d*]pyrimidinediones⁷ had been prepared from substituted 4-amino-5-acetyl-2(1H)-pyrimidinones and isocyanates. Now we synthesized new representatives of pyrimido[4,5-*d*]pyrimidine system **11**, **12** containing oxo-, thio-, and *exo*-methylene groups in 75-82% yields by the reaction of **7** with isocyanates (Scheme 2).



Scheme 2

The process probably involves the addition of **7** to isocyanate with the formation of intermediate ureas, intramolecular cyclization of which gives **11** and **12**. However, compound **7** failed to react with less electrophilic isothiocyanates.

Yellow crystalline compounds **11, 12** are soluble in most organic solvents. The presence of the *exo*-methylene group in their molecules is confirmed by NMR spectroscopy. Thus, in ¹H NMR spectra, methylene protons display the signals of the AB system (δ_A 4.36 and δ_B 4.37 for **11**; δ_A 4.49 and δ_B 4.90 for **12**, *J*=2.5 Hz), while the C atom of the CH₂ group in ¹³C NMR spectra gives a triplet (δ 100.6 for **11** and δ 97.2 for **12**).

Different approaches to the construction of pyrido[2,3-*d*]pyrimidine system were applied. In accordance with Scheme 2, refluxing of **7** with dimethylformamide dimethylacetal (DMF DMA) in benzene results in the condensation product **13**, which in the boiling xylene undergoes cyclization to give the corresponding functionalized pyrido[2,3-*d*]pyrimidine **14**. In a similar manner, 8-benzyl-4-methylthio-2-phenyl-5(8*H*)-pyrido[2,3-*d*]pyrimidinone has earlier been prepared.¹³

Recently we have suggested a method for the synthesis of alkyl 5-oxo-5,8dihydropyrido[2,3-d]pyrimidine-7-carboxylates¹⁴ based on the condensation of 2,6-disubstituted 5-acetyl-4-aminopyrimidines with ethyl oxalate in the presence of MeONa or EtONa. Accordingly, the compound **7** was transformed into methyl ester **15** isolated in moderate yield. Evidently, the process is accompanied by transesterification. No traces of ethyl ester were detected by ¹H NMR spectroscopy.

The structures of pyrido[2,3-*d*]pyrimidine derivatives **14**, **15** were confirmed by spectroscopic methods and microanalysis data (see below in Experimental Section).

Experimental Section

General Procedures. Melting points were determined using a Koffler apparatus and were uncorrected. ¹H NMR (250 MHz) and ¹³C NMR (75 MHz) spectra were recorded on Bruker WM-250 and Bruker AM-300 spectrometers with CDCl₃ and DMSO-d₆ as solvent and TMS as internal standard. Mass spectra were obtained on a Varian MAT-311A instrument (EI, 70eV). IR spectra were recorded on a Specord M-80 spectrometer. Column chromatography was conducted with silica gel, grade 100-160 mesh. Phenyl-, allyl- and benzoylisothiocyanates, DMF DMA, phenyl- and methylisocyanates, and also diethyl oxalate were purchased from Lancaster. Diacetylketene N,S-acetals⁵ were prepared according to published procedures.

