The influence of protecting groups on the diastereoselectivity of catalytic heterogeneous hydrogenation of Baylis-Hillman adducts

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This article is dedicated to Prof. Edmundo Ruveda on occasion of his 70th anniversary (received 30 Aug 03; accepted 15 Oct 03; published on the web 22 Oct 03)

Abstract

In this communication we describe our recent results from a study on catalytic heterogeneous hydrogenation reactions of Baylis-Hillman adducts. Depending on the protecting groups used on the secondary hydroxyl group of these adducts, it is possible to obtain a high degree of diastereoselection. For silylated Baylis-Hillman adducts a high *syn* diastereoselectivity has been obtained. However, when secondary Baylis-Hillman hydroxyl groups were unprotected or protected as acetate a moderate *anti* selectivity was attained. The adducts protected as methyl ether gave poor *syn* diastereoselectivity.

Keywords: Baylis-Hillman, diastereoselectivity, heterogeneous hydrogenation

Introduction

The Baylis-Hillman reaction is a simple and straightforward alternative to generate a new C-C σ bond. In the last few years this reaction has attracted the attention of many organic chemists, mainly due to the great synthetic potentiality of the α -methylene- β -hydroxy compounds (Figure 1) provided by it, associated with its operational simplicity (no special experimental conditions such as low temperature, anhydrous solvent or inert atmosphere are required). This reaction may be broadly defined as a coupling between the α -position of activated alkenes and an electron-deficient sp² carbon atom, for instance an aldehyde carbonyl group. The reaction is usually catalyzed by DABCO (1,4-diazabicyclo [2.2.2.] octane) or a phosphine.¹

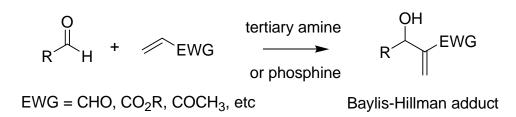


Figure 1. Baylis-Hillman reaction.

The major drawbacks from this reaction are its relative slowness (typically one to 4 weeks to be completed) associated with very low chemical yields (5-10%). To solve this problem various modifications of the experimental protocol have been proposed, e.g., the use of microwaves,² salts and metals³, ionic liquids⁴, an aqueous medium⁵ or the use of a Lewis acid, either as catalyst or as promotor.⁶ We have also demonstrated that the use of ultrasound dramatically accelerates the rate of this reaction.⁷

Recently, we were interested in using a hydrogenation reaction of a Baylis-Hillman adduct as a key step for the total synthesis of a natural pheromone.⁸ To our surprise, the literature contains several papers⁹⁻¹⁵ reporting the utilization of homogeneous catalytic hydrogenation of Baylis-Hillman adducts. Depending on the experimental conditions this hydrogenation reaction provides the *anti* diastereoisomers, with a high degree of diastereoselectivity and the *syn* diastereoisomer with moderate to excellent diastereoselectivity.¹⁵ In contrast there is only one report¹⁶ concerning the use of a heterogeneous catalyst, where no diastereoselectivity has been observed.

In a two preliminary studies on the heterogeneous catalytic hydrogenation of Baylis-Hillman adducts derived from aromatic and aliphatic aldehydes,^{8,17} we demonstrated that it was possible to obtain the *syn* isomer with a high degree of diastereoselectivity. The elevated degree of *syn* diastereoselection seems to be directly related to the presence of a protecting group on the Baylis-Hillman secondary hydroxyl group. These observations have been corroborated by the fact that it was possible to obtain poor to moderate *anti* diastereoselectivity if the Baylis-Hillman adduct were hydrogenated without protection of the secondary hydroxyl group (Scheme 1).

More recently, based on our preliminary results, Bouzide demonstrated that the utilisation of a chelating salt, such as MgBr₂, could also dramatically improve the *syn* diastereoselectivity.¹⁸ Our previous results encouraged us to undertaken an additional study focused on determining the real influence of the protecting group on the diastereoselectivity of the heterogeneous catalytic hydrogenation with Baylis-Hillman adducts originating from both aliphatic and aromatic aldehydes. In this comunication we disclose the results we have obtained with this new study.

R ₁ CCH ₃ —	$\frac{H/C}{1_3CN}$ or TBAF	
1-5	syn (6a-10a)	anti (6b-10b)
Baylis-Hillman adduct	6a/b , R ₁ = Ph; 7a/ t	b , R ₁ = 4-OMePh; 8a/b , R ₁ = 4-CIPh
R= H or TBS	9a/b , R ₁ = 3,4-OCI	H ₂ O-Ph; 10a/b , R ₁ = 4-NO ₂ Ph
R1	Syn : anti ratio (R= H)	Syn : anti ratio (R=TBS)
1 , Ph	1:1	10:1
2 , 4-OCH ₃ Ph	1:3	19:1
3 , 4-Cl-Ph	1:2	5 :1
4 , 3,4-OCH ₂ O-Ph	1:3	10:1
5 , 4-NO ₂ Ph	1:1	16:1

Scheme 1. Diastereoselective heterogeneous hydrogenation of Baylis-Hillman adduct.

Results and Discussion

In order to evaluate the influence of the protecting groups on the diastereoselectivity of the heterogeneous catalytic hydrogenation we have prepared several Baylis-Hillman adducts from aromatic and aliphatic aldehydes, using a method previously described by us.⁷ The Baylis-Hillman reaction between an aldehyde (aromatic or aliphatic) with methyl acrylate (with or without any additional solvent) in the presence of ultrasound radiation furnished the adducts (**1**-**9**) in good to very good chemical yields (Scheme 2).^{7b,8}

$$\begin{array}{c} O \\ R \\ H \end{array}^{+} \\ + \\ CO_2CH_3 \end{array} \xrightarrow{\text{DABCO}} \\ \end{array} \begin{array}{c} O \\ R \\ + \\ \end{array} \begin{array}{c} O \\ CO_2CH_3 \end{array}$$

$$\begin{array}{c} 1, R = Ph; \ \textbf{2}, R = 4 \cdot OCH_3Ph; \\ \textbf{3}, R = 4 \cdot CIPh; \ \textbf{4}, 3.4 \cdot OCH_2O - Ph; \ \textbf{5}, R = 4 \cdot NO_2Ph; \ \textbf{11}, R = \\ methyl \ \textbf{12}, R = ethyl; \ \textbf{13}, R = n - \\ propyl; \ \textbf{14}, R = n \cdot butyl \end{array}$$

Scheme 2. Baylis-Hillman reaction with aromatic and aliphatic aldehydes in the presence of ultrasound.

With the Baylis-Hillman adducts in our hands, the next step of our study was determining which hydroxyl protecting groups should be used. As our interest was centered in evaluating the influence of the spatial volume of protecting groups in the determination of the diastereoisomeric ratio during double bond hydrogenation, we have selected five different silylating reagents in which the substituents attached to the silicon have been gradually increased. Then, the Baylis-Hillman adducts (1-5) were treated with the chloride or the trifluoromethanesulfonate (triflate) of the following silylated protecting groups: trimethylsilyl (TMS), triethylsilyl (TES), *tert*-butyldimethylsilyl (TBDS), tri-isopropylsilyl (TIPS) and *tert*-butyldiphenylsilyl (TBDPS), in the presence of a base, to provide the corresponding silyl ethers. The results and chemical yields obtained in these protecting reactions are summarized in Table 1. We have chosen to start our study with Baylis-Hillman adducts from aromatic aldehydes only because they were less volatile and easier to manipulate than those from aliphatic aldehydes.

Baylis-Hillman adducts	Silylating agents	s Conditions	Time (h)	Product	Yields (%) ^a
	TMSCl	NEt ₃ , THF, rt.	6	15a	61
	TESCI	imidazole, DMF, rt.	12	15b	70
	TIPSOTf	2,6-lutidine, CH ₂ Cl ₂ , rt.	1.5	15c	93
1	TBDSCl	imidazole, DMF, rt.	21	15d	80
	TBDPSCl	imidazole, DMF, rt.	12	15e	76
	TMSCl	NEt ₃ , THF, rt.	6	16a	56
	TESCI	imidazole, DMF, rt.	12	16b	66
	TIPSOTf	2,6-lutidine, CH ₂ Cl ₂ , rt.	1	16c	83
MeO 2	TBDSCl	imidazole, DMF, rt.	21	16d	88
	TBDPSCl	imidazole, DMF, rt.	18	16e	97
ОН	TMSCl	NEt ₃ , THF, rt.	6	17a	99
CO ₂ Me	TESCI	imidazole, DMF, rt.	16	17b	70
	TIPSOTf	2,6-lutidine, CH ₂ Cl ₂ , rt.	1	17c	78
Cl 3	TBDSCI	imidazole, DMF, rt.	21	17d	70
	TBDPSCl	imidazole, DMF, rt.	12	17e	72
ОН	TMSOTf	2,6-lutidine, CH ₂ Cl ₂ , rt.	1/2	18a	70
$0 \sim CO_2 Me$	TESCI	imidazole, DMF, rt.	12	18b	88
	TIPSOTf	2,6-lutidine, CH ₂ Cl ₂ , rt.	2	18c	85
6 4	TBDSCl	imidazole, DMF, rt.	21	18d	77
	TBDPSCl	imidazole, DMF, rt.	12	18e	93
011	TMSCl	NEt ₃ , THF, rt.	6	19a	52
	TESCI	imidazole, DMF, rt.	8	19b	97
	TIPSOTf	2,6-lutidine, CH ₂ Cl ₂ , rt.	1/2	19c	94
O ₂ N 5	TBDSOTf	2,6-lutidine, CH ₂ Cl ₂ , rt	1/2	19d	76
	TBDPSCl	imidazole, DMF, rt.	_b	_ ^b	- ^b

Table 1. Conditions and yields for the secondary hydroxyl protecting reactions

^aYields for isolated and purified products ^bNo protecting product was detected for this adduct under this experimental conditions.

The solution of silvlated ethers **15a-e**/**19a-d** (see Table 1) in ethyl acetate was treated with 5 mol% of Pd/C 5% under an atmosphere of hydrogen at room pressure to provide the corresponding reduced products as a mixture of diastereoisomers (Scheme 3 and Table 2).

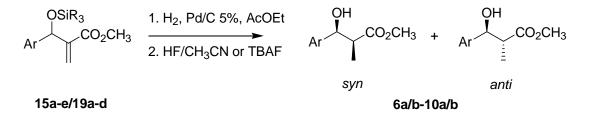
This crude mixture was treated with a solution of 5% HF in acetonitrile or with TBAF in THF to provide the corresponding alcohols **6a/b-10a/b** (Scheme 3).

