# QSAR study for diarylguanidines, noncompetitive NMDA receptor antagonists. A new topological index $A_{A d}$ derived from local invariants of the chemical graphs of diarylguanidines 

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Dedicated to Alexandru Balaban on his $75{ }^{\text {th }}$ Anniversary (received 09 Dec 04; accepted 10 Feb 05; published on the web 01 Mar 05)


#### Abstract

The binding affinities for NMDA receptor ion channel and $\sigma$ receptor sites of four classes of diarylguanidines have been correlated with values of molecular descriptors implemented in CODESSA and PRECLAV programs: twenty-one derivatives of $\mathrm{N}, \mathrm{N}^{\prime}$ - diphenylguanidines type (1), eight $\mathrm{N}, \mathrm{N}^{\prime}$ - dinaphthylguanidines type (2), eighteen N - naphthyl - $\mathrm{N}^{\prime}$ - phenylguanidines type (3) and two miscellaneous guanidines (4). A new topological index $\boldsymbol{A}_{\boldsymbol{A d}}$ based on local invariants of the chemical graphs has been defined and used in QSAR studies generating an increasing of correlation coefficient value $\mathbf{r}$.


Keywords: Diarylguanidines, NMDA receptor ion channel site, $\sigma$ receptors, molecular descriptors, CODESSA, PRECLAV, QSAR

## Introduction

L-Glutamate is an excitatory amino acid (EAA) of the central nervous system (CNS), which is stored in presynaptic vesicles. ${ }^{1-5}$ Postsynaptic glutamate receptors divide into two major classes: i) metabotropic receptors and ii) ionotropic receptors. Ionotropic receptors are ligand-gated ion channels mediating rapid changes in postsynaptic membrane sodium or calcium permeability. It is well known that these receptors are divided ${ }^{6}$ into $N$-methyl-D-aspartate (NMDA) ${ }^{1,7-12}$ and $\alpha$ -amino-3-hydroxy-5-methyl-4-isoxazole propionic acid (AMPA) ${ }^{5,13-15}$ receptors. AMPA is mediating Kainic acid (KA) effects so that most studies refer to AMPA/KA receptor. ${ }^{5,13-15}$

The NMDA receptor ion channel site is the most studied because of its availability for both competitive and noncompetitive antagonists series. ${ }^{1,7-12}$ Through excessive long-term stimulation this receptor seems to be implicated in progressive neuronal losses such as epilepsy ${ }^{1}$, Alzheimer's ${ }^{16-19}$ and Huntington's ${ }^{14-18,20}$ diseases, nerve cell death in stroke ${ }^{1,5}$, amyotrophic lateral sclerosis ${ }^{22}$. Many ligands that interact at NMDA receptor ion channel site ${ }^{10,23-29}$ also bind to $\sigma$ receptors ( $\mathrm{N}, \mathrm{N}^{\prime}$-di-o-tolylguanidine, DTG$)^{9,29} . \sigma$ Receptors do not appear to be associated with the NMDA receptor ion channel site and their physiological function is still unknown. ${ }^{7}$

Diarylguanidines represent a new class of compounds proven to be noncompetitive antagonists for NMDA receptor ion channel site in electrophysiological experiments blocking the response of this receptor. Some of these compounds are neuroprotective agents with therapeutic value in patients suffering from stroke, brain or spinal cord trauma and hypoglycemia. ${ }^{7,9,30,31}$

The present QSAR study includes forty-nine diarylguanidines belonging to four classes: twenty-one compounds of the $\mathrm{N}, \mathrm{N}^{\prime}$ - Diphenylguanidines type (1), eight compounds of the N , $\mathrm{N}^{\prime}$ - Dinaphthylguanidines type (2) , eighteen compounds of the N - Naphthyl - $\mathrm{N}^{\prime}$ phenylguanidines type (3) and two miscellaneous guanidines of type (4). The binding affinity of these diarylguanidines for NMDA receptor ion channel and $\sigma$ receptor was correlated with a variety of molecular descriptors implemented in CODESSA and PRECLAV programs and with a new topological index $\boldsymbol{A}_{\boldsymbol{A d}}$ derived from local invariants of the chemical graphs.


1


2


3


4

Figure 1. Structure of diarylguanidine derivatives.

## QSAR results

In this study, the biological activity for forty-nine diarylguanidines was calculated using equation 1. Experimental values for $\mathrm{IC}_{50}$ were determined by Keana and coworkers. ${ }^{7}$

$$
\mathrm{A}=\log \left(383000 / \mathrm{IC}_{50}\right)
$$

QSAR studies for NMDA receptor ion channel site interaction with the specific radioligand $\left[{ }^{3} \mathrm{H}\right]-(+)-5(\mathrm{~S})$-methyl-10(R),11-dihydro-5H-dibenzo[a,b]cyclohepten-5,10-imine (MK-801, 5) and the $\sigma$ receptor-specific radioligand $\left[{ }^{3} \mathrm{H}\right]$-di-o-tolylguanidine (DTG, 6) interaction were
obtained using CODESSA and PRECLAV programs, cross-validation being made by the "leave-one-out" method in both cases.


5


6

Figure 2. Structure for NMDA receptor ion channel site interaction with the specific radioligand $\left[{ }^{3} \mathrm{H}\right]-(+)-5(\mathrm{~S})$-methyl-10(R),11-dihydro-5H-dibenzo[a,b]cyclohepten-5,10-imine (MK801, 5) and the $\sigma$ receptor-specific radioligand $\left[{ }^{3} \mathrm{H}\right]$-di-o-tolylguanidine (DTG, 6).