5-Acetyl-6-methyl-4-methylthio-1-phenyl-2(1H)-pyrimidinethione (2a). A mixture of 1a (2.60 g, 15 mmol) and phenylisothiocyanate (3.60 mL, 30 mmol) in toluene (25 mL) was heated under reflux for 3 h. After cooling, the precipitate was collected by filtration to give light yellow **2a**: 2.22 g (51%); mp 226-227 °C (from C₆H₆ / *n*-hexane 8:1). ¹H NMR (CDCl₃): δ 1.92 (3H, s, CH₃CO), 2.60 (3H, s, CH₃), 2.68 (3H, s, CH₃), 7.15-7.21 (2H, m, 2H of Ph), 7.43-7.65 (3H, m, 3H of Ph). ¹³C NMR (CDCl₃): δ 13.34 (SCH₃), 19.54 (CH₃), 32.03 (CH₃CO), 122.91 (C-5), 127.17, 129.32, 130.25, 140.84 (Ph), 151.56 (q, C-6, ²J=5.0), 167.45 (q, C-4, ³J=3.0), 181.88 (C-2), 199.48 (CO). MS m/z: 290 (M⁺). IR (CHCl₃) v/cm⁻¹: 1705 (CO), 1580, 1500. Anal. Calcd for C₁₄H₁₄N₂OS₂: C, 57.90; H, 4.86; N, 9.65; S, 22.08. Found: C, 58.28; H, 4.99; N, 9.29; S, 22.06. 5-Acetvl-4-ethylthio-6-methyl-1-phenyl-2(1H)-pyrimidinethione (2b). A mixture of 1b (1.50 g, 8 mmol) and phenylisothiocyanate (1.92 mL, 16 mmol) in toluene (15 mL) was heated under reflux for 4 h. After cooling to 20 °C, hexane (30 mL) was added to the reaction mixture. The precipitate was collected by filtration, subjected to column chromatography, and eluted with C_6H_6 to give yellow solid **2b**: 1.24 g (51%); mp 155-156 °C (from C_6H_6 / *n*-hexane 1:1). ¹H NMR (CDCl₃): δ 1.38 (3H, t, CH₃CH₂), 1.91 (3H, s, CH₃CO), 2.59 (3H, s, CH₃), 3.35 (2H, q, CH₂), 7.16-7.21 (2H, m, 2H of Ph), 7.45-7.62 (3H, m, 3H of Ph). MS m/z: 304 (M⁺). IR (CHCl₃) v/cm⁻¹: 1702 (CO), 1580, 1470. Anal. Calcd for C₁₅H₁₆N₂OS₂: C, 59.18; H, 5.30; N, 9.20; S, 21.07. Found: C, 59.20; H, 5.50; N, 9.39; S, 20.74.

5-Acetyl-1-allyl-6-methyl-4-methylthio-2(1H)-pyrimidinethione (3a). A mixture of 1a (1.04 g,

6 mmol) and allylisothiocyanate (1.17 mL, 12 mmol) in toluene (12 mL) was heated under reflux for 3 h. After cooling to 20 °C, hexane (20 mL) was added to the reaction mixture. The precipitate obtained was filtered off and recrystallized from C₆H₆ / *n*-hexane (1:1) to give yellow-brown solid **3a**: 0.79 g (52%); mp 129-130 °C. ¹H NMR (CDCl₃): δ 2.34 (3H, s, CH₃), 2.54 (3H, s, CH₃), 2.61 (3H, s, CH₃), 5.12-5.37 (4H, m, CH₂=CHCH₂), 5.92-6.08 (1H, m, CH₂=CHCH₂). ¹³C NMR (CDCl₃): δ 13.17 (SCH₃), 17.81 (CH₃), 32.02 (CH₃CO), 54.04 (NCH₂), 118.35 (dd, CH₂=CH, ^{*1*}*J*=154, ^{*1*}*J*=161), 123.64 (C-5), 129,82 (d, CH₂=CH, ^{*1*}*J*=159), 151.17 (q, C-6, ²*J*=5.6), 166.02 (q, C-4, ³*J*=4.3), 181.23 (t, C-2, ³*J*=4.4), 199.76 (CO). MS m/z: 254 (M⁺). IR (CHCl₃) v/cm⁻¹: 1705 (CO), 1580, 1490. Anal. Calcd for C₁₁H₁₄N₂OS₂: C, 51.94; H, 5.55; N, 11.01; S, 25.21. Found: C, 51.84; H, 5.63; N, 10.90; S, 24.82.