Hydrogenation/removal of silyl groups (%) ^a ratio115a-Si(CH_3)_36a/b ; 785.6 : 1215b-Si(Et)_37a/b ; 7015 : 1315c-Si[(CH(CH_3)_2)]_38a/b ; 928 : 1415d-Si(r-Bu)(CH_3)_29a/b ; 9210 : 1515e-Si(r-Bu)(Ph)_210a/b ; 856 : 1616a-Si(CH_3)_36a/b; 827 : 1716b-Si(CH(CH_3)_2)]_38a/b ; 7812 : 1916d-Si(CH(CH_3)_2)]_38a/b ; 8019 : 11016e-Si(CH(CH_3)_2)10a/b ; 7812 : 11117a-Si(CH(CH_3)_2)8a/b ; 8019 : 11217b-Si(Et)_37a/b ; 604.8 : 11317c-Si(CH_3)_36a/b ; 505.2 : 11417d-Si(CH(CH_3)_2)]_38a/b ; 8713 : 11417d-Si(CH(CH_3)_2)]_38a/b ; 683.5 : 11517e-Si(CH_3)_36a/b ; 703.5 : 11618a-Si(CH_3)_36a/b ; 703.5 : 11718b-Si(Et)_37a/b ; 699 : 11918d-Si(CH(CH_3)_2)]_38a/b ; 699 : 11918d-Si(CH_3)_36a/b ; 758 : 12119a ^c -Si(CH_3)_36a/b ; 835.6 : 12219b ^c -Si(CH_3)_36a/b ; 837 : 12319c ^c -Si(CH(CH_3)_2)]_38a/b ; 693.5 : 1	Entry	Substrate	SiR ₃	Products ;	syn : anti ^b
115a-Si(CH_3)_36a/b785.6:1215b-Si(Et)_37a/b; 7015:1315c-Si[(CH(CH_3)_2)]_38a/b; 928:1415d-Si(<i>t</i> -Bu)(CH_3)_29a/b; 9210:1515e-Si(<i>t</i> -Bu)(Ph)_210a/b; 856:1616a-Si(CH_3)_36a/b; 827:1716b-Si(Et)_37a/b; 6019:1816c-Si[(CH(CH_3)_2)]_38a/b; 7812:1916d-Si(<i>t</i> -Bu)(Ph)_210a/b; 7812:11016e-Si(<i>t</i> -Bu)(Ph)_210a/b; 7812:11117a-Si(CH_3)_36a/b; 505.2:11217b-Si(Et)_37a/b; 604.8:11317c-Si[(CH(CH_3)_2)]_38a/b; 8713:11417d-Si(<i>t</i> -Bu)(Ph)_210a/b; 683.5:11517e-Si(<i>t</i> -Bu)(Ph)_210a/b; 683.5:11618a-Si(CH_3)_36a/b; 703.5:11718b-Si((CH(CH_3)_2)]_38a/b; 699:11918d-Si(<i>t</i> -Bu)(CH_3)_29a/b; 9210:12018e-Si(<i>t</i> -Bu)(CH_3)_29a/b; 835.6:12119a ^c -Si(CH_3)_36a/b; 758:12119a ^c -Si(CH_3)_36a/b; 835.6:12219b ^c -Si(CH_3)_36a/b; 835.6:1				Hydrogenation/removal of	
215b $-Si(Et)_3$ $7a/b$; 7015:1315c $-Si[(CH(CH_3)_2)]_3$ $8a/b$; 92 $8:1$ 415d $-Si(r-Bu)(CH_3)_2$ $9a/b$; 92 $10:1$ 515e $-Si(r-Bu)(Ph)_2$ $10a/b$; 85 $6:1$ 616a $-Si(CH_3)_3$ $6a/b$; 82 $7:1$ 716b $-Si(Et)_3$ $7a/b$; 60 $19:1$ 816c $-Si[(CH(CH_3)_2)]_3$ $8a/b$; 78 $12:1$ 916d $-Si(r-Bu)(Ph)_2$ $10a/b$; 78 $12:1$ 1016e $-Si(r-Bu)(Ph)_2$ $10a/b$; 78 $12:1$ 11 $17a$ $-Si(CH_3)_3$ $6a/b$; 50 $5.2:1$ 12 $17b$ $-Si(CH_3)_2$ $8a/b$; 87 $13:1$ 14 $17d$ $-Si(r-Bu)(Ph)_2$ $10a/b$; 68 $3.5:1$ 15 $17e$ $-Si(r-Bu)(Ph)_2$ $10a/b$; 68 $3.5:1$ 16 $18a$ $-Si(CH_3)_3$ $6a/b$; 70 $3.5:1$ 17 $18b$ $-Si(CH_{3})_2$ $9a/b$; 85 $4.8:1$ 15 $17e$ $-Si(r-Bu)(Ph)_2$ $10a/b$; 68 $3.5:1$ 16 $18a$ $-Si(CH_3)_3$ $6a/b$; 70 $3.5:1$ 17 $18b$ $-Si(r-Bu)(CH_{3})_2$ $9a/b$; 92 $10:1$ 20 $18e$ $-Si(r-Bu)(Ph)_2$ $10a/b$; 75 $8:1$ 21 $19a^c$ $-Si(CH_3)_3$ $6a/b$; 83 $5.6:1$ 22 $19b^c$ $-Si(Et)_3$ $7a/b$; 83 $7:1$				silyl groups (%) ^a	
315c $-Si[(CH(CH_3)_2)]_3$ 8a/b; 928:1415d $-Si(t-Bu)(CH_3)_2$ 9a/b; 9210:1515e $-Si(t-Bu)(Ph)_2$ 10a/b; 856:1616a $-Si(CH_3)_3$ 6a/b; 827:1716b $-Si(Et)_3$ 7a/b; 6019:1816c $-Si[(CH(CH_3)_2)]_3$ 8a/b; 7812:1916d $-Si(t-Bu)(CH_3)_2$ 8a/b; 8019:11016e $-Si(t-Bu)(CH_3)_2$ 8a/b; 505.2:11117a $-Si(CH_3)_3$ 6a/b; 505.2:11217b $-Si(Et)_3$ 7a/b; 604.8:11317c $-Si[(CH(CH_3)_2)]_3$ 8a/b; 8713:11417d $-Si(t-Bu)(CH_3)_2$ 9a/b; 683.5:11618a $-Si(CH_3)_3$ 6a/b; 703.5:11718b $-Si(Et)_3$ 7a/b; 699:11918d $-Si(t-Bu)(CH_3)_2$ 9a/b; 9210:12018e $-Si(CH_3)_3$ 6a/b; 758:12119a ^c $-Si(CH_3)_3$ 6a/b; 835.6:12219b ^c $-Si(Et)_3$ 7a/b; 837:1	1	15 a	-Si(CH ₃) ₃	6a/b ; 78	5.6 : 1
415d $-Si(t-Bu)(CH_3)_2$ 9a/b ; 9210 : 1515e $-Si(t-Bu)(Ph)_2$ 10a/b ; 856 : 1616a $-Si(CH_3)_3$ 6a/b; 827 : 1716b $-Si(CH_3)_3$ 7a/b ; 6019 : 1816c $-Si(CH(CH_3)_2)]_3$ 8a/b ; 7812 : 1916d $-Si(t-Bu)(CH_3)_2$ 8a/b ; 8019 : 11016e $-Si(t-Bu)(CH_3)_2$ 8a/b ; 505.2 : 11117a $-Si(CH_3)_3$ 6a/b ; 505.2 : 11217b $-Si(CH_3)_2$ 9a/b ; 854.8 : 11317c $-Si[(CH(CH_3)_2)]_3$ 8a/b ; 8713 : 11417d $-Si(t-Bu)(CH_3)_2$ 9a/b ; 683.5 : 11618a $-Si(CH_3)_3$ 6a/b ; 703.5 : 11718b $-Si(E1)_3$ 7a/b ; 643.5 : 11818c $-Si(CH(CH_3)_2)]_3$ 8a/b ; 699 : 11918d $-Si(t-Bu)(CH_3)_2$ 9a/b ; 9210 : 12018e $-Si(t-Bu)(CH_3)_2$ 9a/b ; 758 : 12119a ^c $-Si(CH_3)_3$ 6a/b ; 835.6 : 12219b ^c $-Si(CH_3)_3$ 6a/b ; 835.6 : 1		15b	-Si(Et) ₃	7a/b ; 70	15:1
515e $-Si(t-Bu)(Ph)_2$ 10a/b ; 856 : 1616a $-Si(CH_3)_3$ 6a/b; 827 : 1716b $-Si(Et)_3$ 7a/b ; 6019 : 1816c $-Si[(CH(CH_3)_2)]_3$ 8a/b ; 7812 : 1916d $-Si(t-Bu)(CH_3)_2$ 8a/b ; 8019 : 11016e $-Si(t-Bu)(CH_3)_2$ 8a/b ; 505.2 : 11117a $-Si(CH_3)_3$ 6a/b ; 505.2 : 11217b $-Si(CH_3)_2$ 9a/b ; 854.8 : 11317c $-Si[(CH(CH_3)_2)]_3$ 8a/b ; 8713 : 11417d $-Si(t-Bu)(Ph)_2$ 10a/b ; 683.5 : 11618a $-Si(CH_3)_3$ 6a/b ; 703.5 : 11718b $-Si(Et)_3$ 7a/b ; 699 : 11818c $-Si[(CH(CH_3)_2)]_3$ 8a/b ; 699 : 11918d $-Si(t-Bu)(Ph)_2$ 10a/b ; 758 : 12018e $-Si(t-Bu)(Ph)_2$ 10a/b ; 758 : 12119a ^c $-Si(CH_3)_3$ 6a/b ; 835.6 : 12219b ^c $-Si(Et)_3$ 7a/b ; 837 : 1	3	15c	-Si[(CH(CH ₃) ₂)] ₃	8a/b ; 92	8:1
616a $-Si(CH_3)_3$ 6a/b; 827:1716b $-Si(Et)_3$ 7a/b; 6019:1816c $-Si[(CH(CH_3)_2)]_3$ 8a/b; 7812:1916d $-Si(t-Bu)(CH_3)_2$ 8a/b; 8019:11016e $-Si(t-Bu)(Ph)_2$ 10a/b; 7812:11117a $-Si(CH_3)_3$ 6a/b; 505.2:11217b $-Si(CH_3)_3$ 6a/b; 8713:11317c $-Si[(CH(CH_3)_2)]_3$ 8a/b; 8713:11417d $-Si(t-Bu)(Ph)_2$ 10a/b; 683.5:11517e $-Si(t-Bu)(Ph)_2$ 10a/b; 683.5:11618a $-Si(CH_3)_3$ 6a/b; 703.5:11718b $-Si(Et)_3$ 7a/b; 747.3:11818c $-Si[(CH(CH_3)_2)]_3$ 8a/b; 699:11918d $-Si(t-Bu)(Ph)_2$ 10a/b; 758:12018e $-Si(t-Bu)(Ph)_2$ 10a/b; 758:12119a ^c $-Si(CH_3)_3$ 6a/b; 835.6:12219b ^c $-Si(Et)_3$ 7a/b; 837:1	4	15d	-Si(<i>t</i> -Bu)(CH ₃) ₂	9a/b ; 92	10:1
716b $-Si(Et)_3$ 7a/b ; 6019 : 1816c $-Si[(CH(CH_3)_2)]_3$ 8a/b ; 7812 : 1916d $-Si(t-Bu)(CH_3)_2$ 8a/b ; 8019 : 11016e $-Si(t-Bu)(Ph)_2$ 10a/b ; 7812 : 11117a $-Si(CH_3)_3$ 6a/b ; 505.2 : 11217b $-Si(Et)_3$ 7a/b ; 604.8 : 11317c $-Si[(CH(CH_3)_2)]_3$ 8a/b ; 8713 : 11417d $-Si(t-Bu)(CH_3)_2$ 9a/b ; 854.8 : 11517e $-Si(t-Bu)(CH_3)_2$ 9a/b ; 683.5 : 11618a $-Si(CH_3)_3$ 6a/b ; 703.5 : 11718b $-Si(Et)_3$ 7a/b ; 699 : 11918d $-Si(t-Bu)(CH_3)_2$ 9a/b ; 9210 : 12018e $-Si(t-Bu)(Ph)_2$ 10a/b ; 758 : 12119a ^c $-Si(CH_3)_3$ 6a/b ; 835.6 : 12219b ^c $-Si(Et)_3$ 7a/b ; 837 : 1	5	15e	$-Si(t-Bu)(Ph)_2$	10a/b ; 85	6:1
816c $-Si[(CH(CH_3)_2)]_3$ 8a/b; 7812:1916d $-Si(t-Bu)(CH_3)_2$ 8a/b; 8019:11016e $-Si(t-Bu)(Ph)_2$ 10a/b; 7812:11117a $-Si(CH_3)_3$ 6a/b; 505.2:11217b $-Si(CH_3)_3$ 7a/b; 604.8:11317c $-Si[(CH(CH_3)_2)]_3$ 8a/b; 8713:11417d $-Si(t-Bu)(CH_3)_2$ 9a/b; 854.8:11517e $-Si(t-Bu)(Ph)_2$ 10a/b; 683.5:11618a $-Si(CH_3)_3$ 6a/b; 703.5:11718b $-Si(Et)_3$ 7a/b; 699:11918d $-Si(t-Bu)(CH_3)_2$ 9a/b; 9210:12018e $-Si(t-Bu)(Ph)_2$ 10a/b; 758:12119a ^c $-Si(CH_3)_3$ 6a/b; 835.6:12219b ^c $-Si(Et)_3$ 7a/b; 837:1	6	16a	-Si(CH ₃) ₃	6a/b ; 82	7:1
916d $-Si(t-Bu)(CH_3)_2$ 8a/b ; 8019 : 11016e $-Si(t-Bu)(Ph)_2$ 10a/b ; 7812 : 11117a $-Si(CH_3)_3$ 6a/b ; 505.2 : 11217b $-Si(Et)_3$ 7a/b ; 604.8 : 11317c $-Si[(CH(CH_3)_2)]_3$ 8a/b ; 8713 : 11417d $-Si(t-Bu)(CH_3)_2$ 9a/b ; 854.8 : 11517e $-Si(t-Bu)(CH_3)_2$ 9a/b ; 683.5 : 11618a $-Si(CH_3)_3$ 6a/b ; 703.5 : 11718b $-Si(Et)_3$ 7a/b ; 747.3 : 11818c $-Si[(CH(CH_3)_2)]_3$ 8a/b ; 699 : 11918d $-Si(t-Bu)(CH_3)_2$ 9a/b ; 9210 : 12018e $-Si(t-Bu)(Ph)_2$ 10a/b ; 758 : 12119a ^c $-Si(CH_3)_3$ 6a/b ; 835.6 : 12219b ^c $-Si(Et)_3$ 7a/b ; 837 : 1	7	16b	-Si(Et) ₃	7a/b ; 60	19:1
1016e $-Si(t-Bu)(Ph)_2$ 10a/b ; 7812 : 11117a $-Si(CH_3)_3$ 6a/b ; 505.2 : 11217b $-Si(Et)_3$ 7a/b ; 604.8 : 11317c $-Si[(CH(CH_3)_2)]_3$ 8a/b ; 8713 : 11417d $-Si(t-Bu)(CH_3)_2$ 9a/b ; 854.8 : 11517e $-Si(t-Bu)(CH_3)_2$ 9a/b ; 683.5 : 11618a $-Si(CH_3)_3$ 6a/b ; 703.5 : 11718b $-Si(Et)_3$ 7a/b ; 747.3 : 11818c $-Si[(CH(CH_3)_2)]_3$ 8a/b ; 699 : 11918d $-Si(t-Bu)(CH_3)_2$ 9a/b ; 9210 : 12018e $-Si(CH_3)_3$ 6a/b ; 758 : 12119a ^c $-Si(CH_3)_3$ 6a/b ; 835.6 : 12219b ^c $-Si(Et)_3$ 7a/b ; 837 : 1	8	16c	-Si[(CH(CH ₃) ₂)] ₃	8a/b ; 78	12:1
1117a $-Si(CH_3)_3$ $6a/b$; 50 $5.2:1$ 1217b $-Si(Et)_3$ $7a/b$; 60 $4.8:1$ 1317c $-Si[(CH(CH_3)_2)]_3$ $8a/b$; 87 $13:1$ 1417d $-Si(t-Bu)(CH_3)_2$ $9a/b$; 85 $4.8:1$ 1517e $-Si(t-Bu)(CH_3)_2$ $9a/b$; 68 $3.5:1$ 1618a $-Si(CH_3)_3$ $6a/b$; 70 $3.5:1$ 1718b $-Si(Et)_3$ $7a/b$; 74 $7.3:1$ 1818c $-Si[(CH(CH_3)_2)]_3$ $8a/b$; 69 $9:1$ 1918d $-Si(t-Bu)(CH_3)_2$ $9a/b$; 92 $10:1$ 2018e $-Si(t-Bu)(CH_3)_2$ $10a/b$; 75 $8:1$ 21 $19a^c$ $-Si(CH_3)_3$ $6a/b$; 83 $5.6:1$ 22 $19b^c$ $-Si(Et)_3$ $7a/b$; 83 $7:1$	9	16d	-Si(<i>t</i> -Bu)(CH ₃) ₂	8a/b ; 80	19:1
1217b $-Si(Et)_3$ 7a/b; 604.8:11317c $-Si[(CH(CH_3)_2)]_3$ 8a/b; 8713:11417d $-Si(t-Bu)(CH_3)_2$ 9a/b; 854.8:11517e $-Si(t-Bu)(Ph)_2$ 10a/b; 683.5:11618a $-Si(CH_3)_3$ 6a/b; 703.5:11718b $-Si(Et)_3$ 7a/b; 747.3:11818c $-Si[(CH(CH_3)_2)]_3$ 8a/b; 699:11918d $-Si(t-Bu)(CH_3)_2$ 9a/b; 9210:12018e $-Si(t-Bu)(Ph)_2$ 10a/b; 758:12119a ^c $-Si(CH_3)_3$ 6a/b; 835.6:12219b ^c $-Si(Et)_3$ 7a/b; 837:1	10	16e	$-Si(t-Bu)(Ph)_2$	10a/b ; 78	12:1
1317c $-Si[(CH(CH_3)_2)]_3$ 8a/b ; 8713 : 11417d $-Si(t-Bu)(CH_3)_2$ 9a/b ; 854.8 : 11517e $-Si(t-Bu)(Ph)_2$ 10a/b ; 683.5 : 11618a $-Si(CH_3)_3$ 6a/b ; 703.5 : 11718b $-Si(Et)_3$ 7a/b ; 747.3 : 11818c $-Si[(CH(CH_3)_2)]_3$ 8a/b ; 699 : 11918d $-Si(t-Bu)(CH_3)_2$ 9a/b ; 9210 : 12018e $-Si(t-Bu)(Ph)_2$ 10a/b ; 758 : 12119a ^c $-Si(CH_3)_3$ 6a/b ; 835.6 : 12219b ^c $-Si(Et)_3$ 7a/b ; 837 : 1	11	17a	-Si(CH ₃) ₃	6a/b ; 50	5.2:1
1417d $-Si(t-Bu)(CH_3)_2$ 9a/b ; 854.8 : 11517e $-Si(t-Bu)(Ph)_2$ 10a/b ; 683.5 : 11618a $-Si(CH_3)_3$ 6a/b ; 703.5 : 11718b $-Si(Et)_3$ 7a/b ; 747.3 : 11818c $-Si[(CH(CH_3)_2)]_3$ 8a/b ; 699 : 11918d $-Si(t-Bu)(CH_3)_2$ 9a/b ; 9210 : 12018e $-Si(t-Bu)(Ph)_2$ 10a/b ; 758 : 12119a ^c $-Si(CH_3)_3$ 6a/b ; 835.6 : 12219b ^c $-Si(Et)_3$ 7a/b ; 837 : 1	12	17b	-Si(Et) ₃	7a/b ; 60	4.8:1
15 $17e$ $-Si(t-Bu)(Ph)_2$ $10a/b$; 68 $3.5:1$ 16 $18a$ $-Si(CH_3)_3$ $6a/b$; 70 $3.5:1$ 17 $18b$ $-Si(Et)_3$ $7a/b$; 74 $7.3:1$ 18 $18c$ $-Si[(CH(CH_3)_2)]_3$ $8a/b$; 69 $9:1$ 19 $18d$ $-Si(t-Bu)(CH_3)_2$ $9a/b$; 92 $10:1$ 20 $18e$ $-Si(t-Bu)(Ph)_2$ $10a/b$; 75 $8:1$ 21 $19a^c$ $-Si(CH_3)_3$ $6a/b$; 83 $5.6:1$ 22 $19b^c$ $-Si(Et)_3$ $7a/b$; 83 $7:1$	13	17c	-Si[(CH(CH ₃) ₂)] ₃	8a/b ; 87	13:1
1618a $-Si(CH_3)_3$ 6a/b ; 703.5 : 11718b $-Si(CH_3)_3$ 6a/b ; 747.3 : 11818c $-Si[(CH(CH_3)_2)]_3$ 8a/b ; 699 : 11918d $-Si(t-Bu)(CH_3)_2$ 9a/b ; 9210 : 12018e $-Si(t-Bu)(Ph)_2$ 10a/b ; 758 : 12119a ^c $-Si(CH_3)_3$ 6a/b ; 835.6 : 12219b ^c $-Si(Et)_3$ 7a/b ; 837 : 1	14	17d	-Si(<i>t</i> -Bu)(CH ₃) ₂	9a/b ; 85	4.8:1
1718b $-Si(Et)_3$ 7a/b ; 747.3 : 11818c $-Si[(CH(CH_3)_2)]_3$ 8a/b ; 699 : 11918d $-Si(t-Bu)(CH_3)_2$ 9a/b ; 9210 : 12018e $-Si(t-Bu)(Ph)_2$ 10a/b ; 758 : 12119a ^c $-Si(CH_3)_3$ 6a/b ; 835.6 : 12219b ^c $-Si(Et)_3$ 7a/b ; 837 : 1	15	17e	$-Si(t-Bu)(Ph)_2$	10a/b ; 68	3.5 : 1
1818c-Si[(CH(CH_3)_2)]_38a/b ; 699 : 11918d-Si(t-Bu)(CH_3)_29a/b ; 9210 : 12018e-Si(t-Bu)(Ph)_210a/b ; 758 : 12119a ^c -Si(CH_3)_36a/b ; 835.6 : 12219b ^c -Si(Et)_37a/b ; 837 : 1	16	18 a	-Si(CH ₃) ₃	6a/b ; 70	3.5:1
1918d $-Si(t-Bu)(CH_3)_2$ 9a/b ; 9210 : 12018e $-Si(t-Bu)(Ph)_2$ 10a/b ; 758 : 12119a ^c $-Si(CH_3)_3$ 6a/b ; 835.6 : 12219b ^c $-Si(Et)_3$ 7a/b ; 837 : 1	17	18b	-Si(Et) ₃	7a/b ; 74	7.3:1
2018e $-Si(t-Bu)(Ph)_2$ 10a/b ; 758 : 12119a ^c $-Si(CH_3)_3$ 6a/b ; 835.6 : 12219b ^c $-Si(Et)_3$ 7a/b ; 837 : 1	18	18c	-Si[(CH(CH ₃) ₂)] ₃	8a/b ; 69	9:1
21 $19a^{c}$ $-Si(CH_3)_3$ $6a/b$; 83 $5.6:1$ 22 $19b^{c}$ $-Si(Et)_3$ $7a/b$; 83 $7:1$	19	18d	-Si(<i>t</i> -Bu)(CH ₃) ₂	9a/b ; 92	10:1
22 $19b^{c}$ -Si(Et) ₃ 7a/b ; 83 7:1	20	18e	$-Si(t-Bu)(Ph)_2$	10a/b ; 75	8:1
	21	19a ^c	-Si(CH ₃) ₃	6a/b ; 83	5.6:1
23 19c ^c $-Si[(CH(CH_3)_2)]_3$ 8a/b ; 69 3.5:1	22	19b ^c	-Si(Et) ₃	7a/b ;83	7:1
	23	19c ^c	-Si[(CH(CH ₃) ₂)] ₃	8a/b ; 69	3.5:1