The main differences between the CODESSA and PRECLAV programs are:
a) Fifty per cent of molecular descriptors calculated and taken into account by these programs are different.
b) The PRECLAV program calculates and use grid-field descriptors while the CODESSA program does not use such descriptors.
c) For the selection of descriptors the PRECLAV program uses CLASS function instead of the criteria used by CODESSA program (only when it is working with testing sets); due to the use of CLASS functions PRECLAV program generates different QSAR equations for different testing sets.
d) In the case of PRECLAV the main selection criterion of QSAR equations is the value of a cross-validation function while CODESSA does not use this criterion.
e) The PRECLAV program uses only BMLR (Best MultiLinear Regression) while CODESSA allow the use of a wide range of statistical methods (Heuristic, PLS, PCA)

## Results obtained with CODESSA program

The first set of QSAR correlations were done using the CODESSA program developed by Katritzky, Karelson and Lobanov. ${ }^{32-34}$

For $\left[{ }^{3} \mathrm{H}\right]-5$ (MK-801) the regression equation has the form:

$$
\begin{equation*}
\log A=a_{0}+\sum a_{i} x_{a i} \tag{2}
\end{equation*}
$$

Values of coefficients obtained with the CODESSA program and definition of variables are given in Table 1. In Figure 3 the experimental versus estimated $\log A$ data correlation is presented.

Table 1. Coefficients and variables in equation 2

| i | $\mathrm{a}_{i}$ | $\mathrm{X}_{\text {ai }}$ | Definition of $\mathrm{x}_{\mathrm{a} i}$ |
| :---: | :---: | :---: | :---: |
| 0 | 2.882 |  |  |
| 1 | -5.377 | $\mathrm{NBO}_{\text {Avg }}$ | - Average bond order of a N atom |
| 2 | -38.58 | $\mathrm{MI}_{\mathrm{A}}$ | - Principal moment of inertia A |
| 3 | 0.121 | $\mathrm{NASE}_{\text {Min }}$ | - Minimum atomic state energy for a N atom |
| 4 | -91.43 | FHDCA / HDCA | - FHDCA Fractional HDCA (HDCA/TMSA) [Quantum-Chemical PC] |
| 5 | - 26.95 | $\mathrm{CERI}_{\text {Max }}$ | - Maximum electrophilic reactivity index for a C atom |
| 6 | -83.72 | $\mathrm{CPC}_{\text {Min }}$ | - Minimum partial charge for a C atom [Zefirov's PC] |
| 7 | -11.30 | FNSA-2 / PNS A | - FNSA-2 Fractional PNSA (PNSA-2/TMSA) [Zefirov's PC] |
| 8 | $-9.751 \cdot 10^{-4}$ | WPSA-2 / PPSA | - WPSA-2 Weighted PPSA <br> (PPSA2*TMSA/1000) [Quantum-Chemical PC] |
| 9 | 3.644 | SIGMA-PI ${ }_{\text {Max Bo }}$ | - Maximum SIGMA-PI bond order |
| 10 | 6.147 | RNDB | - Relative number of double bonds |

The statistical parameters of the fit are:
$\mathrm{N}=49 \quad \mathrm{r}=0.880 \quad \mathrm{~s}=0.451 \quad \mathrm{~F}=13.05 \quad \mathrm{Q}=1.951 \quad \mathrm{R}^{2}=0.774 \quad$ cross validated

$$
R^{2}=0.672
$$

where N is the number of data points, R and $\mathrm{R}^{2}$ denote correlation coefficients, s is the standard deviation of the fit, F is the Fisher test and Q is the quality factor.

## Results obtained with the PRECLAV program

The second set of QSAR correlations was obtained using the PRECLAV program developed by L. Tarko. ${ }^{35,36}$

Results with PRECLAV program for [ $\left.{ }^{3} \mathrm{H}\right]-5$ (MK-801)

$$
\begin{equation*}
\log A=b_{0}+\sum b_{i} x_{b i} \tag{3}
\end{equation*}
$$

Values of coefficients obtained with the PRECLAV program and definition of variables are given in Table 2. In Figure 4 the experimental versus estimated $\log A$ data correlation is presented.

Table 2. Coefficients and variables in equation 3

| i | $\mathrm{b}_{\mathrm{i}}$ | $\mathrm{x}_{\mathrm{bi}}$ | Definition of $\mathrm{x}_{\mathrm{bi}}$ |
| :---: | :---: | :---: | :--- |
| 0 | 2.674 |  |  |
| 1 | 121.6 | nhs | Number of NH single or faint bonds / Number of <br> bonds |
| 2 | -34.74 | F 63 | Resultant electrostatic force on probe atom 63 <br> 3 |
| 16.27 | hd | E(lumo+1) - E(homo-1) gap |  |
| 4 | -3.769 | M 36 | Average chemical bonds parallax for probe atom 36 |
| 5 | -1.319 | zgg | Zagreb topologic index / Heavy atoms number |
| 6 | 13.62 | R 29 | Rejection force sum on probe atom 29 |
| 7 | -13.72 | A 109 | Attraction force sum on probe atom 109 |
| 8 | 32.30 | A 60 | Attraction force sum on probe atom 60 |
| 9 | -1.377 | P 26 | Maximum chemical bonds parallax for probe atom 26 |
| 10 | 2.138 | P 97 | Maximum chemical bonds parallax for probe atom 97 |

The statistical parameters of the fit are:

$$
\begin{array}{llllll}
\mathrm{N}=49 & \mathrm{r}=0.932 & \mathrm{~s}=0.308 & \mathrm{~F}=25.56 & \mathrm{Q}=3.026 & \mathrm{R}^{2}=0.868 \\
& \text { cross validated } \\
& \mathrm{R}^{2}=0.802
\end{array}
$$

where N is the number of data points, R and $\mathrm{R}^{2}$ denote correlation coefficients, s is the standard deviation of the fit, F is the Fisher test and Q is the quality factor.

In Table 3 are listed the experimental and estimated biological activities of diarylguanidines obtained with CODESSA program (eq.2) and PRECLAV program (eq.3) for NMDA receptor ion channel.