5-Acetyl-6-methyl-4-methylthio-2(1*H***)-pyrimidinethione (5).** A mixture of **1a** (1.04 g, 6 mmol) and benzoylisothiocyanate (0.97 mL, 7.2 mmol) in C₆H₆ (30 mL) was stirred for 3 h at 20 °C. Hexane (60 mL) was added to the reaction mixture. The precipitate obtained was filtered off to give 1.49 g (74%) of crude urea **4** (¹H NMR (CDCl₃): δ 2.28 (6H, s, 2CH₃CO), 2.56 (3H, s, SCH₃), 7.50-7.57 (2H, m, 2H of Ph), 7.62-7.68 (1H, m, 1H of Ph), 7.85-7.92 (2H, m, 2H of Ph), 9.45 (1H, s, NH), 16.64 (1H, s, NH).) A mixture of **4** (0.67 g, 2 mmol) and 2.4 mmol MeONa in MeOH (20 mL) was heated under reflux for 1.5 h. The solvent was evaporated *in vacuo*, the residue was triturated with H₂O (30 mL) and extracted with CHCl₃ (2 x 30 mL). The aqueous solution was separated and treated with AcOH. The precipitate obtained was filtered off and washed with ether (2 x 30 mL) to give colorless solid **5**: 0.274 g (64%); mp 210-211 °C. ¹H NMR (DMSO-d₆): δ 2.31 (3H, s, CH₃), 2.46 (3H, s, CH₃), 2.49 (3H, s, CH₃), 13.40 (1H, s, NH). ¹³C NMR (DMSO-d₆): δ 13.21 (SCH₃), 17.47 (CH₃), 31.71 (CH₃CO), 120.60 (C-5), 153.64 (q, C-6, ²*J*=6.3), 170.02 (q, C-4, ³*J*=4.3), 178.73 (C-2), 198.85 (CO). MS m/z: 214 (M⁺). IR (KBr) v/cm⁻¹: 3425, 3150 (NH), 1660 (CO), 1585, 1540. Anal. Calcd for C₈H₁₀N₂OS₂: C, 44.83; H, 4.70; N, 13.07; S, 29.92. Found: C, 44.68; H, 4.96; N, 12.75; S, 29.75.

5-Acetyl-6-methyl-2,4-dimethylthiopyrimidine (6). A mixture of crude **4** (0.67 g, 2 mmol) and MeONa (2.4 mmol) in MeOH (20 mL) was heated under reflux for 1.5 h. After cooling to 20 °C, MeI (0.25 mL, 4 mmol) was added and the mixture was stirred for 30 min. The solvent was evaporated *in vacuo*. The residue was subjected to column chromatography (silica gel) and eluted with hexane / C₆H₆ (1:1) and then C₆H₆ to afford the pure **6**: 0.32 g (71%); mp 73-74 °C (from hexane). ¹H NMR (CDCl₃): δ 2.36 (3H, s, CH₃), 2.56 (3H, s, CH₃), 2.58 (6H, s, 2 SCH₃). ¹³C NMR (CDCl₃): δ 13.89 (2 SCH₃), 22.23 (CH₃), 31.32 (CH₃CO), 127.50 (C-5), 160.72 (q, C-6, ²*J*=6.3), 166.06 (q, C-4, ³*J*=4.5), 170.97 (q, C-2, ³*J*=4.3), 201.78 (CO). MS m/z: 228 (M⁺). IR (CHCl₃) v/cm⁻¹: 1698 (CO), 1530, 1518. Anal. Calcd for C₉H₁₂N₂OS₂: C, 47.34; H, 5.30; N, 12.27; S, 28.09. Found: C, 47.33; H, 5.56; N, 11.94; S, 28.04.

5-Acetyl-4-benzylamino-6-methyl-1-phenyl-2(1*H*)**-pyrimidinethione** (7). A mixture of **2a** (2.03 g, 7.0 mmol) and benzylamine (1.14 mL, 10.5 mmol) in toluene (20 mL) was heated under reflux for 3 h. The precipitate obtained after cooling to 20 °C was filtered off to give colorless solid **7**: 1.03 g (42%); mp 217-218 °C (from C₆H₆). ¹H NMR (CDCl₃): δ 2.18 (3H, s, CH₃), 2.49 (3H, s, CH₃), 4.83 (2H, d, CH₂, *J*=5.5), 7.19-7.60 (10H, m, 10H of 2 Ph), 8.47 (1H, t, NH,