Table 2. Diastereoselectivity for the hydrogenation of double bond of silylated Baylis-Hillman adducts



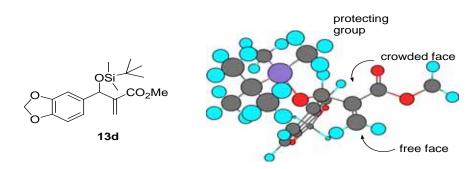
^aYields for isolated and purified products. ^bThe diastereoisomeric ratio was determined by GC (HP column) and measured by ¹H NMR. ^cIn the hydrogenation reaction of the nitro derivative, we have also detected the reduction of the nitro groups, which was transformed to the corresponding amine.



Scheme 3. Heterogeneous catalytic hydrogenation reaction of silylated Baylis-Hillman adducts.

The relative stereochemistry (C2/C3) of the products was determined using ¹H NMR and ¹³C NMR spectroscopy, the vicinal constant coupling (*J*) for *syn* diastereoisomer varying between 5.9-6.32 Hz and 7-9 Hz for the *anti* isomer.¹⁸

The results obtained in the hydrogenation step are summarized in Table 2. The preliminary observations indicate that there was a dependence on the size of the substituents directly bonded to the silicon atom and the diastereoselectivity. Any silylated protecting groups led to the *syn* diastereoisomer as the major isomer. However, when huge silylated protecting groups such as triethylsilyl, tri-isopropylsilyl or *tert*-butyldimethyl silyl were used, it is possible to obtain the greatest proportion of the *syn* isomer. The minimized conformations for two silylated Baylis-Hillman adducts are shown in Figure 2.



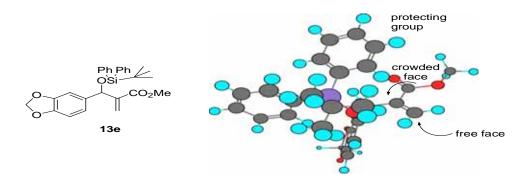
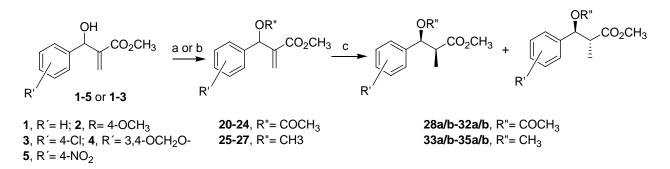


Figure 2. Minimized conformations of silylated Baylis-Hillman adducts.

A preliminary analysis pointed out that there was a difference between the faces of the double bond exerted by the protecting groups. The presence of a silyl group seems to be a sufficient condition to get a moderate to good degree of stereoselection in the heterogeneous hydrogenation reaction of Baylis-Hillman adducts. For silylated groups with large substituents on the silicon the diastereoselectivity seems to be better, however it is clear that there is a limit to the size of these substituents. For example , the *syn* diastereoselectivity for the *tert*-butyldiphenyl silyl groups (entries 5, 10, 15 and 20 of Table 2) is worse than that obtained for triethyl- (entries 2, 7, 12, 17 and 22 of Table 2), triisopropyl- (entries 3, 8, 13, 18 and 23 of Table 2) and *tert*-butyldimethylsilyl (entries 4, 9, 14, 19, 24 of Table 2) groups. Apparently, the diastereoselectivity was guided only by steric effects (see Figure 2). To evaluate the accuracy of these experimental observations we decided to test some additional protecting groups. Thus, we have acetylated and methylated the secondary hydroxyl group of some Baylis-Hillman adducts, which were submitted to experimental hydrogenation conditions similar to those described above.

Thus, the Baylis-Hillman adducts **1-5** were treated wih acetyl chloride in dicloromethane in the presence of triethylamine to give the corresponding acetates **20-24**. To prepare the alkoxy derivatives, adducts **1-3** were treated with Ag_2O and CH_3I in dichloromethane to give the corresponding methyl ethers **25-27** (Scheme 4).¹⁹



Scheme 4. Reagents and conditions: a. CH₃COCl, Et₃N, CH₂Cl₂, rt.; b. Ag₂O, CH₃I, CH₂Cl₂, rt.; c. Pd/C 5%, AcOEt, rt.

Hydrogenation of both acetylated and methylated adducts gave a mixture of diastereoisomers with very poor *anti* diastereoselectivity for the first case and poor *syn* diastereoselectivity for the methyl ethers. The results are summarized in Table 3.

The chemical yields obtained, both for the protecting reactions and for the hydrogenation step, could be considered from moderate to very good. For the methylated adducts we have observed a discrete *syn* diastereoselectivity. Probably the spatial volume occupied by methyl groups is not enough to guide the hydrogen addition selectivity to only one of the two faces of the double bond. This could justify the low degree of diastereoselection obtained in the cases where the protecting group was methyl. On the other hand, the heterogeneous catalytic hydrogenation reaction of the acetylated Baylis-Hillman adducts led to a mixture in which an inversion of the distribution of diastereoisomers has occurred. In these cases, despite the very poor diastereoselectivity attained, the major isomer was *anti*.

From our point of view, these results could be based on an electronic effect. Alternatively, a configuration where the double bond is freed from the influence of the substituents could be also responsible for these results.

Entry	Acetylated and Methylated Baylis- Hillman adducts	Protection Yield (%) ^{a,b}	syn : anti ratio	Hydrogenation (%) ^a
1	20 , R'=H, R"=Ac	82	28a/b ; 1 : 3	65 ^b
2	21 , $R' = OCH_3$, $R'' = Ac$	78	29a/b ; 1 : 1	70 ^b
3	22 , R'= Cl, R''= Ac	80	30a/b ; 1 : 1.7	63 ^b
4	23 , R'= 3,4-OCH ₂ O, R''= Ac	89	31a/b ; 1 : 2.3	60^{b}
5	24 , $R' = NO_2$, $R'' = Ac$	76	32a/b ; 1 : 1.5	78^{b}
7	25 , $R' = H$; $R'' = CH_3$	85	33a/b ; 3 : 1	90
8	26 , $R' = OCH_3$, $R'' = CH_3$	78	34a/b ; 2.1 : 1	92
9	27 , R'= Cl, R''= CH ₃	84	35a/b ; 2.3 : 1	89

Table 3. Hydrogenation of the acetylated and methylated Baylis-Hillman adducts

^a Yields are for isolated and purified products. ^b For all cases we have detected hydrogenolysis reaction which ratio varied from 10 to 20%.

At this point in the work it was clear to us that it was possible to obtain an elevated degree of *syn* diastereoselection if we protect the secondary hydroxyl group of the Baylis-Hillman adducts as a silylated ether (preferentially using TES, TIPS or TBDS groups). To check the generality of these effects on diastereoselectivity we decided to hydrogenate some additional Baylis-Hillman adducts from aliphatic aldehydes. Solutions of Baylis-Hillman adducts (**11-14**) in

dimethylformamide were treated with *tert*-butyldimethylsilyl chloride in the presence of imidazole to provide the corresponding silyl ethers (**36-39**) in chemical yields varying from 77-80%. We decided to choose *tert*-butyldimethylsilyl as the standard protecting only due to the ease of working with it.

The results obtained with the heterogeneous hydrogenation of this new set of protected Baylis-Hillman adducts are summarized in Table 4.

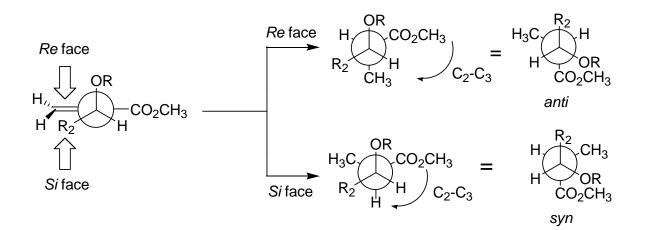
In all silvlated esters (Table 2 and 4) the presence of the silvl group blocks the Re face, directing the delivery of the hydrogen preferentially to the Si face to yield the *syn* diastereoisomer (Scheme 5).