Table 3. Experimental vs. estimated biological activity of diarylguanidines for NMDA receptor ion channel site obtained with CODESSA program (eq.2) and PRECLAV program (eq.3)

| No. | Structure | X | R | $\mathrm{R}_{1}$ | $\operatorname{logA}{ }^{7}$ <br> (obs.) | $\operatorname{logA}$ <br> (calc.) <br> eq.2 | $\log \mathrm{A}$ <br> (calc.) <br> eq. |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1 |  | $2-\mathrm{CH}_{3}$ | H | - | 1.623 | 2.181 | 1.319 |
| 2 | H | H | - | 2.054 | 1.599 | 1.931 |  |
| 3 | $\mathbf{1}$ | $2-\mathrm{C}_{2} \mathrm{H}_{5}$ | H | - | 2.669 | 2.586 | 3.018 |
| 4 |  | $2-\mathrm{CH}_{\left(\mathrm{CH}_{3}\right)_{2}}$ | H | - | 3.247 | 3.187 | 3.268 |
| 5 |  | $2-\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}$ | H | - | 1.000 | 0.906 | 1.098 |
| 6 | $2-\mathrm{I}$ | H | - | 3.225 | 2.658 | 3.144 |  |
| 7 |  | $2-\mathrm{OCH}_{3}$ | H | - | 2.379 | 2.508 | 2.487 |
| 8 |  | $2-\mathrm{C}_{6} \mathrm{H}_{5}$ | H | - | 1.583 | 2.166 | 1.115 |

Table 3. Continued

| No. | Structure | X | R | R 1 | logA7 <br> (obs.) | logA <br> (calc.) |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  |  | logA <br> (calc.) |  |  |
| 9 | $3-\mathrm{CH}_{3}$ | H | - | 3.058 | 2.748 | 3.293 |
| eq. |  |  |  |  |  |  |

Table 3. Continued

| No. | Structure | X | R | R1 | $\log \mathrm{A} 7$ <br> (obs.) | $\log A$ (calc.) eq. 2 | $\log A$ (calc.) eq. 3 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 41 |  | $3-\mathrm{N}_{3}$ | H | $\mathrm{CH}_{3}$ | 3.770 | 4.035 | 3.876 |
| 42 |  | $3-\mathrm{NO}_{3}$ | CH3 | H | 2.136 | 2.248 | 2.041 |
| 43 |  | $3-\mathrm{NH}_{2}$ | CH3 | H | 1.583 | 2.798 | 1.983 |
| 44 |  | $3-\mathrm{N}_{3}$ | CH3 | H | 2.355 | 2.702 | 2.826 |
| 45 |  | H | H | CH3 | 2.570 | 2.616 | 3.059 |
| 46 |  | $3-\mathrm{C}_{2} \mathrm{H}_{5}$ | $\mathrm{CH}_{3}$ | CH3 | 3.583 | 4.026 | 3.595 |
| 47 |  | $3-\mathrm{C}_{2} \mathrm{H}_{5}$ | $\mathrm{C}_{2} \mathrm{H}_{5}$ | CH3 | 3.596 | 3.363 | 3.939 |
| 48 | 4 |  |  |  | 2.309 | 2.254 | 2.247 |
| 49 |  |  |  |  | 2.121 | 1.847 | 2.356 |



Figure 3. Experimental vs. estimated biological activity of diarylguanidines for NMDA receptor ion channel site obtained with CODESSA.


Results with CODESSA program for $\left[{ }^{3} H\right]-6$ (DTG)
For $\left[{ }^{3} \mathrm{H}\right]-6$ (DTG) the regression equation has the form:

$$
\begin{equation*}
\log A=c_{0}+\sum c_{i} x_{c i} \tag{4}
\end{equation*}
$$

Values of coefficients obtained with the CODESSA program and definition of variables are given in Table 4. In Figure 5 the experimental versus estimated $\log A$ data correlation is presented.

Table 4. Coefficients and variables in equation 4

| i | $\mathrm{c}_{i}$ | $\mathrm{x}_{\mathrm{c} i}$ | Definition of $\mathrm{x}_{\mathrm{ci}}$ |
| :---: | :---: | :---: | :---: |
| 0 | 0.799 |  |  |
| 1 | 103.3 | HDCA-2/SQRT | - HA dependent HDCA-2/SQRT(TMSA) [Zefirov's PC] |
| 2 | 2.935 | FNSA-2/PNSA | - FNSA-2 Fractional PNSA (PNSA-2/TMSA) [QuantumChemical PC] |
| 3 | -5.732 | SIGMA-PI ${ }_{\text {Max }}$ bo | - Maximum SIGMA-PI bond order |
| 4 | -329.4 | FPSA-3 / PPSA | - FPSA-3 Fractional PPSA (PPSA-3/TMSA) [Zefirov's $\mathrm{PC}]$ |
| 5 | -0.093 | HDSA-2 | - HA dependent HDSA-2 [Quantum-Chemical PC] |
| 6 | 4.522 | TEI | - Topographic electronic index (all bonds) [Zefirov's PC] |
| 7 | 80.13 | $\mathrm{NPCA}_{\text {Min }}$ | - Minimum partial charge for a N atom [Zefirov's PC] |
| 8 | 5.125 | $\mathrm{CBO}_{\text {Avg }}$ | - Average bond order of a C atom |
| 9 | -0.127 | KSI | - Kier shape index (order 1) |
| 10 | 2.437 | $\mathrm{NVA}_{\text {Avg }}$ | - Average valence of a N atom |

The statistical parameters of the fit are:

$$
\begin{array}{lllllll}
\mathrm{N}=49 & \mathrm{r}=0.914 & \mathrm{~s}=0.423 & \mathrm{~F}=19.32 & \mathrm{Q}=2.160 & \mathrm{R}^{2}=0.836 & \text { cross } \quad \text { validated } \\
& & &
\end{array}
$$

where N is the number of data points, R and $\mathrm{R}^{2}$ denote correlation coefficients, s is the standard deviation of the fit, F is the Fisher test and Q is the quality factor.