J=5.5). ¹³C NMR (CDCl₃): δ 22.65 (CH₃), 32.99 (CH₃CO), 45.22 (CH₂), 109.48 (C-5), 127.66, 128.03, 128.12, 128.78,129.09, 129.95, 137.22, 141.45 (2 Ph), 154.22 (C-4), 157.46 (q, C-6, ²*J*=5.4), 182.89 (C-2), 200.08 (CO). MS m/z: 349 (M⁺). IR (CHCl₃) v/cm⁻¹: 3340 (NH), 1655 (CO), 1588. Anal. Calcd for C₂₀H₁₉N₃OS: C, 68.74; H, 5.48; N, 12.03; S, 9.18. Found: C, 69.05; H, 5.63; N, 12.39; S, 9.02.

The filtrate was subjected to column chromatography (silica gel) and eluted with C_6H_6 to afford pure N-benzyl-N'-phenylthiourea (0.57 g): mp 155-156 °C; lit.¹⁵ mp 153-154 °C. ¹H NMR (CDCl₃): δ 4.91 (2H, d, CH₂, *J*=5.0), 6.32 (1H, t, NH, *J*=5.0), 7.20-7.50 (10H, m, 10H of 2 Ph), 8.08 (1H, s, NH). MS m/z: 242 (M⁺).

5-Acetyl-4-butylamino-6-methyl-1-phenyl-2(1*H***)-pyrimidinethione** (8). A mixture of **2a** (0.44 g, 1.5 mmol) and BuNH₂ (0.23 mL, 2.3 mmol) in toluene (10 mL) was heated under reflux for 3 h. After cooling to 20 °C, hexane (10 mL) was added to the reaction mixture. The precipitate obtained was filtered off to give solid **8**: 0.19 g (41%); mp 219-220 °C. (from C₆H₆ / *n*-hexane 1:1). ¹H NMR (CDCl₃): δ 0.97 (3H, t, CH₃CH₂), 1.38-1.48 (2H, m, CH₂), 1.57-1.67 (2H, m, CH₂), 2.16 (3H, s, CH₃CO), 2.50 (3H, s, CH₃), 3.62-3.68 (2H, m, CH₂N), 7.18-7.24 (2H, m, 2H of Ph), 7.42-7.60 (3H, m, 3H of Ph), 8.10 (1H, t, NH, *J*=5.5). MS m/z: 315 (M⁺). IR (CHCl₃) v/cm⁻¹: 3320 (NH), 1648 (CO), 1580. Anal. Calcd for C₁₇H₂₁N₃OS: C, 64.73; H, 6.71; N, 13.32; S, 10.17. Found: C, 64.70; H, 6.76; N, 13.03; S, 10.38.

5-Acetyl-6-methyl-4-morpholino-1-phenyl-2(1*H*)-**pyrimidinethione** (**9**). A mixture of **2a** (1.74 g, 6.0 mmol) and morpholine (1.04 mL, 12 mmol) in toluene (20 mL) was heated under reflux for 6 h. The solvent and excess morpholine were evaporated *in vacuo*. The residue obtained was recrystallized from C₆H₆ to give colorless solid **9**: 1.36 g (69%); mp 228-229 °C. ¹H NMR (CDCl₃): δ 1.99 (3H, s, CH₃), 2.42 (3H, s, CH₃), 3.75 (8H, s, 4 CH₂), 7.19-7.23 (2H, m, 2H of Ph), 7.40-7.62 (3H, m, 3H of Ph). ¹³C NMR (CDCl₃): δ 18.85 (CH₃), 31.08 (CH₃CO), 41.81 (CH₂), 66.52 (CH₂), 111.67 (C-5), 127.95, 129.06, 129.80, 141.31 (Ph), 155.51 (q, C-6, ²*J*=6.0), 156.89 (C-4), 181.58 (C-2), 200.08 (CO). MS m/z: 329 (M⁺). IR (CHCl₃) v/cm⁻¹: 1690 (CO), 1575. Anal. Calcd for C₁₇H₁₉N₃O₂S: C, 61.98; H, 5.81; N, 12.76; S, 9.73. Found: C, 62.22; H, 5.96; N, 12.51; S, 9.36.