Entry	Baylis-Hillman adducts	Product/SiR ₃	Reduction yield (%)	Product/Ratio syn : anti ^b
1		36 , TBDS	78	40a/b , 6 : 1
2	OH CO ₂ CH ₃ 12	37 , TBDS	82	41a/b , 16 : 1
3	OH CO ₂ CH ₃ 13	38 , TBDS	76	42a/b , 16 : 1
4	OH CO ₂ CH ₃ 14	39 , TBDS	76	43a/b , 17 : 1

Table 4. Hydrogenation of silylated Baylis-Hillman adducts from aliphatic aldehydes

^a Yields for isolated and purified products. ^b The diastereoisomeric ratio was determined by GC (HP column) and measured by ¹H NMR.

The stereochemical outcome of these hydrogenation reactions can be rationalized based on a conformation in which the alkyl substituent occupies the inside position and the carbon-oxygen bond is perpendicular to the π bond. This reduces any destablizing interactions between the alkyl substituents and the carboxyl group (Scheme 5).²⁰⁻²¹



Scheme 5. Stereochemical rationalisation for the heterogeneous hydrogenation of Baylis-Hillman adducts.

For the cases we have studied, we observed the same stereochemical behaviour of silylated Baylis-Hillman adducts, which gave *syn* diastereoisomers as major products when submitted to the heterogeneous hydrogenation experimental conditions. For methyl ether derivatives the same behaviour, e.g. *syn* diastereoselectivity, has been observed, however the diastereoselectivity obtained in these cases was not as good as those described for the silylated ether derivatives. For acetylated derivatives, when diastereoselectivity was observed, it was very poor and the isomer ratios were inverted. In these cases the *anti* diastereoisomer were the major one.

Conclusions

This simple heterogeneous hydrogenation method of silylated Baylis-Hillman adducts could be considered as an alternative strategy to directly provide products which otherwise could only be obtained through an aldol condensation with propionate derivatives. However, the aldol condensation, despite its widespread utilisation, requires special experimental conditions, such as low temperatures and sophisticated bases. As it is possible to obtain enantiomerically pure Baylis-Hillman adducts (by resolution or through asymmetric base-catalysed reaction) their hydrogenation could also provide an easy and straighforward access to enantiomerically pure propionate derivatives.

Finally, this method should also be considered as an interesting, easier alternative to the homogeneous hydrogenation reaction of Weinreb amides obtained from Baylis-Hillman adducts.¹⁵

Experimental Section

General Procedures. The ¹H and ¹³C spectra were recorded on a Varian GEMINI BB-300 at 300 and 75.4 MHz, respectively, or on an Inova instrument at 500 and 125 MHz, respectively. The mass spectra were recorded using a HP model 5988A GC/MS with a High Resolution Autospec-Micromass/EBE. IR were obtained with a Nicolet model Impact 410. Melting points were measured in open capillary tubes using an Electrothermal model 9100, and are uncorrected. Yields were determined from GC analyses on a HP6890 equipment with a flame ionization detector, using a HP-5 capillary (crosslinked 5% Phenyl methyl siloxanes, 28 m) column. Manipulations and reactions were not performed under dry atmospheres or employing dry solvents, unless otherwise specified. Purification and separations by column chromatography were performed on silica gel, using normal or flash chromatography. TLC visualization was achieved by spraying with 5% ethanolic phosphomolybdic acid and heating. All Baylis–Hillman reactions were sonicated in an ultrasonic cleaner UNIQUE model GA 1000 (1000 W, 25 kHz). Aromatic aldehydes and aliphatic aldehydes were purchased from Aldrich, Acros or Lancaster and were used without previous purification.

General procedure for the protection of Baylis-Hillman adducts as trimethylsilyl ethers (Preparation of 15a, 16a, 17a, 19a)

A mixture of 1 mmol of the Baylis-Hillman adduct (1, 2, 3 and 5), 1.5 equivalents of trimethylsilylchloride (freshly distilled) and 1.2 equivalents of triethylamine (dried over CaH₂) in 4mL of dry tetrahydrofuran was stirred at room temperature, under nitrogen, for 6 h (the end of the reaction was confirmed by tlc). Then, the reaction medium was extracted with hexane (3 x 20 mL). After washing with distilled water and drying over Na₂SO₄, the organic layer was evaporated under reduced pressure to furnish an oily residue which was purified by silica gel column chromatography.

Methyl 2-trimethylsilyloxy(phenyl)methylacrylate (15a). Eluent: hexane : ethyl acetate 90:10 v/v, (0.18 g of a volatile and fluid liquid, 61%). IR (v_{max} /Film) 3087, 2954, 2856, 1722 (C=O), 1634 (C=C) cm⁻¹; ¹H NMR (300MHz, CDCl₃) δ 0.05 (s, 9H, OSi(C<u>H</u>₃)₃), 3.66 (s, 3H, OC<u>H</u>₃ ester), 5.59 (s, 1H, C<u>H</u>OSiR₃), 6.20 (s, 1H, double bond), 6.27 (s, 1H, double bond), 7.22-7.33 (m, 5H aromatics); ¹³C NMR (125MHz, CDCl₃) δ 170.8, 148.2, 142.8, 128, 126.8, 126.4, 123.1, 73.9, 52.2, 0.7; Anal. calcd for C₁₄H₂₀O₃Si: C, 63.60%; H, 7.62%. Found C, 63.58%; H, 7.60%. **Methyl 2-[4-methoxyphenyl(trimethylsilyloxy)methyl]acrylate (16a).** Eluent: hexane : ethyl acetate 95:5 v/v, (0.20 g of a volatile and fluid liquid, 56%). IR (v_{max} /Film) 3087, 2954, 2856, 1722 (C=O), 1632 (C=C) cm⁻¹; ¹H NMR (300MHz, CDCl₃) δ 0.05 (s, 9H, OSi(C<u>H</u>₃)₃), 3.67 (s, 3H, OCH₃ ester), 3.70 (s, 3H, OCH₃ aromatic), 5.56 (s, 1H, C<u>H</u>OSiR₃), 6.0 (s, 1H, double bond), 6.25 (s, 1H, double bond), 6.83 (d, *J* = 8.5Hz, 2H aromatics), 7.24 (d, *J* = 8.5Hz, 2H aromatics); ¹³C NMR (125MHz, CDCl₃) δ 170.8, 157.7, 148.2, 136.6, 129.9, 123.1, 112.1, 73.9, 55.0, 52.2, 0.52.2, 0.52.2, 0.52.2, 0.53 (C=C) (C=C)

0.7; Anal. calcd for C₁₅H₂₂O₄Si: C, 61.19%; H, 7.53%. Found C, 61.17%; H, 7.52%.

Methyl 2-[4-chlorophenyl(trimethylsilyloxy)methyl]acrylate (17a). Eluent: hexane : ethyl acetate 90:10 v/v, (0.13 g of a volatile and fluid liquid, 99%). IR (ν_{max} /Film) 3087, 2954, 2856, 1720 (C=O), 1630 (C=C) cm⁻¹; ¹H NMR (300MHz, CDCl₃) δ 0.02 (s, 9H, OSi(C<u>H</u>₃)₃), 3.61 (s, 3H, OC<u>H</u>₃ ester), 5.12 (s large, C<u>H</u>OSiR₃), 5.96 (s large, 1H, double bond), 6.22 (s large, 1H), 7.21-7.20 (m, 4H aromatics); ¹³C NMR (125MHz, CDCl₃) δ 170.8, 148.2, 141.8, 132.5, 129.0, 126.7, 123.1, 73.9, 52.2, 0.8; Anal. calcd for C₁₄H₁₉ClO₃Si: C, 56.27%; H, 6.41%. Found C, 56.27%; H, 6.40%.

Methyl 2-trimethylsilyloxy(4-nitrophenyl)methylacrylate (19a). Eluent: hexane : ethyl acetate 85:15 v/v, (0.25 g of a fluid liquid, 52%). IR (v_{max} /Film) 3087, 2954, 2856, 1725 (C=O), 1632 (C=C) cm⁻¹; ¹H NMR (300MHz, CDCl₃) δ 0.09 (s, 9H, OSi(CH₃)₃), 3.70 (s, 3H, OCH₃ ester), 5.67 (s, 1H, CHOSiR₃), 6.09 (s, 1H, double bond), 6.34 (s, 1H, double bond), 7.53 (d, *J* = 8.8Hz, 2H aromatics); ¹³C NMR (125MHz, CDCl₃) δ 170.8, 148.2, 141.8, 132.5, 129.0, 126.7, 123.1, 73.9, 52.2, 0.8; Anal. calcd for C₁₄H₁₉NO₅Si: C, 54.35%; H, 6.19%. Found C, 54.33%; H, 6.17%.

Preparation of methyl 2-benzo[d][1.3]dioxol-5-yl(trimethylsilyloxy)methylacrylate (18a).

A mixture of 1 mmol of Baylis-Hillman adducts (**4**), 1.5 equivalents of trimethylsilyl trifluoromethanesulfonate and 2 equivalents of anhydrous 2,6-lutidine in 0.5 mL of dichloromethane was stirred at room temperature, under a nitrogen atmosphere, for 30 minutes. After this time, the reaction medium was diluted with 60 mL of ethyl ether and the organic layer was successively washed with a 0.1 mol/L solution of HCl, a saturated solution of NaHCO₃ and finally brine. Then, the organic layer was dried over anhydrous Na₂SO₄ and evaporated under reduced pressure to provide an oily residue, which was purified by silica gel column chromatography. Eluent: hexane : ethyl acetate 94:6 v/v, (0.18 g of a volatile and fluid liquid, 70%). IR (v_{max}/Film) 3087, 2954, 2856, 1722 (C=O), 1634 (C=C) cm⁻¹; ¹H NMR (300MHz, CDCl₃) δ 0.05 (s, 9H, OSi(CH₃)₃), 3.66 (s, 3H, OCH₃ ester), 5.95 (s, 1H, CHOSiR₃), 6.01 (s, 2H, -OCH₂O-), 6.20 (s, 1H, double bond), 6.27 (s, 1H, double bond), 7.22-7.33 (m, 5H aromatics); ¹³C NMR (125MHz, CDCl₃) δ 170.8, 148.2, 142.8, 146.8, 145.6, 136.8, 120.4, 109.8, 107, 75.6, 52.2, 0.7; Anal. calcd for C₁₅H₂₀O₅Si: C, 58.42%; H, 6.54%. Found C, 58.41%; H, 6.52%.

General procedure for the protection of Baylis-Hillman adducts as triethylsilyl, *tert*butyldimethylsilyl and *tert*-butyldiphenysilyl ethers. (Preparation of 15b, 15d and 15e, 16b, 16d and 16e, 17b, 17d-e, 18b, 18d-e, 19b, 36-39)

A mixture of 1 mmol of the Baylis-Hillman adduct, 1.3 mmol of triethylsilyl chloride, *tert*butyldimethylsilyl chloride or *tert*-butyldiphenylsilylchloride, 2.5 mmol of imidazole and 0.3 cm³ of dry N,N-dimethylformamide was stirred at room temperature, under nitrogen, for 12-18 h (the end of reaction was confirmed by tlc). Then, the reaction medium was quenched with 10 cm³ of hexane. After washing with brine (3 x 5 cm³), drying over MgSO₄ and evaporation, the residue obtained was purified by silica gel column chromatography (eluent indicated for each adduct).

Methyl 2-triethylsilyloxy(phenyl)methylacrylate (15b). Eluent: hexane: ethyl acetate 92:8 v/v, (0.2 g of an oil, 70%). IR (ν_{max} /Film) 3032, 2954, 2875, 1454, 1089, 1064, 1722 (C=O), 1630 (C=C) cm⁻¹; ¹H NMR (500MHz, CDCl₃) δ 0.48-0.6 (m, 6H, OSiC<u>H</u>₂CH₃), 0.82-0.9 (m, 9H, OSiCH₂C<u>H</u>₃), 3.66 (s, 3H, OCH₃ ester), 6.1 (s large, 1H, C<u>H</u>OSiR₃), 6.25 (s, 1H, double bond), 6.26 (s, 1H, double bond), 7.2-7.4 (m, 5H aromatics); ¹³C NMR (125MHz, CDCl₃) δ 171, 147.4, 141.3, 127.7, 126.8, 126.4, 122.6, 76, 52.2, 6.3, 6.1; Anal. calcd for C₁₇H₂₆O₃Si: C, 66.62%; H, 8.55%. Found C, 66.59%; H, 8.53%.

Methyl 2-*tert*-**butyldimethylsilyloxy(phenyl)methylacrylate** (**15d**). Eluent: hexane: ethyl acetate 90:10 v/v, (0.2 g of a viscous oil, 80%). IR (v_{max} /Film) 3087, 2954, 2856, 1722 (C=O), 1634 (C=C), 1257, 1088 cm⁻¹; ¹H NMR (300MHz, CDCl₃) δ -0.1 (s, 6H, OSi(C<u>H</u>₃)₂), 0.9 (s, 9H, OSi(C(C<u>H</u>₃)₃), 3.65 (s, 1H, OC<u>H</u>₃ ester), 5.58 (s, 1H, C<u>H</u>OSiR₃), 5.94 (s, 1H, double bond), 6.11 (s, 1H, double bond), 7.10-7.33 (m, 5H aromatics); ¹³C NMR (75.4MHz, CDCl₃) δ 168.8, 138.3, 134.9, 130.9, 130.6, 127.1, 127.8, 125.8, 52.2, 45.6, 26.1, 20.8, -6.3; Anal. calcd for C₁₇H₂₆O₃Si: C, 66.62%; H, 8.55%. Found C, 66.60%; H, 8.54%.