Results with PRECLAV program for [ $\left.{ }^{3} \mathrm{H}\right]-6$ (DTG)

$$
\begin{equation*}
\log A=d_{0}+\sum d_{i} x_{d i} \tag{5}
\end{equation*}
$$

Values of coefficients obtained with the PRECLAV program and definition of variables are given in Table 5. In Figure 6 the experimental versus estimated $\log A$ data correlation is presented.

Table 5. Coefficients and variables in equation 5

| i | $\mathrm{d}_{\mathrm{i}}$ | $\mathrm{x}_{\mathrm{di}}$ | Definition of $\mathrm{x}_{\mathrm{di}}$ |
| :---: | :---: | :---: | :--- |
| 0 | 8.593 |  |  |
| 1 | -15.98 | cns | Number of CN single or faint bonds / Number of bonds |
| 2 | -2060 | F 60 | Resultant electrostatic force on probe atom 60 |
| 3 | -1.810 | P 87 | Maximum chemical bonds parallax for probe atom 87 |
| 4 | -0.122 | lga | Number of aromatic bonds |
| 5 | -187.6 | R 40 | Rejection force sum on probe atom 40 |
| 6 | -17.46 | R 58 | Rejection force sum on probe atom 58 |
| 7 | -0.427 | nri | Number of I atoms |
| 8 | 5.436 | P 7 | Maximum chemical bonds parallax for probe atom 7 |
| 9 | -1.509 | lup | E(lumo+1) energy |
| 10 | 0.394 | cna | Percent of carbon • Minimum charge for C atoms |

The statistical parameters of the fit are:
$\begin{array}{llllll}\mathrm{N}=49 & \mathrm{r}=0.938 & \mathrm{~s}=0.319 & \mathrm{~F}=28.45 & \mathrm{Q}=2.940 & \mathrm{R}^{2}=0.880\end{array} \quad \begin{aligned} & \text { cross validated } \\ & \mathrm{R}^{2}=0.814\end{aligned}$
where N is the number of data points, R and $\mathrm{R}^{2}$ denote correlation coefficients, s is the standard deviation of the fit, F is the Fisher test and Q is the quality factor.

In Table 6 are listed the experimental and estimated biological activities of diarylguanidines obtained with CODESSA program (eq.4) and PRECLAV program (eq.5) for $\sigma$ receptor.

Table 6. Experimental vs. estimated biological activity of diarylguanidines for $\sigma$ receptor obtained with CODESSA program (eq.4) and PRECLAV program (eq.5)

| No | Com pounds | X | R | $\mathrm{R}_{1}$ | $\begin{aligned} & \log \mathrm{A}^{7} \\ & \text { (obs.) } \end{aligned}$ | $\begin{gathered} \log \mathrm{A} \\ \text { (calc.) } \\ \text { eq. } 4 \end{gathered}$ | $\begin{gathered} \log \mathrm{A} \\ \text { (calc.) } \\ \text { eq. } 5 \\ \hline \end{gathered}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1 |  | $2-\mathrm{CH}_{3}$ | H | - | 4.091 | 3.626 | 2.742 |
| 2 |  | H | H | - | 2.984 | 3.151 | 3.222 |
| 3 |  | $2-\mathrm{C}_{2} \mathrm{H}_{5}$ | H | - | 4.437 | 4.080 | 4.285 |
| 4 |  | $2-\mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ | H | - | 3.638 | 3.573 | 3.232 |
| 5 | 1 | $2-\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}$ | H | - | 3.031 | 2.623 | 2.832 |
| 6 |  | 2-I | H | - | 4.469 | 4.159 | 4.410 |
| 7 |  | $2-\mathrm{OCH}_{3}$ | H | - | 2.284 | 3.069 | 2.377 |
| 8 |  | $2-\mathrm{C}_{6} \mathrm{H}_{5}$ | H | - | 1.674 | 1.942 | 1.720 |
| 9 |  | $3-\mathrm{CH}_{3}$ | H | - | 3.949 | 3.086 | 4.290 |