5-Acetyl-6-methyl-2,4-dimorpholinopyrimidine (**10**). A mixture of **5** (0.21 g, 1 mmol) and morpholine (9 mL, 102 mmol) was heated under reflux for 6 h. A morpholine excess was evaporated *in vacuo*. The residue obtained was subjected to column chromatography (silica gel) and eluted with C₆H₆ to afford colorless pyrimidine **10**: 0.24 g (79%); mp 159-160 °C (from hexane). ¹H NMR (CDCl₃): δ 2.30 (3H, s, CH₃), 2.38 (3H, s, CH₃), 3.40 (4H, t, 2 CH₂), 3.72 (8H, t, 4 CH₂), 3.79 (4H, t, 2 CH₂). ¹³C NMR (CDCl₃): δ 23.23 (CH₃), 30.10 (CH₃CO), 44.14 (CH₂), 49.22 (CH₂), 66.49 (CH₂), 66.85 (CH₂), 111.01 (C-5), 159.55 and 164.36 (C-2 and C-4), 166.19 (q, C-6, ²*J*=6.0), 202.45 (CO). MS m/z: 306 (M⁺). IR (CHCl₃) v/cm⁻¹: 1675 (CO), 1558, 1535, 1520. Anal. Calcd for C₁₅H₂₂N₄O₃: C, 58.80; H, 7.24; N, 18.29. Found: C, 59.07; H, 7.43; N, 18.04.

1-Benzyl-5-methyl-4-methylene-3,6-diphenyl-7-thioxo-3,4,6,7-tetrahydro-2(1*H***)-pyrimido[4,5-***d*]**pyrimidinone (11).** A mixture of **7** (0.17 g, 0.5 mmol) and PhNCO (0.11 mL, 1 mmol) in toluene (6 mL) was heated under reflux for 3 h. The solvent was evaporated *in vacuo*. The residue obtained was subjected to column chromatography (silica gel) and eluted with C₆H₆ and then C₆H₆ / CHCl₃ (1:1) to afford the oil, which was dissolved in C₆H₆. Hexane (6 mL) was added to the solution, and the precipitate obtained was filtered off to give yellow solid **11**: 0.18 g (82%); mp 137-138 °C. ¹H NMR (CDCl₃): δ 2.22 (3H, s, CH₃), 4.36 and 4.43 (both for 1H, both d, CH₂=, *J*=2.5), 5.45 (2H, s, CH₂), 7.20-7.78 (15H, m, 15H of 3 Ph). ¹³C NMR (CDCl₃): δ 21.93 (CH₃), 45.19 (CH₂), 100.62 (t, CH₂=, ^{*I*}*J*=164.0), 104.40 (C-4a), 127.57, 127.66, 128.34, 128,66, 128.80, 129.37, 129.95, 130.16, 130.35, 136.76, 137.69, 137.89 (3 Ph), 141.89 (t, C-4, ²*J*=8.0), 149.94, 152.31 (C-2 and C-8a), 154.01 (q, C-5, ²*J*=6.0), 182.73 (C-7). MS m/z: 450 (M⁺). IR (CHCl₃) v/cm⁻¹: 1708 (CO), 1625, 1605, 1590, 1520. Anal. Calcd for C₂₇H₂₂N₄OS: C, 71.97; H, 4.92; N, 12.44; S, 7.12. Found: C, 71.89; H, 5.00; N, 12.12; S, 6.92.