Methyl 2-*tert*-**butyldiphenylsilyloxy(phenyl)methylacrylate** (15e). Eluent: hexane: ethyl acetate 95:5 v/v, (0.19g of a viscous oil, 76%). IR (v_{max} /Film) 3087, 2954, 2856, 1724 (C=O), 1632 (C=C) cm⁻¹; ¹H NMR, 300MHz, CDCl₃) δ 1.0 (s, 9H, OSi-<u>*t*-Bu</u>), 3.58 (s, 3H, OCH₃ ester), 5.52 (s large, 1H, CHOSiR₃), 6.23 (s large, 2H, double bond), 7.15-7.62 (m, 15H aromatics, Si(Ph)₂ + aromatics); ¹³C NMR (125MHz, CDCl₃) δ 170.8, 148.2, 136.1, 133.1, 129.9, 127.8, 127.7, 126.4, 126.8, 123.1, 80.9, 52.2, 22.6; Anal. calcd for C₂₇H₃₀O₃Si: C, 75.31%; H, 7.02%. Found C, 75.29%; H, 7.0%.

Methyl 2-[triethylsilyloxy(4-methoxyphenyl)methyl]acrylate (16b). Eluent: hexane: ethyl acetate 95 : 5 v/v, (0.22 g of a viscous oil, 66%). IR (v_{max} /Film) 2954, 2912, 2875, 2835, 1722 (C=O), 1610 (C=C), 1512, 1458, 1248, 1078 cm⁻¹; ¹H NMR (500MHz, CDCl₃) δ 0.48-0.6 (m, 6H, OSiCH₂CH₃), 0.82-0.9 (m, 9H, OSiCH₂CH₃), 3.71 (s, 3H, OCH₃ ester), 3.79 (s, 3H, OCH₃ aromatic), 5.56 (s, 1H, CHOSiR₃), 6.08 (s large, 1H, double bond), 6.23 (s large, 1H, double bond), 6.81 (d, *J* = 8.8Hz, 2H aromatics), 7.26 (d, *J* = 8.8Hz, 2H aromatics); ¹³C NMR (125MHz, CDCl₃) δ 171, 157.5, 148.2, 134.8, 129.6, 123.1, 112.1, 75.8, 55, 52.2, 6.34, 6.12; Anal. calcd for C₁₈H₂₈O₄Si: C, 64.25%; H, 8.39%. Found C, 64.23%; H, 8.39%.

Methyl 2-[*tert*-butyldimethylsilyloxy(4-methoxyphenyl)methyl]acrylate (16d). Eluent: hexane : ethyl acetate 95:5 v/v, (0.25 g of a viscous oil, 88%). IR (v_{max} /Film) 2954, 2875, 2835, 1720 (C=O), 1632 (C=C), cm⁻¹; ¹H NMR (300MHz, CDCl₃) δ -0.12 (s, 3H, OSiC<u>H₃</u>), 0.04 (s, 3H, OSiC<u>H₃</u>), 0.86 (s, 9H, *t*-BuSi), 3.66 (s, 3H, OCH₃, ester), 3.78 (s, 3H, OCH₃ aromatic), 5.54 (s large, 1H, C<u>H</u>OSiR₃), 6.05 (s large, 1H, double bond), 6.21 (s large, 1H, double bond), 6.8 (d, J = 8.7Hz, 2H aromatics), 7.23 (d, J = 8.7Hz, 2H aromatics); ¹³C NMR (75.4MHz, CDCl₃) 171, 157.5, 149, 135, 129.6, 123, 112.1, 75.8, 55, 52.2, 25.8, 16.9, -4.5; Anal. calcd for C₁₈H₂₈O₄Si: C, 64.25%; H, 8.39%. Found C, 64.25%; H, 8.39%. **Methyl 2-**[*tert*-butyldiphenylsilyloxy(4-methoxyphenyl)methyl]acrylate (16e). Eluent: hexane: ethyl acetate 95:5 v/v, (0.15 g of a viscous oil, 97%). IR (v_{max} /Film) 1722 (C=O), 1629 (C=C), cm⁻¹; ¹H NMR (300MHz, CDCl₃) δ 1.0 (s, 9H), 3.5 (s, 3H, OCH₃ ester), 3.78 (s, OCH₃ aromatic), 5.54 (s large, 1H, C<u>H</u>OSiR₃), 6.28 (s large, 2H, double bond), 6.69 (d, *J* = 8.8Hz, 2H aromatics), 7.0 (d, *J* = 8.8Hz, 2H aromatics), 7.20-7.71 (m, 10H aromatics, -Si(Ph)₂); ¹³C NMR (75.4MHz, CDCl₃) δ 170.9, 157, 147.9, 136.1, 133.1, 129.9, 129.6, 127.8, 112.1, 80.9, 55, 52.2, 22.6, 20; Anal. calcd for C₂₈H₃₂O₄Si: C, 73.01%; H, 7.00%. Found C, 72.99%; H, 6.98%.

Methyl 2-[4-chlorophenyl(triethylsilyoxy)methyl]acrylate (17b). Eluent: hexane: ethyl acetate 95:5 v/v, (0.15 g of a fluid oil, 70%). IR (v_{max} /Film) 1723 (C=O), 1632 (C=C), cm⁻¹; ¹H NMR (300MHz, CDCl₃) δ 0.48-0.6 (m, 6H, OSiC<u>H</u>₂CH₃), 0.82-0.9 (m, 9H, OSiCH₂C<u>H</u>₃), 3.71 (s, 3H, OCH₃ ester), 5.57 (s large, 1H, C<u>H</u>OSiR₃), 6.1 (s large, 1H, double bond), 6.26 (s large, 1H, double bond), 7.25-7.31 (m, 4H aromatics); ¹³C NMR (75.4MHz, CDCl₃) δ 171, 148, 140, 128.7, 126.7, 123.1, 75.8, 52.2, 6.34, 6.12; Anal. calcd for C₁₇H₂₅ClO₃Si: C, 59.89%; H, 7.39%. Found C, 59.86%; H, 7.37%.

Methyl 2-[4-chlorophenyl(*tert*-butyldimethylsilyloxy)methyl]acrylate (17d). Eluent: hexane: ethyl acetate 95:05 v/v, (0.15 g of a viscous oil, 70%). IR (v_{max} /Film) 1717 (C=O), 1634 (C=C) cm⁻¹; ¹H NMR (300MHz, CDCl₃) δ 0.0 (s, 6H, OSi(C<u>H</u>₃)₂), 0.9 (s, 9H, OSi(CH₃)₂*t*-Bu), 3,67 (s, 3H, OCH₃ ester), 5.55 (s, 1H, C<u>H</u>OSiR₃), 6.13 (s, 1H, double bond), 6.32 (s, 1H, double bond), 7.20-8.24 (m, 4H, aromatics); ¹³C NMR (75.4MHz, CDCl₃) δ 170.8, 148.2, 140.2, 128.7, 126.7, 123.1, 75.8, 52.2, 25.8, 16.9, -4.56; Anal. calcd for C₁₇H₂₅ClO₃Si: C, 59.89%; H, 7.39%. Found C, 59.89%; H, 7.38%.

Methyl 2-[4-chlorophenyl(*tert*-butyldiphenylsilyloxy)methyl]acrylate (17e). Eluent: hexane: ethyl acetate 95:05 v/v, (0.12 g of a viscous oil, 72%). IR (v_{max} /Film) 1722 (C=O), 1630 (C=C) cm⁻¹; ¹H NMR (300MHz, CDCl₃) δ 1.0 (s, 9H, OSi-*t*-Bu), 3.54 (s, 3H, OC<u>H</u>₃ ester), 5.52 (s large, 1H, C<u>H</u>OSiR₃), 6.3 (s large, 2H, C=C<u>H</u>₂), 7.0-7.12 (m, 4H aromatics), 7.21-7.8 (m, 10H aromatics, Si(C₆<u>H</u>₅)₂; ¹³C NMR (75.4MHz, CDCl₃) δ 170.8, 148.2, 136.1, 133.1, 129.9, 127.8, 127.7, 126.8, 126.4, 80.9, 52.2, 22.6; Anal. calcd for C₂₇H₂₉ClO₃Si: C, 69.73%; H, 6.29%. Found C, 69.71%; H, 6.27%.

Methyl 2-benzo[*d*][1,3]dioxol-5-yl(triethylsilyloxy)methylacrylate (18b). Eluent: hexane: ethyl acetate 90:10 v/v, (0.12 g of a viscous oil, 88%). IR (v_{max} /Film) 2959, 2877, 1722, 1630, 1489, 1442, 1238, 1078 cm⁻¹; ¹H NMR (300MHz, CDCl₃) δ 0.5-0.58 (m, 2H), 0.95-0.9 (m, 3H), 3.67 (s, 3H, OCH₃ ester), 5.52 (s large, 1H, C<u>H</u>OSiR₃), 5.91 (s, 2H, -OC<u>H</u>₂O-), 6.0 (s large, 1H, double bond), 6.23 (s large, 1H, double bond), 6.70-6.84 (m, 3H aromatics); ¹³C NMR (75.4MHz, CDCl₃) δ 171, 148.2, 146, 145.6, 135.3, 123.1, 120.2, 109.8, 107, 52.2, 6.34, 6.1; Anal. calcd for C₁₈H₂₆O₅Si: C, 61.68%; H, 7.48%. Found C, 61.66%; H, 7.48%..

Methyl 2-benzo[*d*][1,3]dioxol-5-yl(*tert*-butyldimethylsilyloxy)methylacrylate (18d). Eluent: hexane: ethyl acetate 95:5 v/v, (0.35 g of a viscous oil, 77%). IR (v_{max} /Film) 1722 (C=O), 1630 (C=C) cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ -0.10 (s, OSiC<u>H</u>₃), 0.05 (s, OSiC<u>H</u>₃), 0.85 (s, OSi-<u>*t*</u>-<u>Bu</u>), 3.7 (s, OCH₃, ester), 5.5 (s, 1H, C<u>H</u>OSiR₃), 5.9 (s, 2H, -OC<u>H</u>₂O-), 6.0 (s large, 1H, double bond), 6.2 (s large, 1H, double bond), 6.7-6.9 (m, 3H aromatics); MS (70 eV, m/e, %): 350 (M⁺, ~0%), 293 (100%), 231 (10%), 187 (13%), 89 (23%); HRMS (M⁺) Calcd. for $C_{18}H_{26}O_5Si$ 293,08429 – loss of *t*-butyl (C₄H₉); Found: 350,154952 - 57,070425 = 293,084527.

Methyl 2-benzo[*d*][1,3]dioxol-5-yl(*tert*-butyldiphenylsilyloxy)methylacrylate (18e). Eluent: hexane: ethyl acetate 90:10 v/v, (0.2 g of a viscous oil, 93%). IR (v_{max} /Film) 1722 (C=O), 1630 (C=C) cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 1.06 (s, 9H, OSi-<u>*t*-Bu</u>), 3.53 (s, 3H, OCH₃ ester), 5.52 (s large, 1H, C<u>H</u>OSiR₃), 5.87 (m, 2H, -OC<u>H</u>₂O-), 6.27 (s, 1H, double bond), 6.56 (s large, 1H, double bond), 6.7 (s large, 1H aromatic), 7.23-7.72 (m, 12H, aromatics); ¹³C NMR (75.4MHz, CDCl₃) δ 169, 148.2, 146.8, 145.6, 136.2, 133.1, 129.9, 127.8, 123.1, 120.2, 109.6, 107, 82.3, 52.2, 22.6, 20; Anal. calcd for C₂₈H₃₀O₅Si: C, 70.86%; H, 6.37%. Found C, 70.86%; H, 7.37%.

Methyl 2-triethylsilyloxy(4-nitrophenyl)methylacrylate (19b). Eluent: hexane: ethyl acetate 90:10 v/v, (0.18 g of a fluid oil, 97%). IR (v_{max} /Film) 1722 (C=O), 1630 (C=C) cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 0.48-0.6 (m, 6H, OSiCH₂CH₃), 0.82 -0.9 (m, 9H, OSiCH₂CH₃), 3.75 (s, 3H, OCH₃ ester), 5.68 (s, 1H, CHOSiR₃), 6.19 (s, 1H, double bond), 6.33 (s, 1H, double bond), 7.56 (d, *J* = 8.8 Hz, 2H aromatics), 8.15 (d, *J* = 8.8 Hz, 2H aromatics); ¹³C NMR (75.4MHz, CDCl₃) δ 169, 150.3, 148.2, 145.1, 130, 123.1, 120.8, 75.8, 51.5, 6.3, 6.1; Anal. calcd for C₁₇H₂₅NO₅Si: C, 58.09%; H, 7.17%. Found C, 58.07%; H, 7.15%.

Methyl 3-*tert*-butyldimethylsilyloxy-2-methylenebutanoate (36). Eluent: hexane-ethyl acetate, 90:10 v/v. (0.25 g of a viscous oil, 78% yield). IR (v_{max}/Film) 1717 (C=O), 1630 (C=C) cm⁻¹; ¹H NMR (CDCl₃, 300 MHz): δ 0.05 (s, 6H, OSi(C<u>H</u>₃)₂), 0.89 (s, 9H, OSi-*t*-Bu), 1.6 (d, J = 3.0Hz, 3H), 3.74 (s, 3H, OC<u>H</u>₃ ester), 4.4 (s large, 1H, C<u>H</u>OSiR₃), 6.2 (s large, 2H, double bond); ¹³C NMR (75MHz, CDCl₃): δ 171, 148.9, 121.3, 69.9, 52.2, 25.8, 21.5, -4.89; Anal. calcd for C₁₂H₂₄O₃Si: C, 58.97%; H, 9.90%. Found C, 58.95%; H, 9.90%.

Methyl 3-*tert***-butyldimethylsilyloxy-2-methylenepentanoate** (**37**). Eluent: hexane-ethyl acetate, 90:10 v/v. (0.2 g of a viscous oil, 78% yield). ¹H NMR (CDCl₃, 300 MHz): δ 0.05 (s, 6H, Si(C<u>H</u>₃)₂), 0.89 (s, 9H, OSi-<u>*t*-Bu</u>), 0.91 (t, *J* = 6.2 Hz, 3H, CH₃ terminal), 1.52 – 1.41 (m, 1H), 1.70 – 1.79 (m, 1H), 3.74 (s, 3H, OC<u>H</u>₃ ester), 4.56 (s large, 1H, C<u>H</u>OSiR₃), 5.91 (s large, 1H, double bond), 6.23 (s large, 1H, double bond); ¹³C NMR (75.4MHz, CDCl₃): δ 166.8, 143.7, 124.6, 71.0, 51.6, 30.3, 25.8, 25.7, 25.6, 18.1, 9.1, -4.89,-4.83; Anal. calcd. for C₁₃H₃₆O₃Si C 58.20; H 13.43%; Found C 58.17; H 13.41.