Table 6. Continued

| No | Com pounds | X | R | R1 | $\begin{aligned} & \log A 7 \\ & \text { (obs.) } \end{aligned}$ | $\begin{gathered} \log \mathrm{A} \\ \text { (calc.) } \\ \text { eq. } 4 \end{gathered}$ | $\begin{gathered} \log \mathrm{A} \\ \text { (calc.) } \\ \text { eq. } 5 \end{gathered}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 10 |  | 3-C2H5 | H | - | 4.680 | 3.885 | 4.110 |
| 11 |  | $3-\mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ | H | - | 3.600 | 3.957 | 3.941 |
| 12 |  | $\begin{gathered} 3- \\ \left(\mathrm{CH}_{2}\right)_{2} \mathrm{CH}_{3} \end{gathered}$ | H | - | 3.959 | 3.828 | 4.072 |
| 13 |  | 3-I | H | - | 3.486 | 4.129 | 3.676 |
| 14 |  | $3-\mathrm{OCH}_{3}$ | H | - | 3.037 | 2.904 | 3.245 |
| 15 |  | $4-\mathrm{CH}_{3}$ | H | - | 2.854 | 3.607 | 3.368 |
| 16 |  | $4-\mathrm{C}_{2} \mathrm{H}_{5}$ | H | - | 3.194 | 2.983 | 3.060 |
| 17 |  | 4- $\mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ | H | - | 3.199 | 3.426 | 3.334 |
| 18 |  | $4-\mathrm{Br}$ | H | - | 4.064 | 3.971 | 3.979 |
| 19 |  | $2-\mathrm{CH}_{3}$ | $\mathrm{CH}_{3}$ | - | 1.785 | 2.064 | 2.323 |
| 20 |  | $3-\mathrm{CH}_{3}$ | $\mathrm{CH}_{3}$ | - | 3.190 | 2.862 | 2.988 |
| 21 |  | $3-\mathrm{C}_{2} \mathrm{H}_{5}$ | $\mathrm{CH}_{3}$ | - | 3.669 | 3.618 | 3.416 |
| 22 | 2 | - | H | H | 3.365 | 2.948 | 3.153 |
| 23 |  | - | H | $\mathrm{CH}_{3}$ | 1.902 | 2.030 | 1.595 |
| 24 |  | - | H | $\mathrm{C}_{2} \mathrm{H}_{5}$ | 1.795 | 2.400 | 2.053 |
| 25 |  | - | H | $\mathrm{C}_{6} \mathrm{H}_{5}$ | 1.683 | 1.498 | 1.668 |
| 26 |  | - | $\mathrm{CH}_{3}$ | $\mathrm{CH}_{3}$ | 1.553 | 1.223 | 1.647 |
| 27 |  | - | $\mathrm{CH}_{3}$ | $\mathrm{C}_{2} \mathrm{H}_{5}$ | 1.654 | 1.616 | 1.293 |
| 28 |  | - | $\mathrm{CH}_{3}$ | $\mathrm{C}_{6} \mathrm{H}_{5}$ | 1.541 | 1.074 | 1.256 |
| 29 |  |  | $E$ |  | 3.619 | 2.980 | 3.342 |
| 30 | 3 | $3-\mathrm{C}_{2} \mathrm{H}_{5}$ | H | H | 3.850 | 4.048 | 4.101 |
| 31 |  | $2-\mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ | H | H | 3.624 | 3.508 | 3.147 |
| 32 |  | 2-I | H | H | 3.981 | 3.567 | 3.719 |
| 33 |  | $3-\mathrm{CH}_{3}$ | H | H | 3.708 | 3.788 | 3.705 |
| 34 |  | $3-\mathrm{C}_{2} \mathrm{H}_{5}$ | H | $\mathrm{CH}_{3}$ | 2.178 | 2.718 | 2.323 |
| 35 |  | $3-\mathrm{C}_{2} \mathrm{H}_{5}$ | H | $\mathrm{C}_{2} \mathrm{H}_{5}$ | 2.242 | 2.161 | 2.539 |
| 36 |  | $3-\mathrm{C}_{2} \mathrm{H}_{5}$ | $\mathrm{CH}_{3}$ | H | 2.583 | 2.724 | 2.643 |
| 37 |  | $2-\mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ | $\mathrm{CH}_{3}$ | H | 1.722 | 1.947 | 2.247 |
| 38 |  | $3-\mathrm{CH}_{3}$ | H | $\mathrm{CH}_{3}$ | 2.313 | 2.533 | 2.418 |
| 39 |  | $3-\mathrm{NO}_{3}$ | H | $\mathrm{CH}_{3}$ | 1.786 | 2.037 | 2.027 |
| 40 |  | $3-\mathrm{NH}_{2}$ | H | $\mathrm{CH}_{3}$ | 2.151 | 2.533 | 1.961 |
| 41 |  | $3-\mathrm{N}_{3}$ | H | $\mathrm{CH}_{3}$ | 2.103 | 2.037 | 1.765 |

Table 6. Continued

| 42 |  | $3-\mathrm{NO} 3$ | CH 3 | H | 2.161 | 2.437 | 2.100 |
| :--- | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 43 |  | $3-\mathrm{NH} 2$ | CH 3 | H | 1.469 | 1.944 | 2.369 |
| 44 | 3 | $3-\mathrm{N} 3$ | CH 3 | H | 2.145 | 2.053 | 2.179 |
| 45 |  | H | H | CH 3 | 2.120 | 1.706 | 2.635 |
| 46 |  | $3-\mathrm{C}_{2} \mathrm{H}_{5}$ | $\mathrm{CH}_{3}$ | $\mathrm{CH}_{3}$ | 2.500 | 1.994 | 2.482 |
| 47 | $3-\mathrm{C}_{2} \mathrm{H}_{5}$ | $\mathrm{C}_{2} \mathrm{H}_{5}$ | $\mathrm{CH}_{3}$ | 2.223 | 2.590 | 2.245 |  |
| 48 |  |  |  |  |  |  |  |



Figure 5. Experimental vs. estimated biological activity of diarylguanidines for $\sigma$ receptor obtained with CODESSA.

Figure 6. Experimental vs. estimated biological activity of diarylguanidines for $\sigma$ receptor obtained with PRECLAV.

## New topological index $A_{\text {Ad }}$ derived from local invariants

A graph is a pair $G=(V, E)$ of sets satisfying $E \subseteq[V]^{2}$; thus the elements of $E$ are 2-element subsets of V . The elements $V$ are the vertices and $E$ are the edges of the graph $G$. When $V$ represents the atoms of a molecule and elements of $E$ symbolize covalent bonds between pairs of atoms, then $G$ becomes a molecular graph.$^{37}$ A graph invariant is a topological property that is conserved by molecular isomorphism. An inclusive graph invariant is one, which is the same for two or more graphs (i. e. molecules) and represents a degree of molecular similarity. ${ }^{38-41,43}$

To obtain a local invariant set $\boldsymbol{X}$ one method is to solve a linear system of equations:

$$
\begin{equation*}
Q \cdot X=R \tag{6}
\end{equation*}
$$

where $\boldsymbol{Q}$ is a matrix derived from the adjacency matrix, $\boldsymbol{R}$ is a column vector and $\boldsymbol{X}$ is the column vector of local invariants $x_{i} .{ }^{42,43}$

In the present QSAR study, we considered the distance matrix $(\boldsymbol{D})$ obtained using MOPAC program as $\boldsymbol{Q}$ matrix, the diagonal terms being replaced with the value of the topological index $\boldsymbol{J}$ (Balaban) ${ }^{39,40,44-47}$ for the considered molecule. Column vector $\boldsymbol{R}$ was obtained multiplying the adjacency matrix ( $\boldsymbol{A}$ ) corresponding to the chemical graph with a column vector $\mathbf{Z}$ whose elements are the atomic numbers $\boldsymbol{z}_{\boldsymbol{i}}$. The equations (eq. 7) becomes:

$$
D \cdot X=A \cdot Z
$$

Using the local invariant set $\boldsymbol{X}$ we built a new topological index $\boldsymbol{A}_{\boldsymbol{A d}}$ named Beteringhe-FilipTarko index, defined in equation 8, which has been included in the QSAR study of the interaction of diarylguanidines with the two types of receptors:

$$
\begin{equation*}
\boldsymbol{A}_{\boldsymbol{A d}}=(q / q+N) \Sigma \log \left(x_{i}\right)^{2} \tag{8}
\end{equation*}
$$

where $\boldsymbol{q}$ is the number of edges in the molecular graph, $\boldsymbol{N}$ is the number of vertices in the molecular graph and $x_{i}$ are the local vertex invariants.