1-Benzyl-3,5-dimethyl-4-methylene-6-phenyl-7-thioxo-3,4,6,7-tetrahydro-2(1H)-

pyrimido[4,5-*d*]**pyrimidinone** (12). A mixture of 7 (0.17 g, 0.5 mmol) and MeNCO (0.06 mL, 1 mmol) in toluene (6 mL) was heated in a sealed tube in an oil bath (110-115 °C) for 6 h. The further procedure was analogous to the above experiment and afforded solid 12: 0.146 g (75%); mp 216-217 °C. ¹H NMR (CDCl₃): δ 2.25 (3H, s, CH₃), 3.29 (3H, s, NCH₃), 4.49 and 4.90 (both for 1H, both d, CH₂=, *J*=2.5), 5.45 (2H, s, CH₂), 7.20-7.40 (5H, m, 5H of 2 Ph), 7.48-7.68 (5H, m, 5H of 2 Ph). ¹³C NMR (CDCl₃): δ 21.74 (CH₃), 32.54 (NCH₃), 45.04 (CH₂), 97.25 (CH₂=), 104.29 (C-4a), 127.60, 128.32, 129.28, 129.55, 130.28, 136.74, 136.95 (2 Ph), 141.96 (C-4), 150.39 and 152.16 (C-2 and C-8a), 153.79 (C-5), 182.76 (C-7). MS m/z: 388 (M⁺). IR (CHCl₃) v/cm⁻¹: 1698 (CO), 1624, 1605, 1590, 1522. Anal. Calcd for C₂₂H₂₀N₄OS: C, 68.02; H, 5.19; N, 14.42; S, 8.25. Found: C, 67.88; H, 5.02; N, 14.48; S, 8.01.

8-Benzyl-4-methyl-3-phenyl-2-thioxo-2,3-dihydro-5(8H)-pyrido[2,3-d]pyrimidinone (14). A mixture of 7 (0.35 g, 1 mmol) and DMF DMA (0.26 mL, 2 mmol) in C₆H₆ (6 mL) was heated under reflux for 1 h. The solvent was evaporated in vacuo. The residue obtained was subjected to column chromatography (silica gel) and eluted with $CHCl_3$ to give pyrimidine 13: 0.36 g (89%); mp 122-125 °C. ¹H NMR (CDCl₃): δ 2.03 (3H, s, CH₃), 2.88 and 3.15 (both for 3H, both s, N(CH₃)₂), 4.82 (2H, d, CH₂, J=5.5), 5.17 and 7.67 (both for 1H, both d, CH=CH, J=12.8), 7.18-7.60 (11H, m, 10H of 2 Ph and NH). A solution of **13** (0.36 g) in *m*-xylene (20 mL) was heated under reflux for 6 h. The solvent was evaporated in vacuo. The residue obtained was dissolved in C₆H₆ (5 mL). Hexane (8 mL) was added to the solution, and the precipitate obtained was filtered off to give yellow solid 14: 0.23 g, (73%); mp 170-171 °C. ¹H NMR (CDCl₃): δ 2.73 (3H, s, CH₃), 5.43 (2H, s, CH₂), 6.07 and 7.40 (both for 1H, both d, H-6 and H-7, J=6.5), 7.28-7.65 (10H, m, 10H of 2 Ph). ¹³C NMR (CDCl₃): δ 21.61 (CH₃), 52.37 (CH₂), 109.64 (C-4a), 114.52 (d, C-6, ¹J=171), 127.13, 128.50, 128.60, 129.05, 129.53, 130.42, 135.17, 141.18 (2 Ph), 142.01 (d, C-7, ¹J=178), 152.90 (C-8a), 166.90 (q, C-4, ²J=6.5), 179.06 (C-5), 181.59 (C-2). MS m/z: 359 (M⁺). IR (CHCl₃) v/cm⁻¹: 1652 (CO), 1567, 1560. Anal. Calcd for C₂₁H₁₇N₃OS: C, 70.17; H, 4.77; N, 11.69; S, 8.92. Found: C, 69.85; H, 4.89; N, 11.37; S, 8.69.