Methyl 3-*tert*-**butyldimethylsilyloxy-2**-methylenehexanoate (38). Eluent: hexane-ethyl acetate, 95:5 v/v. (0.2 g of a viscous oil, 77% yield).¹H NMR (CDCl₃, 300 MHz) δ -0.01 (s, 3H), 0.06 (s, 3H), 0.92–0.83 (m, 12H, CH₃ terminal + OSi-<u>*t*-Bu</u>), 1.69 – 1.45 (m, 4H), 3.71 (s, 3H, OCH₃ ester), 4.57 (dd, J = 5.9 and 4.8 Hz, 1H, CHOSiR₃), 5.86 (s large, 1H, double bond), 6.22 (s large, 1H, double bond); ¹³C NMR (CDCl₃, 75.4MHz) δ 166.6 (C=O), 144.0, 124.2, 70.1, 51.6, 40.2, 25.9, 18.4, 18.2, 14.1, -4.5, -4.9; Anal. calcd for C₁₄H₂₈O₃Si: C, 61.72%; H, 10.36%. Found C, 61.70%; H, 10.35%.

Methyl 3-*tert***-butyldimethylsilyloxy-2-methyleneheptanoate (39).** Eluent: hexane-ethyl acetate, 95:5 v/v. (0.23 g of a viscous oil, 80% yield). ¹H NMR (CDCl₃, 300MHz) δ -0.01 (s, 3H), 0.06 (s, 3H), 0.91 (s, 12H, OSi-<u>*t*-Bu</u> + CH₃ terminal), 1.64-1.24 (m, 6H), 3.76 (s, 3H, OC<u>H</u>₃ ester), 4.49 (m, 1H, C<u>H</u>OSiR₃), 5.91 (s large, 1H, double bond), 6.22 (s large, 1H, double bond); ¹³C NMR (CDCl₃, 300MHz) δ 166.6 (C=O), 144.0, 124.2, 70.2, 51.6, 37.6, 27.2, 25.9, 25.7, 22.6, 18.2, 17.1, 14.1, -4.5, -4.9; Anal. calcd for C₁₅H₃₀O₃Si: C, 62.89%; H, 10.55%. Found C, 62.89%; H, 10.54%.

Alternative general procedure for the preparation of silylated Baylis-Hillman adducts (trialkyltriflate as silylating agent)

A mixture of 1 mmol of Baylis-Hillman adduct (1-5), 1.5 equivalents of trimethylsilyl trifluoromethanesulfonate, tri-isopropyl trifluoromethanesulfonate or *tert*-butyldimethylsilyl trifluoromethanesulfonate and 2 equivalents of anhydrous 2,6-lutidine in 0.5 mL of dichloromethane was stirred at room temperature, under a nitrogen atmosphere, for 30 minutes. After this time, the reaction medium was diluted with 60 mL of ethyl ether and the organic layer was successively washed with a 0.1 mol/L solution of HCl, a saturated solution of NaHCO₃ and finally brine. Then, the organic layer was dried over anhydrous Na₂SO₄ and evaporated under reduced pressure to provide an oily residue, which was purified by silica gel column chromatography.

Methyl 2-tri-isopropylsilyloxy(phenyl)methylacrylate (15c). Eluent: hexane: ethyl acetate 95: 5 v/v (0.2 g of a viscous colorless oil, 93%). IR (v_{max} /Film) 1719 (C=O), 1630 (C=C) cm⁻¹; ¹H NMR (300MHz, CDCl₃) δ 0.93-1.1 9 (m, 21H, OSi($\underline{i-Pr}$)₃), 3.66 (s, 3H, OCH₃ ester), 5.70 (s, 1H, CHOSiR₃), 6.24 (s large, 2H, double bond), 7.22-7.40 (m. 5H aromatics); ¹³C NMR (125MHz, CDCl₃) δ 170.8, 148.2, 139.7, 127.4, 126.8, 126.4, 123.1, 77.5, 52.2, 15.5, 9.9; Anal. calcd for C₂₀H₃₂O₃Si: C, 68.92%; H, 9.25%. Found C, 68.90%; H, 9.24%.

Methyl 2-[4-methoxyphenyl(tri-isopropylsilyloxy)methyl]acrylate (16c). Eluent: hexane: ethyl acetate 92: 8 v/v (0.12 g of a viscous colorless oil, 83%). IR (v_{max} /Film) 1717 (C=O), 1630 (C=C) cm⁻¹; ¹H NMR (300MHz, CDCl₃) δ 0.93-1.1 (m, 21H, OSi(<u>*i*-Pr</u>)₃), 3.66 (s, 3H, OC<u>H</u>₃ ester), 3.77 (s, 3H, OCH₃, aromatic), 5.65 (s, 1H, C<u>H</u>OSiR₃), 6.21 (s large, 2H, double bond), 6.79 (d, *J* = 8.8Hz, 2H aromatics), 7.30 (d, *J* = 8.78Hz, 2H aromatics); ¹³C NMR (75.4 MHz, CDCl₃) δ 170, 157.5, 148.2, 133.3, 129.4, 123.1, 112.2, 55, 52.2, 15.5, 9.9; Anal. calcd for C₂₁H₃₄O₄Si: C, 66.62%; H, 9.05%. Found C, 66.62%; H, 9.04%.

Methyl 2-[4-chlorophenyl(tri-ispropylsilyloxy)methyl]acrylate (17c). Eluent: hexane: ethyl acetate 98 : 2 v/v (0.17 g of a viscous colorless oil, 78%). IR (v_{max} /Film) 1721 (C=O), 1632 (C=C) cm⁻¹; ¹H NMR (300MHz, CDCl₃) δ 0.93-1.1 (m, 21H, OSi(*i*-Pr)₃), 3.67 (s, 3H, OCH₃ ester), 5.67 (s, 1H, C<u>H</u>OSiR₃), 6.23-6.26 (s large, 2H, double bond), 7.24 (d, *J* = 8.4 Hz, 2H aromatics), 7.3 (d, *J* = 8.4Hz, 2H aromatics); ¹³C NMR (75.4MHz, CDCl₃) δ 166.7, 144, 141.7, 132.9, 128.2, 128.1, 123.5, 71.9, 51.6, 17.9, 17.82, 12.1; Anal. calcd for C₂₀H₃₁ClO₃Si: C, 62.72%; H, 8.16%. Found C, 62.70%; H, 8.16%.

Methyl 2-benzo[*d*][1,3]dioxol-5-yl(tri-isopropylsilyloxy)methylacrylate (18c). Eluent: hexane: ethyl acetate 90 : 10 v/v (0.19 g of a viscous colorless oil, 85%); IR (v_{max} /Film) 1720 (C=O), 1632 (C=C) cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ 0.86-1.11 (m, 21H, OSi(*i*-Pr)₃), 3.67 (s, 3H, OCH₃ ester), 5.60 (s large, 1H, CHOTIPS), 5.91 (s large, 2H, OCH₂O), 6,21 (s large, 2H, C=CH₂), 6.69 (s large, 1H, aromatic), 6.78 (m, 2H, aromatics); ¹³C NMR (75.4MHz, CDCl₃) δ 170, 148.2, 146.8, 146, 132, 123.1, 119.9, 109.3, 107, 79.1, 52.2, 15.5, 9.9; Anal. calcd for C₂₁H₃₂O₅Si: C, 64.25%; H, 8.22%. Found C, 64.25%; H, 8.21%.

Methyl 2-tri-isopropylsilyloxy(4-nitrophenyl)methylacrylate (19c). Eluent: hexane: ethyl acetate 98 : 2 v/v (0.22 g of a viscous colorless oil, 94%). IR (v_{max} /Film) 1723 (C=O), 1634 (C=C) cm⁻¹; ¹H NMR (300MHz, CDCl₃) δ 1.0 (m, 21H, OSi (<u>*i*-Pr</u>)₃), 3.66 (s, 3H, OC<u>H</u>₃ ester), 5.65 (s large, 1H, C<u>H</u>OSiR₃), 6.26 (s large, 2H, C<u>H</u>₂ – double bond), 7.5 (d, *J* = 8.0Hz, 2H aromatics), 7.92 (d, *J* = 8.0Hz, 2H aromatics); ¹³C NMR (75.4MHz, CDCl₃) δ 166.0, 148.2, 147.8, 145.1, 129.9, 120.8, 52.2, 15.5, 9.9; Anal. calcd for C₂₀H₃₁NO₅Si: C, 61.04%; H, 7.94%. Found C, 61.01%; H, 7.93%.

Methyl 2-*tert*-**butyldimethylsilyloxy**(**4**-**nitrophenyl)methylacrylate** (**19d**). Eluent: hexane: ethyl acetate 98: 2 v/v (0.13 g of a viscous colorless oil, 76%). IR (v_{max} /Film) 1717 (C=O), 1634 (C=C) cm⁻¹; ¹H NMR (300MHz, CDCl₃) δ -0. 06 (s, 3H, OSiC<u>H</u>₃), 0.07 (s, 3H, OSiCH₃), 0.88 (s, 9H, Si-<u>*t*-Bu</u>), 3.69 (s, 3H, OC<u>H</u>₃ ester), 5.68 (s, 1H, C<u>H</u>OSiR₃), 6.17 (s, 1H, double bond), 6.33 (s, 1H, double bond), 7.56 (d, *J* = 8.5 Hz, 2H aromatics), 8.15 (d, *J* = 8.7 Hz, 2H aromatics); ¹³C NMR (75.4MHz, CDCl₃) 165.8, 150.1, 147.2, 142.7, 127.7, 125.0, 123.4, 71.8, 51.8, 29.6, 25.6, 18.1, -4.9, -5.0; Anal. calcd for C₁₇H₂₅NO₅Si: C, 58.09%; H, 7.17%. Found C, 58.08%; H, 7.15%.

General procedure for acetylation of Baylis-Hillman adducts 20-24

To a mixture of 1 mmol of the Baylis-Hillman adducts (1-5) and 2.1 equivalents of triethylamine (dried over CaH₂ and distilled) in 4 mL of dry dichloromethane were added, at 0° C, 1.5 equivalents of acetyl chloride (freshly distilled). The reaction medium was stirred, at room temperature, under nitrogen, for 3-6 h (the end of the reaction was confirmed by tlc). Then, cold water (3 mL) was added and the reaction medium was diluted with dichloromethane (15 mL). After washing with distilled water, saturated solution of NaHCO₃ (2 x 15 mL), brine (20 mL) and drying over Na₂SO₄, the organic layer was evaporated under reduced pressure to furnish an oily residue which was purified by silica gel column chromatography.

Methyl 2-methylcarbonyloxy(phenyl)methylacrylate (20). Eluent: hexane: ethyl acetate 80: 20 v/v (0.2 g of a tinged yellow oil, 82%). IR (v_{max} /Film) 1740 (C=O), 1726 (C=O), 1634 (C=C) cm⁻¹; ¹H NMR (300MHz, CDCl₃) δ 2.31 (s, 3H, CH₃CO), 3.91 (s, 3H, OCH₃ ester), 5.89 (s, 1H, CHOCOCH₃), 6.41 (s, 1H, =CH₂), 6.70 (s, 1H, =CH₂), 7.51-7.56 (m, 5H aromatics); ¹³C NMR (75.4MHz, CDCl₃) δ 169.1, 165.1, 139.4, 137.5, 128.2, 127.5, 125.5, 73, 51.9, 21.1; MS (70eV, m/z): 191 (M⁺ - 43, 100%), 175 (73%), 159 (62%), 132 (35%), 115 (68%), 105 (70%), 77 (32%); Anal. calcd for C₁₃H₁₄O₄: C, 66.66%; H, 6.02%. Found C, 66.64%; H, 6.01%.

Methyl 2-[4-methoxyphenyl(methylcarbonyloxy)methyl]acrylate (21). Eluent: hexane: ethyl acetate 70 : 30 v/v (0.08 g of a colorless oil, 78%). IR (v_{max} /Film) 1733 (C=O), 1719 (C=O), 1641 (C=C) cm⁻¹; ¹H NMR (300MHz, CDCl₃) δ 2.09 (s, 3H, CH₃CO), 3.69 (s, 3H, OCH₃ ester), 3.79 (s, 3H, OCH₃ aromatic), 5.87 (s, 1H, CHOCOCH₃), 6.37 (s, 1H, double bond), 6.62 (s, 1H, double bond), 6.84 (d, *J* = 8.0Hz, 2H aromatics), 7.31 (d, *J* = 8.0Hz, 2H aromatics); ¹³C NMR (75,4MHz, CDCl₃) δ 169.2, 163.2, 159.4, 139.5, 129.6, 129, 125, 113.7, 72.8, 55.2, 52, 21.2; MS (70eV, m/z): 143 (M⁺-COCH₃, 77%), 127 (100%), 115 (85%), 111 (80%), 95 (60%), 83 (48%), 67 (90%); Anal. calcd for C₁₄H₁₆O₅: C, 63.63%; H, 6.10%. Found C, 63.60%; H, 6.09%.

Methyl 2-[4-chlorophenyl(methylcarbonyloxy)methyl]acrylate (22). Eluent: hexane: ethyl acetate 70: 30 v/v (0.11 g of a colorless oil, 80%). IR (ν_{max} /Film) 1730 (C=O), 1722 (C=O), 1634 (C=C) cm⁻¹; ¹H NMR (300MHz, CDCl₃) δ 2.0 (s, 3H, CH₃CO), 3.74 (s, 3H, OCH₃ ester), 6.25 (s large, 1H), 6.44 (s large, 1H), 6.6 (s, 1H), 7.20-7.33 (m, 4H aromatics); ¹³C NMR (CDCl₃, 75.4MHz) δ 169, 168.5, 143.5, 136.8, 133.6, 129.2, 127.8, 125.6, 74, 52.2, 21; Anal. calcd for C₁₃H₁₃ClO₄: C, 58.11%; H, 4.88%. Found C, 58.08%; H, 4.84%.