A better correlation of binding affinity $(\log \mathrm{A})$ of diarylguanidines for $\sigma$ receptor was observed including the new Beteringhe-Filip-Tarko topological index $\boldsymbol{A}_{\boldsymbol{A d}}$ in the QSAR equation.

Results with CODESSA program for $\left[{ }^{3} \mathrm{H}\right]-6$ (DTG) including $\boldsymbol{A}_{\boldsymbol{A d}}$ in the correlation

$$
\begin{equation*}
\log A=e_{0}+\sum e_{i} X_{e i} \tag{9}
\end{equation*}
$$

Values of coefficients and definition of variables are given in Table 7. In Table 8 are listed the experimental and estimated biological activities of diarylguanidines for $\sigma$ receptor. In Figure 7 the experimental versus estimate $\log A$ data correlation is presented.

Table 7. Coefficients and variables in equation 9

| i | $\mathrm{e}_{\mathrm{i}}$ | $\mathrm{X}_{\text {ei }}$ | Definition of $\mathrm{x}_{\mathrm{ei}}$ |
| :---: | :---: | :---: | :---: |
| 0 | 31.56 |  |  |
| 1 | 129.9 | HDCA-2 / SQRT | HA dependent HDCA-2/SQRT(TMSA) [Zefirov's PC] |
| 2 | 2.332 | FNSA-2 / <br> PNSA | FNSA-2 Fractional PNSA (PNSA-2/TMSA) [Quantum-Chemical $\mathrm{PC}]$ |
| 3 | -. 023 | HDSA-1 | HA dependent HDSA-1 [Quantum-Chemical PC] |
| 4 | - 262.5 | FPSA-3 / PPSA | FPSA-3 Fractional PPSA (PPSA-3/TMSA) [Zefirov's PC] |
| 5 | 2.596 | TEI | Topographic electronic index (all bonds) [Zefirov's PC] |
| 6 | 0.130 | Min n-nR $\mathrm{C}_{\text {- }}$ | Minimum n-n repulsion for a C-C bond |
| 7 | 3.613 | $\mathrm{CBO}_{\text {Avg }}$ | Average bond order of a C atom |
| 8 | -0.179 | $\boldsymbol{A}_{\text {Ad }}$ | Beteringhe-Filip-Tarko topological index |
| 9 | -0.246 | $\mathrm{NASE}_{\text {Max }}$ | Maximum atomic state energy for a N atom |
| 10 | -6.633 | $\begin{gathered} \text { FHDSA / } \\ \text { HDSA } \end{gathered}$ | FHDSA Fractional HDSA (HDSA/TMSA) [Quantum-Chemical $\mathrm{PC}]$ |

The statistical parameters of the fit are:
$\mathrm{N}=49 \quad \mathrm{r}=0.942 \quad \mathrm{~s}=0.412 \quad \mathrm{~F}=24.4 \quad \mathrm{Q}=2.286 \quad \mathrm{R}^{2}=0.887 \quad$ cross validated $\mathrm{R}^{2}=0.795$ where N is the number of data points, R and $\mathrm{R}^{2}$ denote correlation coefficients, s is the standard deviation of the fit, F is the Fisher test and Q is the quality factor.

Table 8. Values of topological index $\mathrm{A}_{\mathrm{Ad}}$ and experimental vs. estimated biological activity of diarylguanidines for $\sigma$ receptor

| No. | Compounds | X | R | $\mathrm{R}_{1}$ | $\mathbf{A}_{\mathbf{A d}}$ | LogA ${ }^{7}$ <br> (obs.) | $\operatorname{logA}$ <br> (calc.) |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| 1 | $2-\mathrm{CH}_{3}$ | H | - | -0.205 | 4.091 | 4.073 |  |
| 2 | H | H | - | 0.515 | 2.984 | 3.359 |  |
| 3 |  | $2-\mathrm{C}_{2} \mathrm{H}_{5}$ | H | - | 0.264 | 4.437 | 3.902 |
| 4 |  | $2-\mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ | H | - | 0.229 | 3.638 | 3.342 |
| 5 |  | $2-\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}$ | H | - | 0.127 | 3.031 | 2.910 |
| 6 | $\mathbf{1}$ | $2-\mathrm{I}$ | H | - | 1.214 | 4.469 | 4.243 |
| 7 |  | $2-\mathrm{OCH}_{3}$ | H | - | 0.646 | 2.284 | 2.857 |
| 8 |  | $2-\mathrm{C}_{6} \mathrm{H}_{5}$ | H | - | 2.002 | 1.674 | 2.138 |
| 9 |  | $3-\mathrm{CH}_{3}$ | H | - | 0.881 | 3.949 | 3.475 |
| 10 |  | $3-\mathrm{C}_{2} \mathrm{H}_{5}$ | H | - | 0.667 | 4.680 | 4.268 |
| 11 | $3-\mathrm{CH}^{2}\left(\mathrm{CH}_{3}\right)_{2}$ | H | - | 1.870 | 3.600 | 4.014 |  |