Methyl 8-benzyl-4-methyl-5-oxo-3-phenyl-2-thioxo-2,3,5,8-tetrahydropyrido[2,3-*d*]-pyrimidine-7-carboxylate (15). A mixture of 7 (0.14 g, 0.4 mmol), diethyl oxalate (0.16 mL, 1.2 mmol), and MeONa (1.2 mmol) in MeOH (8 mL) was heated under reflux for 2 h. After cooling to 20 °C, AcOH was added, and the solvent was evaporated *in vacuo*. The residue was subjected to column chromatography (silica gel) and eluted with C₆H₆ and then C₆H₆ / MeOH (50:0.2). The solvents were removed and diethyl ether (3 mL) was added. The precipitate obtained was filtered off to afford yellow solid **15**: 0.07 g (42%); mp 155-156 °C. ¹H NMR (CDCl₃): δ 2.77 (3H, s, CH₃), 3.71 (3H, s, CH₃O), 5.92 (2H, s, CH₂), 6.41 (1H, s, H-6), 7.15-7.25 (4H, m, 4H of 2 Ph), 7.25-7.38 (3H, m, 3H of Ph), 7.50-7.68 (3H, m, 3H of Ph). ¹³C NMR (CDCl₃): δ 21.71 (CH₃), 52.45 (CH₂), 53.65 (CH₃O), 109.77 (C-4a), 116.04 (C-6), 126.94, 128.53, 128.64, 129.10, 129.57, 130.42, 135.28, 140.97 (2 Ph), 143.96 (C-7), 153.30 (C-8a), 162.89 (COO), 166.78 (C-4), 178.48 (C-5), 181.55 (C-2). MS m/z: 417 (M⁺). IR (CHCl₃) v/cm⁻¹: 1740 (CO), 1644 (CO), 1560. Anal. Calcd for C₂₃H₁₉N₃O₃S: C, 66.17; H, 4.59; N, 10.07; S, 7.68. Found: C, 66.12; H, 4.61; N, 9.86; S, 7.80.

References

- 1. Takahata, H.; Yamazaki, T. Heterocycles 1988, 27, 1953.
- 2. Yokoyama, M.; Togo, H.; Kondo, S. Sulfur Reports 1990, 10, 23.
- 3. Dieter, R.K. Tetrahedron 1986, 42, 3029.
- 4. Junjappa, H.; Ila, H.; Asokan, C.V. *Tetrahedron* **1990**, *46*, 5423.
- 5. Dorokhov, V.A.; Gordeev, M.F.; Shashkova, E.M.; Komkov, A.V.; Bogdanov, V.S. Bull. Acad. Sci. USSR. Div. Chem. Sci. 1991, 40, 2274 [Engl. Transl.].
- 6. Dorokhov, V.A.; Komkov, A.V.; Ugrak, B.I. Russian Chem. Bull. 1993, 42, 1364.
- 7. Dorokhov, V.A.; Gordeev, M.F.; Komkov, A.V.; Bogdanov, V.S. Bull. Acad. Sci. USSR. Div. Chem. Sci. 1991, 40, 2267 [Engl. Transl.].
- 8. Dorokhov, V.A.; Prezent, M.A.; Bogdanov, V.S. Russian Chem. Bull. 1995, 44, 1080.
- 9. For a review, see: Katoh, A.; Nishio, T.; Kashima, C. Heterocycles 1987, 26, 2223.
- 10. Aggarwal, A.; Ila, H.; Junjappa, H. Synthesis 1982, 65.
- 11. Lue, P.; Greenhill, J.V. Adv. In Heterocycl. Chem. 1997, 67, 707.
- 12. Kashima, C.; Katoh, A.; Yokota, Y.; Omote, Y. Synthesis 1983, 151.
- 13. Dorokhov, V.A.; Komkov, A.V.; Shashkova, E.M.; Bogdanov, V.S.; Bochkareva, M.N. *Russian Chem. Bull.* **1993**, *42*, 1848.
- 14. Komkov, A.V.; Dorokhov, V.A. Russian Chem. Bull. 2002, 51, 1875.
- 15. Dixon, A.E. J. Chem. Soc. 1889, 55, 300.