Methyl 2-benzo[*d*][1,3]dioxol-5-yl(methylcarbonyloxy)methylacrylate (23). Eluent: hexane: ethyl acetate 85: 15 v/v (0.23 g of a colorless oil, 89%). IR (v_{max} /Film) 1731 (C=O), 1717(C=O), 1630 (C=C) cm⁻¹; ¹H NMR (300MHz, CDCl₃) δ 2.0 (s, 3H, CH₃CO), 3.7 (s, 3H, OCH₃ ester), 5.83 (s, 1H, CHOCOCH₃), 6.0 (s, 2H, -OCH₂O-), 6.42 (s large, 1H, double bond), 6.44 (s large, 1H, double bond), 7.0-7.12 (m, 3H aromatics); ¹³C NMR (CDCl₃, 75.4MHz) δ 170, 168.5, 147.9, 146.7, 143.5, 132, 125.6, 120.7, 110, 108.1, 101.3, 76, 52.2, 21; Anal. calcd for C₁₄H₁₄O₆: C, 60.43%; H, 5.07%. Found C, 60.39%; H, 4.99%.

Methyl 2-methylcarbonyloxy(4-nitrophenyl)methylacrylate (24). Eluent: hexane: ethyl acetate 80: 20 v/v (0.15 g of a viscous colorless oil, 76%). IR (v_{max} /Film) 1735 (C=O), 1719 (C=O), 1632 (C=C) cm⁻¹; ¹H NMR (300MHz, CDCl₃) δ 2.05 (s, 3H, CH₃CO), 3.74 (s, 3H, OCH₃ ester), 6.1 (s, 1H, CHOCOCH₃), 6.21 (s large, 1H, double bond), 6.4 (s, 1H, double bond), 7.59 (d, *J* = 8.7Hz, 2H aromatics), 7.82 (d, *J* = 8.7Hz, 2H aromatics); ¹³C NMR (CDCl₃, 75.4MHz) δ 169.7, 168, 146.8, 146, 143.6, 130.7, 125.6, 121.9, 73.9, 52.2, 21.1; Anal. calcd for C₁₃H₁₃NO₆: C, 55.91%; H, 4.69%. Found C, 55.87%; H, 4.65%.

General procedure for methylation of Baylis-Hillman adducts 25-27. To a stirred solution of 1 mmol of Baylis-Hillman adduct (1-3) in dichloromethane (4 mL) was added Ag₂O (1.5-2 mmol) and methyl iodide (3 mmol). The resulting suspension was then stirred at 40°C until complete consumption of the Baylis-Hillman adduct. The reaction was filtered and the solvent evaporated. The residue was further purified by column chromatography to provide the corresponding methyl ethers.

Methyl 2-methoxy(phenyl)methylacrylate (25). Eluent: hexane: ethyl acetate 70: 30 v/v (0.19 g of a viscous colorless oil, 86%). IR (v_{max} /Film) 1722 (C=O), 1634 (C=C) cm⁻¹; ¹H NMR (300MHz, CDCl₃) δ 3.34 (s, 3H, OC<u>H</u>₃ ether), 3.71 (s, 3H, OC<u>H</u>₃ ester), 5.15 (s, 1H, C<u>H</u>OCH₃), 5.94 (s, 1H, double bond), 6.35 (s, 1H, double bond), 7.29-7.36 (m, 5H aromatics); ¹³C NMR

 $(CDCl_3, 75.4MHz)$ δ 165.4, 140.9, 139.1, 127.1, 123.9, 80.5, 56.2, 51; Anal. calcd for $C_{12}H_{14}O_3$: C, 69.88%; H, 6.84%. Found C, 69.87%; H, 6.84%.

Methyl 2-methoxy(4-methoxyphenyl)methylacrylate (26). Eluent: hexane: ethyl acetate 70 : 30 v/v (0.198 g , 84%). IR (v_{max} /Film) 1720 (C=O), 1630 (C=C), 1503 cm⁻¹; (300MHz, CDCl₃) δ 3.32 (s, 3H, OC<u>H</u>₃ methyl ether), 3.74 (s, 3H, OC<u>H</u>₃ ester), 3.79 (s, 3H, OC<u>H</u>₃ aromatic), 5.2 (s, 1H, C<u>H</u>OCH₃), 6.15 (s, 1H, double bond), 6.26 (s, 1H, double bond), 6.73 (d, *J* = 8.8Hz, 2H aromatics), 7.23 (d, *J* = 8.8Hz, 2H aromatics); ¹³C NMR (125MHz, CDCl₃) δ 166.9, 157.7, 143, 133.5, 130.2, 126.5, 113.5, 77.7, 56.6, 55, 52.2; Anal. calcd for C₁₃H₁₆O₄: C, 66.09%; H, 6.83%. Found C, 66.04%; H, 6.80%.

Methyl 2-[4-chlorophenyl(methoxy)methyl]acrylate (27). Eluent: hexane: ethyl acetate 70 : 30 v/v (0.187 g, 78%). IR (v_{max} /Film) 1720 (C=O), 1633 (C=C) cm⁻¹; (300MHz, CDCl₃) δ 3.3 (s, 3H, OC<u>H₃</u> methyl ether), 3.67 (s, 3H, OC<u>H₃</u> ester), 4.83 (s, 1H, C<u>H</u>OCH₃), 6.15 (s, 1H, double bond), 6.2 (s, 1H, double bond), 7.17-7.36 (m, 4H aromatics); ¹³C NMR (125MHz, CDCl₃) δ 166.7, 143, 138.9, 129.3, 126.5, 77.6, 56.6, 52.3; Anal. calcd for C₁₂H1₃ClO₃: C, 59.88%; H, 5.44%. Found C, 59.85%; H, 5.42%.

General procedure for the hydrogenation of Baylis-Hillman adducts. To a suspension of 5% Pd-C (10 mol%) in ethyl acetate (5 cm³) was added, under nitrogen, a solution of the adduct (1 mmol) in ethyl acetate (5 cm³). Then, the reaction atmosphere was changed to hydrogen and the reaction mixture was stirred at room temperature. After a period for reaction completion (30 min. to 2 h), the suspension was filtered over a pad of Celite and the solvent was removed under reduced pressure (see chemical yields in Table 2). The residue was then treated with a solution of 5% HF in acetonitrile (1-3 mL) or TBAF in THF (1-3 mL) at room temperature for 2 h. For the silylated adducts (15a-e/19a-d), after this time, the reaction mixture was diluted with ethyl acetate (10-15 mL) and water (10-15mL). The phases were separated and the the organic layer was washed with distilled water (10 mL) and brine (10 mL), dried over Na₂SO₄ and evaporated to provide an oily residue sufficiently pure to be analysed without additional purification.

Methyl 3-hydroxy-2-methyl-3-phenylpropanoate (6a/b). **Diastereoisomer** *syn* (6a): ¹H NMR (300MHz, CDCl₃) δ 1.15 (d, J = 7.69Hz, 3H, CHC<u>H₃</u>) 2.69-2.82 (m, 1H, C<u>H</u>CH₃), 3.6 (s, 3H, OC<u>H₃</u> ester), 5.05 (d, J = 5.92Hz, 1H, C<u>H</u>OH), 7.27-7.37 (m, 5H aromatics); ¹³C NMR (300MHz, CDCl₃) δ 173.6, 136.7, 128.1, 127.3, 126.8, 73.4, 51, 38.5, 15; **diastereoisomer** *anti* (6b): ¹H NMR (300MHz, CDCl₃) δ 0.90 (d, J = 6.9Hz, 3H, CHC<u>H₃</u>), 2.69-2.82 (m, 1H, C<u>H</u>CH₃), 3.76 (s, 3H, OC<u>H₃</u> ester), 4.73 (d, J = 8.1Hz, 1H, C<u>H</u>OH), 7.27-7.37 (m, 5H aromatics); ¹³C NMR (300MHz, CDCl₃) δ 175.6, 137.6, 128.1, 127.3, 126.8, 73.4, 51, 44.3, 16.4; Anal. calcd for C₁₁H₁₄O₃: C, 68.02%; H, 7.27%. Found C, 68.00%; H, 7.24%. Methyl 3-hydroxy-3-(4-methoxyphenyl)-2-methylpropanoate (7a/b). Diastereoisomer syn (7a): ¹H NMR (300MHz, CDCl₃) δ 1.14 (d, J = 6.9Hz, 3H, CHC<u>H</u>₃), 2.61-2.72 (m, 1H, C<u>H</u>CH₃), 3.54 (s, 3H, OC<u>H</u>₃ ester), 3.79 (s, 3H, OCH₃ ether), 4.91 (d, J = 6.2Hz, 1H, C<u>H</u>OH), 6.82 (d, J = 8.79Hz, 2H aromatics), 7.2 (d, J = 8.4Hz, 2H aromatics); ¹³C NMR (300MHz, CDCl₃) δ 173, 158.1, 134.4, 129.4, 113.1, 73.4, 55, 51, 38.6, 15; diastereoisomer anti (7b). ¹H NMR (300MHz, CDCl₃)δ 0.85 (d, J = 7.3Hz, 3H, CHC<u>H</u>₃), 2.61-2.72 (m, 1H, C<u>H</u>CH₃), 3.54 (s, 3H, OC<u>H</u>₃ ester), 3.79 (s, 3H, OC<u>H</u>₃ ether), 4.64 (d, J = 9.8Hz, 1H, C<u>H</u>OH), 6.82 (d, J = 8.8Hz, 2H aromatics), 7.2 (d, J = 8.4Hz, 2H aromatics); ¹³C NMR (300MHz, CDCl₃) δ 175.6, 158.1, 134.4, 129.4, 113.2, 73.4, 55, 51, 44.3, 16.4; Anal. calcd for C₁₂H₁₆O₄: C, 64.27%; H, 5.19%. Found C, 64.23%; H, 5.16%.

Methyl 3-(4-chlorophenyl)-3-hydroxy-2-methylpropanoate (8a/b). Diastereoisomer *syn* (8a): ¹H NMR (300MHz, CDCl₃) δ 1.12 (d, J = 6.0Hz, 3H), 2.55-2.74 (m, 1H), 3.55 (s, 3H), 4.95 (d, J = 6.1Hz, 1H), 7.2-7.4 (m, 4H aromatics); ¹³C NMR (300MHz, CDCl₃) δ 173.6, 139.3, 133.2, 128.8, 127.7, 73.4, 51, 38.5, 15; **diastereoisomer** *anti* (8b): ¹H NMR (300MHz, CDCl₃) δ 0.92 (d, J = 6.7Hz, 3H), 2.55-2.74 (m, 1H), 3.55 (s, 3H), 4.67 (d, J = 9.2Hz, 1H), 7.2-7.4 (m, 4H aromatics); ¹³C NMR (300MHz, CDCl₃) δ 175.6, 139.3, 133.2, 128.8, 127.7, 73.4, 51, 44.3, 16.4; Anal. calcd for C₁₁H₁₃ClO₃: C, 57.78%; H, 5.73%. Found C, 57.75%; H, 5.74%.

Methyl 3-benzo[*d*][1,3]dioxol-5-yl-3-hydroxy-2-methylpropanoate (9*a*/b). Diastereoisomer *syn* (9*a*): ¹H NMR (300MHz, CDCl₃) δ 1.14 (d, J = 6.95Hz, 3H), 2.57-2.71 (m, 1H, C<u>H</u>CH₃), 3.55 (s, 3H, OC<u>H</u>₃ ester), 4.87 (d, J = 6.2Hz, 1H, C<u>H</u>OH), 5.93-6.07 (m, 2H, -OC<u>H</u>₂O-), 6.71-6.77 (m, 1H aromatic), 6.81-6.84 (m, 2H aromatics); ¹³C NMR (300MHz, CDCl₃) δ 173, 147.8, 146.3, 135.9, 120.7, 110.3, 108, 101.3, 74, 50.9, 38.5, 15; diastereoisomer *anti* (9*b*): ¹H NMR (300MHz, CDCl₃) δ 0.05 (s, 9H, OSi(CH₃)₃), 1.7 (d, J = 6.9Hz, 3H, CH₃), 2.57-2.71 (m, 1H, C<u>H</u>CH₃), 3.71 (s, 3H, OC<u>H</u>₃ ester), 4.64 (d, J = 9.1 Hz, 1H, C<u>H</u>OH), 5.93-6.0 (m, 2H, -OC<u>H</u>₂O-), 6.71-6.77 (m, 1H aromatic), 6.81-6.84 (m, 2H aromatics); ¹³C NMR (300MHz, CDCl₃) δ 175.5, 147.8, 146.3, 135.5, 120.7, 110.3, 108, 101.3, 74, 51, 44.3, 16.3; Anal. calcd for C₁₂H₁₄O₅: C, 60.50%; H, 5.92%. Found C, 60.49%; H, 5.90%.

Methyl 3-(4-nitrophenyl)-3-hydroxy-2-methylpropanoate (10a/b). Diastereoisomer syn (10a): ¹H NMR (300MHz, CDCl₃) δ 1.15 (d, J = 6.61Hz, 3H), 2.53-2.72 (m, 1H), 3.55 (s, 3H), 4.72 (d, J = 6.6 Hz, 1H), 6.62 (d, J = 8.0Hz, 2H aromatics), 7.07 (d, J = 8.4Hz, 2H aromatics); ¹³C NMR (300MHz, CDCl₃) δ 173.6, 149.5, 145.7, 130.1, 121.8, 73.4, 51, 38.5, 15; diastereoisomer anti (10b): ¹H NMR (300MHz, CDCl₃) δ 0.86 (d, J = 6.98 Hz, 3H, CH₃), 2.53-2.72 (m, 1H), 3.55 (s, OCH₃ ester), 4.58 (d, J = 9.1Hz, 1H), 6.62 (d, J = 8.0Hz, 2H aromatics); ¹³C NMR (300MHz, CDCl₃) δ 176, 149.5, 145.7, 130.2, 121.8, 73, 49.9, 44.3, 16.4; Anal. calcd for C₁₁H₁₃NO₅: C, 55.23%; H, 5.48%. Found C, 55.20%; H, 5.47%.