Table 8. Continued
$\left.\begin{array}{lllllll}\hline \text { No. } & \text { Compounds } & \mathrm{X} & \mathrm{R} & \mathrm{R} 1 & \mathrm{AAd} & \begin{array}{l}\text { LogA7 } \\ \text { (obs.) }\end{array} \\ \hline 12 & 3-\left(\mathrm{CH}_{2}\right)_{2} \mathrm{CH}_{3} & \mathrm{H} & - & -1.199 & 3.959 & 3.828 \\ \text { (calc.) }\end{array}\right]$

Table 8. Continued

| No. Compounds | X | R | R 1 | AAd | LogA7 <br> (obs.) | $\log \mathrm{A}$ <br> (calc.) |  |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| 47 | $3-\mathrm{C}_{2} \mathrm{H}_{5}$ |  | 0.608 | 2.261 | 2.584 |  |  |
| 48 |  |  |  |  |  |  |  |



Figure 7. Experimental vs. estimated biological activity of diarylguanidines for $\sigma$ receptor when in the QSAR equation $\boldsymbol{A}_{\boldsymbol{A d}}$ is used.

## Discussion

In Tables 1-8 QSAR correlation equations are presented, either with experimental and calculated values for the binding affinity $(\log A)$ of forty-nine diarylguanidines for the NMDA receptor ion channel site and the $\sigma$ receptor. The diagrams presented in Figures 3-7 show a linear dependence between experimental and calculated values of binding affinity $(\log \mathrm{A})$, the corresponding linear correlation equation is also presented. Comparing the values of correlation coefficient $\mathbf{r}$ obtained using the CODESSA and PRECLAV programs for the same number of descriptors, somewhat higher values were obtained when using PRECLAV program ( $\mathrm{r}=0.932$ in case of NMDA
receptor ion channel site and 0.938 in case of $\sigma$ receptor) than when CODESSA program was used ( $\mathrm{r}=0.880$ in case of NMDA receptor ion channel site and 0.914 in case of $\sigma$ receptor).

In the study performed with CODESSA, quantum-chemical descriptors (NBOAvg - Average bond order of a N atom - being retained as specific descriptor in the models for both receptors) have the highest weight in QSAR models for both receptors while in the study performed with PRECLAV, grid-field descriptors have the highest weight.

Grid electrostatic descriptors are resultant forces (F), repulsive (R) and attractive (A), between net atomic charges of the studied molecules and probe atoms placed in the evenly distributed points of a virtual network. Forces F, R and A represent a measure of the interaction of the molecule with the active site of the receptor in a certain zone of space.

Grid geometric descriptors are the parallaxes of atom pairs meaning the angle under which can be seen that pair of atoms from a point of the virtual network; for every network point the parallaxes are calculated for all atom pairs, then maximum parallaxes $(\mathrm{P})$ and average parallaxes $(\mathrm{M})$; the parallaxes are a measure of the section area of the molecule seen from a certain point of space.

The fact that in the final QSAR equations obtained with PRECLAV more grid descriptors are included than global descriptors, shows that electrostatic interactions and host-guest like interactions of the molecules with the active site are more important than other molecular characteristics (molecular mass, number of atoms of a certain type, type of chemical bonds).

When the new Beteringhe-Filip-Tarko topological index $\boldsymbol{A}_{\boldsymbol{A d}}$, derived from local invariants of the chemical graphs of the diarylguanidines, is used an increase of the correlation coefficient $(\mathrm{r}=$ 0.942 ) is observed in the QSAR model corresponding to the $\sigma$ receptor done using CODESSA, leading to the idea that a better structural characterization of diarylguanidines is represented in this new index. This is accountable to the fact that the new Beteringhe-Filip-Tarko index $\boldsymbol{A}_{\boldsymbol{A d}}$ was devised using the geometric characteristics of the molecule.

## Conclusions

In this study the binding affinities for the NMDA receptor ion channel and $\sigma$ receptor sites of four classes of diarylguanidines have been correlated with values of molecular descriptors implemented in CODESSA and PRECLAV programs: twenty-one derivatives of $\mathrm{N}, \mathrm{N}^{\prime}-$ Diphenylguanidines type (1), eight $N, N^{\prime}$ - Dinaphthylguanidines type (2), eighteen $N$ - Naphthyl - $\mathrm{N}^{\prime}$ - phenylguanidines type (3) and two miscellaneous guanidines (4).

Correlation coefficient, $\mathbf{r}$, has a higher value when the PRECLAV program is used for building the QSAR model for the NMDA receptor ion channel site and $\sigma$ receptor, due to the use of grid/field descriptors which show that electrostatic/host-guest interactions of the molecules with a certain site are more important than other molecular specific features.

Introduction of the new Beteringhe-Filip-Tarko topological index $\mathbf{A}_{\mathbf{A d}}$ in QSAR model for $\sigma$ receptor increased the predictive value of the model.

## References

1. Muir, K. W.; Lees, K. R. Stroke 1995, 26, 503.
2. Curtis, D. R.; Watkins, J. C. J. Neurochem. 1960, 6, 117.
3. Krnjevic, K.; Phillis, J. W. J. Physiol. 1963, 165, 274.
4. Takeuchi, A.; Takeuchi, N. J. Physiol. 1964, 170, 296.
5. Ohmori, J.; Sakamoto, S.; Kubota, H.; Sasamata, M. S.; Okada, M.; Kawasaki, S.; Hidaka, K.; Togami, J.; Furuya, T.; Murase, K. J. Med. Chem. 1994, 37, 467.
6. Monaghan, D. T.; Bridges, R. J.; Cotman, C. W. Annu. Rev. Pharmacol. Toxicol. 1989, 29, 365.
7. Reddy, N. L.; Hu, L. Y.; Cotter, R. E.; Fischer, J. B.; Wong, W. J.; McBurney, R. N.; Weber, E.; Holmes, D. L.; Wong, S. T.; Presad, R.; Keana, J. F. W. J. Med. Chem. 1994, 37, 260.
8. Salituro, F. G.; Tomlinson, R. C.; Baron, B. M.; Palfreyman, M. G.; McDonald, I. A. J. Med. Chem. 1994, 37, 334.
9. Keana, J. F. W.; McBurney, R. N.; Scherz, M. W.; Fischer, J. B.; Hamilton, P. N.; Smith, S. M.; Server, A. C.; Finkbeiner, S.; Stevens, C. F.; Jahr, C.; Weber, E. Proc. Natl. Acad. Sci. USA 1989, 86, 5631.
10. Sonders, M. S.; Keana, J. F. W.; Weber, E. Trends Neurosci. 1988, 11, 37.
11. Anis, N. A.; Berry, S. C.; Burton, N. R.; Lodge, D. Br. J. Pharmacol. 1983, 79, 565.
12. Rothman, S. M.; Olney, J. W. Trends Neurosci. 1987, $10,299$.
13. Smith, S. E.; Durmuller, N.; Meldrum, B. S. Eur. J. Pharmacol. 1991, 201, 179.
14. Barreca, M. L. ; Gitto, R. ; Quartarone, S. ; De Luca, L. ; De Sarro, G. ; Chimirri, A. J. Chem. Inf. Comput. Sci. 2003, 43, 651.
15. Rogawski, M. A.; Donevan, S. D. Drugs. Adv. Neurol. 1999, 79, 947.
16. Lipton, S. A.; Rosenberg, P. A. N. Engl. J. Med. 1994, 330, 613.
17. Greenmayer, J. T. Arch. Neurol. 1986, 43, 1058.
18. Wieloch, T. Science 1985, 230, 681.
19. Margos, W. F.; Greenamyre, J. T.; Penney, J. B. Jr.; Young, A. B. Trends Neurosci. 1987, 10, 65.
20. Beal, M. F.; Kowall, N. W.; Ellison, D. W.; Mazurek, M. F.; Swartz, K. J.; Martin, J. B. Nature(London) 1986, 321, 168.
21. Choi, D. W. Neurosci. Lett. 1985, 58, 293.
22. Kurland, L. T. Trends Neurosci. 1988, 11, 51.
23. Honey, C. R.; Miljkovic, Z.; McDonald, J. F. Neurosci. Lett. 1985, 61, 135.
24. Benavides, J.; Rivy, J. P.; Carter, C.; Scatton, B. Eur. J. Pharmacol. 1988, 149, 67.
25. Wong, E. H. F.; Kemp, J. A.; Priestley, T. Proc. Natl. Acad. Sci. USA 1986, 83, 7104.
26. Gee, K. R.; Barmettler, P.; Rhodes, M. R.; McBurney, R. N.; Reddy, N. L.; Hu, L. Y.; Cotter, R. E.; Hamilton, P. N.; Weber, E.; Keana, J. F. W. J. Med. Chem. 1993, 36, 1938.
27. Weber, E.; Sonders, M.; Quarum, M.; McLean, S.; Pou, S.; Keana, J. F. W. Proc. Natl. Acad. Sci. USA 1986, 83, 8784.
28. Vincent, J. P.; Kartalovski, B.; Geneste, P.; Kamenka, J. M.; Lazdunski, M. Proc. Natl. Acad. Sci. USA 1979, 76, 4678.
29. Zukin, S. R. ; Zukin, R. S. Proc. Natl. Acad. Sci. USA 1979, 76, 5372.
30. Safir, S. R.; Kushner, S.; Brancone, L. M.; Subbarow, Y. J. Org. Chem. 1948, 13, 924.
31. Gee, K. R.; Durant, G. J.; Holmes, D. L.; Magar, S. S.; Weber, E.; Wong, S. T.; Keana, J. F. W. Bioconjug. Chem. 1993, 4, 226.
32. Katritzky, A. R.; Perumal, S.; Petrukhin, R.; Kleinpeter, E. J. Chem. Inf. Comput. Sci. 2001, 41, 569.
33. Katritzky, A. R.; Petrukhin, R.; Perumal, S.; Karelson, M.; Prakash, I.; Desai, N. Croat. Chem. Acta 2002, 75, 475.
34. Sild, S.; Karelson, M. J. Chem. Inf. Comput. Sci. 2002, 42, 360.
35. Tarko, L. Rev. Roum. Chim. 2000, 45(7-8), 809.
36. Tarko, L. ; Ivanciuc, O. MATCH (Comm. Math. Comp. Chem.) 2001, 44, 201.
37. Basak, S. A.; Balaban, A. T.; Grunwald, G. D.; Gute, B. D. J. Chem. Inf. Comput. Sci. 2000, 40, 891.
38. Balaban A. T., Ed.; Chemical Applications of Graph Theory, Academic Press: London, 1976.
39. Balaban, A. T. "QSAR and computational methods in drug discovery". In: Encyclopedia of Analytical Chemistry Meyers, R. A., Ed.; Wiley: Chichester, 2000; Vol. 8, p7288.
40. Balaban A. T. "A personal view about topological indices for QSAR/QSPR". In: QSAR/QSPR Studies by Molecular Descriptors. Diudea, M, Ed.; Huntington: New York, 2000; pp1-25.
41. Trinajstić, N. Chemical Graph Theory, $2^{\text {nd }}$ Ed.; CRC Press: Boca Raton, FL, 1992; pp 225274.
42. Filip, P. A.; Balaban, T. S.; Balaban, A. T. J. Math. Chem. 1987, 1, 61.
43. Toropov, A. A.; Schultz, T. W. J. Chem. Inf. Comput. Sci. 2003, 43, 560.
44. Balaban, A. T. Chem. Phys. Lett. 1982, 89, 399.
45. Balaban A. T. Pure Appl. Chem. 1983, 55, 199.
46. Beteringhe, A.; Baciu, I.; Caproiu, M. T., Constantinescu, T., Balaban, A. T. J. Planar Chromatogr. 2003, 16, 268.
47. Devillers, J.; Balaban, A. T., Eds; Topological Indices and Related Descriptors in QSAR and QSPR. Gordon and Breach: The Netherlands, 1999.