Methyl 2-methyl-3-methylcarbonyloxy-3-phenylpropanoate (28a/b). Diastereoisomer *syn* (28a): ¹H NMR (300MHz, CDCl₃) δ 1.16 (d, J = 6.5Hz, 3H), 2.0 (s, 3H, COC<u>H</u>₃), 2.86-2.90 (m, 1H), 3.65 (s, 3H), 5.71 (d, J = 7.0Hz, 1H), 7.21-7.30 (m, 5H aromatics); ¹³C NMR (300MHz, CDCl₃) δ 174.3, 171.3, 136.5, 128.4, 127.2. 126.7, 80.5, 51, 36.4, 21.4, 16.8; **diastereoisomer** *anti* (28b): ¹H NMR (300MHz, CDCl₃) δ 1.05 (d, J = 6.5Hz, 3H), 2.05 (s, 3H), 2.87-2.93 (m, 1H), 3.62 (s, 3H), 5.92 (d, J = 4.6Hz, 1H), 7.26 (s large, 5 H aromatics) ; ¹³C NMR (300MHz, CDCl₃) δ 176.3, 171.3, 136.5, 128.4, 127.2, 126.9, 80.5, 51, 42.2, 21.3, 18.2; Anal. calcd for C₁₃H₁₆O₄: C, 66.09%; H, 6.83%. Found C, 66.05%; H, 6.82%.

Methyl 3-(4-methoxyphenyl)-2-methyl-3-methylcarbonyloxypropanoate (29a/b). Diastereoisomer syn (29a): ¹H NMR (300MHz, CDCl₃) δ 1.14 (d, J = 6.3Hz, 3H), 2.05 (s, 3H, COC<u>H</u>₃), 2.81-2.93 (m, 1H), 3.59 (s, 3H), 3.73 (s, 3H), 5.71 (d, J = 7.0Hz, 1H), 6.79 (d, J = 8.8Hz, 2H aromatics), 7.05 (d, J = 8.8Hz, 2H aromatics); ¹³C NMR (75.4MHz, CDCl₃) δ 174.3, 171.3, 158.2, 134, 129.6, 113.1, 80.5, 55, 51, 36.4, 21.4, 16.8; diastereoisomer *anti* (29b): ¹H NMR (300MHz, CDCl₃) δ 1.14 (d, J = 6.5Hz, 3H), 2.0 (s, 3H), 3.62 (s, 3H), 3.75 (s, 3H), 5.92 (d, J = 5.3Hz, 1H), 6.79 (d, J = 8.8Hz, 2H aromatics), 7.05 (d,

Methyl 3-(4-chlorophenyl)-2-methyl-3-methylcarbonyloxypropanoate (30a/b). Diastereoisomer *syn* (30a): ¹H NMR (300MHz, CDCl₃) δ 1.15 (d, J = 6.5Hz, 3H), 2,05 (s, 3H), 2.86-2.93 (m, 1H), 3.65 (s, 3H), 5.71 (d, J = 7.0Hz, 1H), 7.28-7.31 (m, 4H aromatics); ¹³C NMR (75.4MHz, CDCl₃) δ 174.3, 171.2, 137.2, 133.3, 128.9, 127.6, 80.5, 51, 36.4, 21.3, 16.7; **diastereoisomer** *anti* (30b): ¹H NMR (300MHz, CDCl₃) δ 1.12 (d, J = 6.4Hz, 3H), 2.0 (s, 3H), 2.84-2.90 (m, 1H), 3.63 (s, 3H), 5.92 (d, J = 5.2 Hz, 1H), 7.28-7.31 (m, 4H aromatics); ¹³C NMR (75.4MHz, CDCl₃) δ 176.2, 171.3, 137.2, 133.3, 128.9, 127.7, 80.5, 51, 42.2, 21.4, 17.8; Anal. calcd for C₁₃H₁₅ClO₄: C, 57.68%; H, 5.59%. Found C, 57.65%; H, 5.57%.

Methyl 3-benzo[*d*][1,3]dioxol-5-yl-2-methyl-3-methylcarbonyloxypropanoate (31a/b). Diastereoisomer *syn* (31a): ¹H NMR (300MHz, CDCl₃) δ 1.13 (d, J = 6.5Hz, 3H), 2.05 (s, 3H), 2.80-2.93 (m, 1H), 3.68 (s, 3H), 5.84 (d, J = 7.0Hz, 1H), 6.05 (s, 2H), 6.68-6.79 (m, 3H aromatics); ¹³C NMR (75.4MHz, CDCl₃) δ 174.3, 171.3, 147.7, 146.5, 132.7, 119.7, 110.5, 107.9, 101.3, 81.8, 50.9, 36.5, 21.3, 16.8; **diastereoisomer** *anti* (31b): ¹H NMR (300MHz, CDCl₃) δ 1.16 (d, J = 6.5Hz, 3H), 2.05 (s, 3H), 2.87-2.93 (m. 1H), 3.64 (s, 3H), 6.05 (s, 2H), 6.08 (d, J = 5.3Hz, 1H), 6.68-6.79 (m, 3H aromatics); ¹³C NMR (75.4MHz, CDCl₃) δ 176.3, 171.3, 147.7, 146.4, 132.7, 119.7, 119.7, 110.5, 107.9, 101.3, 81.8, 51, 42.2, 21.4, 18; Anal. calcd for C₁₄H₁₆O₆: C, 59.99%; H, 5.75%. Found C, 59.99%; H, 5.83%.

Methyl 2-methyl-3-methylcarbonyloxy-3-(4-nitrophenyl)propanoate (32a/b). Diastereoisomer *syn* (32a): ¹H NMR (300MHz, CDCl₃) δ 1.16 (d, J = 6.4Hz, 3H), 2.05 (s, 3H), 2.83-2.91 (m, 1H), 3.60 (s, 3H), 5.71 (d, J = 7.0Hz, 1H), 7.0 (d, J = 8.9Hz, 2H aromatics), 7.81 (d, J = 8.9Hz, 2H

aromatics); ¹³C NMR (75.4MHz, CDCl₃) δ 174.3, 171.3, 146.4, 145.9, 130.3, 121.7, 80.5, 51, 36.4, 21.3, 16.7; **diastereoisomer** *anti* (**32b**): ¹H NMR (300MHz, CDCl₃) δ 1.17 (d, *J* = 6.0Hz, 3H), 2.0 (s, 3H), 2.86-2.93 (m, 1H), 3.67 (s, 3H), 5.92 (d, *J* = 5.3Hz, 1H), 7.0 (d, *J* = 8.9Hz, 2H aromatics), 7.81 (d, *J* = 8.9Hz, 2H aromatics); ¹³C NMR (75.4MHz, CDCl₃) δ 176.3, 171.3, 146.4, 145.9, 130.3, 121.7, 80.5, 51, 42.2, 21.3, 18.2. Anal. calcd for C₁₃H₁₅NO₆: C, 55.51%; H, 5.38%. Found C, 55.51%; H, 5.38%.

Methyl-3-methoxy-2-methyl-3-phenylpropanoate (33a/b). Diastereoisomer *syn* (33a): ¹H NMR (500MHz, CDCl₃) δ 1.23 (d, J = 6.95Hz, 3H), 2.78 (dq, J = 10 and 6.95Hz, 1H), 3.24 (s, 3H), 3.55 (s, 3H), 4.44 (d, J = 6.95Hz, 1H), 7.29-7.36 (m, 5 H aromatics); ¹³C NMR (125MHz, CDCl₃) δ 175.8, 138.9, 128.5, 128.3, 127.7, 86, 56.8, 51.8, 47, 14; **diastereoisomer** *anti* (33b) ¹H NMR (500MHz, CDCl₃) δ 0.88 (d, J = 6.5Hz, 3H), 2.75 (m, 1H), 3.15 (s, 3H), 3.76 (s, 3H), 4.25 (d, J = 10Hz, 1H), 7.29-7.36 (m, 5H aromatics); ¹³C NMR (125MHz, CDCl₃) δ 174.4, 139.7, 128.2, 127.8, 127.0, 84.3, 57.1, 51.4, 47.4, 12.3; Anal. calcd for C₁₄H₁₄O₆: C, 60.43%; H, 5.07%. Found C, 60.39%; H, 4.99%.

Methyl-3-methoxy-3-(4-methoxyphenyl)-2-methylpropanoate (34a/b). Diastereoisomer *syn* **(34a):** ¹H NMR (500MHz, CDCl₃) δ 1.12 (d, J = 6.5Hz, 3H), 2.72-2.79 (m, 1H), 3.35 (s, 3H), 3.63 (s, 3H), 3.72 (s, 3H), 4.41 (d, J = 6.0Hz, 1H), 6.82 (m, 2H aromatics), 7.0 (m, 2H aromatics); ¹³C NMR (125MHz, CDCl₃) δ 172.8, 157.5, 134.8, 129.6, 113.3, 82.3, 56, 55, 51, 36.9, 15; **diastereoisomer** *anti* **(34b)** ¹H NMR (500MHz, CDCl₃) δ 1.12 (d, J = 6.5Hz, 3H), 2.72-2.80 (m, 1H), 3.40 (s, 3H),, 3.65 (s, 3H), 3.75 (s, 3H), 4.42 (d, J = 6.0Hz, 1H), 6.82 (m, 2H aromatics), 7.0 (m, 2H aromatics); ¹³C NMR (125MHz, CDCl₃) δ 174.8, 157.5, 134.8, 129.6, 113.3, 82.3, 56, 55, 51, 42.6, 16.5; Anal. calcd for C₁₃H₁₈O₄: C, 65.53%; H, 7.61%. Found C, 65.50%; H, 7.59%.

Methyl-3-(4-chlorophenyl)-3-methoxy-2-methylpropanoate (35a/b). Diastereoisomer syn (35a): ¹H NMR (500MHz, CDCl₃) δ 1.15 (d, J = 6.5Hz, 3H), 2.73-2.80 (m, 1H), 3.40 (s, 3H), 3.65 (s, 3H), 4.4 (d, J = 7.0Hz, 1H), 7.19 (d, J = 8.0Hz, 2H aromatics), 7.33 (d, J = 8.0Hz, 2H aromatics); ¹³C NMR (125MHz, CDCl₃) δ 172.8, 139.2, 132.6, 129, 127.9, 82.3, 56.6, 51, 36.9, 15; diastereoisomer anti (35b) ¹H NMR (500MHz, CDCl₃) δ 1.15 (d, J = 6.54Hz, 3H), 2.73-2.80 (m, 1H), 3.40 (s, 3H), 3.65 (s, 3H), 4.41 (d, J = 6.5Hz, 1H), 7.19 (d, J = 8.0Hz, 2H aromatics); ¹³C NMR (125MHz, CDCl₃) δ 174.8, 139.3, 129, 127.9, 82.4, 56., 51, 42.6, 16.5; Anal. calcd for C₁₂H₁₅ClO₃: C, 59.39%; H, 6.23%. Found C, 59.39%; H, 6.23%.

Methyl 3-hydroxy-2-methylbutanoate (40a). Reaction time. 10 h [78% of a single diastereomer (we were unable to isolate the *anti* diastereoisomer)]. ¹H NMR (300MHz, CDCl₃) δ 1.12 (t, J = 6.5Hz, 3H, CHC<u>H</u>₃), 1.21 (d, J = 6.42Hz, 3H, C<u>H</u>₃ terminal), 2.35-2.41 (m, 1H), 3.66 (s, 3H, OC<u>H</u>₃ ester), 3.74-3.81 (m, 1H, C<u>H</u>OH); ¹³C NMR (75.4MHz, CDCl₃) δ 172.1, 67, 51, 38.7, 20.4, 14.1; Anal. calcd for C₆H₁₂O₃: C, 54.53%; H, 9.15%. Found C, 54.51%; H, 9.12%.

Methyl 3-hydroxy-2-methylpentanoate (41a). Reaction time. 12 h; [82% of a single diastereomer (we were unable to isolate the *anti* diastereoisomer)]. ¹H NMR (300MHz, CDCl₃) δ 0.97 (t, 3H, CH₃ terminal), 1.12 (d, J = 6.9 Hz, 3H, CH₃CH), 1.58–1.64 (m, 2H, CH₂CH₃), 2.38-2.45 (m, CH₃CHCO), 3.66 (s, 3H, OCH₃ ester), 3.70-3.75 (dt, J = 5.9 and 5.5 Hz,1 H, CHOH); ¹³C NMR (75.4MHz, CDCl₃) δ 173.8 (C=O), 69.8, 51.2, 38, 26.8, 14.3, 11.4. %; Anal. calcd for C₇H₁₄O₃: C, 57.51%; H, 9.65%. Found C, 57.50%; H, 9.63%.

Methyl 3-hydroxy-2-methylhexanoate (42a). Reaction time. 10 h; (76% of a colorless oil, only a single diastereoisomer detected $\geq 95\%$). ¹H NMR (C₆D₆, 300MHz) δ 0.89 (t, 3H, CH₃ terminal), 1.12 (d, J = 6.9Hz, 3H), 1.5-1.8 (m, 4H), 2.54 (dq, J = 4.76 and 6.9Hz, 1H), 3.67 (s, 3H), 4.0 (dt, J = 4.0 and 6.0Hz, 1H, C<u>H</u>OH); ¹³C NMR (CDCl₃, 75.4MHz) δ 174.6 (C=O), 77.4, 69.5, 51, 39.9, 35.8, 21.2, 14.2; Anal. calcd. for C₈H₁₆O₃ C, 59.97%; H, 10.07%; Found C, 59.97%; H, 10.05%.

Methyl 3-hydroxy-2-methylheptanoate (43a). Reaction time. 10 h; (76% of a fluid oil, only a single diastereoisomer detected $\ge 95\%$). 1H NMR (300MHz, C₆D₆) δ 0.85 (t, 3H, CH₃ terminal), 1.23 (d, J = 7.3Hz, 3H), 1.33-1.26 (m, 2H), 1.59-1.49 (m, 4H), 2.52 (dq, J = 4.4 and 6.95Hz, 1H), 3.44 (s, 3H), 4.14 (ddd, J = 1.8, 4.3 and 6.2Hz, 1H, CHOH); ¹³C NMR (75.4MHz, C₆D₆) δ 174.6 (C=O), 73.6, 51.1, 44.6, 35.4, 27.9, 14.4, 11.03; HRMS (M⁺) Calcd. for C₇H₁₄O₃ 174.12559; Found 174.12550.

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